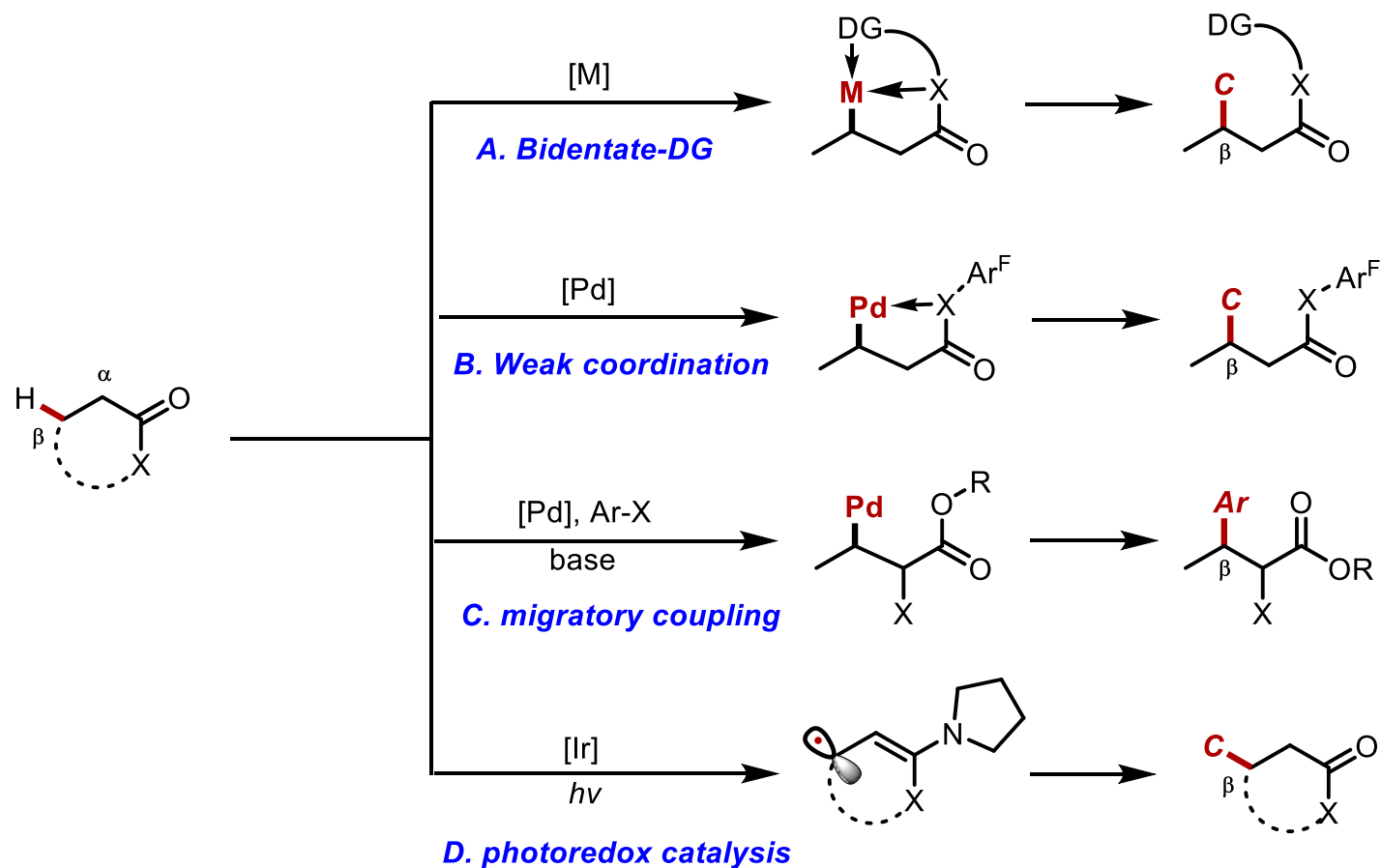


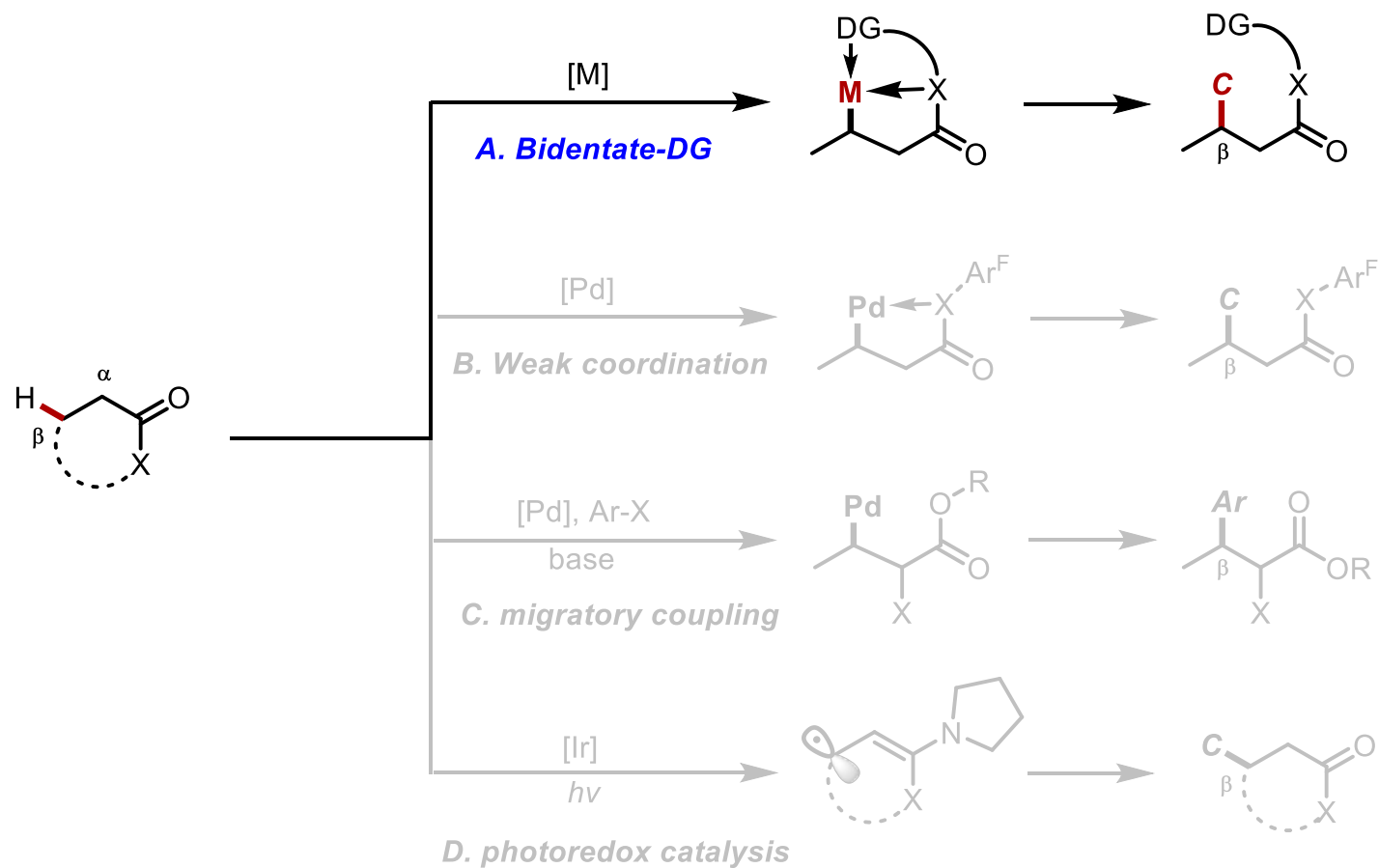
*Catalytic C-C Bond Forming Transformations
via β -C-H Functionalization of Carbonyl
Compounds*

Zhongxing Huang
May 7st, 2014

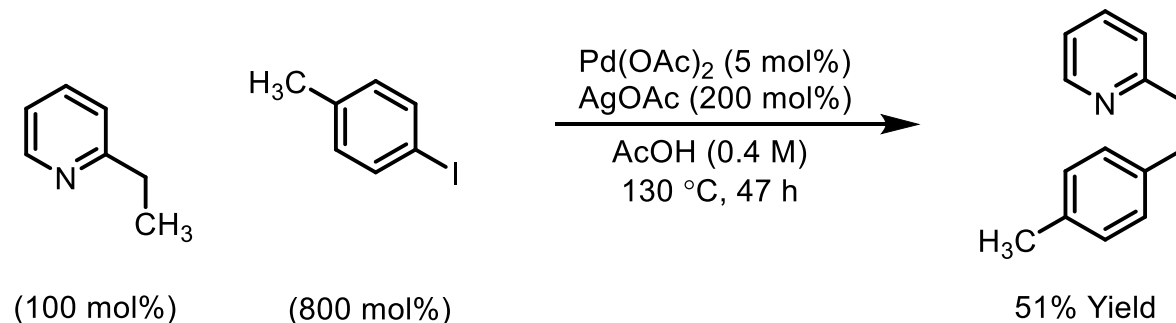


Not covered today:

- NHC catalysis
- Dehydrogenative C-C coupling
- Enamine catalysis



▪ Daugulis's Novel Discovery

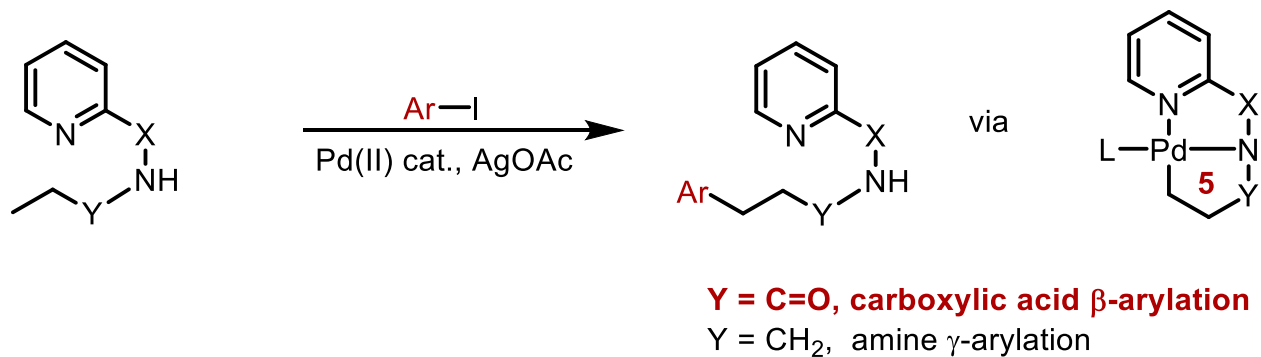


Features:

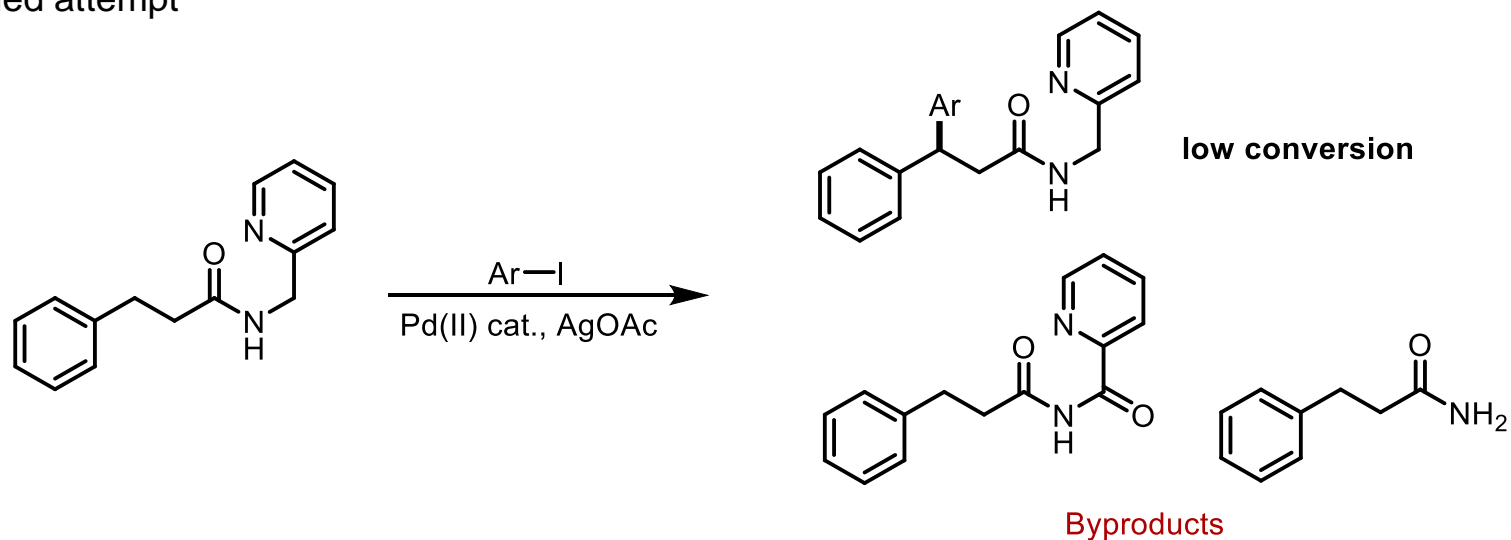
- Aryl iodide/silver salt combination (Sanford: diaryliodonium salt)
- Unactivated sp^3 C-H bonds
- Methylene C-H bonds cannot be arylated
- Electron-rich aryl iodide \gg electron-poor ones

Design of detachable auxiliary

Design

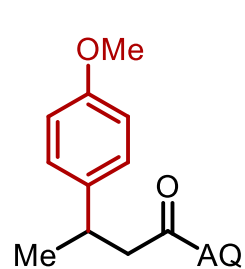
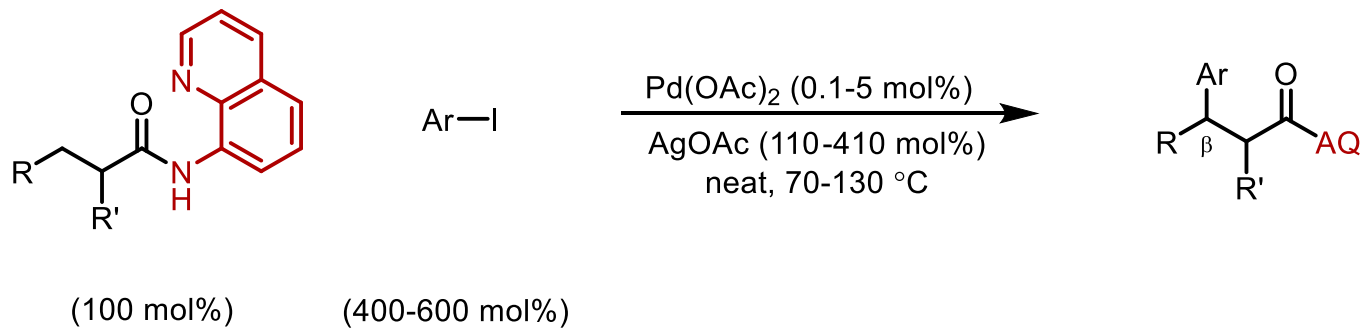


Failed attempt

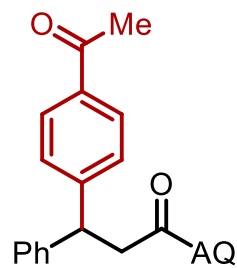


Design of detachable auxiliary

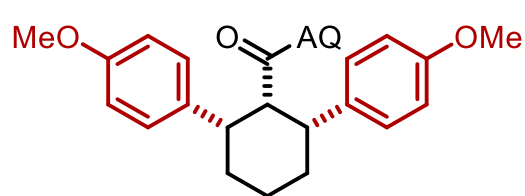
- 8-aminoquinoline (AQ)



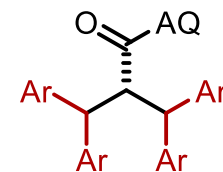
5 min, 110 °C
92% Yield



20 min, 120 °C
60% Yield



5 h, 70 °C
61% Yield

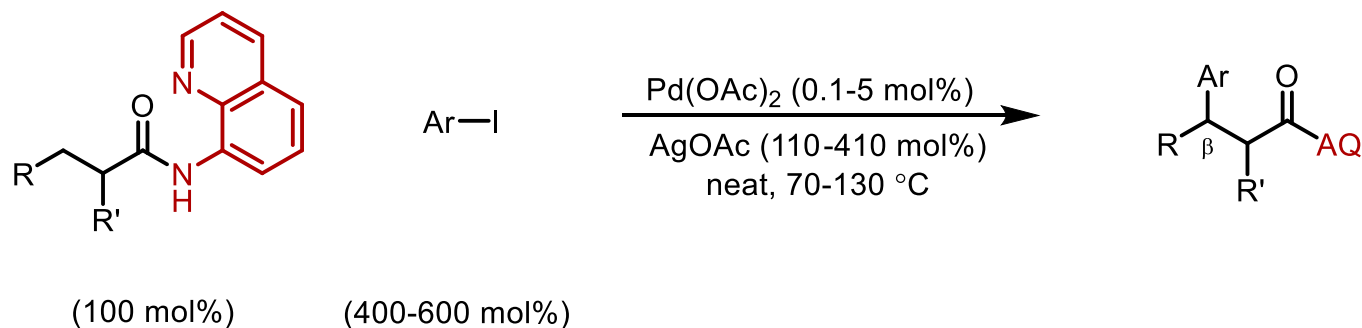


$Ar = p-C_6H_4Me$

30 min, 120 °C
60% Yield

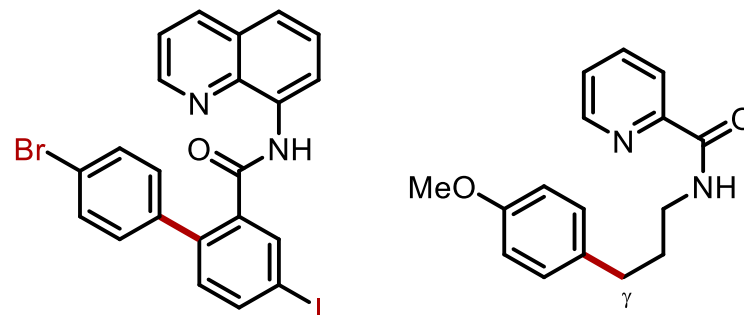
Design of detachable auxiliary

- 8-aminoquinoline (AQ)

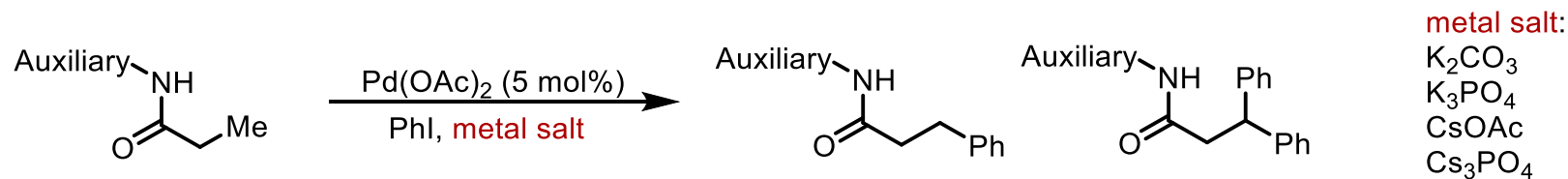


Features:

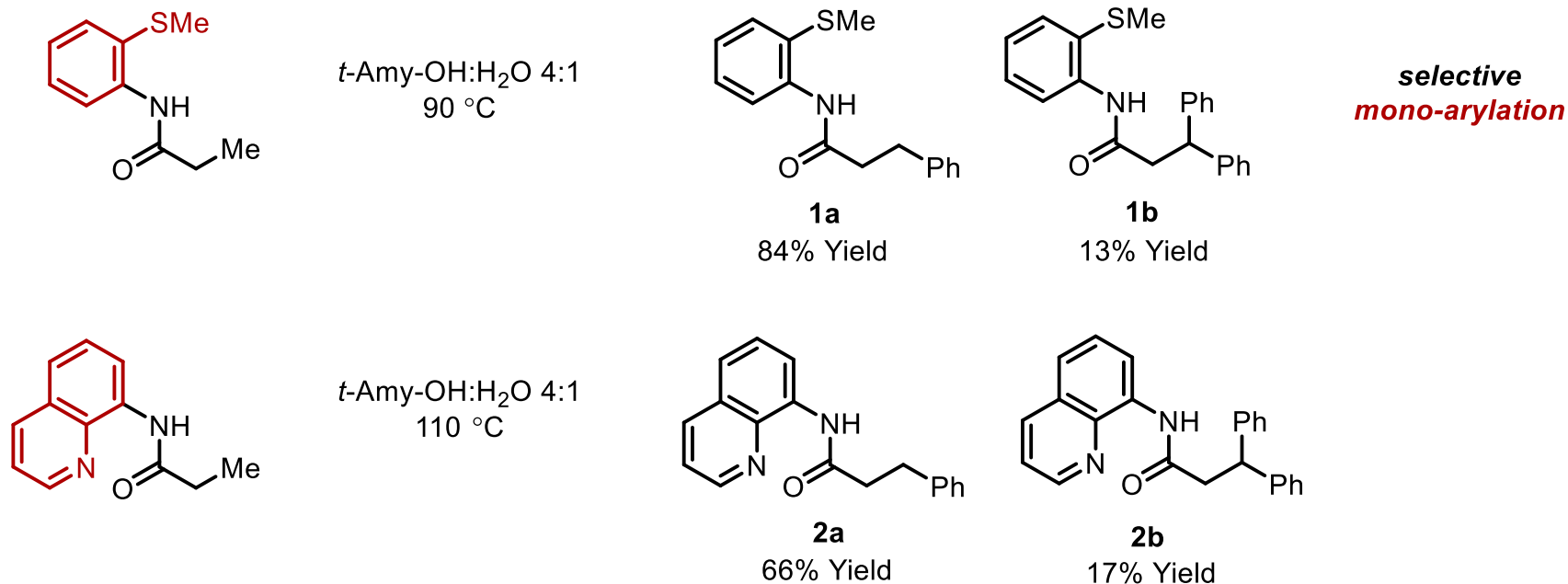
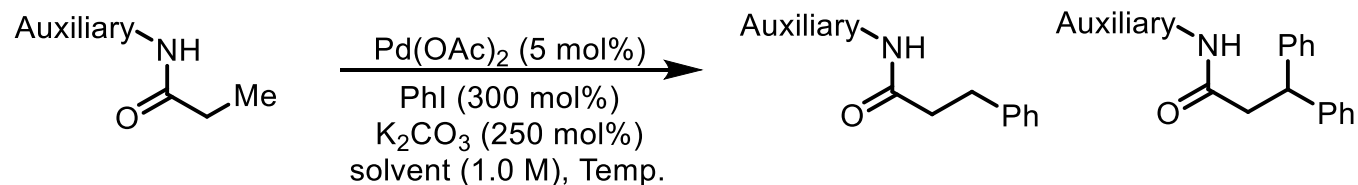
- Aryl iodide/silver salt combination
- Methylene C-H activation enabled
- Rate: secondary > primary C-H bonds
- Free NH required
- Neat condition



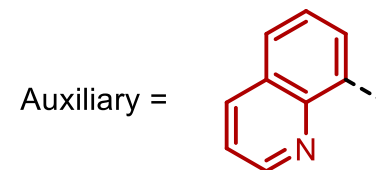
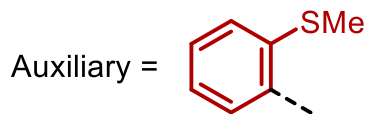
▪ 'Silver-free' conditions



