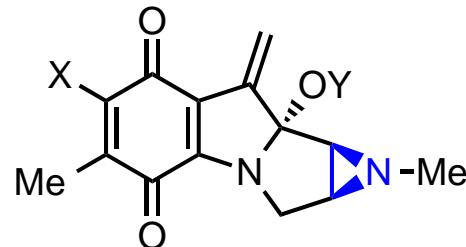
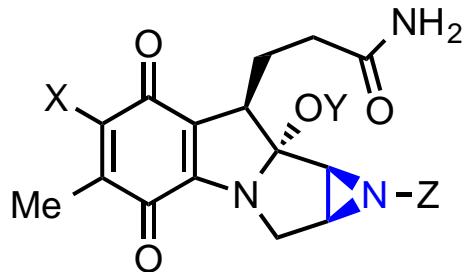
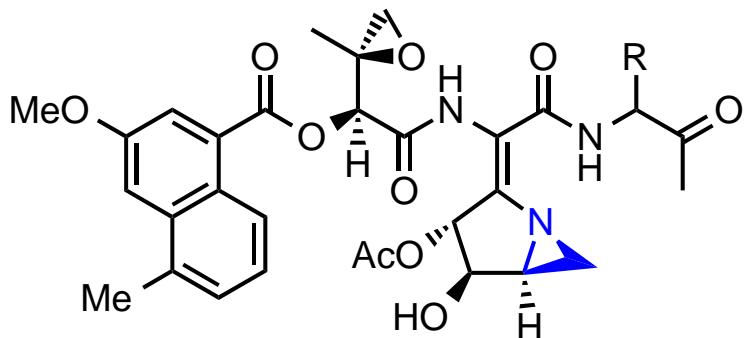


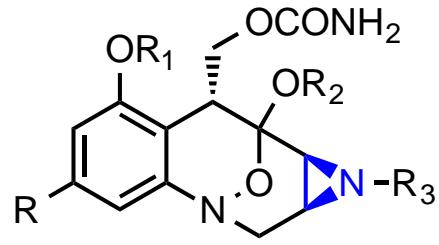
# *Syntheses of Aziridines*



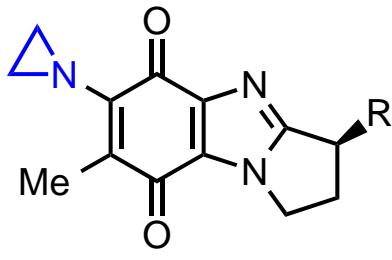
Mitomycin



Azinomycin



FR-xxxx



PBI-xxxx

2014. 03. 05  
Haye Min Ko

# *Contents*

1. *Physical properties*
2. *Biological properties*
3. *Reactions of aziridines (for organic chemists)*
4. *Syntheses of aziridines from olefins*
  - *Nitrene methods*
  - *Cu*
  - *Rh*
  - *Other metals*
  - *Carbene and ylide methods*
5. *Conclusion*

# *Physical Properties*

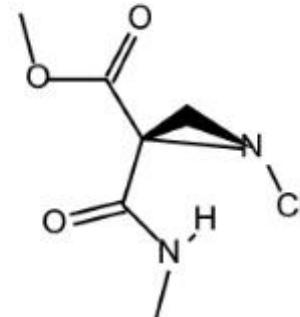
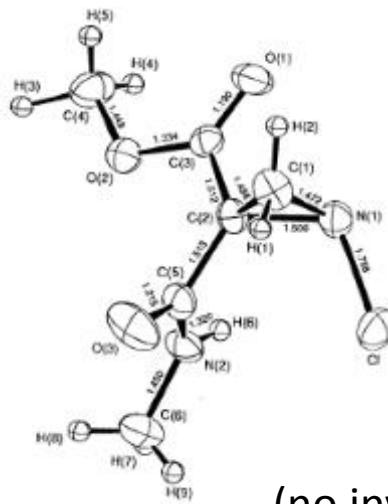
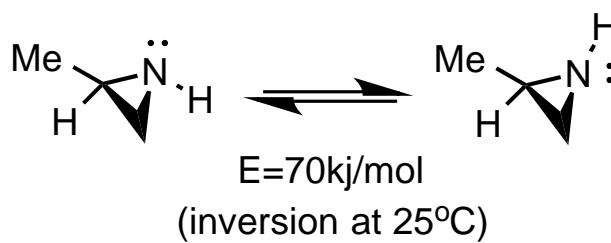


ethyleneimine  
water-soluble,  
colorless, distillable

liquid

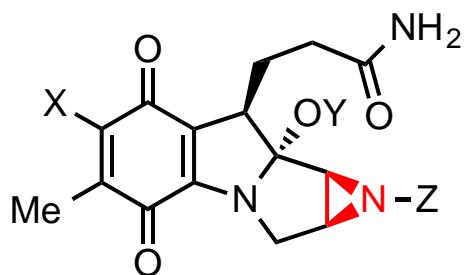
b.p. 57°C

pKa 7.98



(no inversion at 50°C)

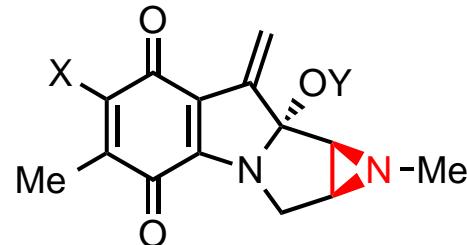
# *Biological Properties (Natural Products)*



## Mitomycin A X=OMe, Y=Me, Z=H

### **Mitomycin B X=OMe, Y=H, Z=Me**

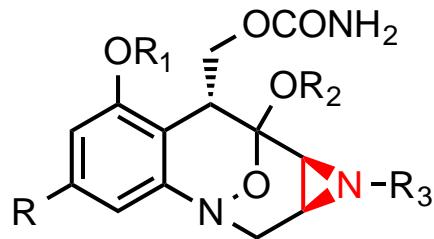
### **Mitomycin C X=NH<sub>2</sub>, Y=Me, Z=H**