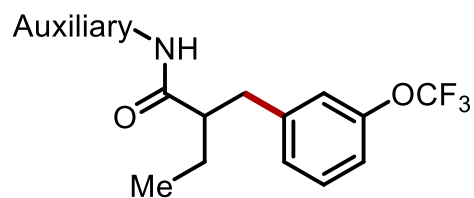
Optimization of Auxiliary



Primary vs secondary C-H activation

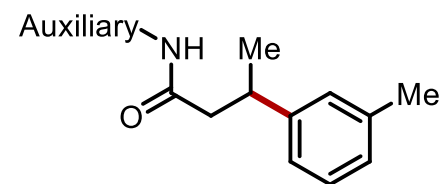


t-Amy-OH:H₂O
90 °C, K₂CO₃



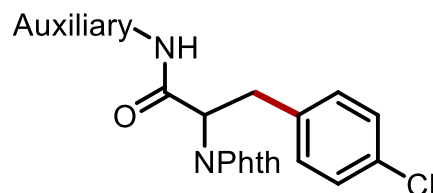
60% Yield

t-Amy-OH
90 °C, Cs₃PO₄



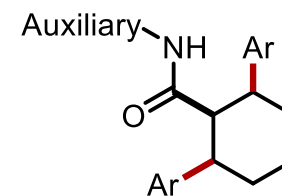
79% Yield

Toluene
110 °C, CsOAc



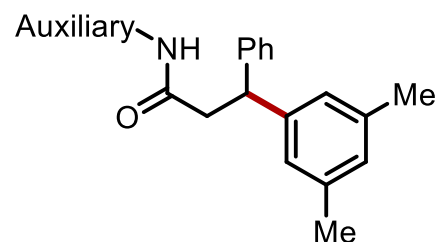
79% Yield

t-Amy-OH
90 °C, Cs₃PO₄
Ar = *m*-C₆H₄Br



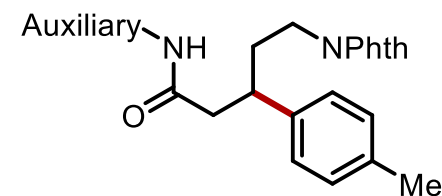
52% Yield

t-Amy-OH
90 °C, K₂CO₃
20 mol% PivOH



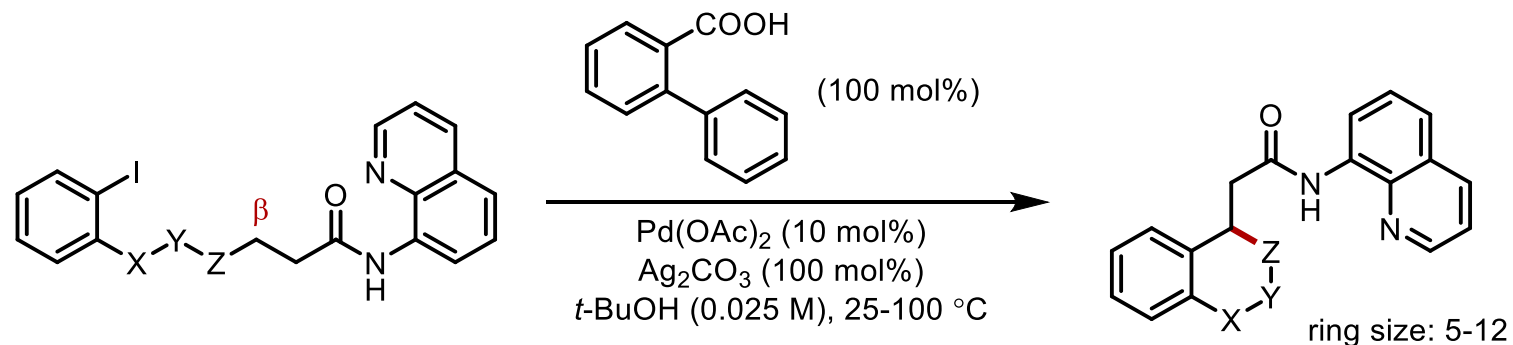
47% Yield

t-Amy-OH
90 °C, Cs₃PO₄

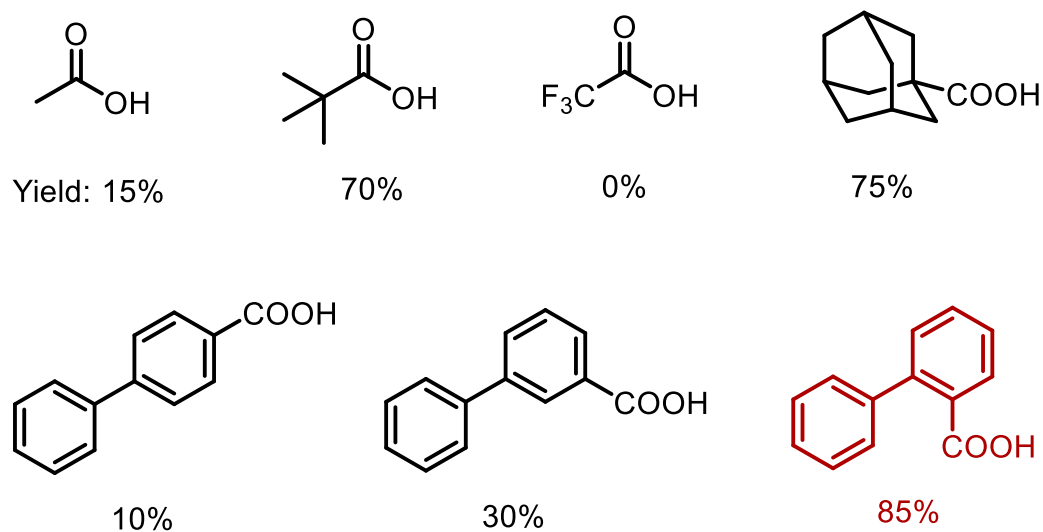


76% Yield

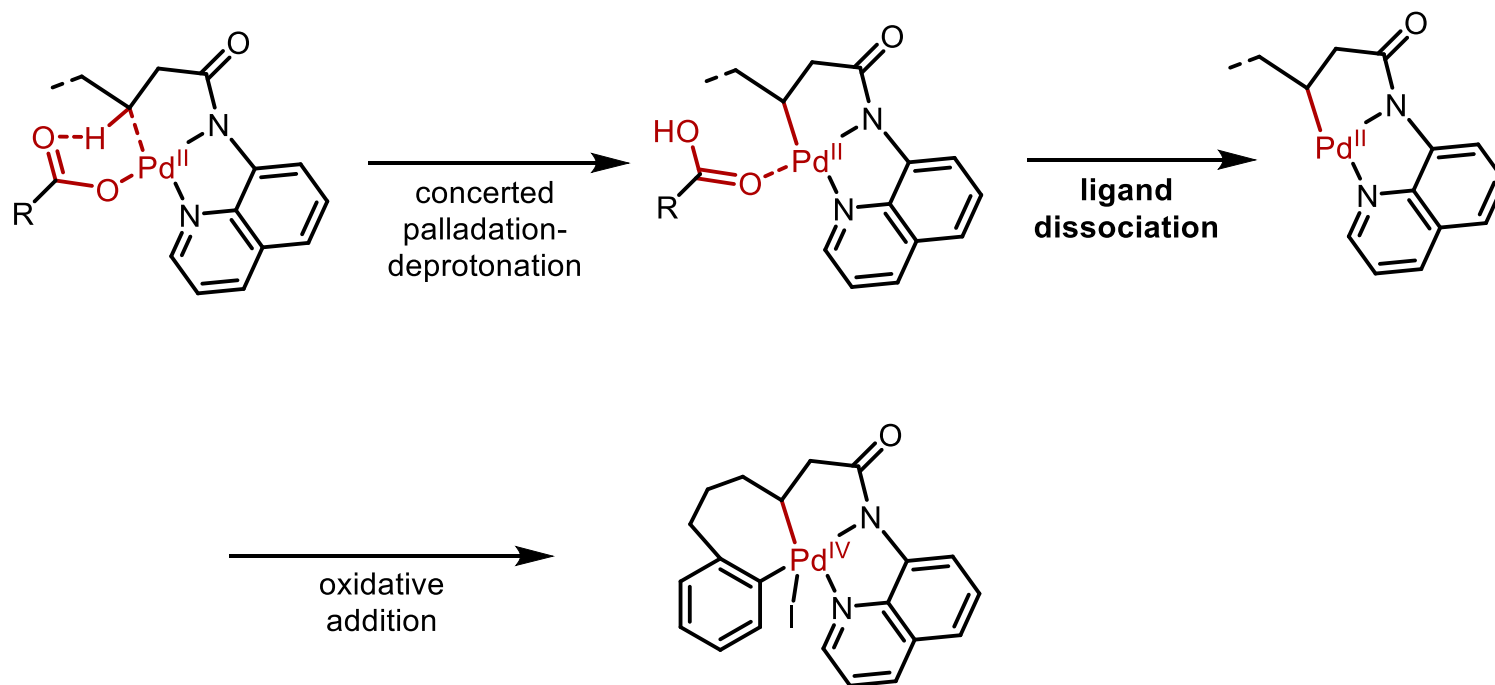
▪ Intramolecular arylation



- Acid screen (50 °C, $X=Y=Z=\text{CH}_2$)

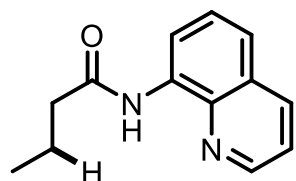


- Reasoning

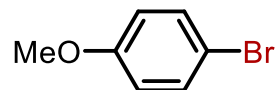


Other aryl source

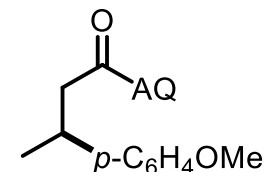
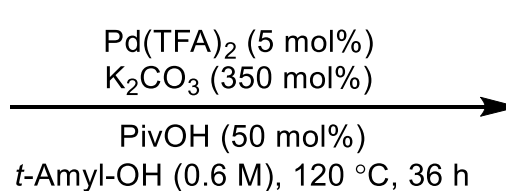
Zeng



(100 mol%)

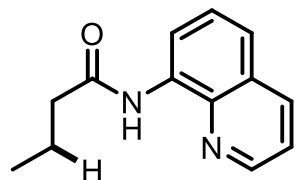


(400 mol%)

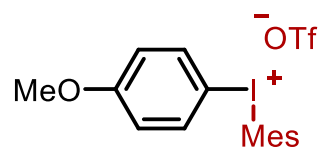


80% Yield

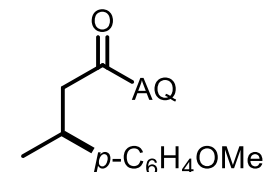
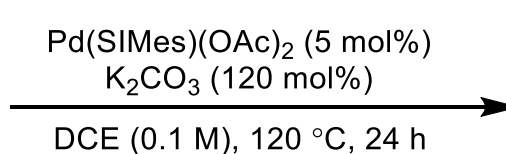
Shi



(100 mol%)



(120 mol%)



68% Yield

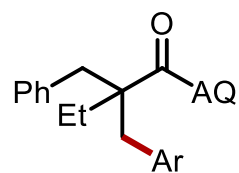
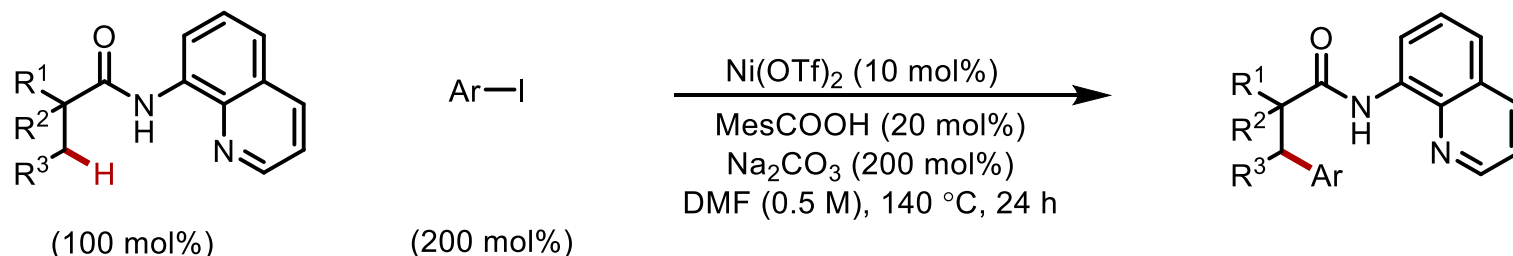
Zeng's work

- OA as rate-determining step
- Na_2CO_3 and K_3PO_4 gave lower yield
- Cs_2CO_3 and AgOAc inhibit the reaction

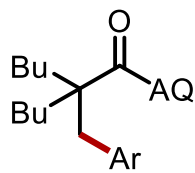
Shi's work

- ArI gave diminished yield
- Largely lowered Ar equivalence
- Lowered reactivity of catalyst

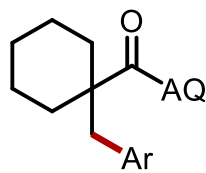
Nickel Catalysis



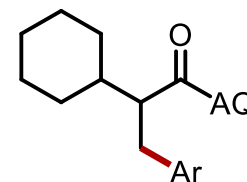
56%



50%

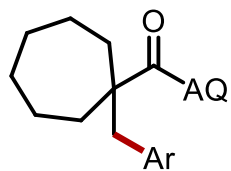


61%



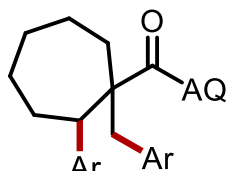
30% (40%
recovered)

$\text{Ar} = p\text{-C}_6\text{H}_4\text{OMe}$

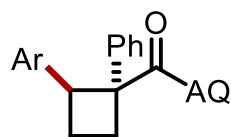


41%

and

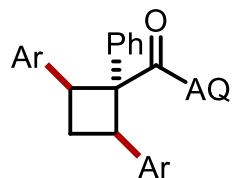


33%



50%

and

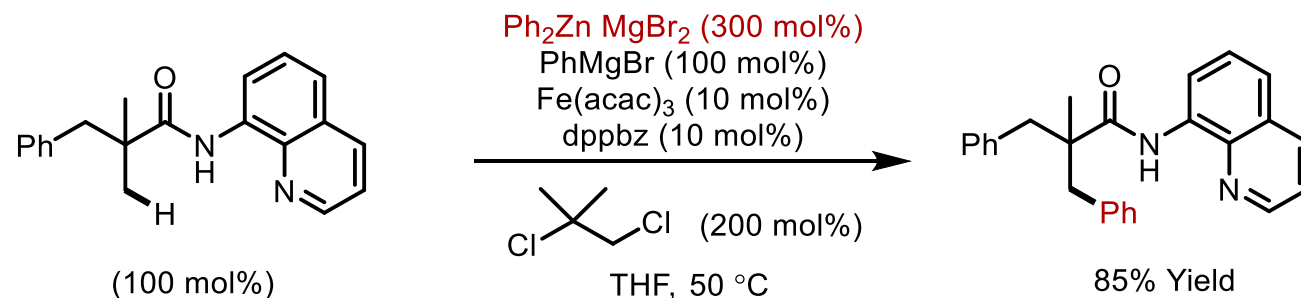


33%

Features:

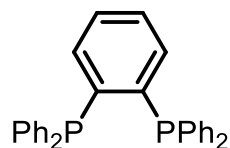
- Primary > secondary C-H bonds
- No second arylation
- α -C-H bond not tolerated
- Thio DG doesn't work
- radical pathway not supported

Iron Catalysis



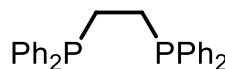
no
ligand

0%



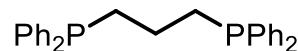
dppbz

80%



dppe

9%

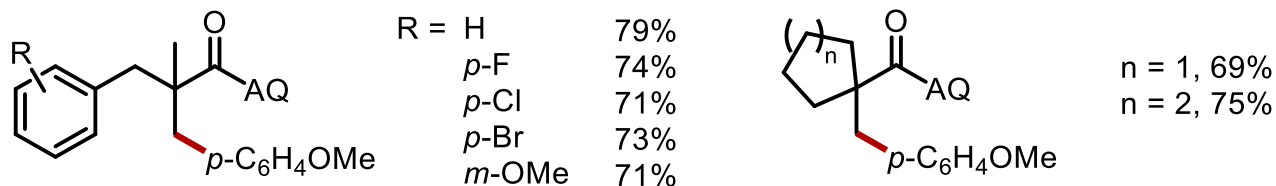


dppe

0%

Features:

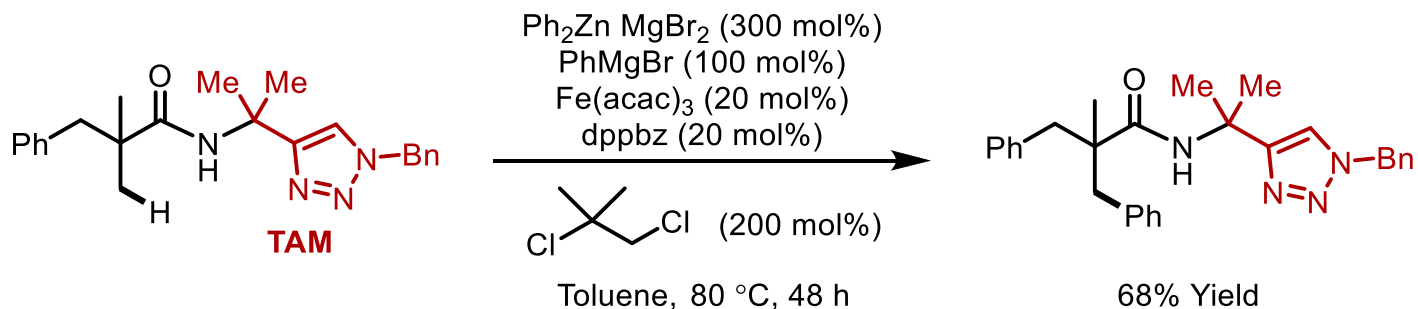
- 7 equiv. Grignard mixed with 3 equiv. zinc
- 1 equiv. Grignard for deprotonation
- Sensitive to ligand structure



Features:

- α -C-H bond not tolerated
- Primary > secondary (benzylic) C-H bonds
- No second arylation

TAM as directing group

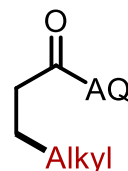
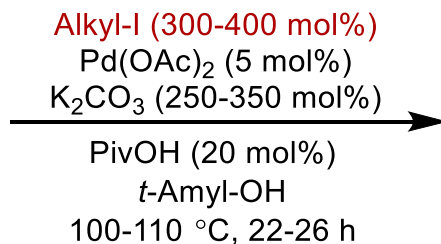
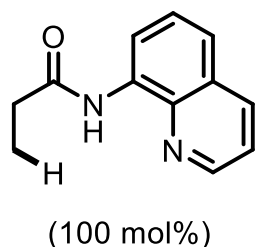


▪ Pd catalysis

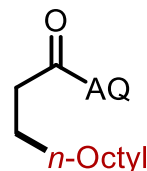
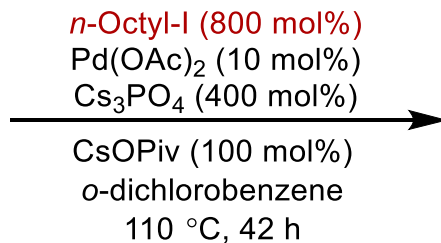
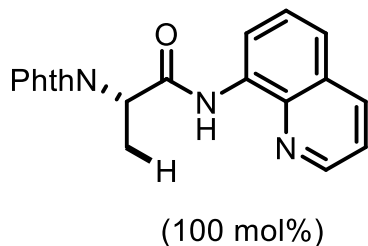
Challenges:

- Difficult C-H insertion
- Difficult oxidative addition
- Difficult alkyl-alkyl reductive elimination

▪ Three examples from Daugulis



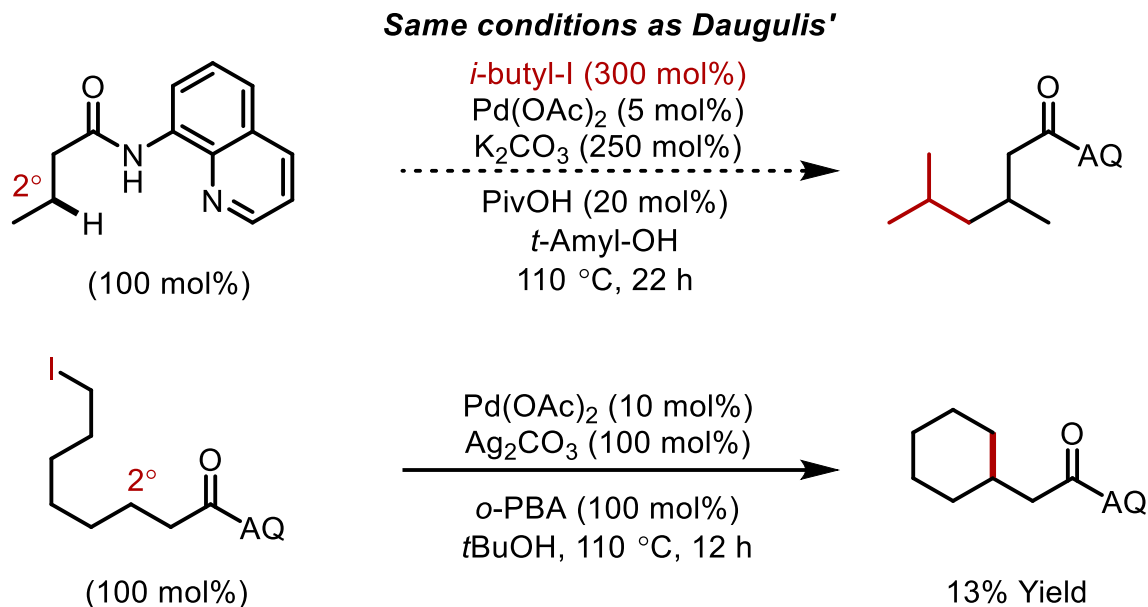
Alkyl = *i*-butyl 58% Yield
Alkyl = *n*-octyl 47% Yield



42% Yield

- Primary C-H bond only
- Low efficiency
- Ag considered bad

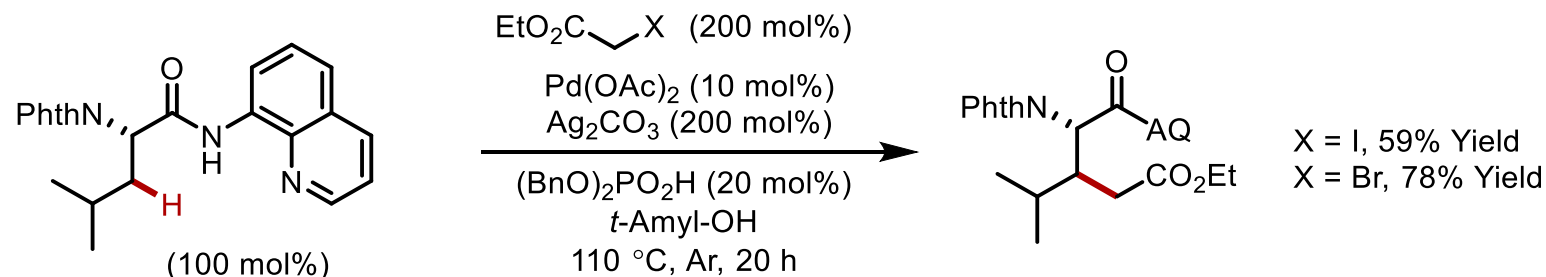
Chen



- Secondary C-H palladation should not be a problem
- Difficulty of OA and RE (**RI's problem**)
- Side reactions: esterification, E2...(**RI's problem**)

Better electrophile required

Chen

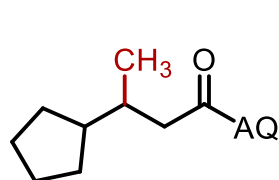


Ag^+ :

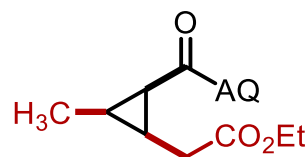
- Promotes OA via $\text{S}_\text{N}2$
- Halide scavenger

$(\text{BnO})_2\text{PO}_2\text{H}$:

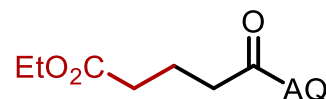
- 1.0 equiv. gave lower yield
- Form soluble complex with Ag salt



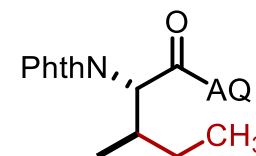
92%



First w/ ester: 80%
second w/ MeI: 66%



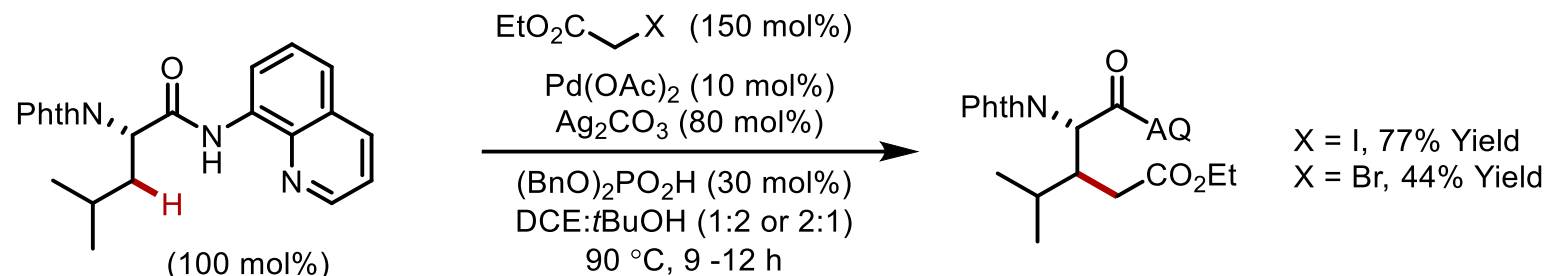
90%



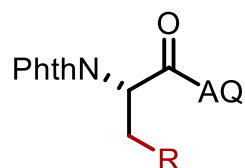
62%, dr ~8:1

Secondary alkyl halide didn't work

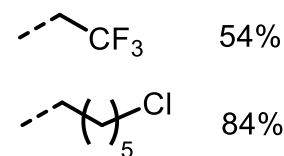
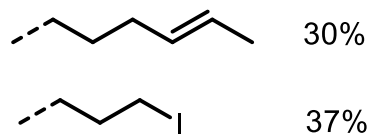
Shi



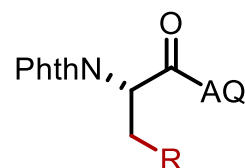
w/ alkyl iodide



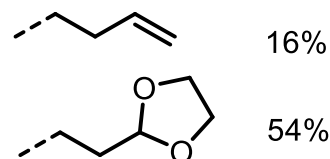
$\text{R} =$	Me	55%
	Et	79%
	<i>n</i> Pr	84%
	<i>i</i> PrCH ₂ CH ₂	85%



w/ alkyl bromide

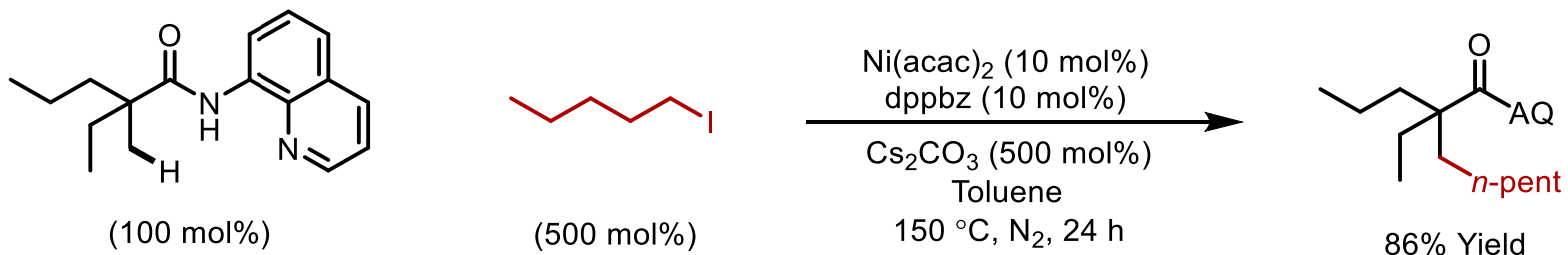


$\text{R} =$	Me	79%
	Et	78%
	<i>n</i> Pr	54%
	<i>i</i> PrCH ₂ CH ₂	43%

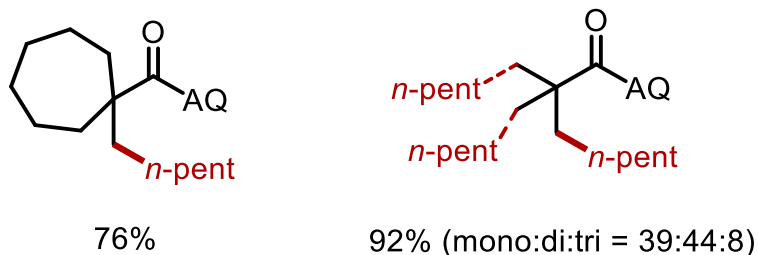


Secondary alkyl halide didn't work

▪ Ni catalysis

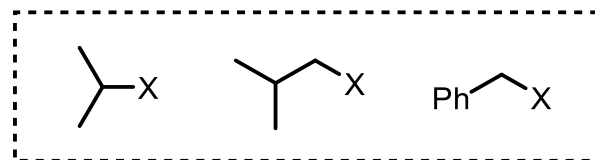


Amide scope

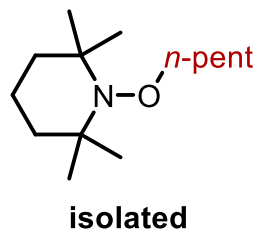


- Quaternary α -carbon required
- No alkylation for secondary C-H

Alkyl halide scope

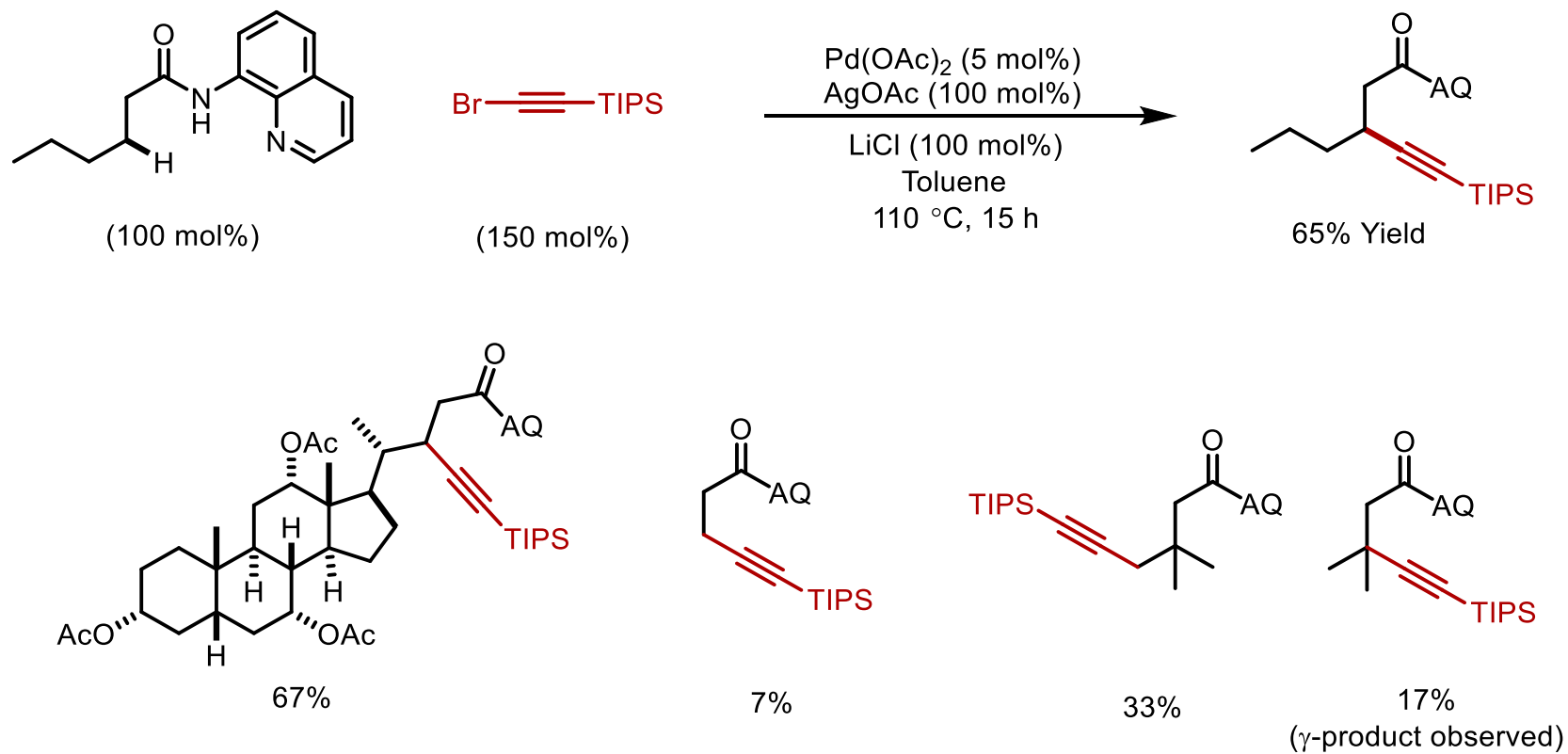


- Alkene, cyano tolerated
- Alkyl-Br and -Cl worked with CsI



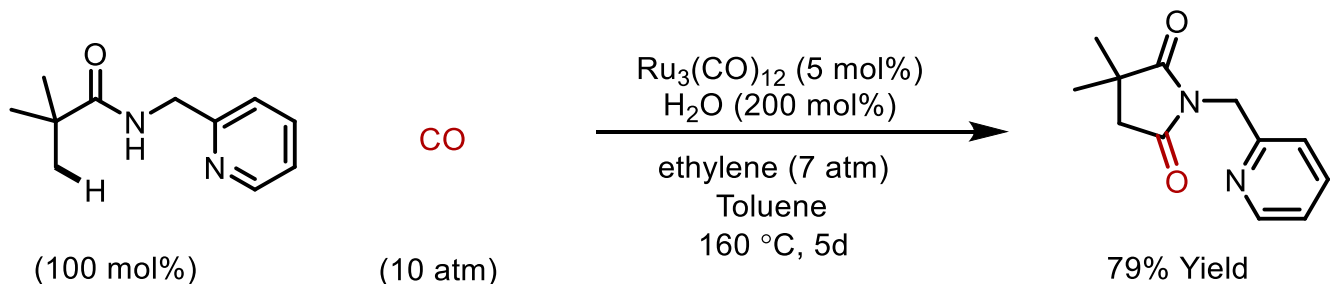
- TEMPO inhibit the reaction
- C-H activation as rds

DG Strategy-Bidentate-Alkynylation

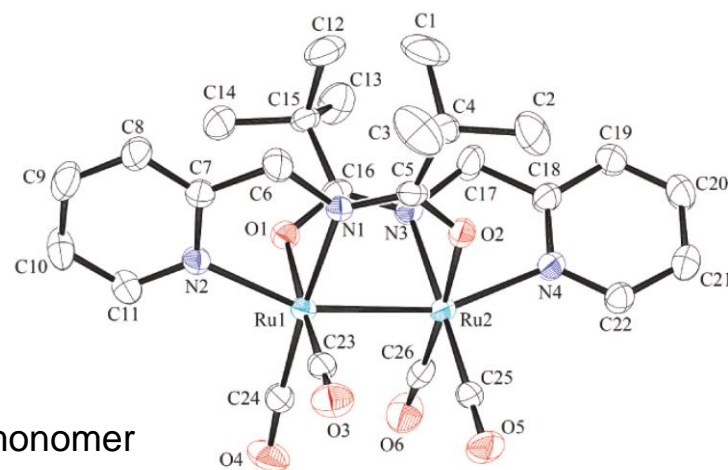
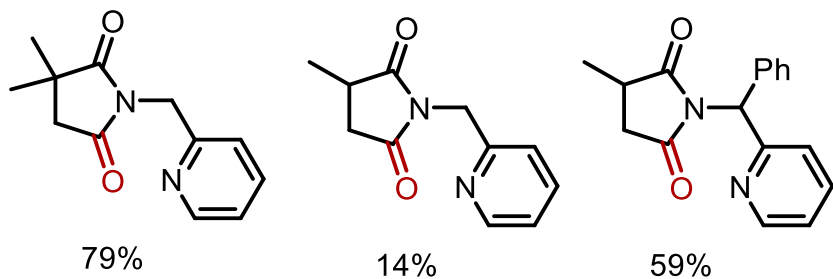


- $-\text{NHC}_6\text{F}_5$ not working as directing group
- Only TIPS acetylene bromide as electrophile
- Selective for secondary C-H bonds

Ru catalysis



- No ethylene, no carbonylation
- Water increases the efficiency of the carbonylation
- Methyl C-H bond > methylene C-H bond

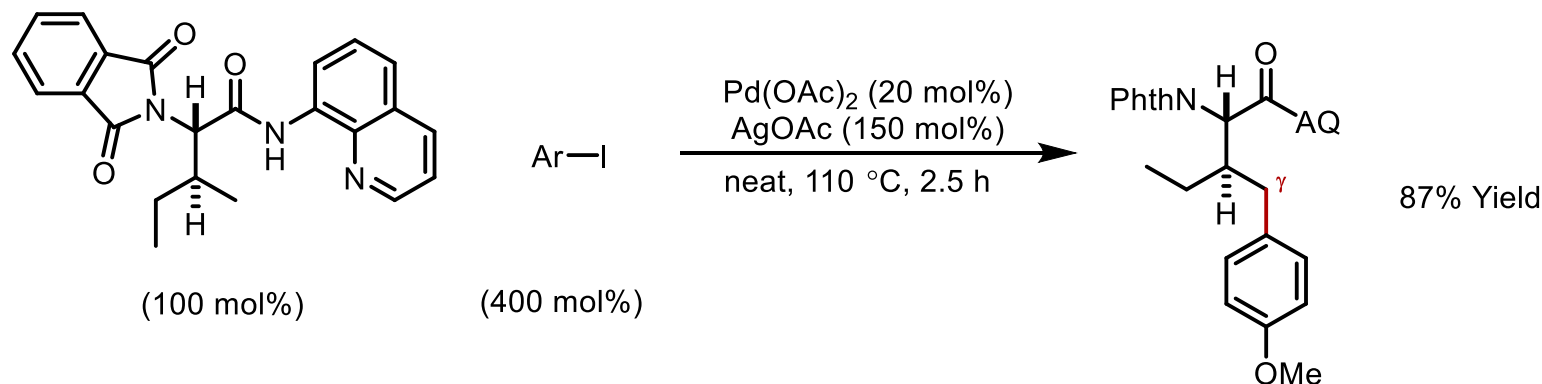
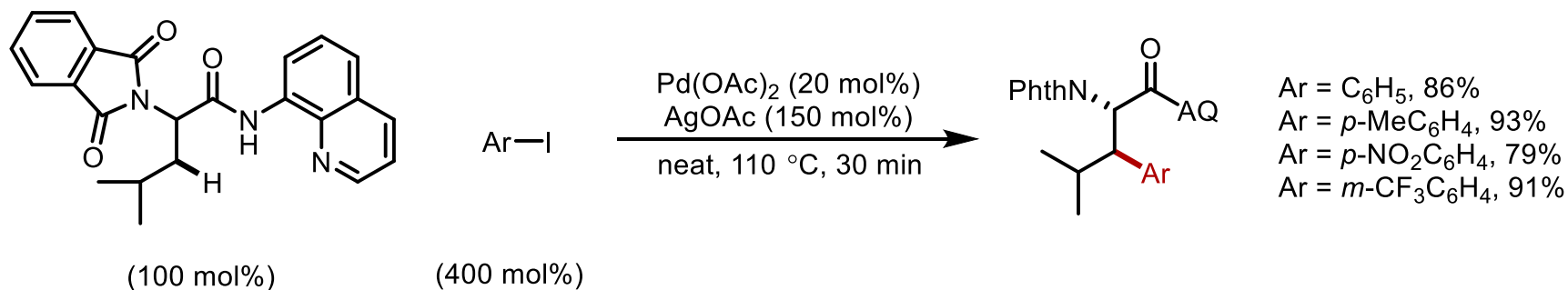


Ru(I) dimer

- Ru(I) dimer as catalyst resting state
- React with water to release CO_2 and Ru(II)-hydride monomer
- Ru(II)-hydride is in the catalytic cycle

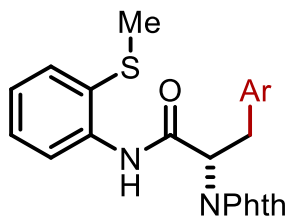
■ Unnatural amino acid

Corey



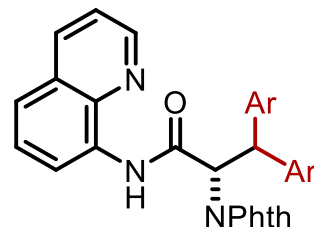
Unnatural amino acid

Daugulis

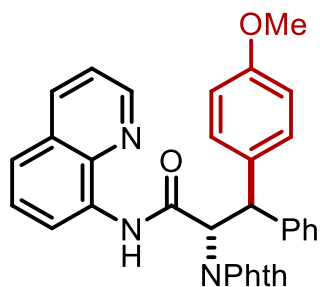


Ar = Ph
 Ar = *p*-OMeC₆H₄
 Ar = 2-naphthyl
 Ar = 2-benzothiophenyl
 Ar = 3-(1-methylindolyl)
 Ar = 3-azidophenyl

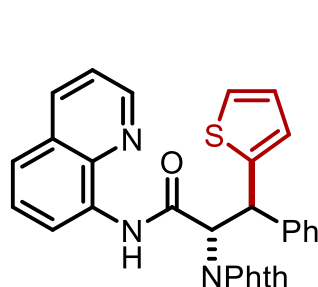
78%
 68%
 60%
 74%
 61%
 81%



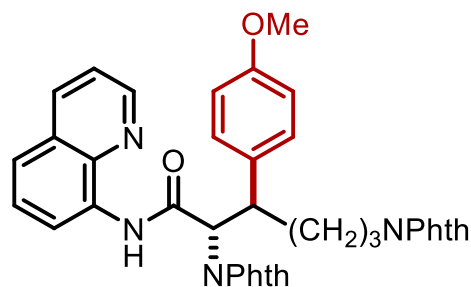
Ar = 3,4-(CH₃)₂C₆H₃ 92%
 Ar = 4-EtO₂CC₆H₄ 84%



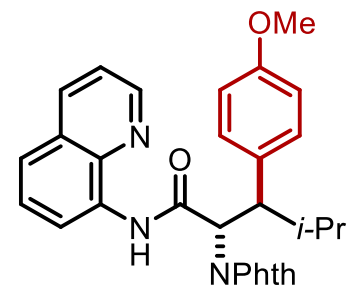
91%, dr 24:1



95%, dr >50:1



85%, dr 16:1

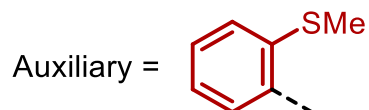
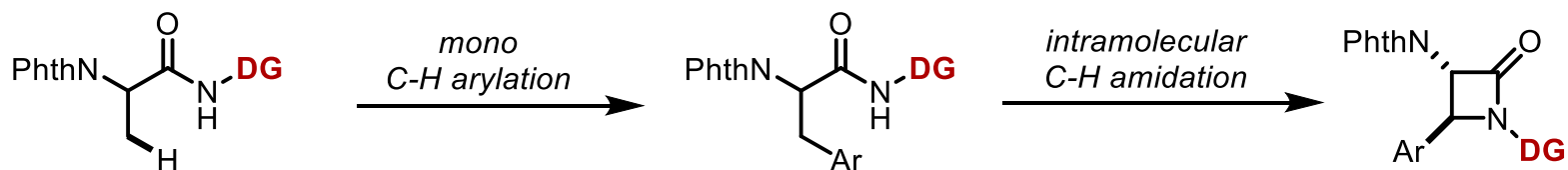


77%, dr >50:1

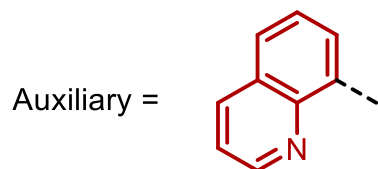
- ee eroded by less than 10%

▪ α -amino- β -lactams

Designed Strategy

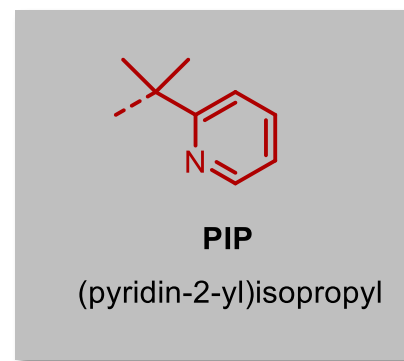


- selective mono-arylation
- fragile in oxidative amidation

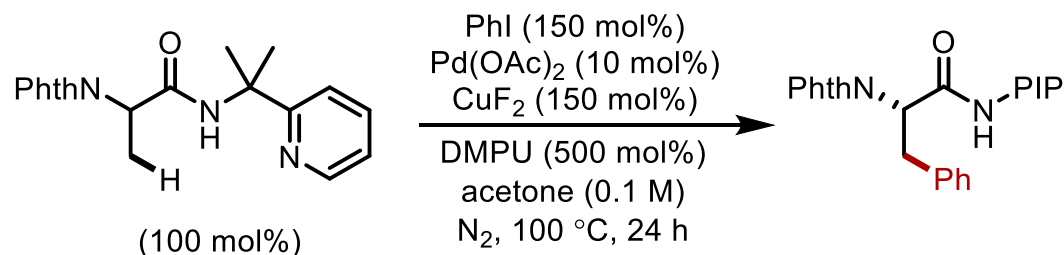


- problematic mono-selectivity

New DG employed

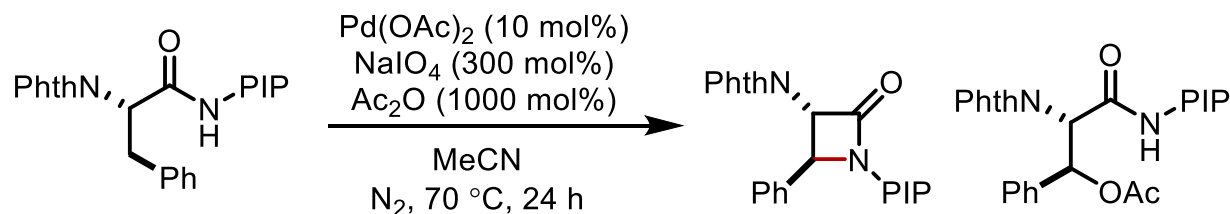


▪ α -amino- β -lactams



- CuF₂ as iodide scavenger
- Cu(OAc)₂ gave 43% yield
- DMPU improves selectivity

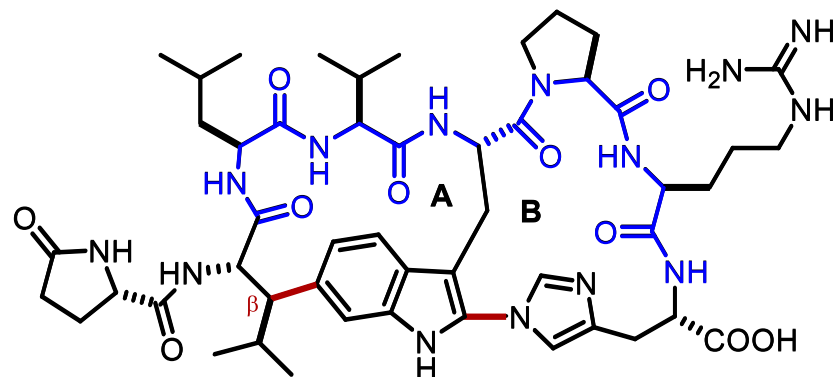
(81% Yield, 99% ee, mono:di 25:1)



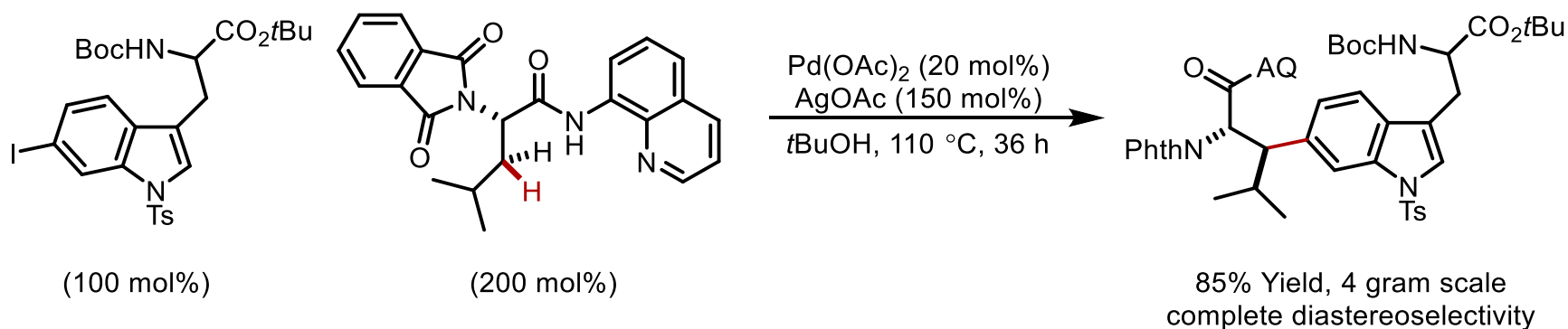
(74% Yield, amidation:acetoxylation 12.5:1)