## Mitomycin G X=NH<sub>2</sub>, Y=Me

## Mitomycin H X=OMe, Y=H

## Mitomycin K X=OMe, Y=Me



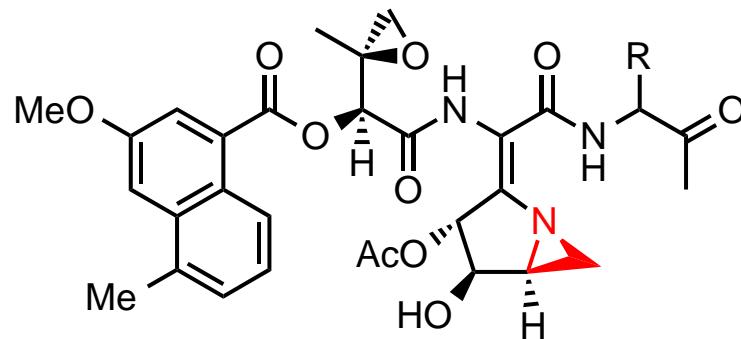
## **FR-900482 R=CHO, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>=H**

**FR-66979** R=CH<sub>2</sub>OH, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>=H

**FR-70496** R=CHO, R<sub>1</sub>=Me, R<sub>2</sub>=H, R<sub>3</sub>=Ac

FK-973      R=CHO, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>=Ac

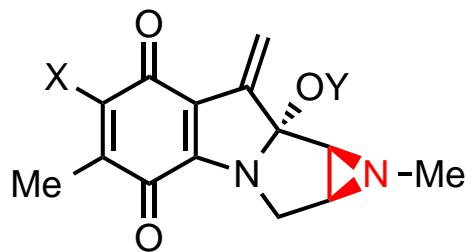
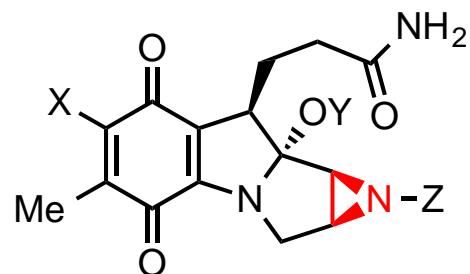
FK-317      R=CHO    R<sub>1</sub>=Me    R<sub>2</sub>=Ac



## Azinomycin A R=H

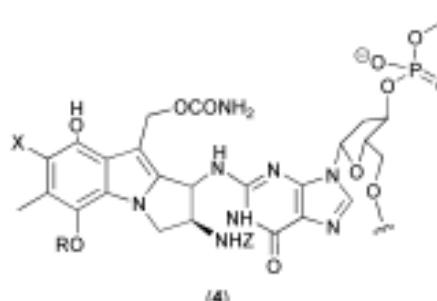
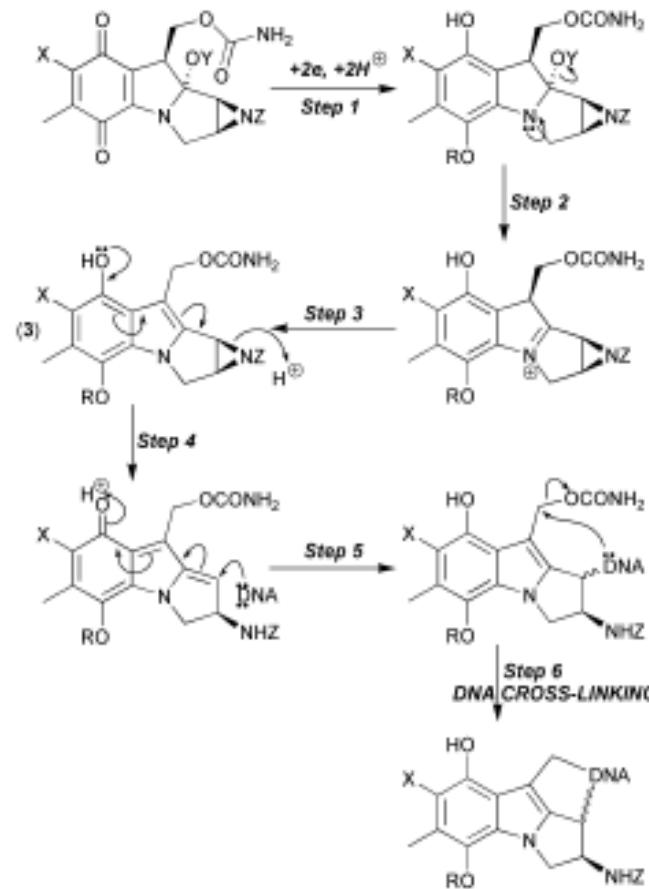
## Azinomycin B R=CHO