- NaIO₄ good with FG, but not soluble
- NaIO₄/Ac₂O gives IO_{4-n}(OAc)_{2n-1}, soluble

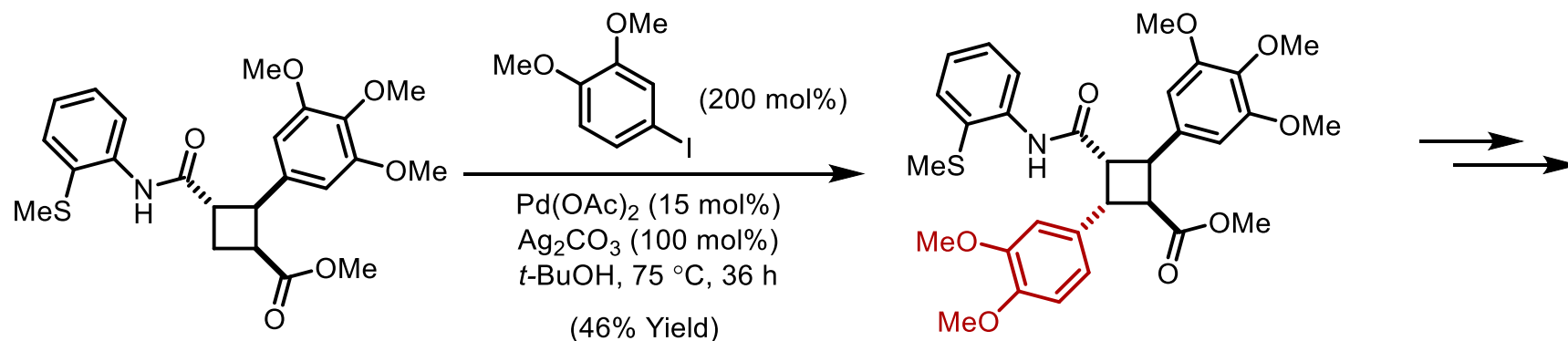
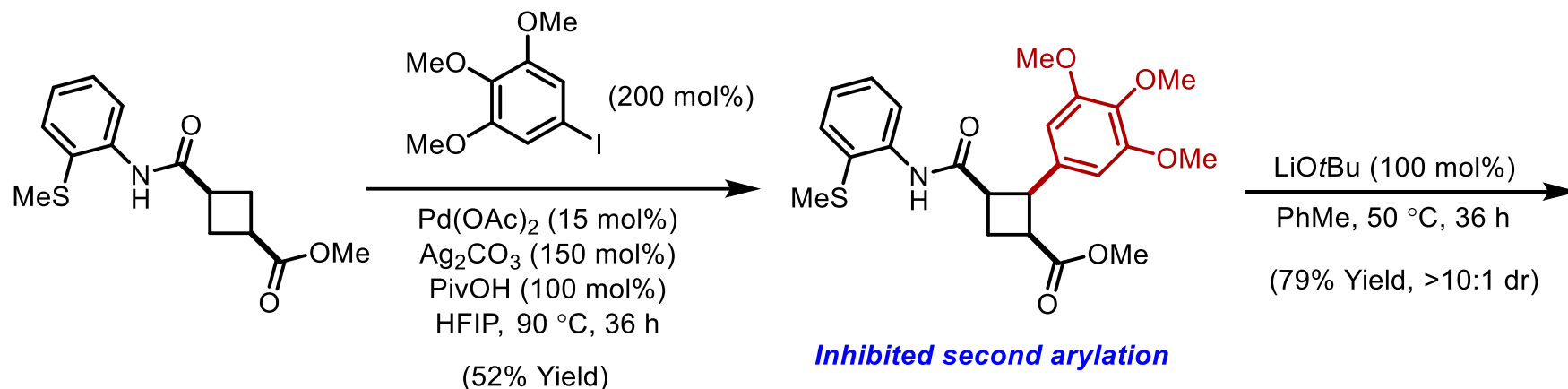
■ Total Synthesis of Celogentin C



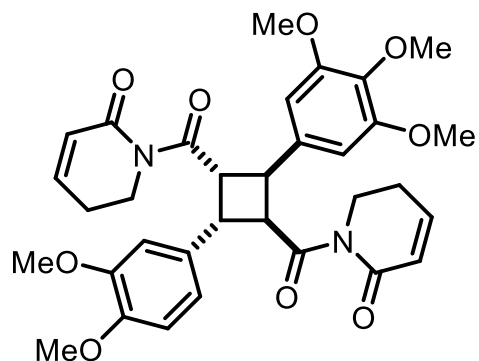
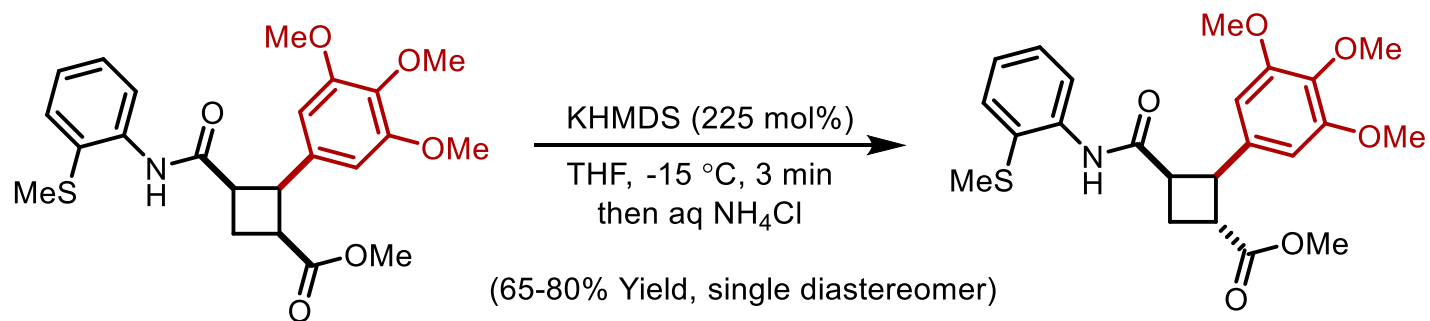
Celogentin C



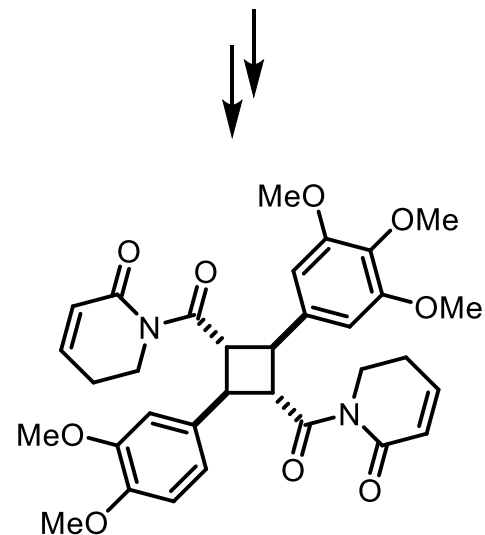
■ Total Synthesis of Piperarborenine B



■ Total Synthesis of Piperarborenine B

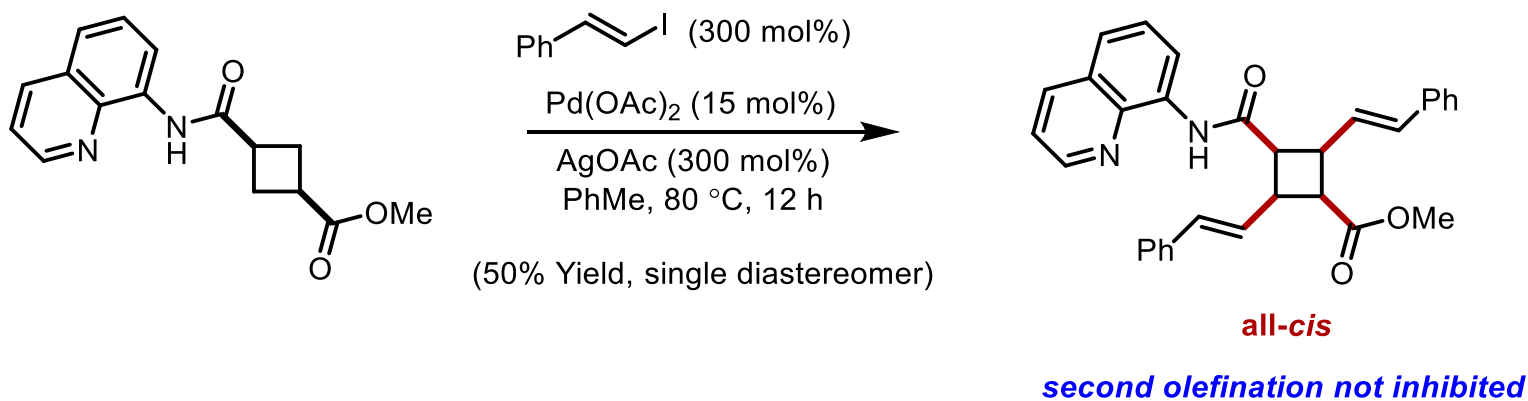


Piperarborenine B

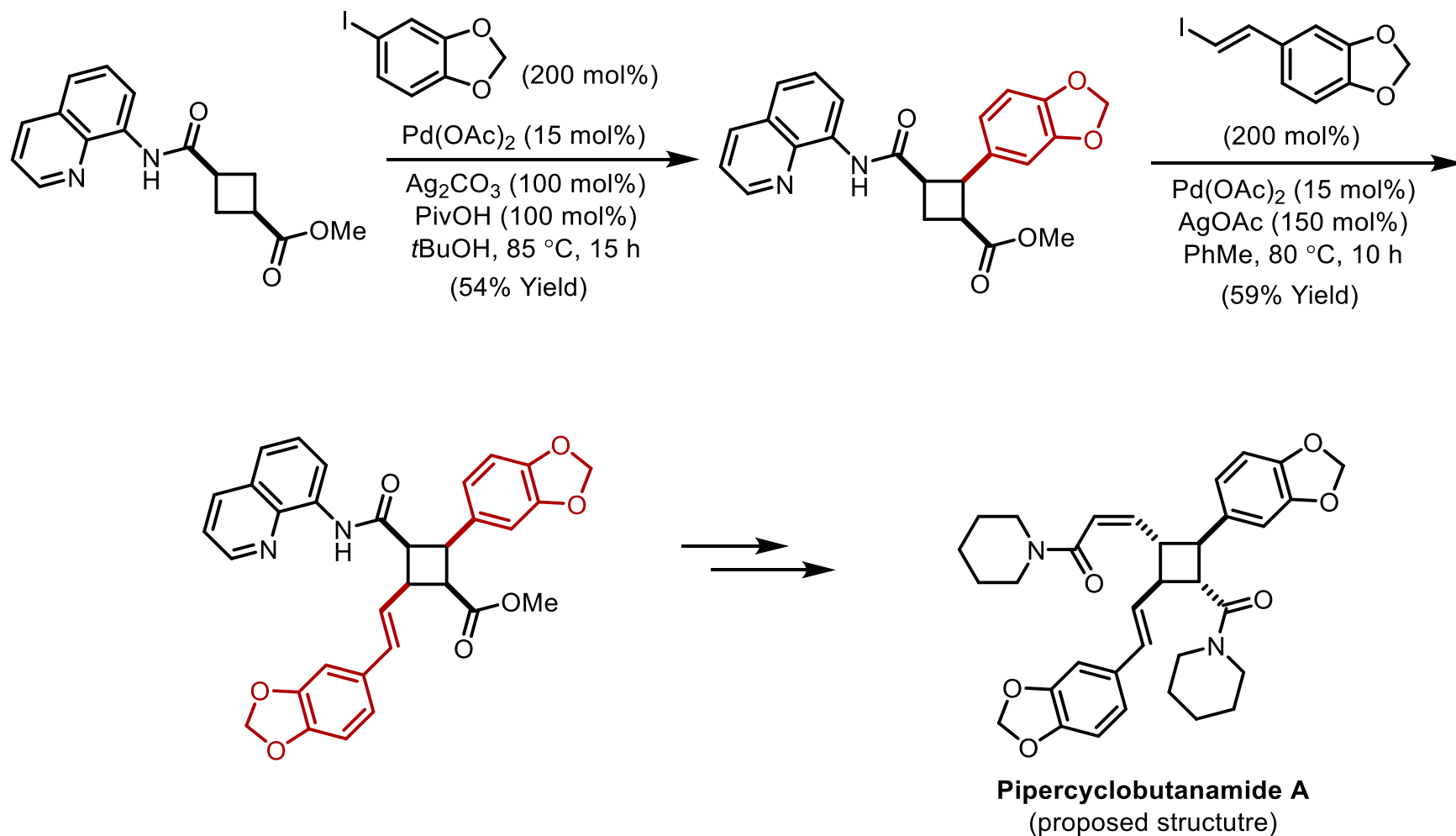


**Piperarborenine D
(proposed structure)**

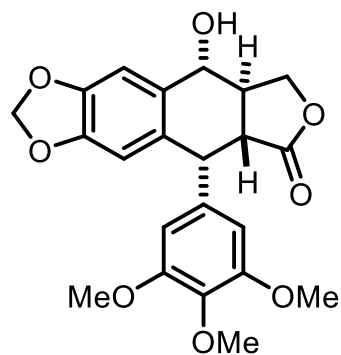
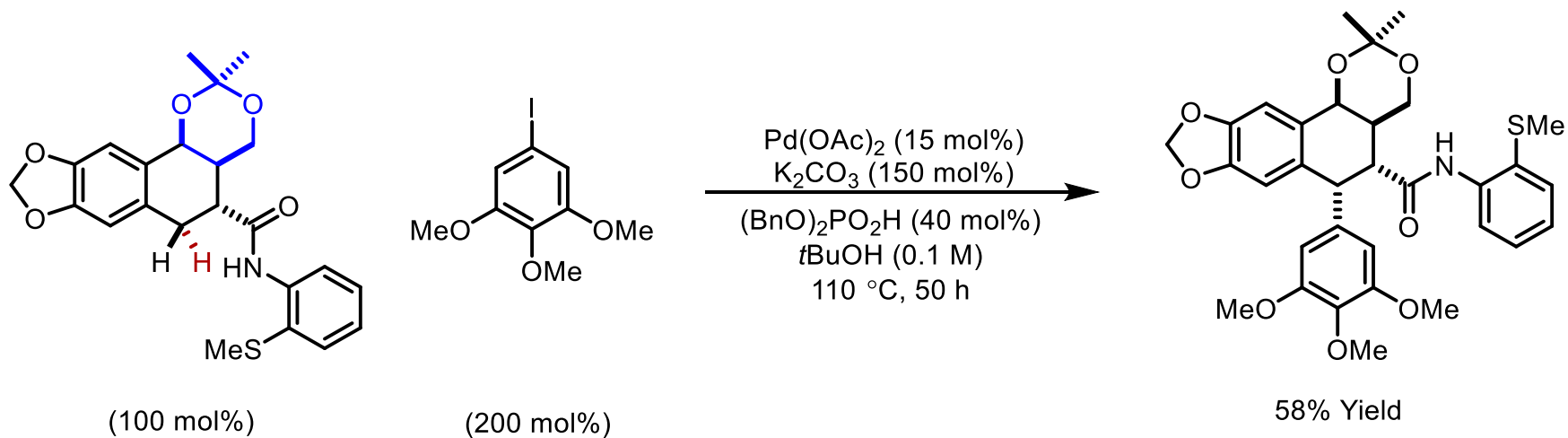
- Pipercyclobutanamide A from sequential arylation and olefination

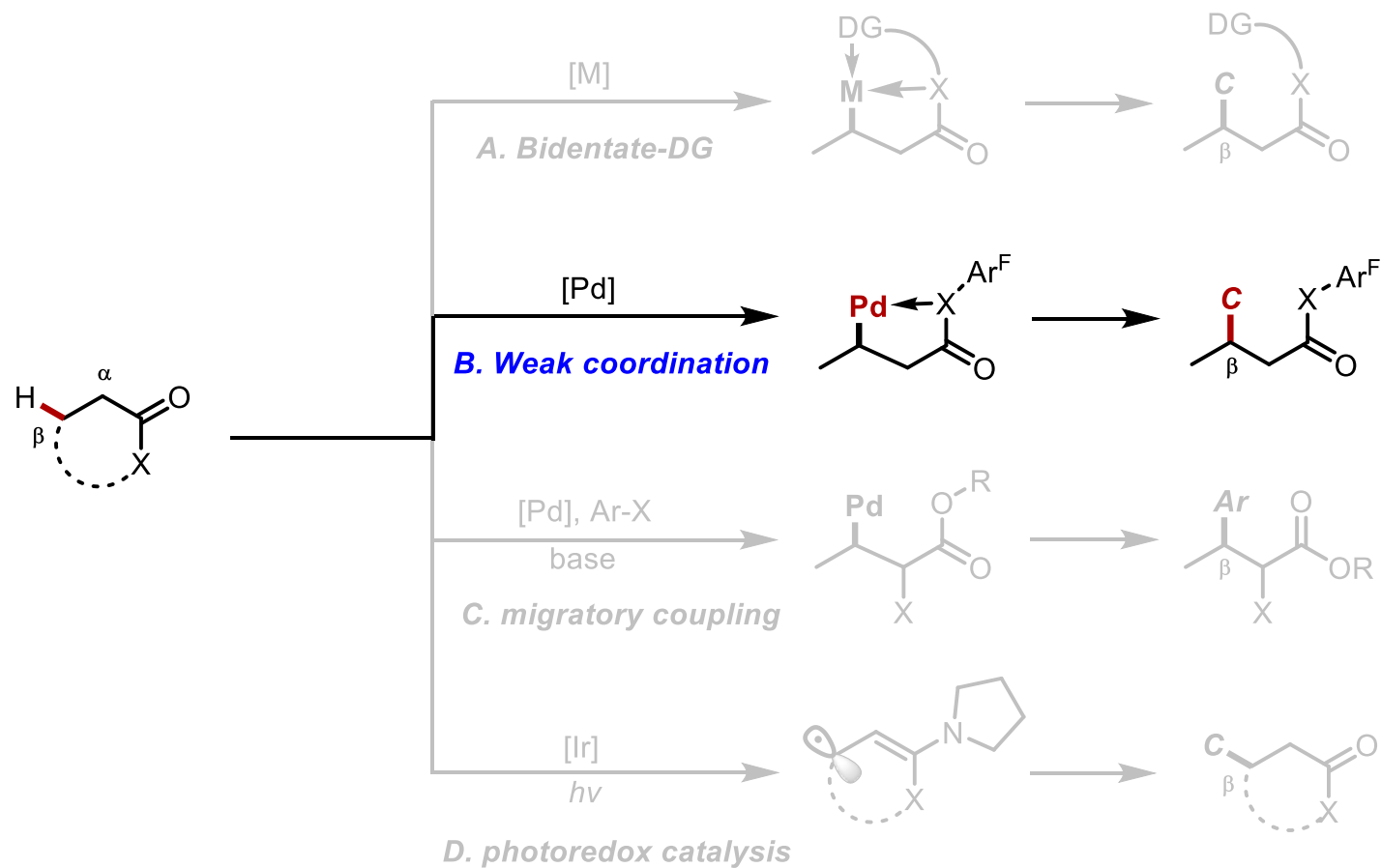


■ Pipericyclobutanamide A from sequential arylation and olefination

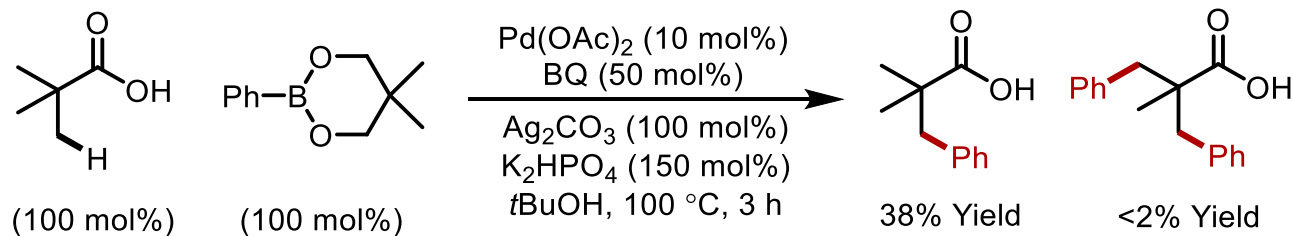


■ Total Synthesis of Podophyllotoxin

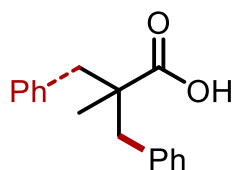
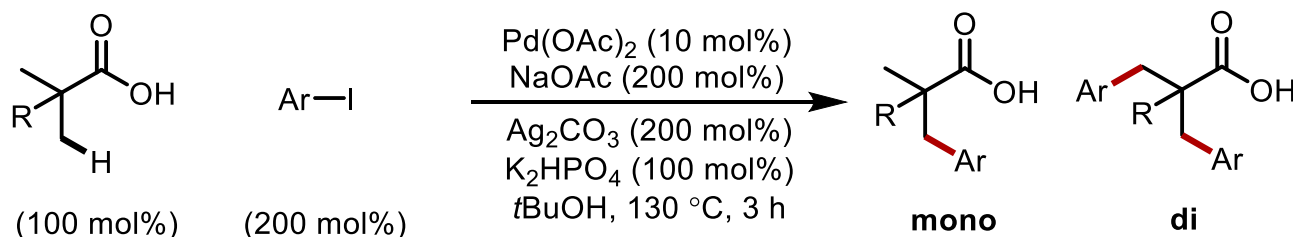




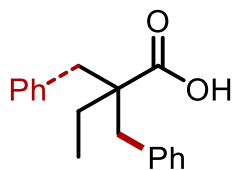
Carboxylic acid as DG



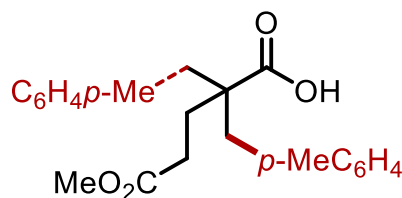
- In-situ generated K salt
- Low efficiency
- Pd(0)/Pd(II) proposed



70%
(mono:di 5:2)



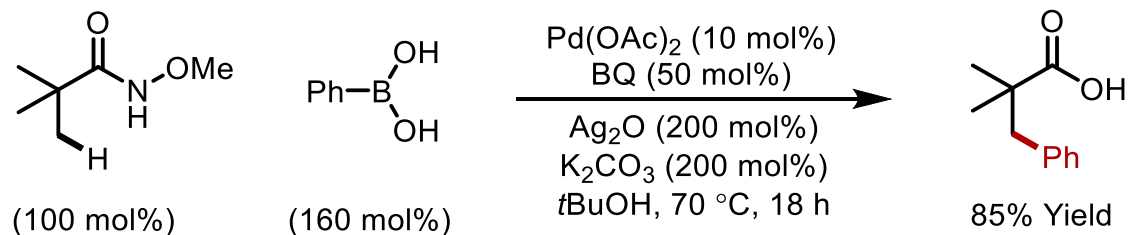
72%
(mono:di 4:2)



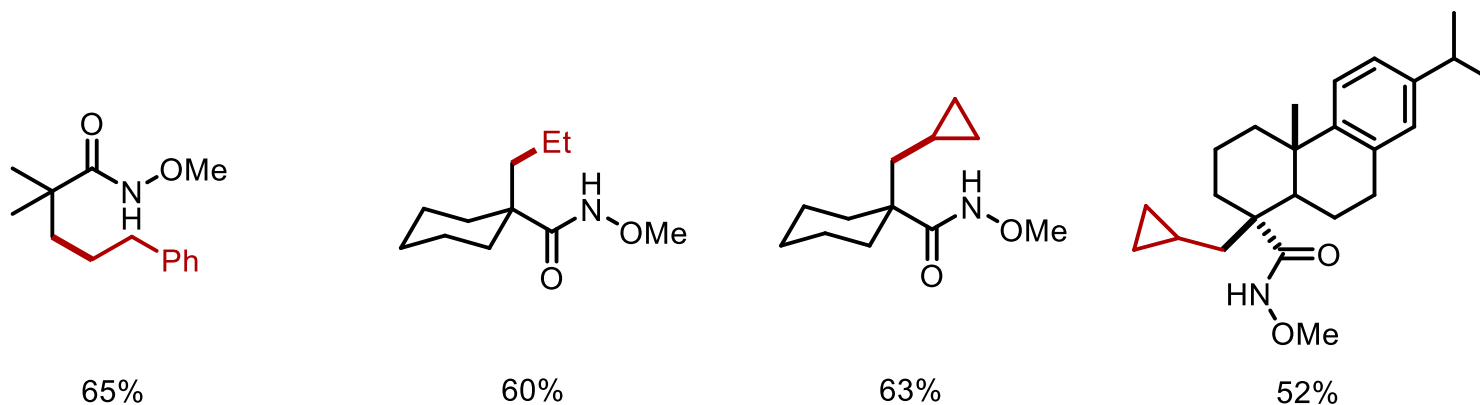
43%
(mono:di 5:1)

- Ag as iodide scavenger
- NaOAc also helps
- Pd(II)/Pd(IV) proposed
- Methyl C-H only
- α -proton not tolerated

Stronger DG

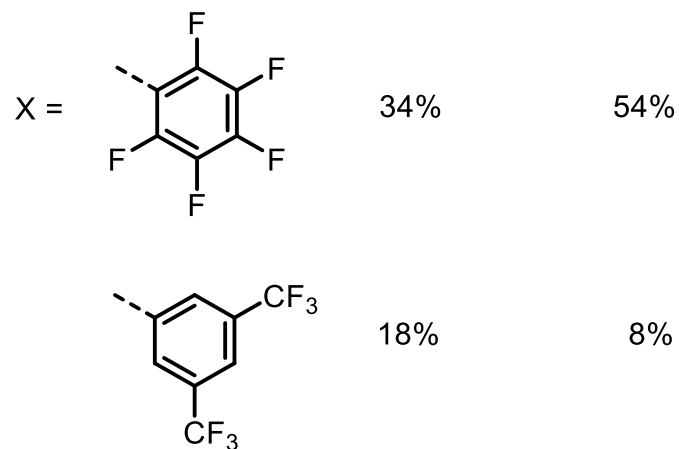
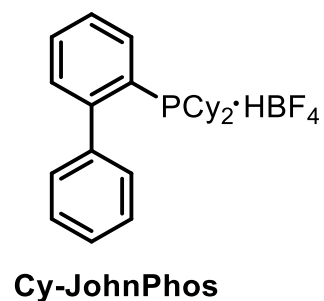
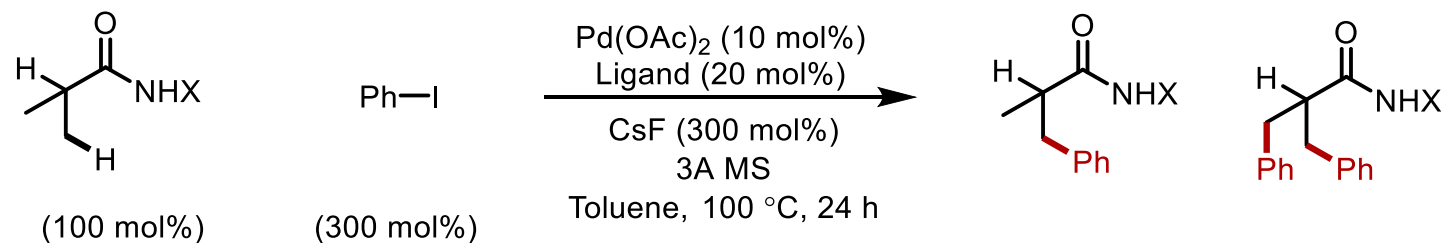
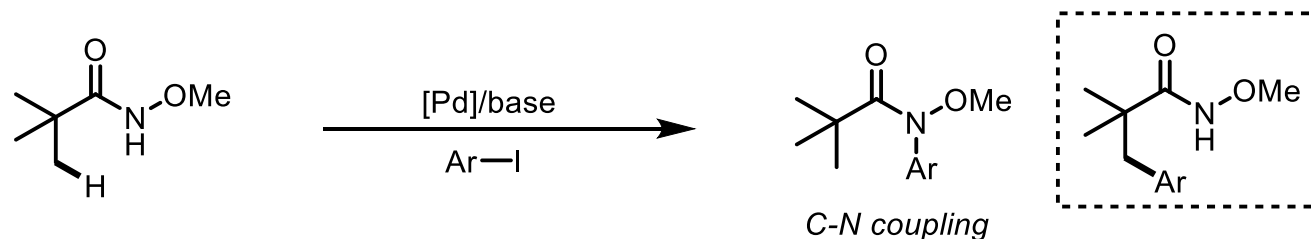


2,2,5,5-tetramethyl THF as solvent

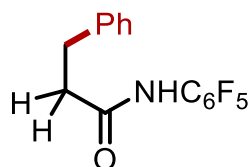


- Pd(0)/Pd(II) proposed
- Air (20 atm) can be used as oxidant
- Methyl C-H only
- α -proton not tolerated

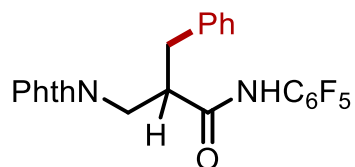
Acidic amide as DG



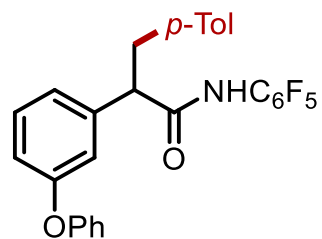
▪ Acidic amide as DG



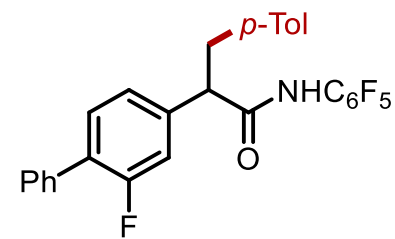
58%



64%

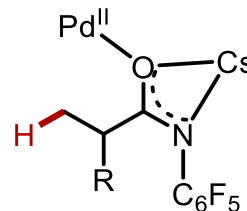
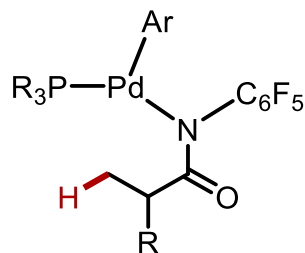


68%

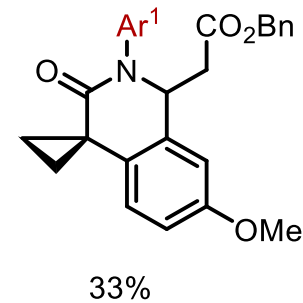
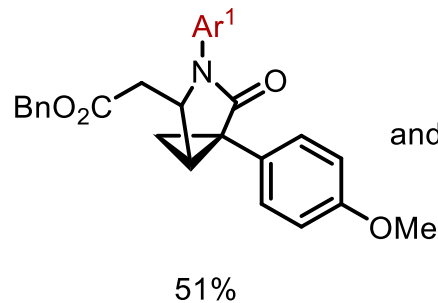
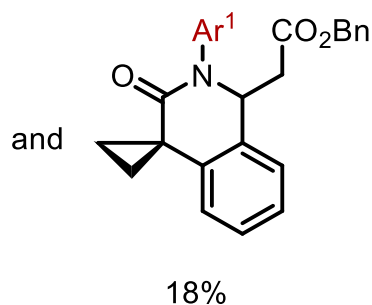
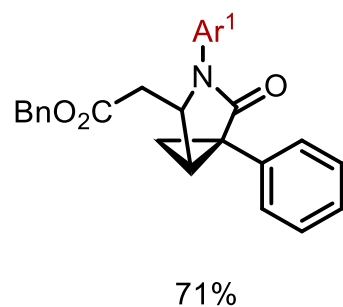
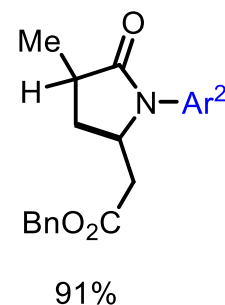
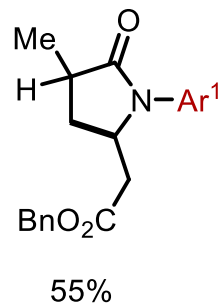
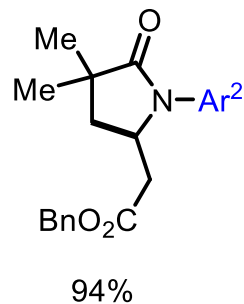
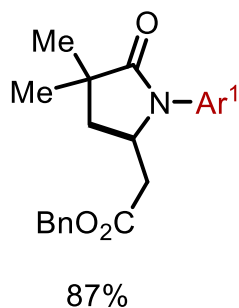
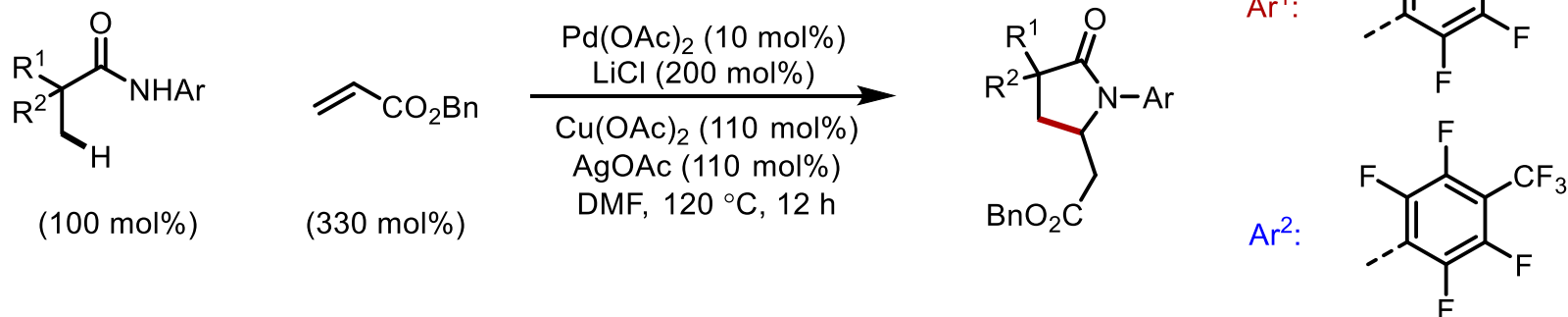


72%

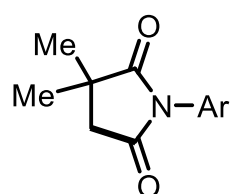
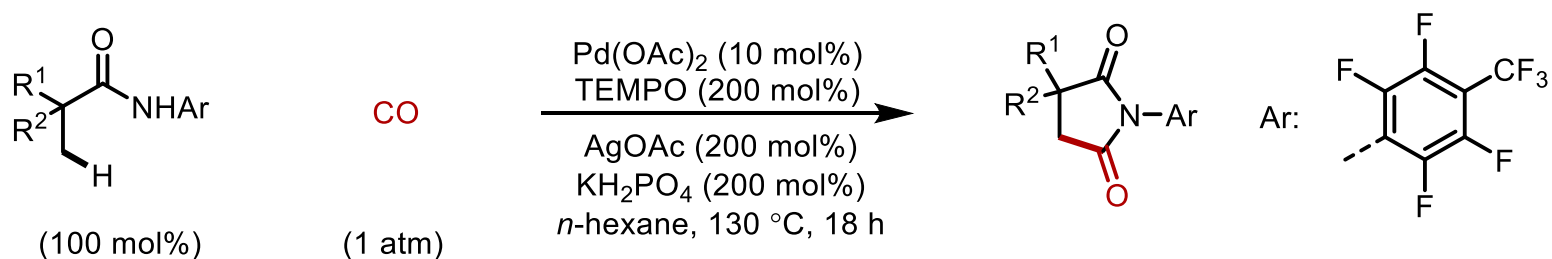
- α -proton tolerated under weaker CsF
- Methyl C-H only
- No C-N cross coupling found



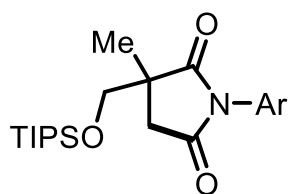
▪ Olefination



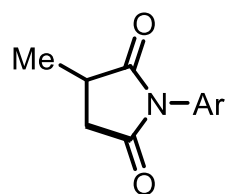
▪ Carbonylation



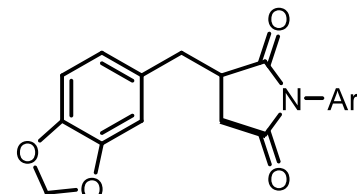
94%



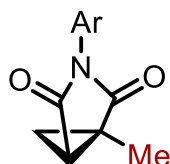
70%



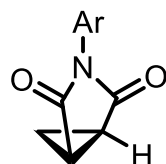
65%



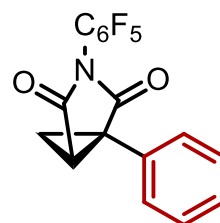
52%



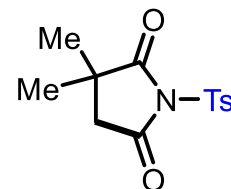
86%



65%



65%

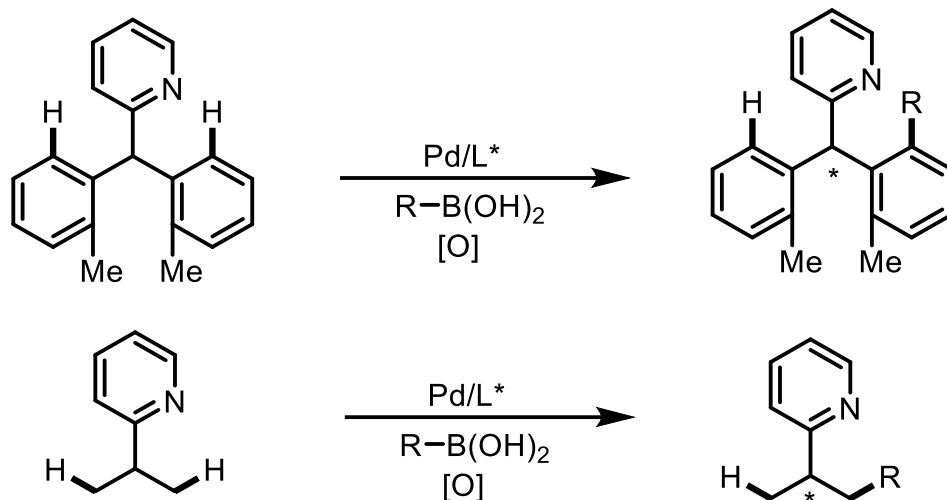


61%

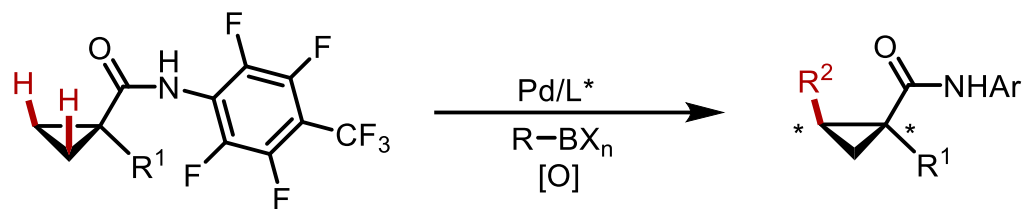
- In this case, cyclopropyl methylene C-H > methyl C-H and sp² C-H

▪ Enantioselective C-H activation of cyclopropanes

Previous work



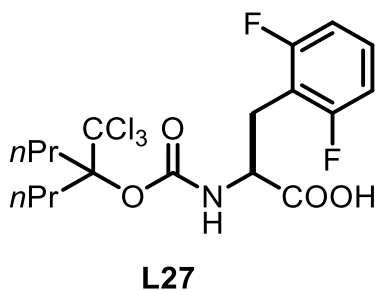
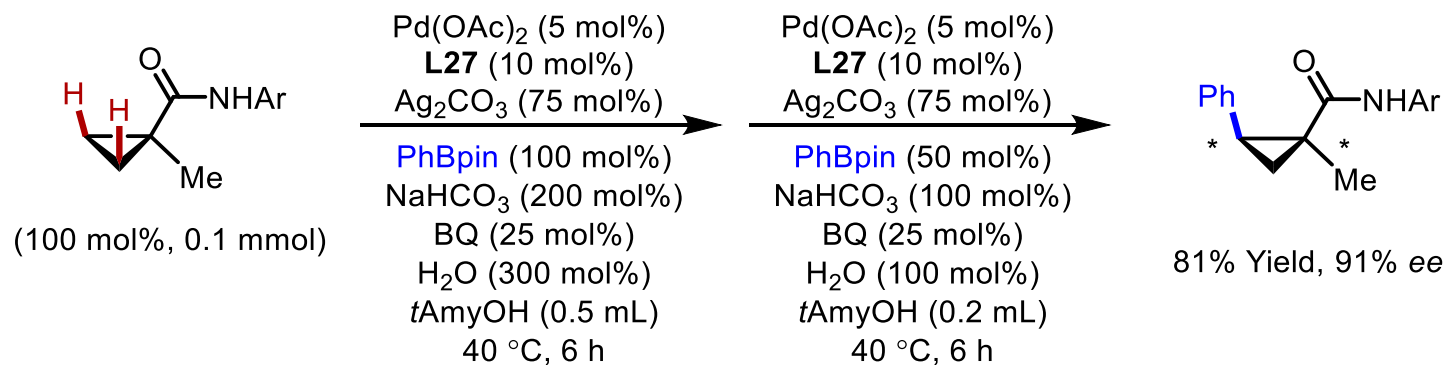
This work



- special acidic methylene
- desymmetrization?

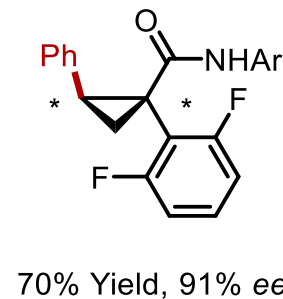
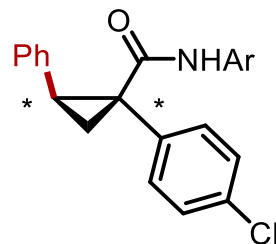
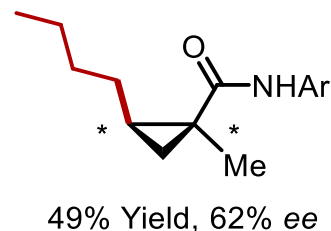
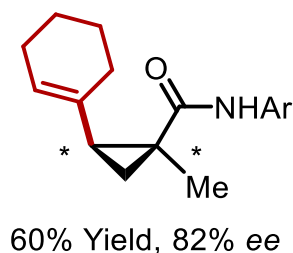
Enantioselective C-H activation of cyclopropanes

- First use of amino acid ligand in the area

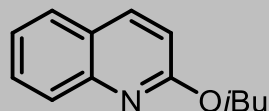
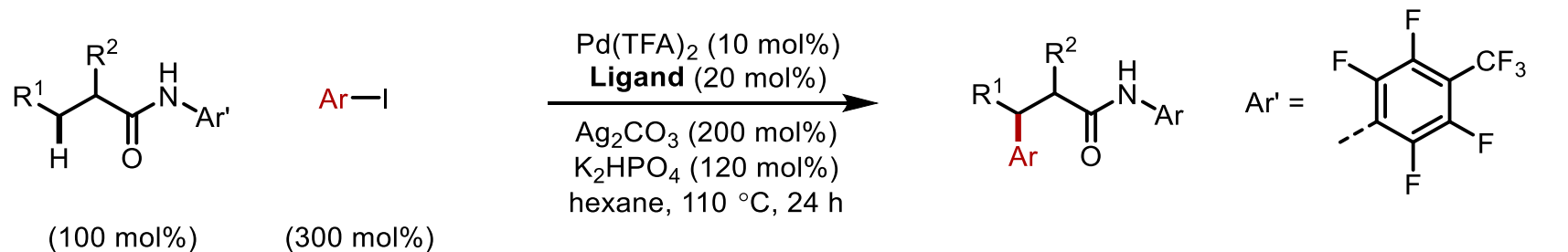


Take-home message

Don't give up!

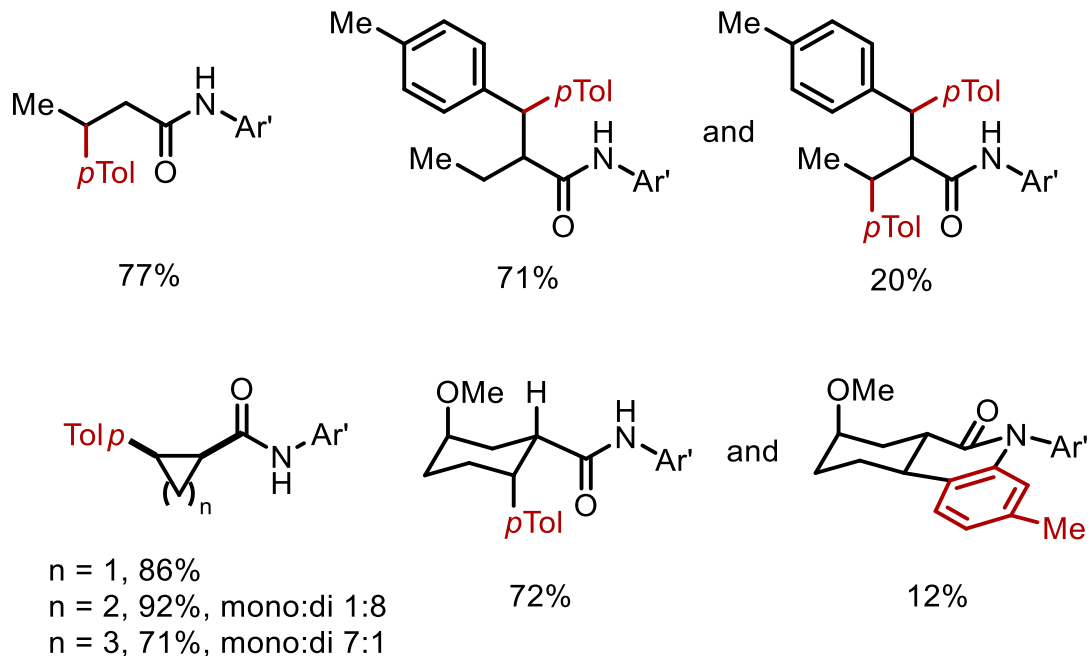


Ligand-enabled 'real' methylene C-H activation

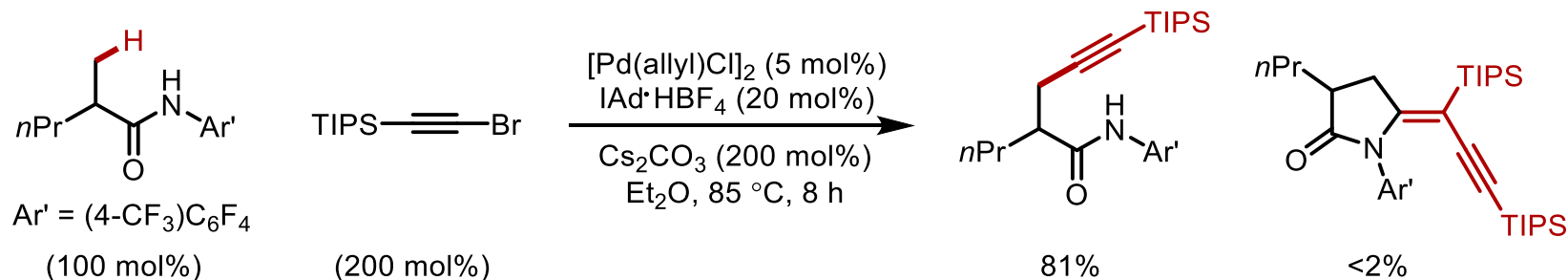


Ligand

- steric bulk (not too bulk)
- electron-donating
- singly bound fashion



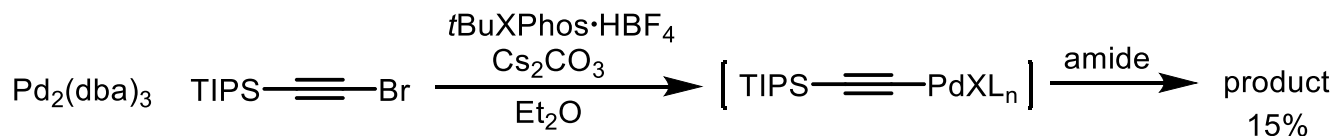
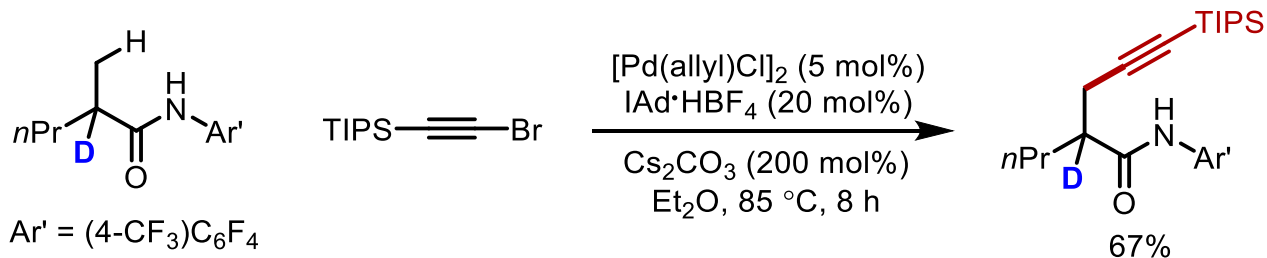
Alkynylation