# Biological Properties(DNA interaction of aziridines)



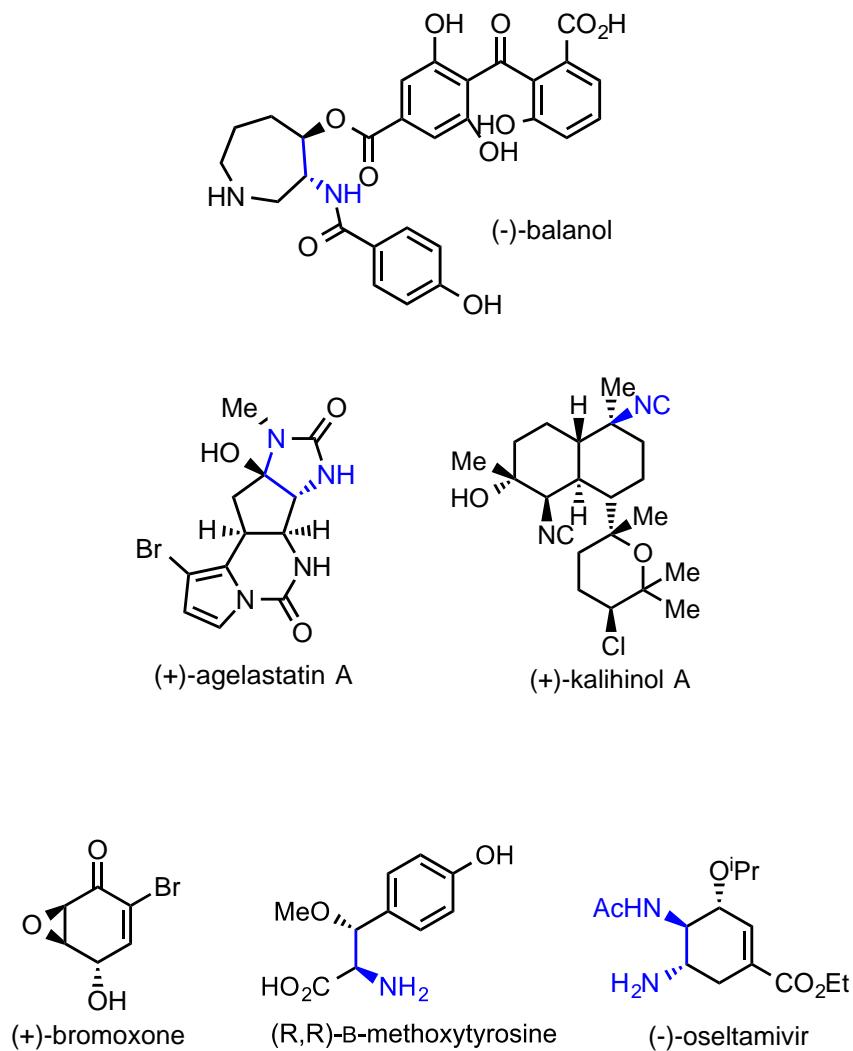
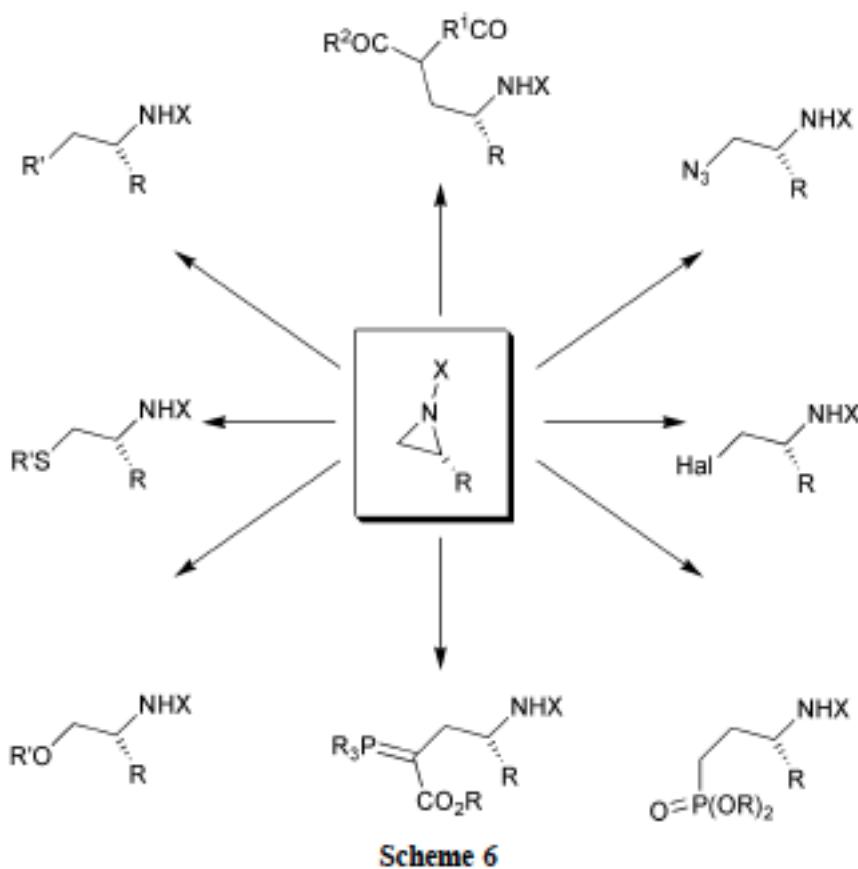
Mitosanes