Scope:

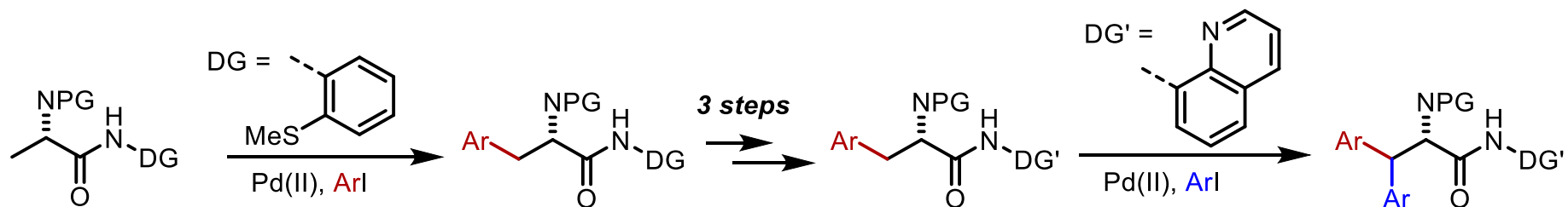
- α-proton required (pivalic acid gave low yield)
- Methyl C-H only

Bulkiness of Pd(II)-alkynyl species

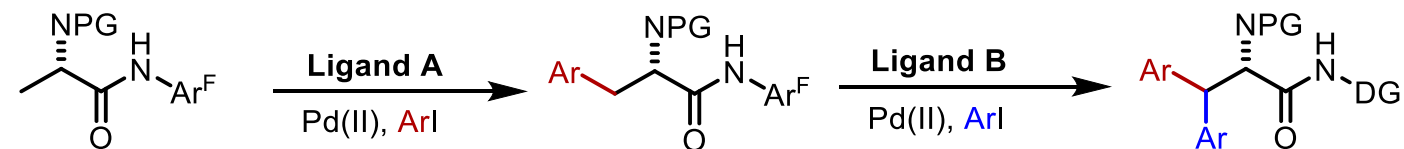


- β,β'-hetero-diarylation of amino acid

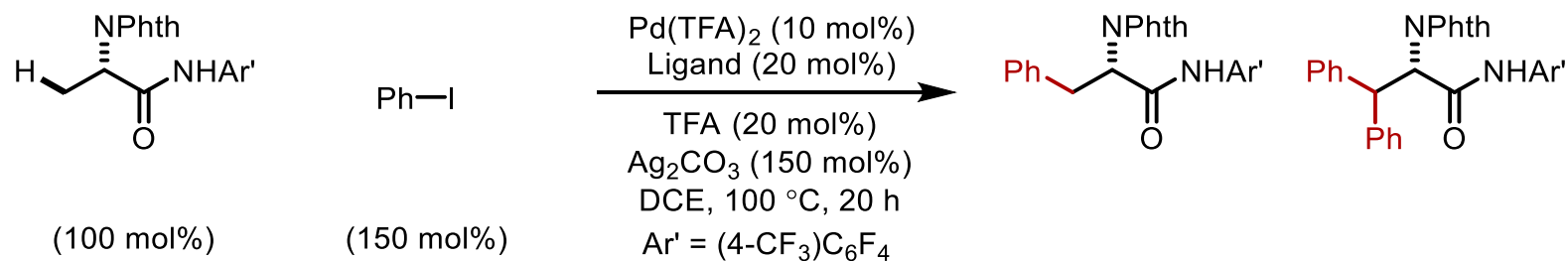
DG-controlled route



Ligand-controlled route

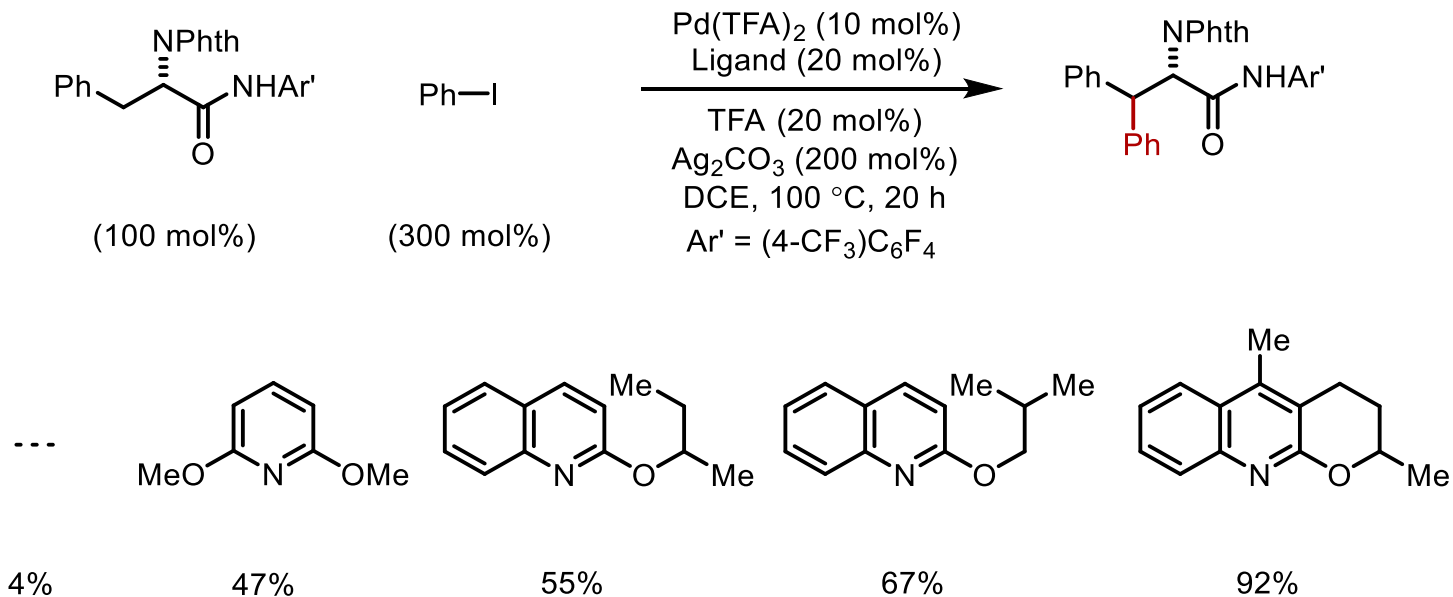


β,β'-hetero-diarylation of amino acid



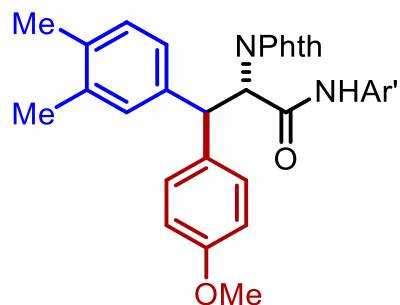
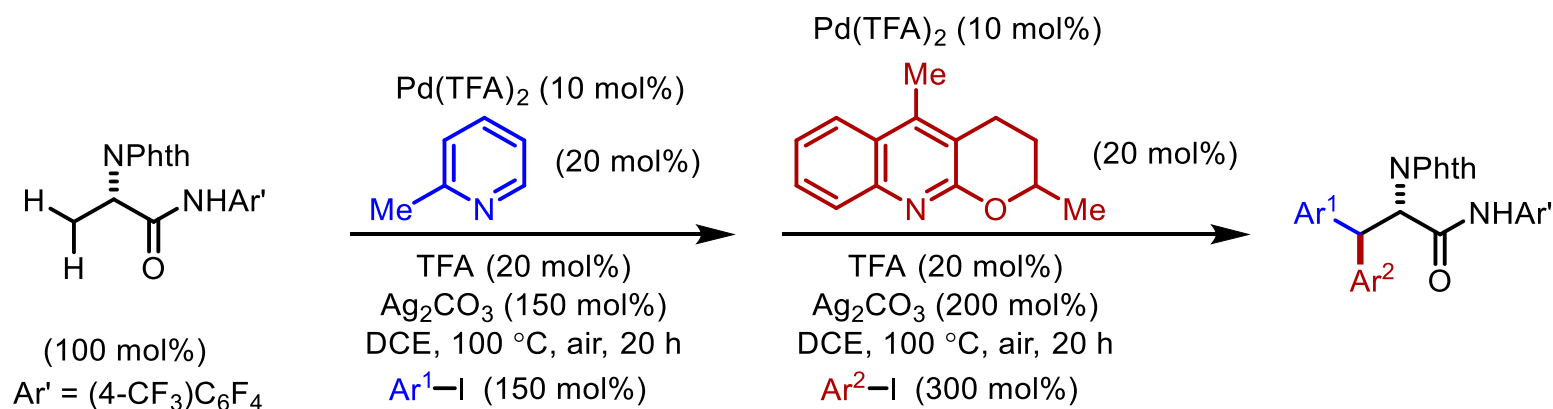
mono:	47%	29%	52%	65%	87%	88%	91%	94%
di:	1%	0%	0%	2%	13%	12%	9%	2%

▪ β,β' -hetero-diarylation of amino acid

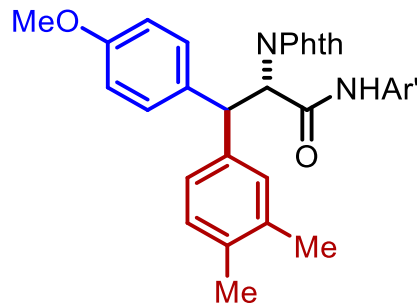


- More 'forcing' conditions than mono-arylation
- fixed conformation facilitates conjugation

β,β'-hetero-diarylation of amino acid



59%, dr > 20:1

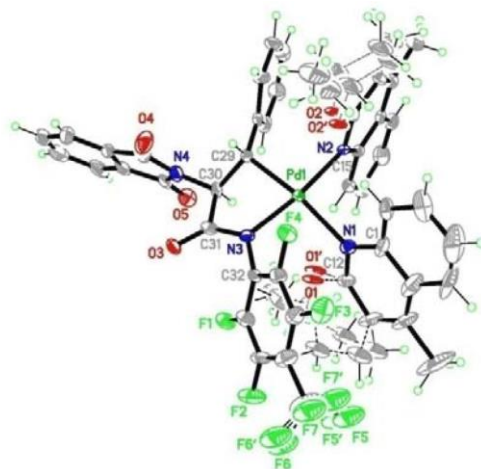
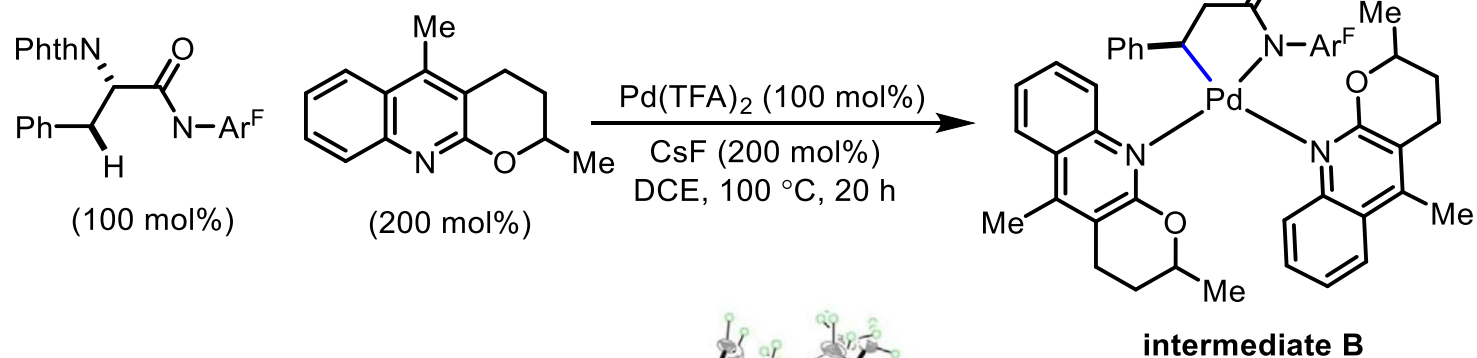
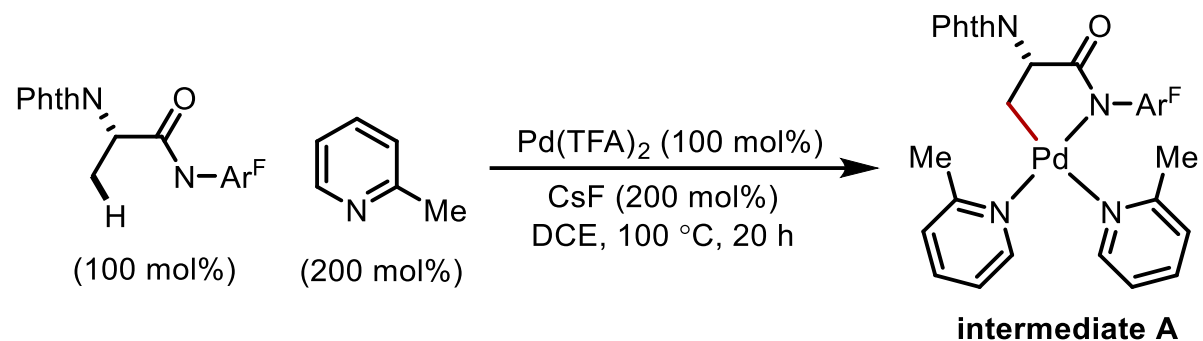


60%, dr 16:1

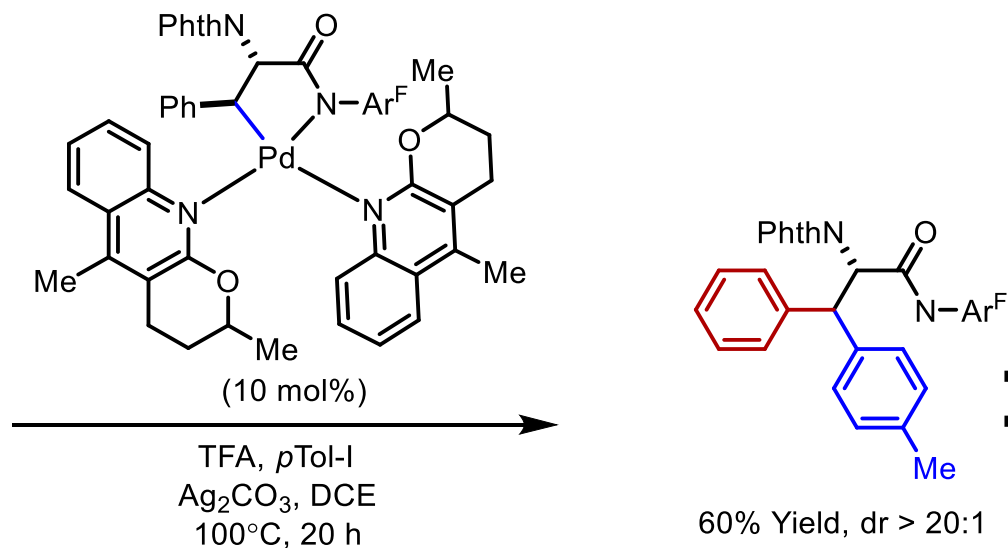
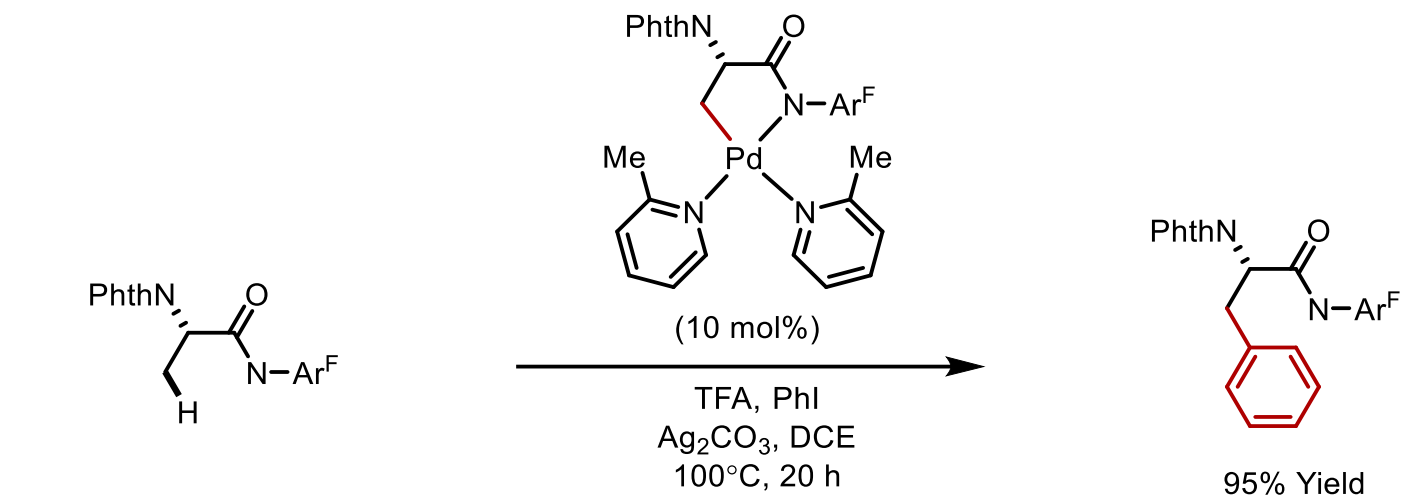
One pot reaction

- Biaryl formation detected
- Remaining $\text{Ar}^1\text{-I}$ outcompeted by $\text{Ar}^2\text{-I}$
- no loss in ee

- **β,β' -hetero-diarylation of amino acid**



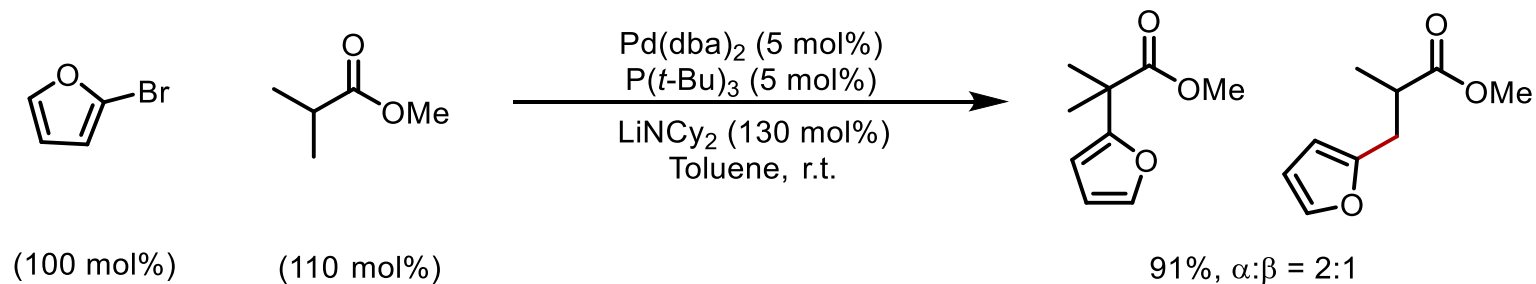
▪ β,β' -hetero-diarylation of amino acid



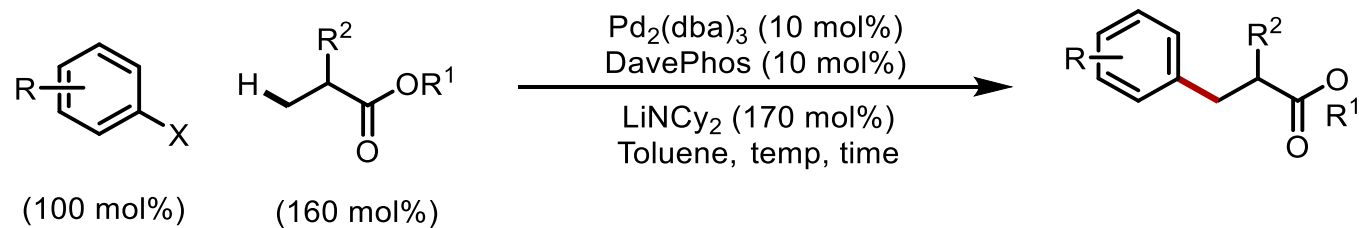
- Pd(II)/Pd(IV) proposed
- TFA facilitates dissociation of one ligand



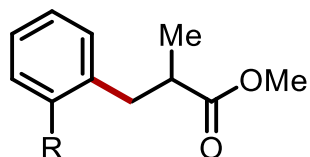
Hartwig's discovery



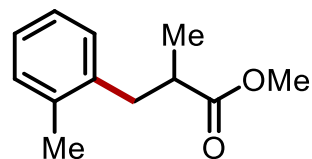
Baudoin's development



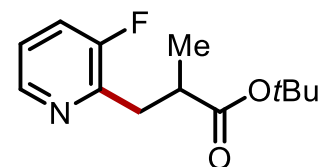
Baudoin's development



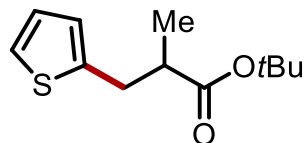
R = Cl 82% (X = Br, 22 °C, 2h)
 Cl 63% (X = Cl, 50 °C, 22 min)
 CF₃ 62% (X = Br, 110 °C 2h)
 OCF₃ 75% (X = Br, 35 °C, 2h)



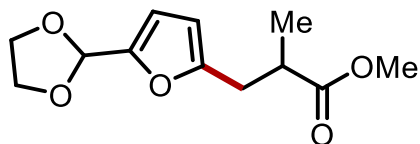
0%, X = Br
 α-arylation only



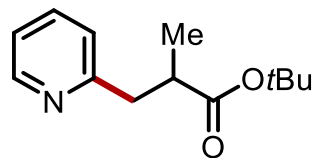
56% β:α 4:1
 (X = Cl, 30 °C, 90 min)



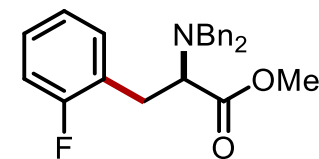
69%
 (X = Cl, 70 °C, 70 min)



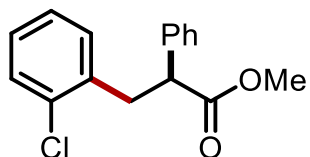
70%
 (X = Cl, 50 °C, 45 min)



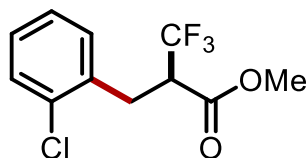
0%, X = Cl
 α-arylation only



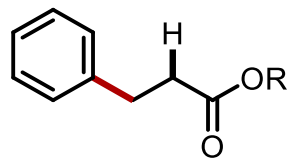
63%
 (X = Br, 50 °C, 3.5 h)