*Streptomyces verticillatus*  
anti-tumour  
antibiotic activity

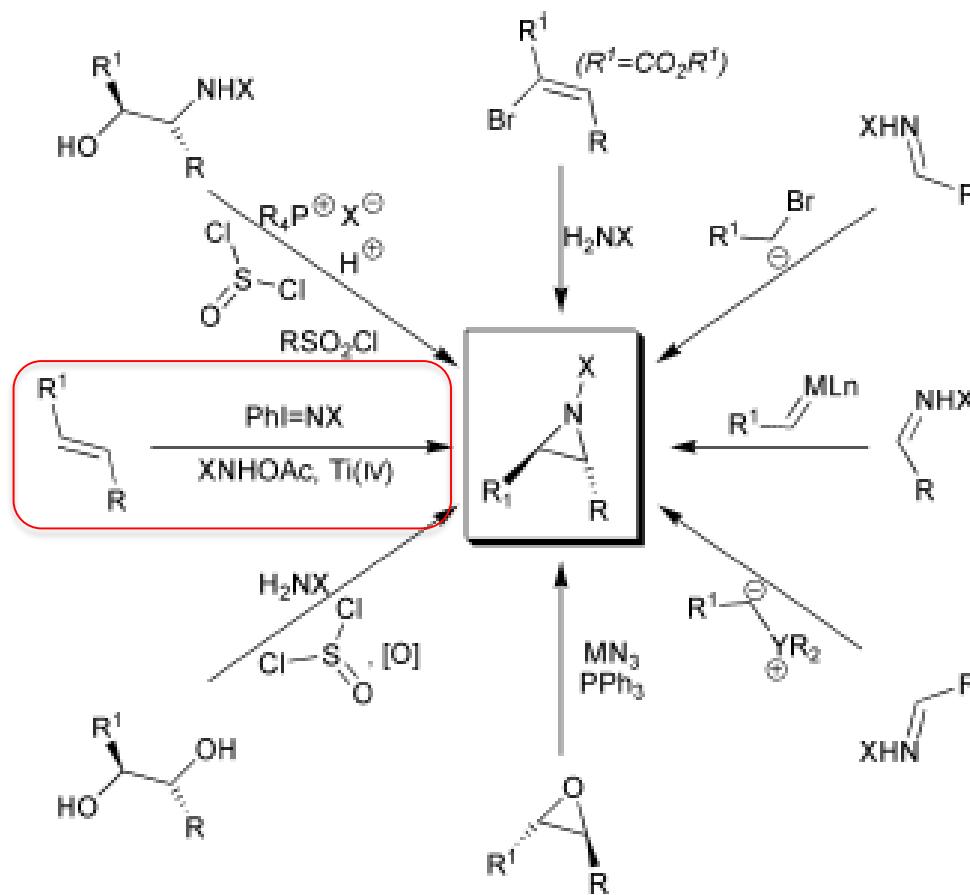


Scheme 1 Mode of action of Mitosanes.

# Reactions of Aziridines (Ring opening processes)

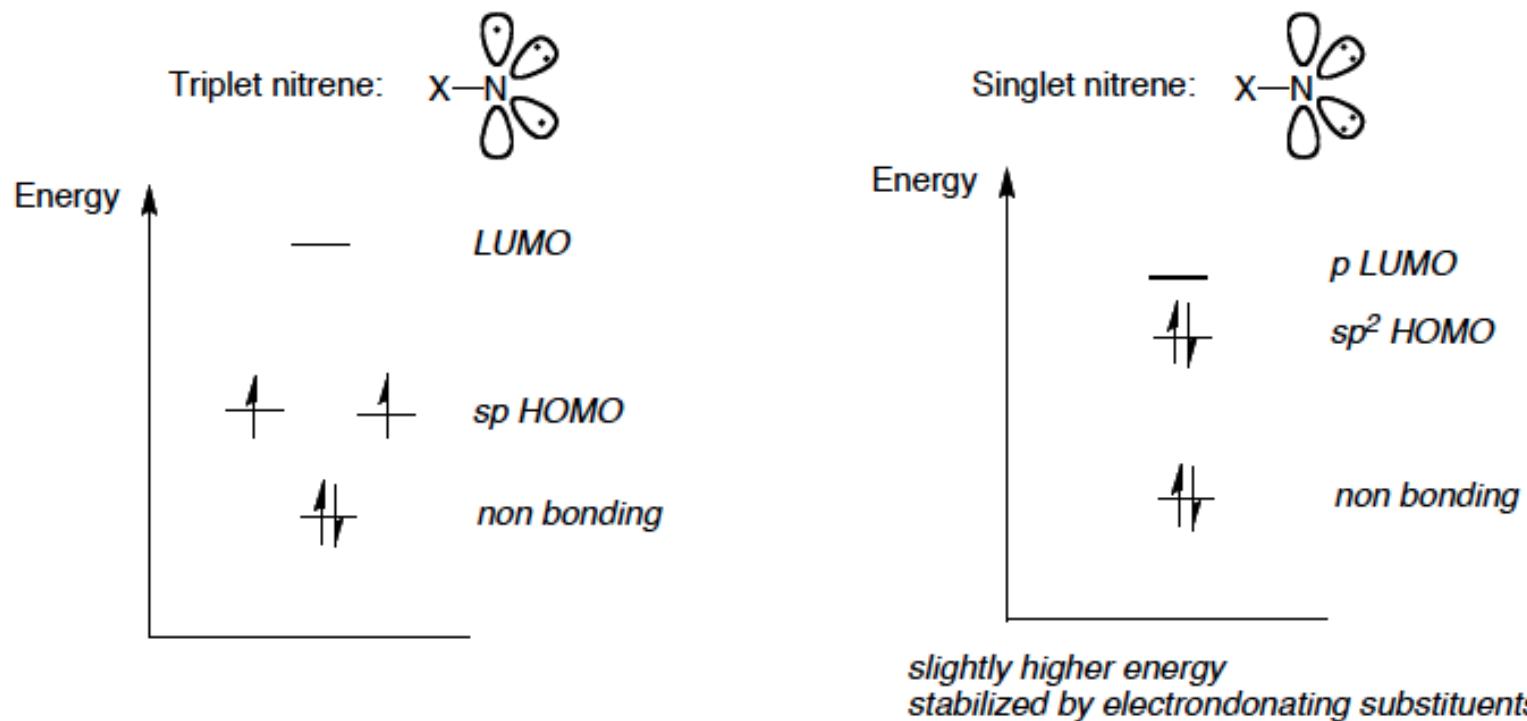


# Syntheses of Aziridines (Overview)

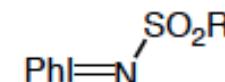
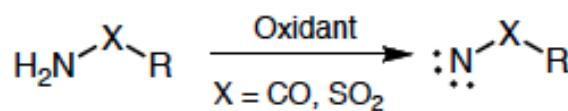
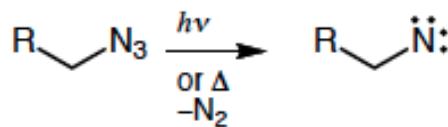


Scheme 19

# *Introduction of nitrenes*



Common nitrene precursors:

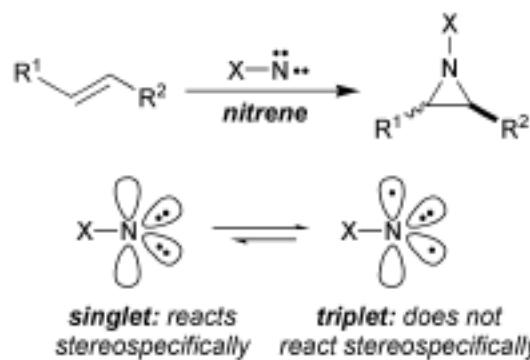


J. B. Sweeney, *Chem. Soc. Rev.*, **2002**, 31, 247–258

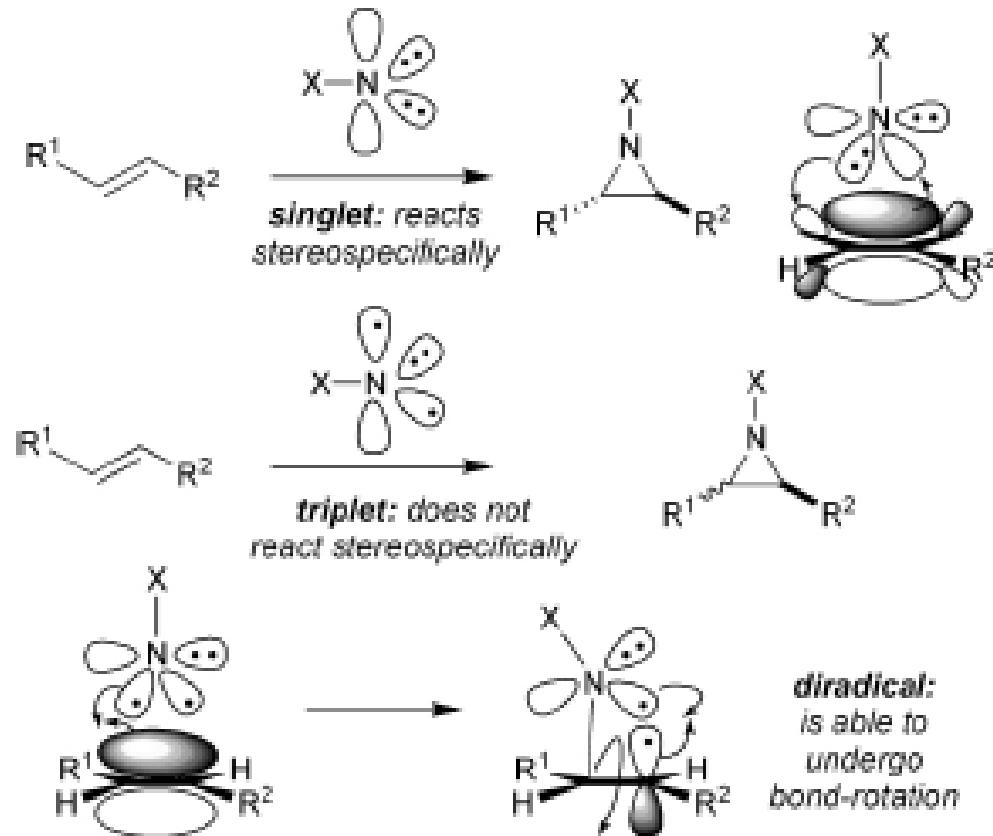
H. Pellissier, *Tetrahedron*, **2010**, 66, 1509-1555

W. Schoeller, A. B. Rozhenko, *Eur. J. Inorg. Chem.* **2001**, 845-850.

# *Stereoselectivity of nitrenes*

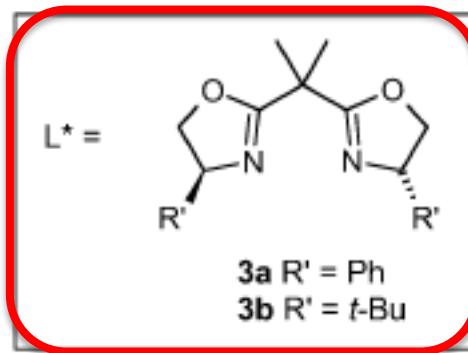
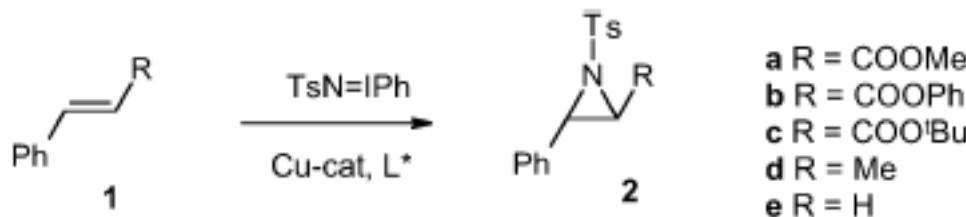


Scheme 21



Scheme 22

## *Cu-catalyzed aziridinations(Evans)*



**Table 1. Aziridination of Olefins with Bis(oxazoline)–Copper Complexes**

olefin	cat.	solvent, conditions	yield, %	ee, %	conf.
methyl cinnamate ( <b>1a</b> )	<b>3a</b>	C <sub>6</sub> H <sub>6</sub> (24 h, 21 °C)	63	94	(S)
phenyl cinnamate ( <b>1b</b> )	<b>3a</b>	C <sub>6</sub> H <sub>6</sub> (24 h, 21 °C)	64	97	(S)
<i>t</i> -butyl cinnamate ( <b>1c</b> )	<b>3a</b>	C <sub>6</sub> H <sub>6</sub> (24 h, 21 °C)	60	96	(S)
<i>trans</i> - $\beta$ -methylstyrene ( <b>1d</b> )	<b>3b</b>	MeCN (3 d, –20 °C)	62	70	(S)
styrene ( <b>1e</b> )	<b>3b</b>	styrene (2.5 h, 0°C)	89	63	(R)