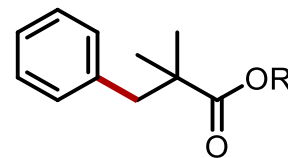
0%



0%

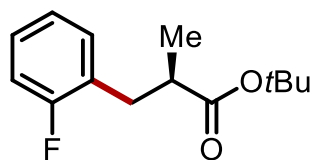
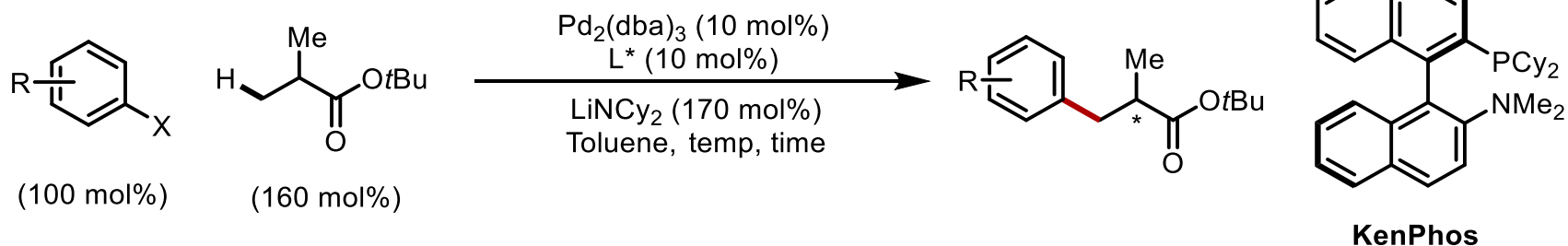
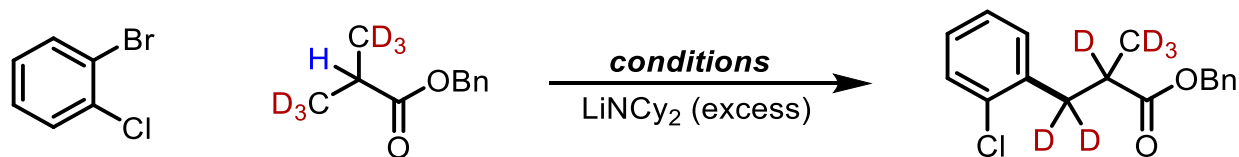


no α or β arylation

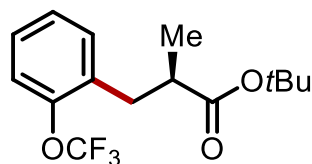


no β arylation

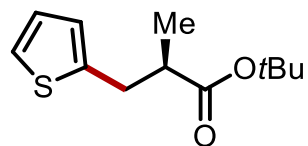
Migratory coupling



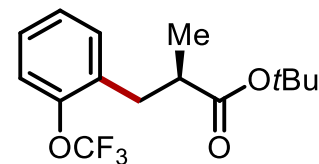
70%, e.r. 75:25
($\text{X} = \text{Br}$, 30 °C, 3h)



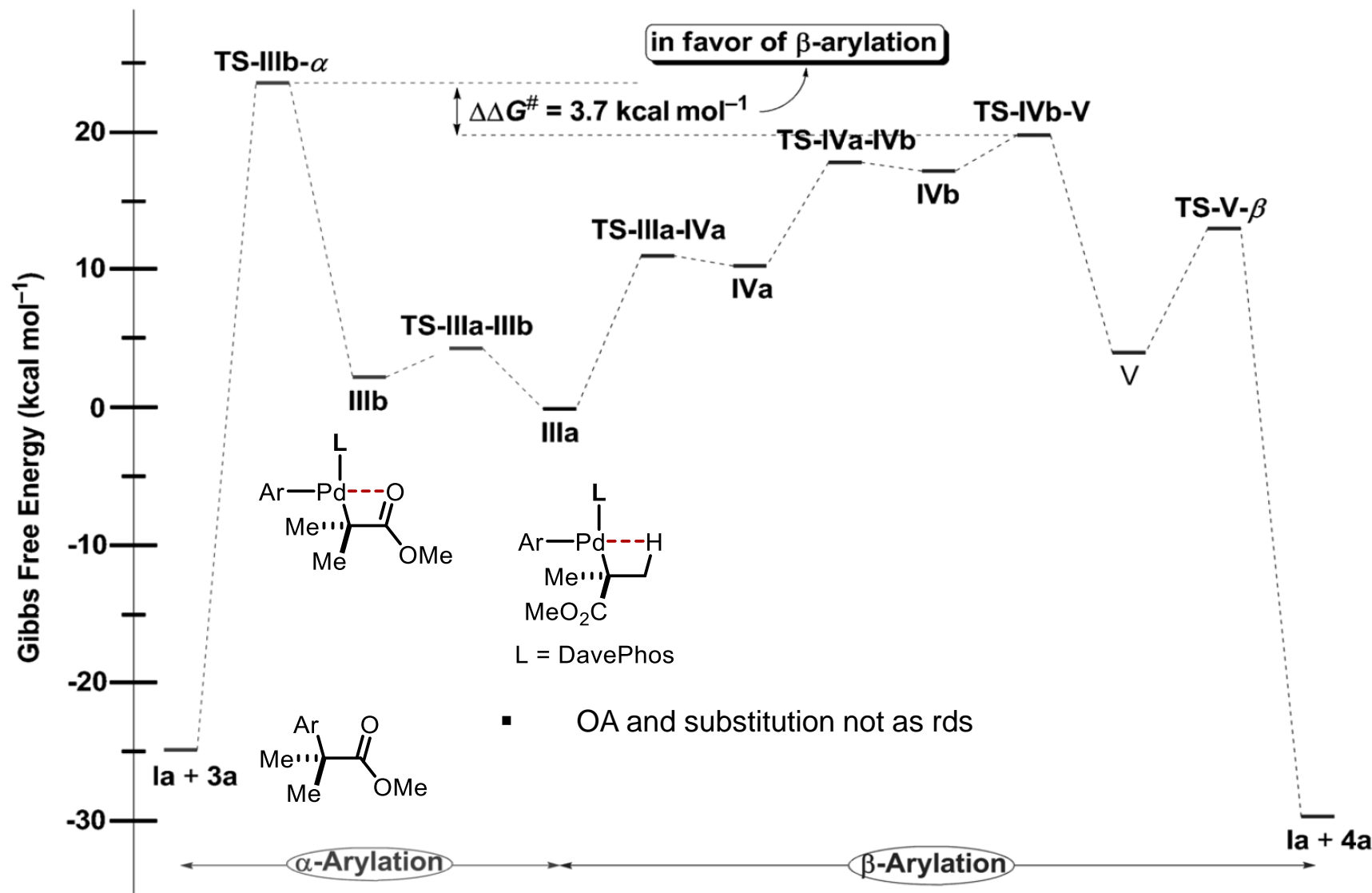
55%, e.r. 76:24
($\text{X} = \text{Br}$, 35 °C, 1.5h)

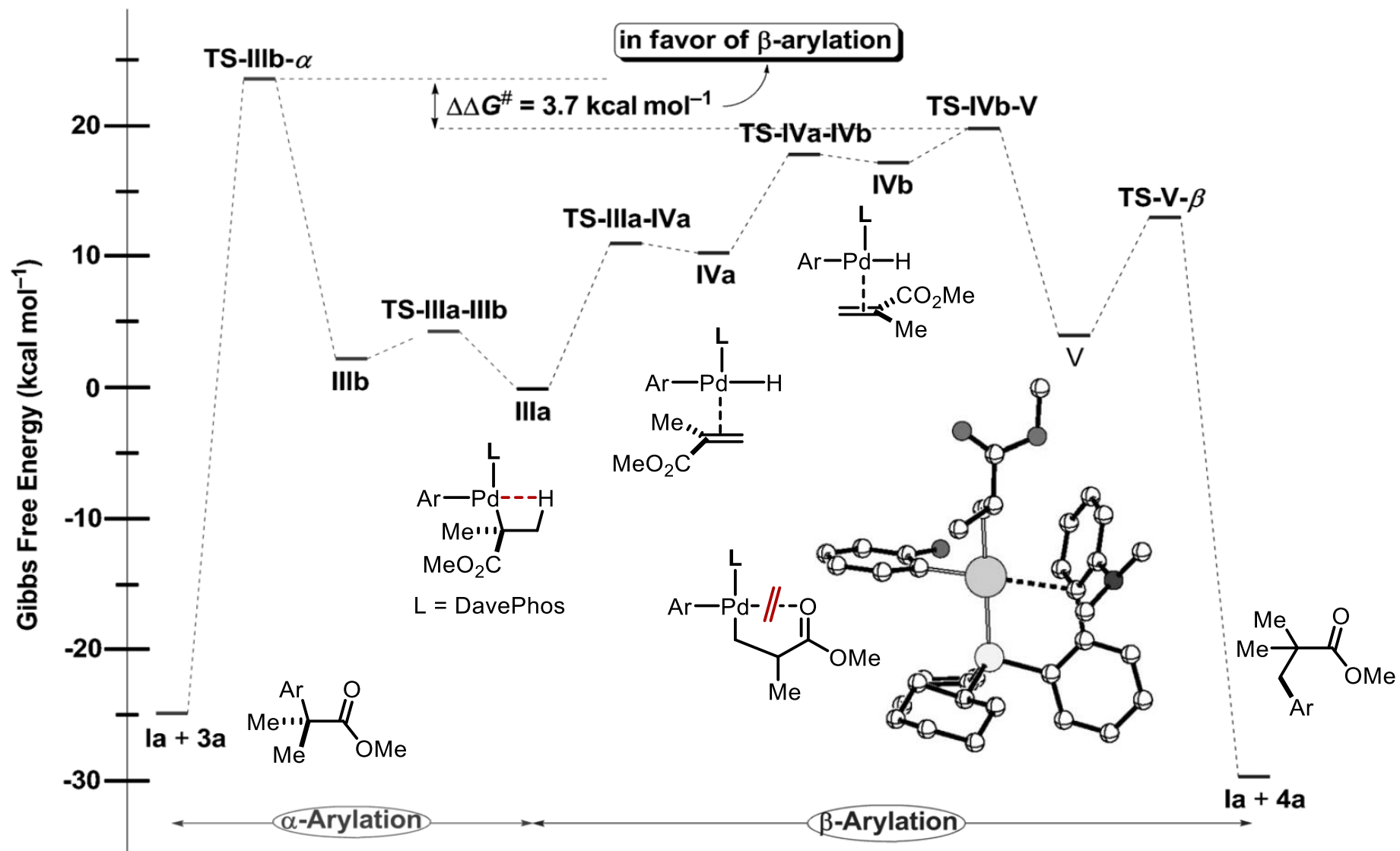


63%, e.r. 67:33
($\text{X} = \text{Cl}$, 70 °C, 1.5h)

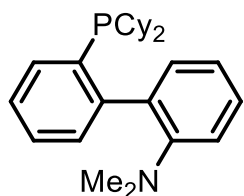
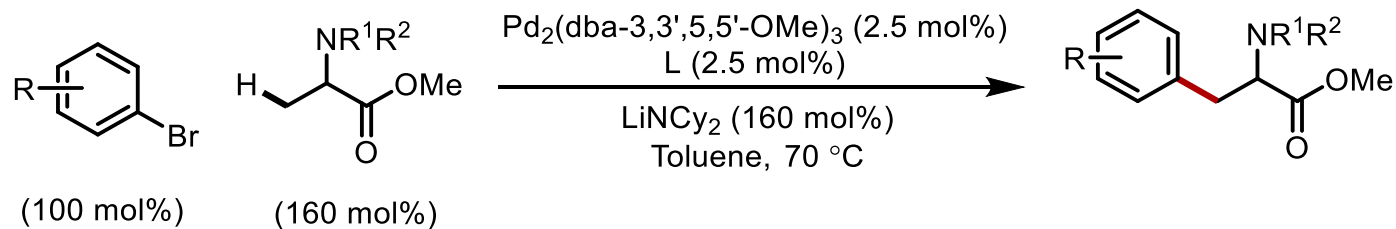


69%, e.r. 77:23
($\text{X} = \text{Br}$, 30 °C, 2.5h)

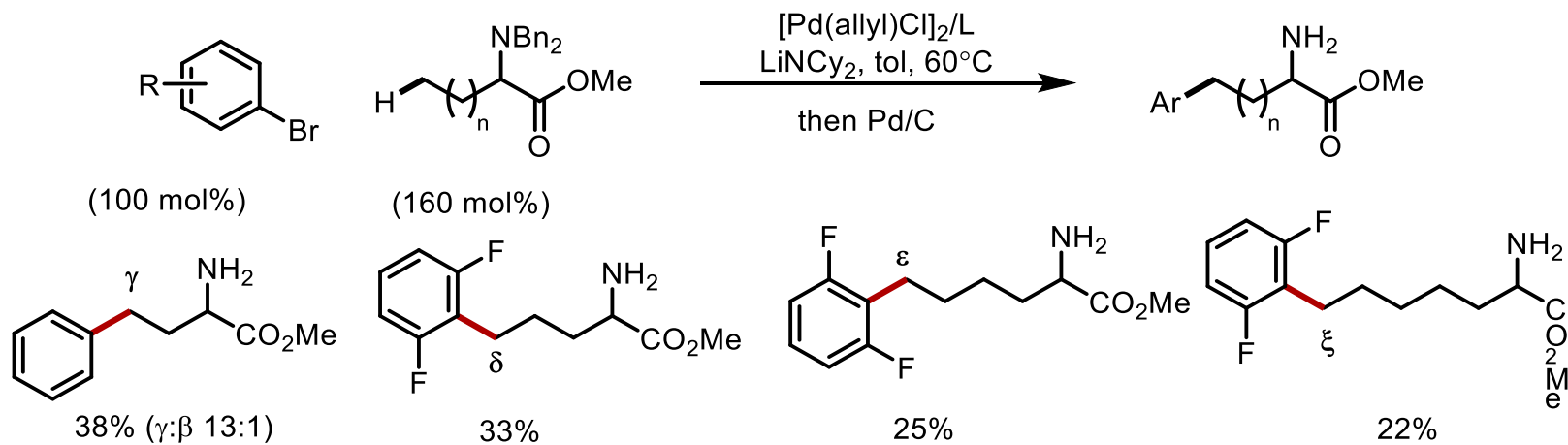
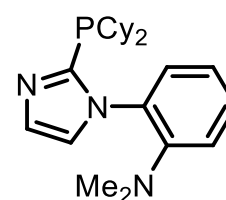
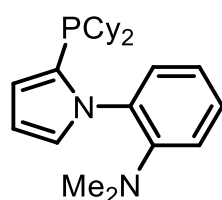




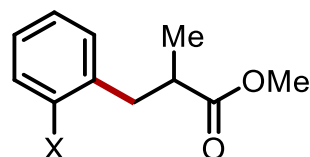
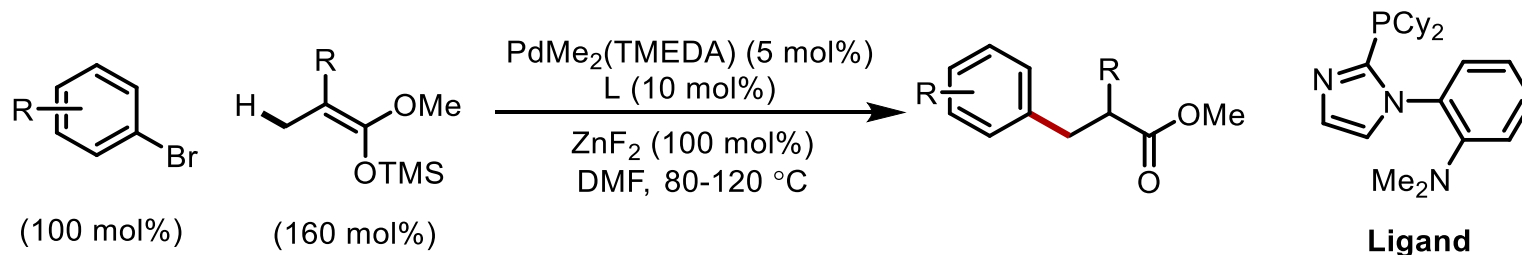
Arylation of amino ester



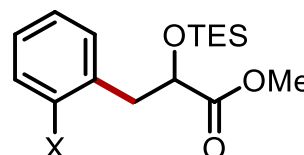
DavePhos



Use of silyl ketene acetals

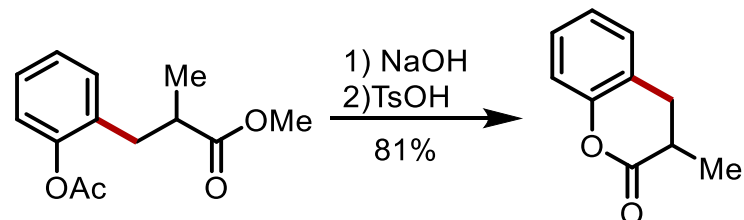
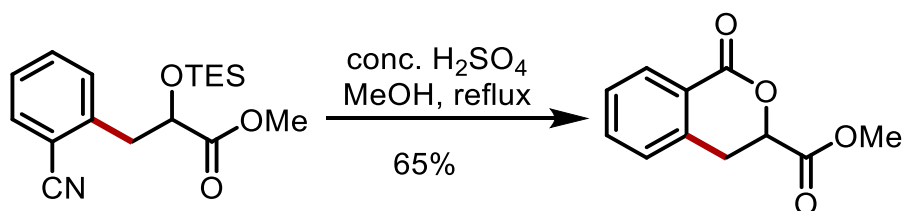


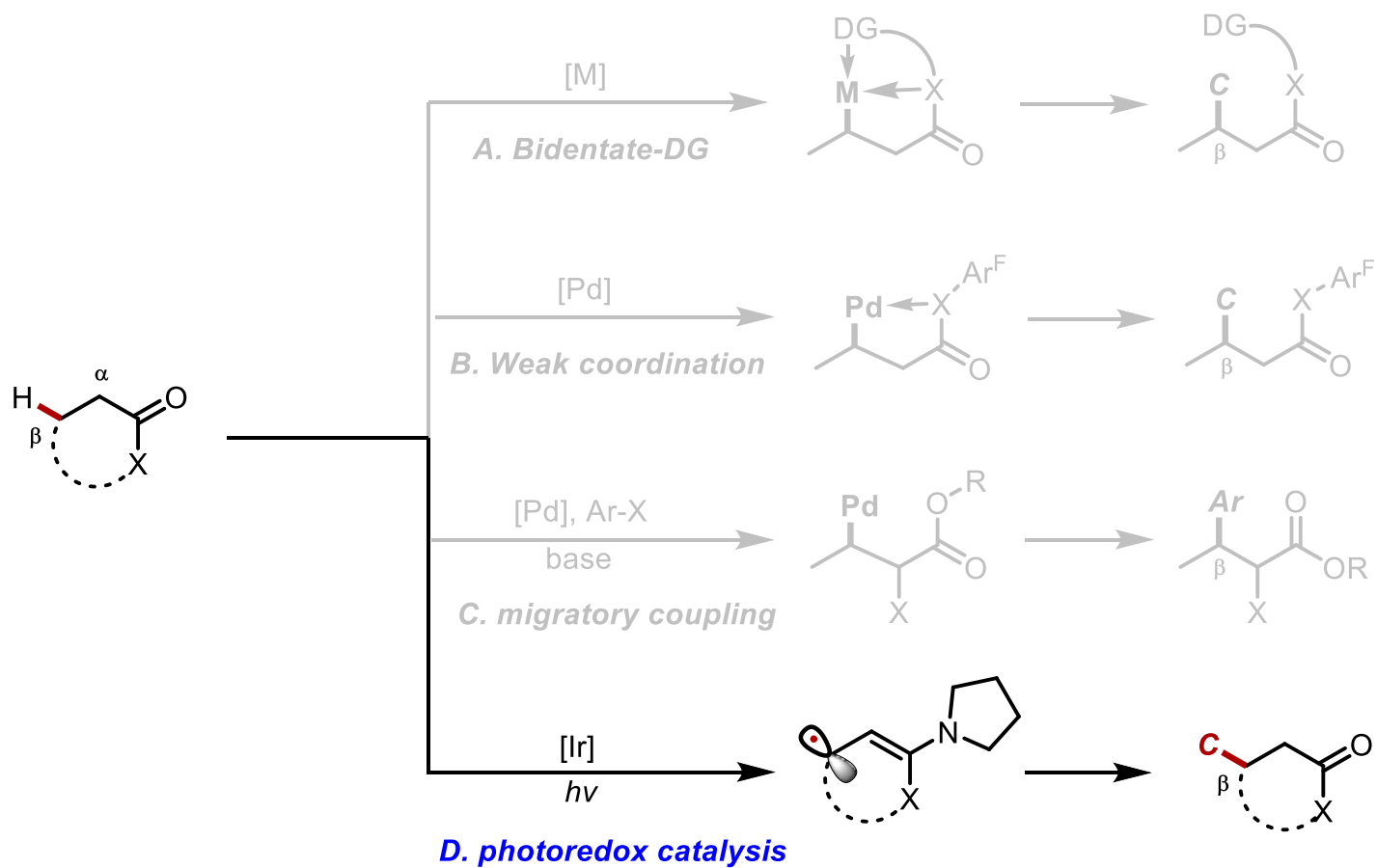
X = CN	72%
X = CO ₂ Me	43%
X = OAc	64%
X = NO ₂	56%



X = F	59%
X = OMe	44%
X = CF ₃	50%

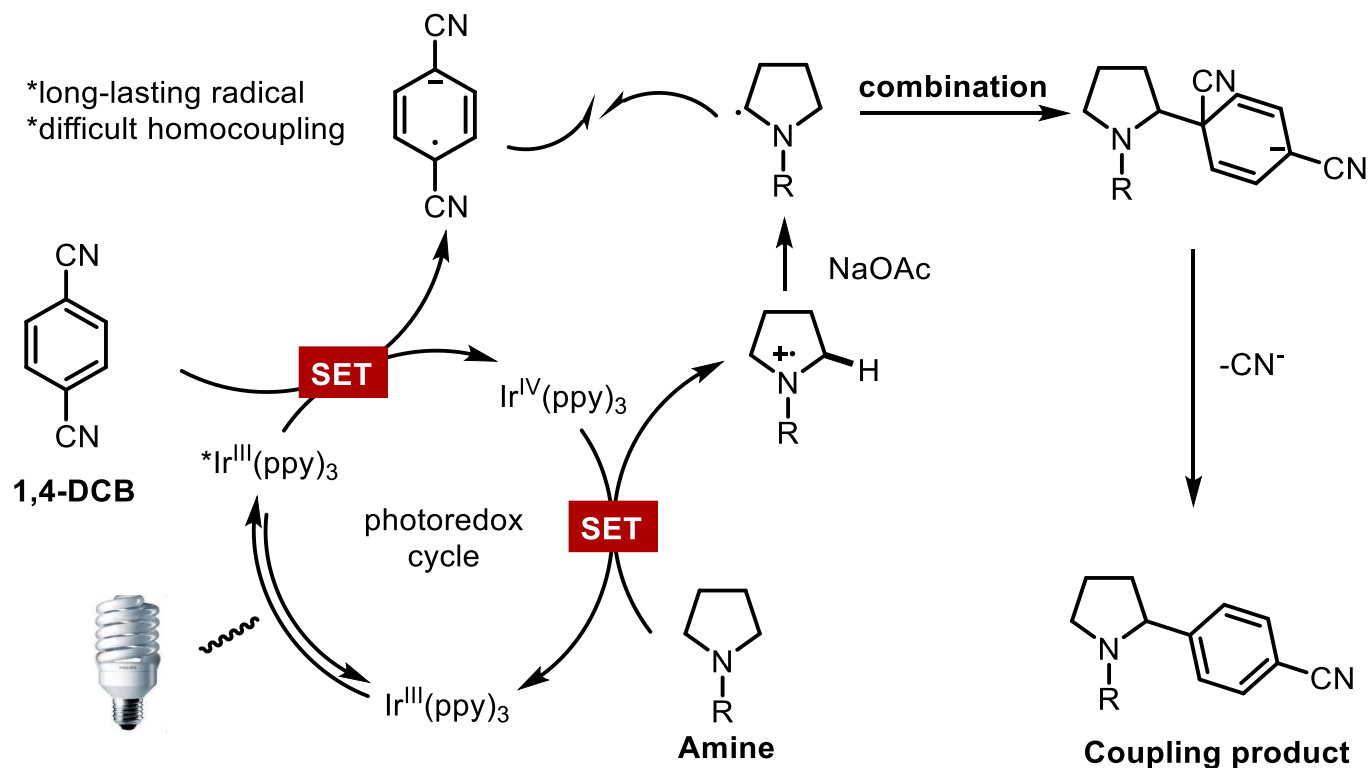
- Lower reactivity
- Higher FG tolerance