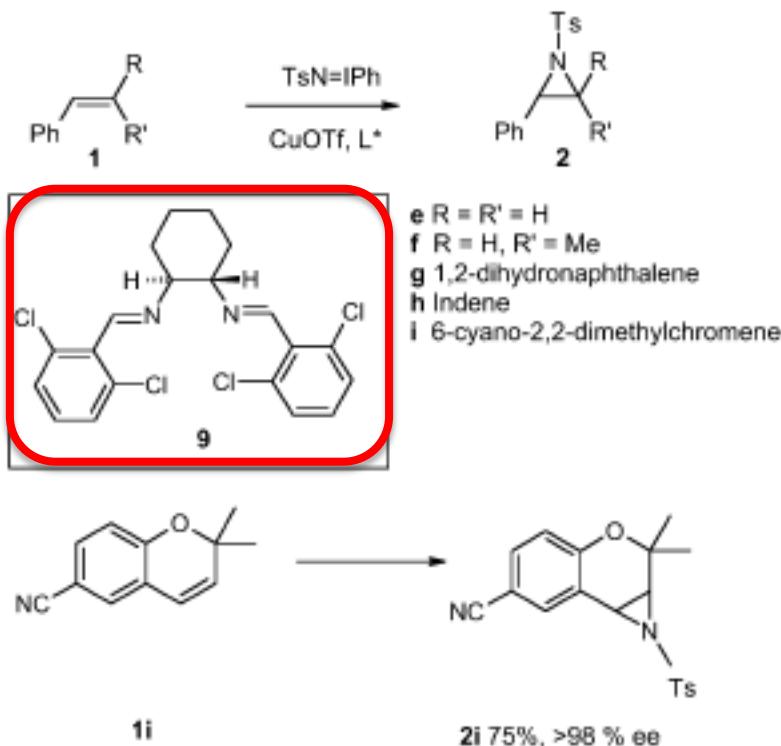
D. A. Evans, M. M. Faul, M. T. Bilodeau, B. A. Anderson, D. M. Barnes, *J. Am. Chem. Soc.* **1993**, *115*, 5328.

D. A. Evans, M. M. Faul, M. T. Bilodeau, *J. Am. Chem. Soc.* **1994**, *116*, 2742.

J. B. Sweeney, *Chem. Soc. Rev.*, **2002**, 31, 247–258

P. Muller, C. Fruit, *Chem. Rev.* **2003**, *103*, 2905-2919

# *Cu-catalyzed aziridinations(Jacobsen)*



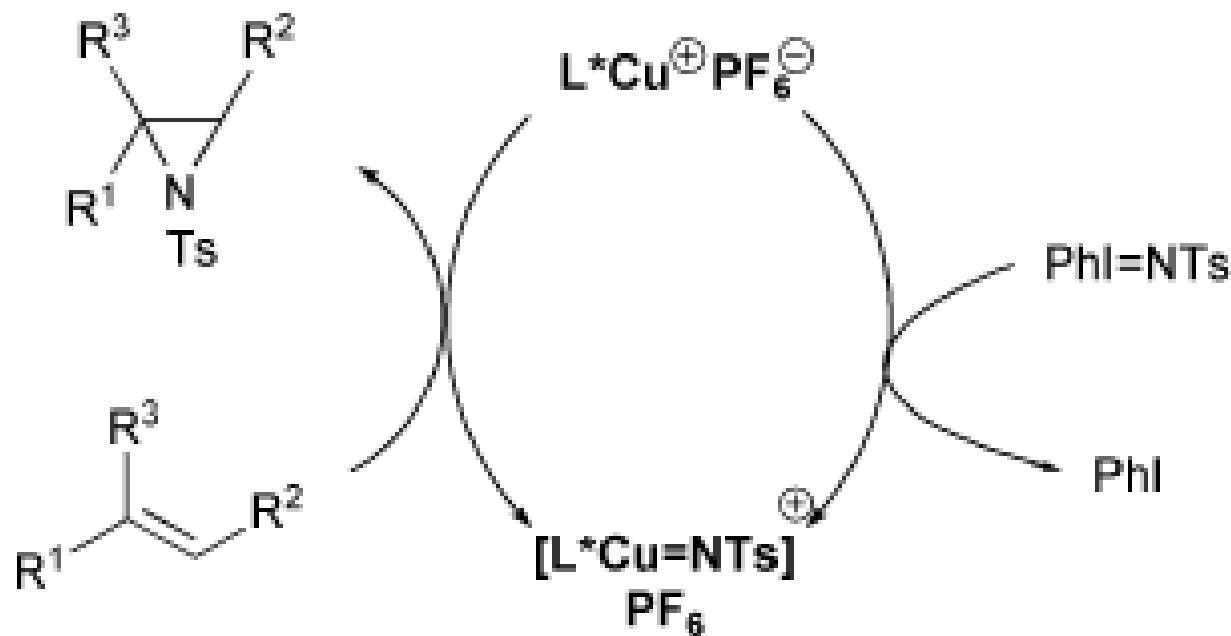
**Table 2. Cu-Catalyzed Aziridination of Olefins with Jacobsen's Ligand 9<sup>a</sup>**

olefin	aziridine, yield, %	ee, %	abs. conf.
styrene ( <b>1e</b> )	79	66	(R)
<i>cis</i> - $\beta$ -methylstyrene ( <b>1f</b> )	79 ( <i>dt</i> = 3:1)	67 ( <i>cis</i> )	(1 <i>R</i> ,2 <i>S</i> )
		81 ( <i>trans</i> )	(1 <i>S</i> ,2 <i>S</i> )
1,2-dihydronaphthalene ( <b>1g</b> )	70	87	(1 <i>R</i> ,2 <i>S</i> )
indene ( <b>1h</b> )	50	58	(1 <i>R</i> ,2 <i>S</i> )
6-cyano-2,2-dimethyl- chromene ( <b>1i</b> )	75	>98	(3 <i>R</i> ,4 <i>R</i> )

<sup>a</sup> –40 to –78 °C, CH<sub>2</sub>Cl<sub>2</sub>, 4 Å sieves, 5–10 mol % of Cu, 6–12 mol % of 9.

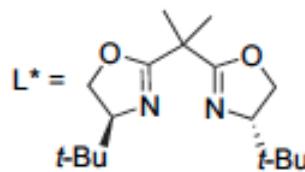
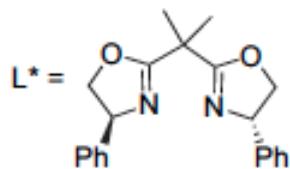
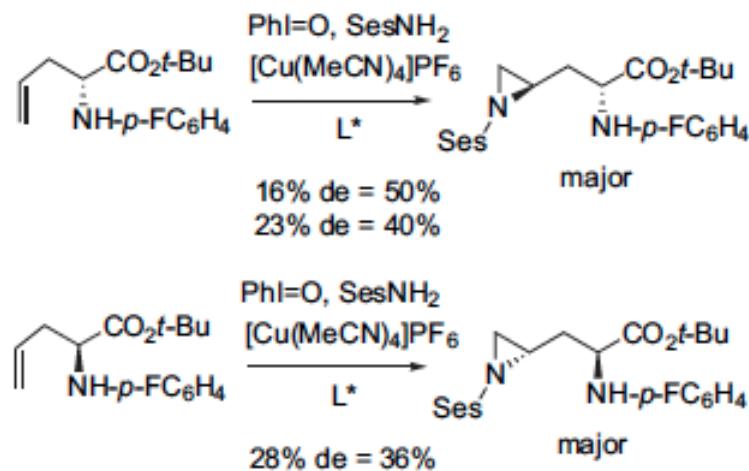
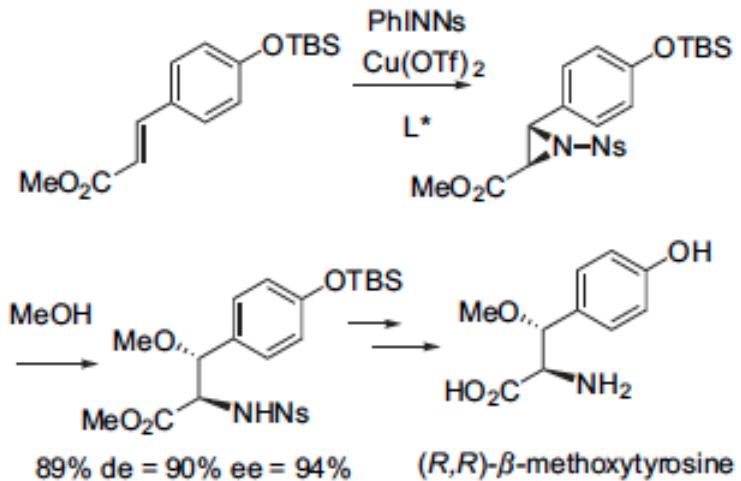
- Z. Li, K. R. Conser, E. N. Jacobsen, *J. Am. Chem. Soc.* **1993**, *115*, 5326.  
 Z. Li, R. W. Quan, E. N. Jacobsen, *J. Am. Chem. Soc.* **1995**, *117*, 5889.  
 J. B. Sweeney, *Chem. Soc. Rev.*, **2002**, *31*, 247–258  
 P. Muller, C. Fruit, *Chem. Rev.* **2003**, *103*, 2905–2919

## *Proposed Mechanism (Jacobsen)*



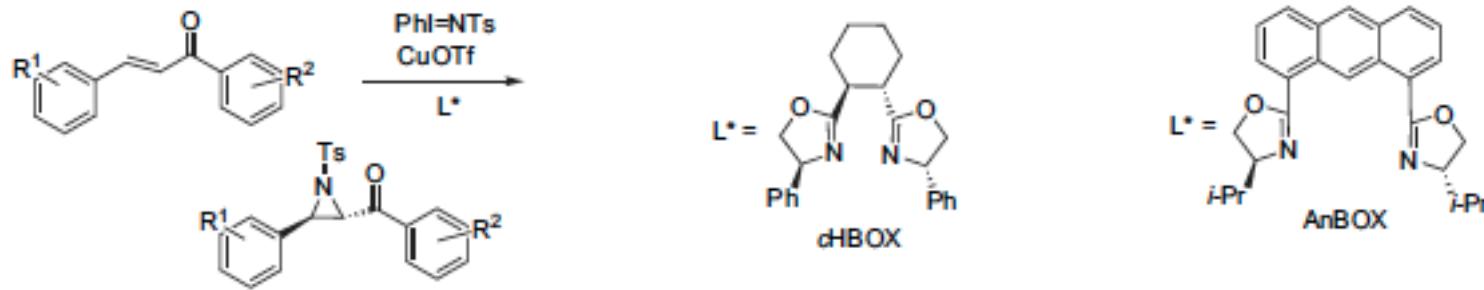
Z. Li, K. R. Conser, E. N. Jacobsen, *J. Am. Chem. Soc.* **1993**, *115*, 5326.  
Z. Li, R. W. Quan, E. N. Jacobsen, *J. Am. Chem. Soc.* **1995**, *117*, 5889.