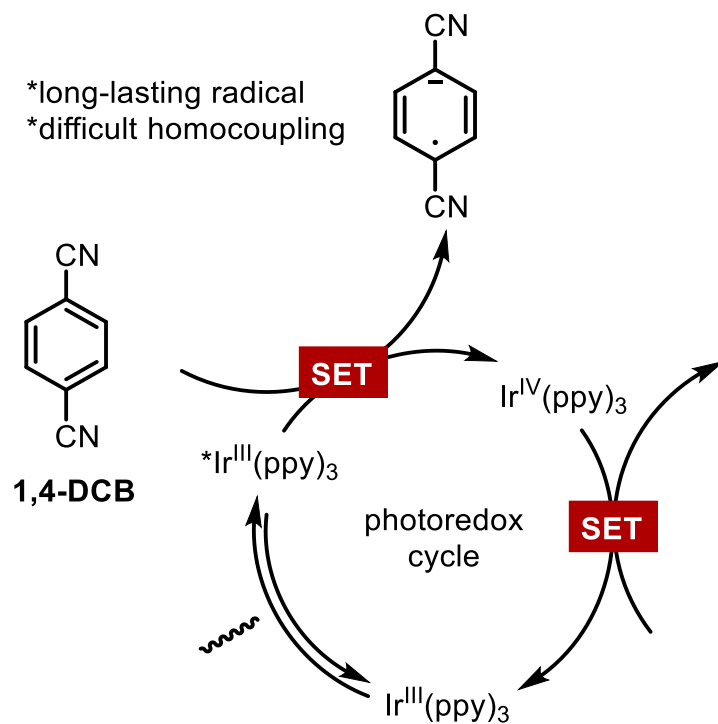


Photoredox Catalysis

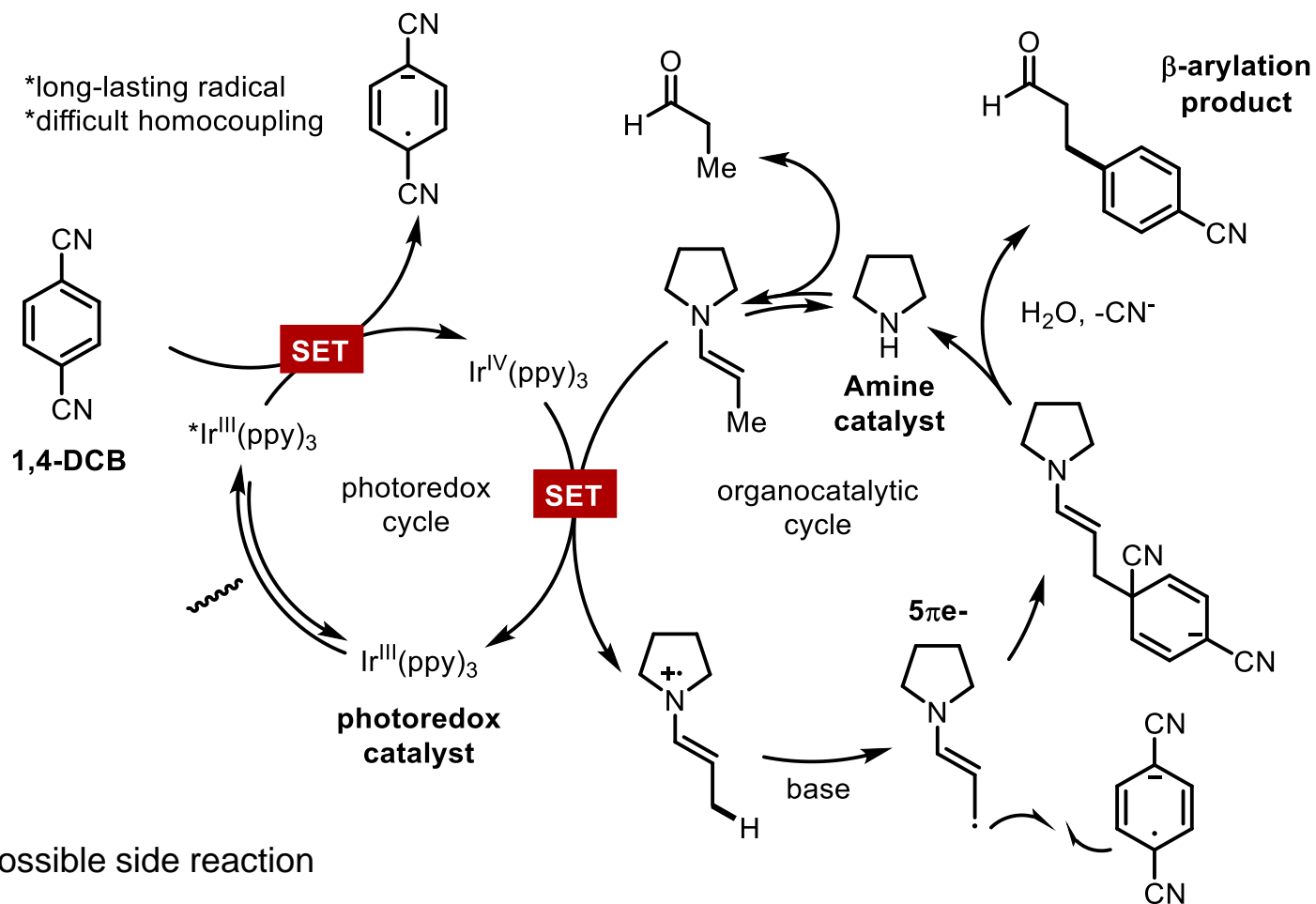
▪ α -arylation of amine



- β -arylation of aldehyde and ketone

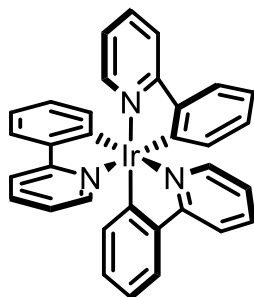
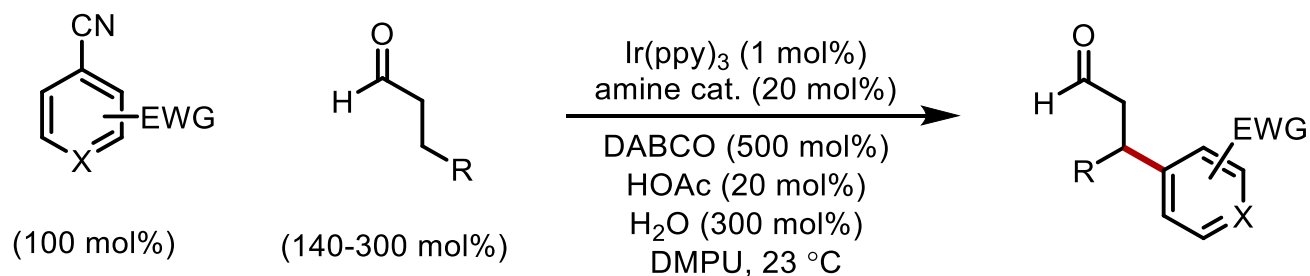


β-arylation of aldehyde and ketone

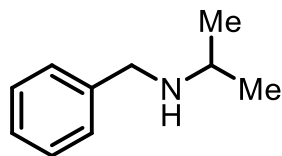


possible side reaction

▪ β -arylation of aldehyde



photoredox catalyst

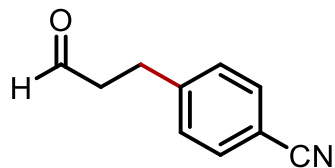


amine catalyst

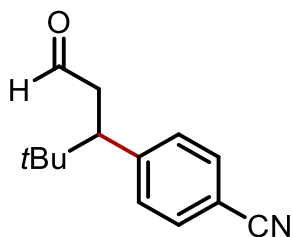
Amine

- easy to condense (e-rich)
- easy to be oxidized (e-rich)
- easy to turn over

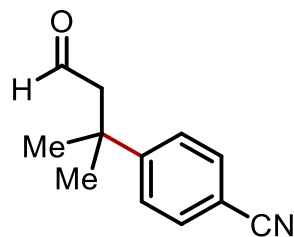
▪ β -arylation of aldehyde



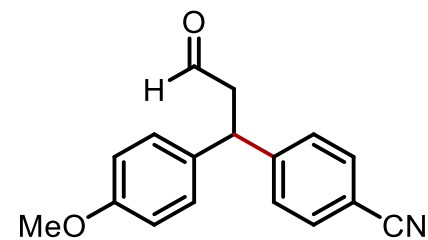
78%



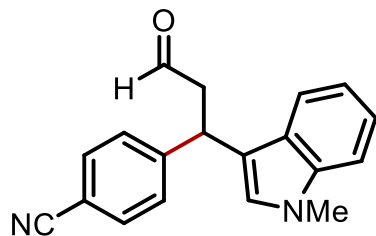
44%



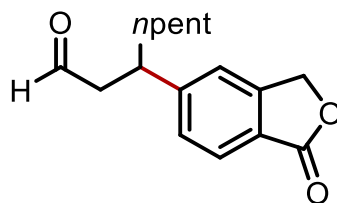
74%



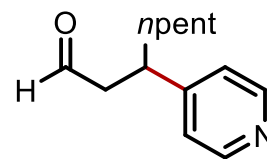
88%



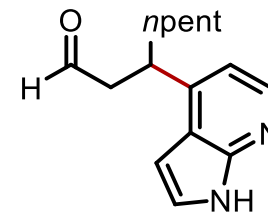
63%



63%

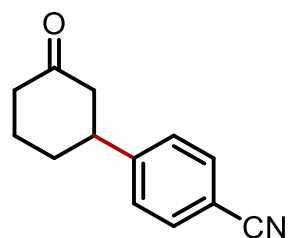
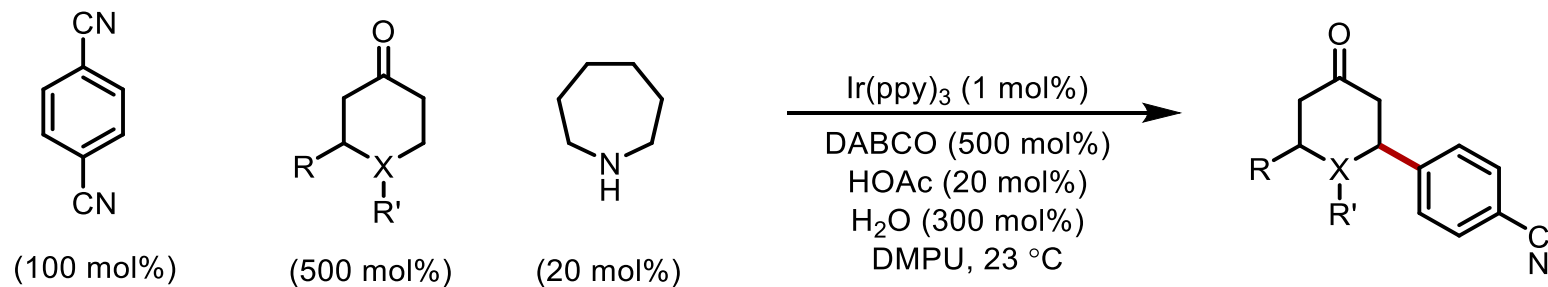


70%

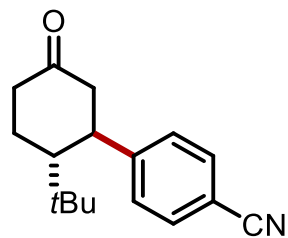


53%

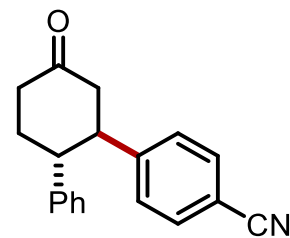
▪ β -arylation of ketone



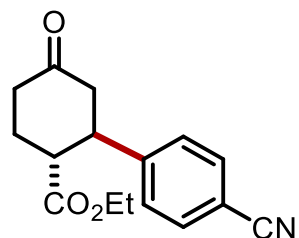
88%



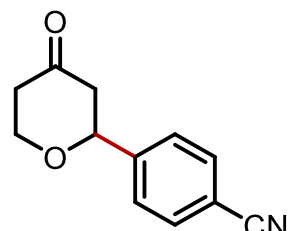
70%, > 20:1 dr



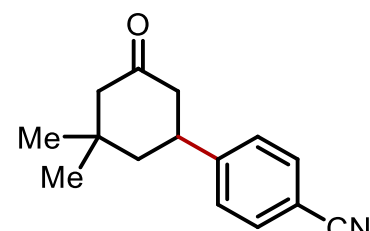
79%, > 20:1 dr



81%, > 20:1 dr

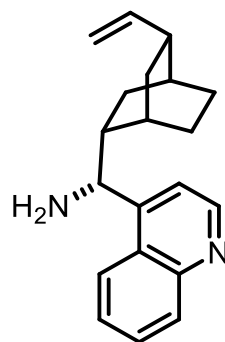
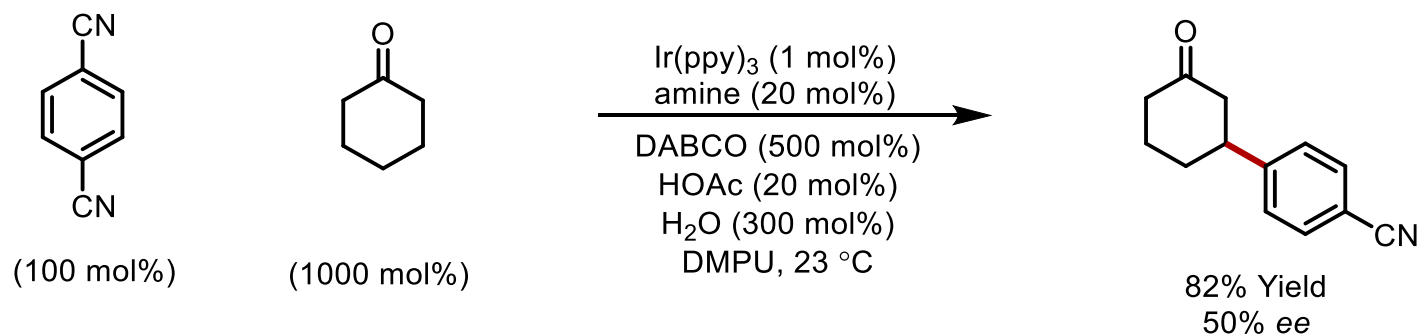


63%



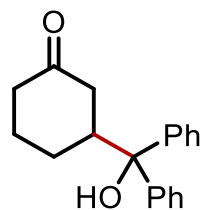
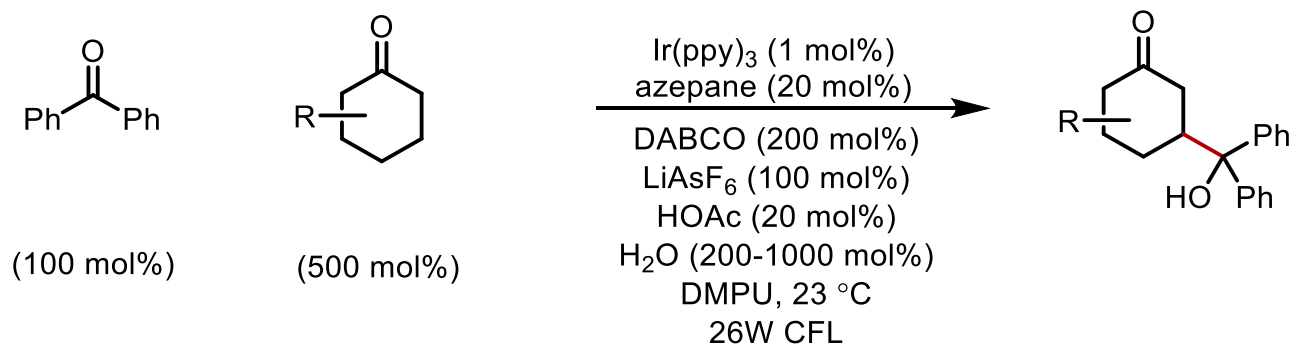
83%

- Enantioselective β -arylation of ketone

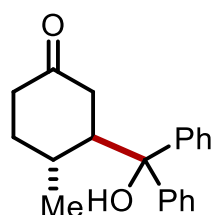


Chiral amine catalyst

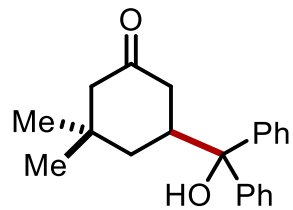
▪ β -aldol of ketone



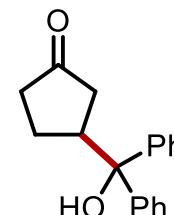
81%



79%, dr > 20:1



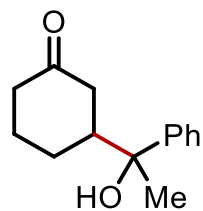
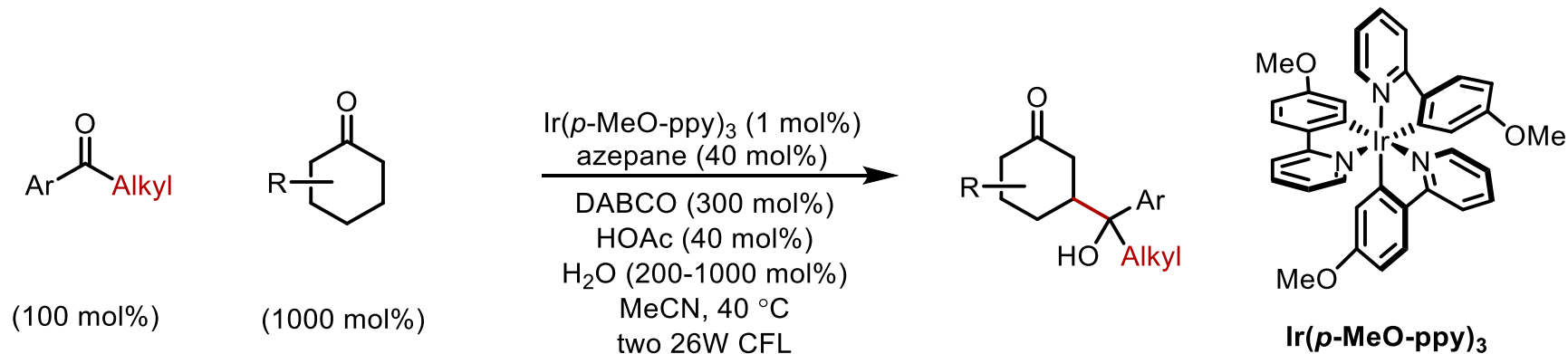
43%, dr > 20:1



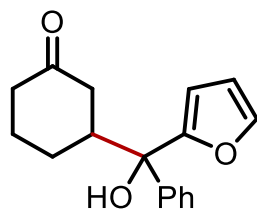
65%



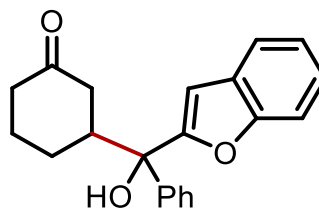
β-aldol of ketone



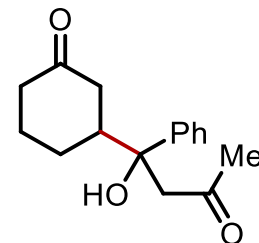
77%



69%

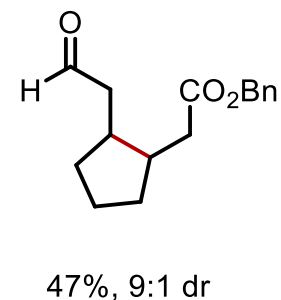
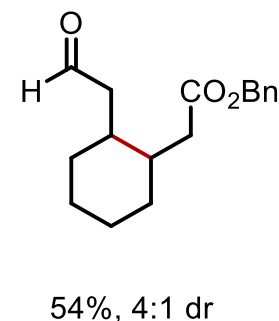
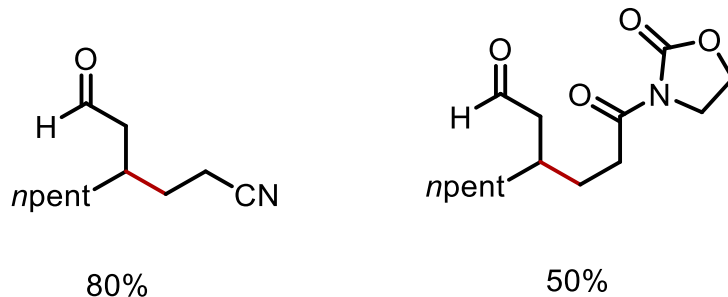
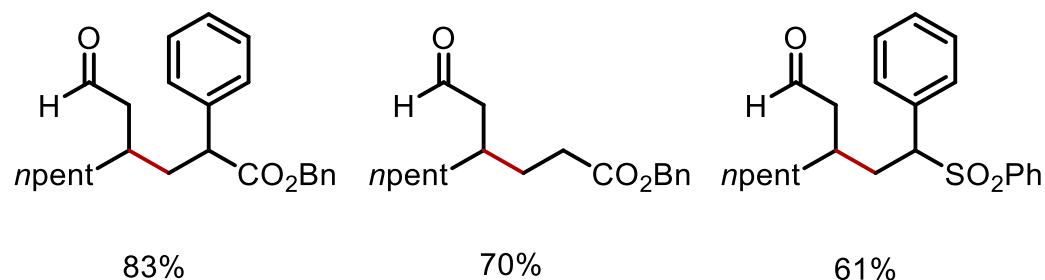
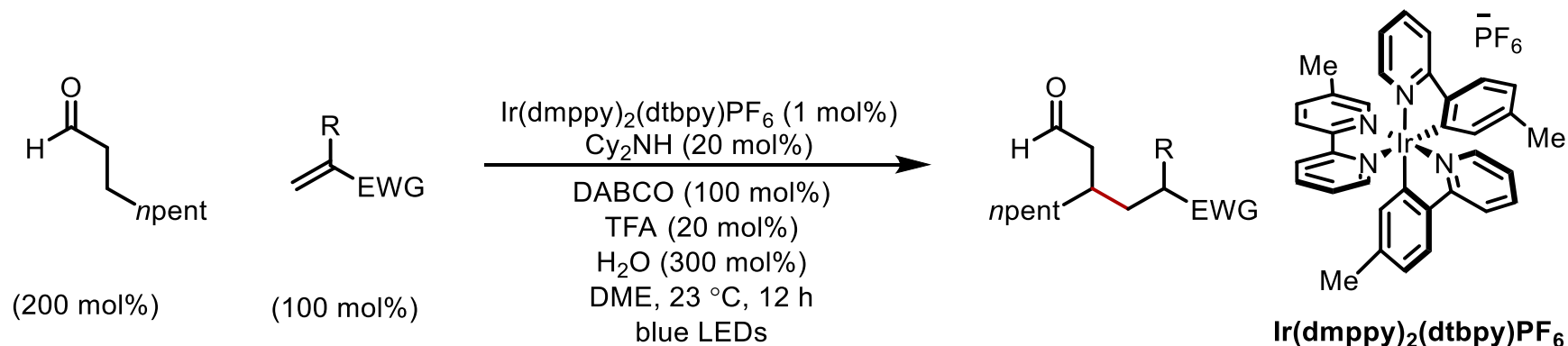


56%



54%

β-alkylation of aldehyde



- **Bidentate DG**
 - Classic DG
 - Various C-C formation and metals
 - Highly applicable
- **Weak coordination**
 - Impressive ligand control
 - Various C-C bond formation
 - Amino acid synthesis
- **Migratory Coupling**
 - Easily available substrate
 - No poly-arylation
 - Amino ester synthesis
- **Photoredox catalysis**
 - Conceptual advance
 - Versatile radical chemistry
 - Cyclic substrates and aldehyde

Thank
you!

The image features the words "Thank you!" in a highly decorative, hand-drawn style. The letters are thick and outlined in black, with various patterns and colors. The word "Thank" is in the top row, and "you!" is in the bottom row. The letters are surrounded by several stylized flowers in blue, pink, and purple. The background is a light blue gradient.