# *Cu-catalyzed aziridinations(nitrenes)*



Cranfill, D. C.; Lipton, M. A. *Org. Lett.* **2007**, *9*, 3511–3513.  
 Leman, L.; Sanie` re, L.; Dauban, P.; Dodd, R. H. *ARKIVOC* **2003**, *vi*, 126–134.  
 H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

# *Cu-catalyzed aziridinations(ligands)*



$R^1 = R^2 = \text{H}$ : 56% ee = 91%

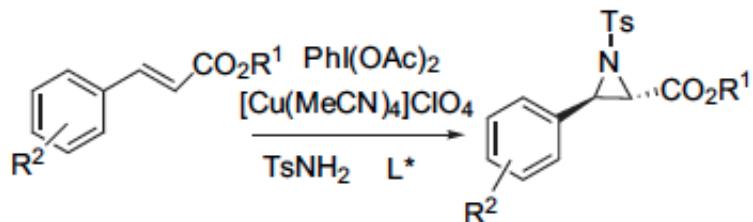
$R^1 = p\text{-Me}$ ,  $R^2 = \text{H}$ : 62% ee = 94%

$R^1 = R^2 = \text{H}$ : 80% ee = 96%

$R^1 = p\text{-Me}$ ,  $R^2 = \text{H}$ : 86% ee = 98%

- Ma, L.; Du, D.-M.; Xu, J. *J. Org. Chem.* **2005**, *70*, 10155–10158.  
Xu, J.; Ma, L.; Jiao, P. *Chem. Commun.* **2004**, 1616–1617  
Ma, L.; Jiao, P.; Zhang, Q.; Xu, J. *Tetrahedron: Asymmetry* **2005**, *16*, 3718–3734.  
H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

# *Cu-catalyzed aziridinations(ligands)*



R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = H: 49% ee = 96%

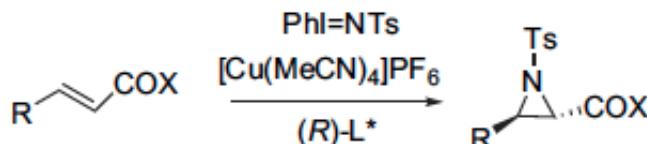
R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = 4-F: 55% ee = 95%

R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = 4-Cl: 45% ee = 95%

R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = 4-Me: 39% ee = 88%

R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = 4-MeO: 41% ee = 74%

R<sup>1</sup> = *t*-Bu, R<sup>2</sup> = 2-NO<sub>2</sub>: 19% ee = 97%



from (R)-BINIM-DC:

R = *p*-ClC<sub>6</sub>H<sub>4</sub>, X = OMe: 82% ee = 81% (2*S*,3*R*)

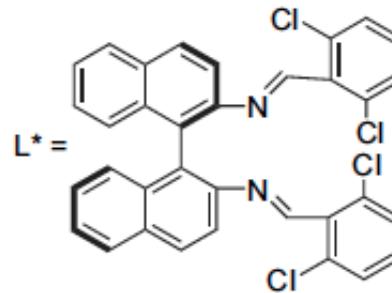
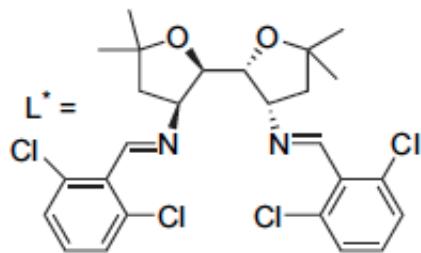
R = 1-Naph, X = OMe: 74% ee = 77% (2*S*,3*R*)

R = 2-Naph, X = OMe: 74% ee = 68% (2*S*,3*R*)

R = Ph, X = OPh: 48% ee = 89% (2*S*,3*R*)

R = Ph, X = *O**t*-Bu: 57% ee = 98% (2*S*,3*R*)

R = X = Ph: 87% ee = 84% (2*S*,3*R*)

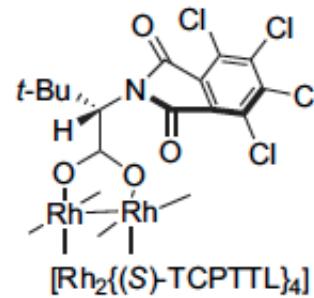
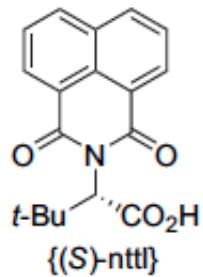
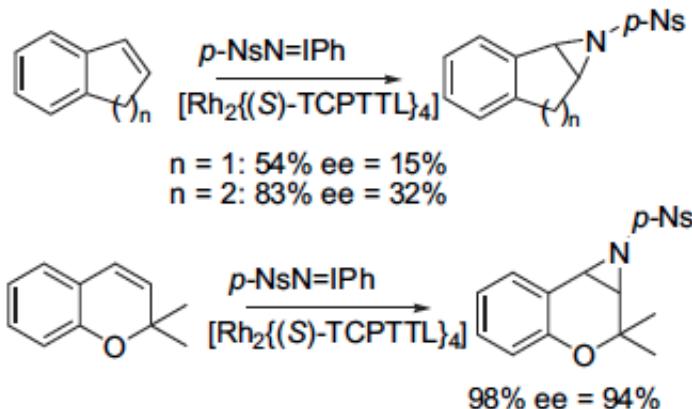
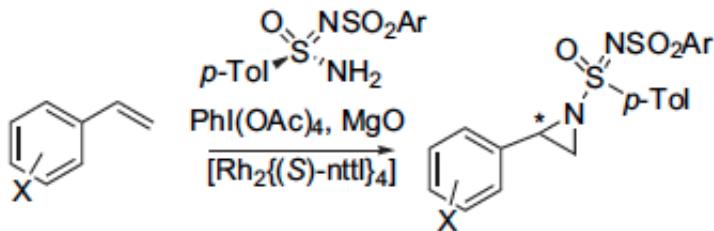


Wang, X.; Ding, K. *Chem.-Eur. J.* **2006**, *12*, 4568–4575.

Suga, H.; Kakehi, A.; Ito, S.; Ibata, T.; Fudo, T.; Watanabe, Y.; Kinoshita, Y. *Bull. Chem. Soc. Jpn.* **2003**, *76*, 189–199

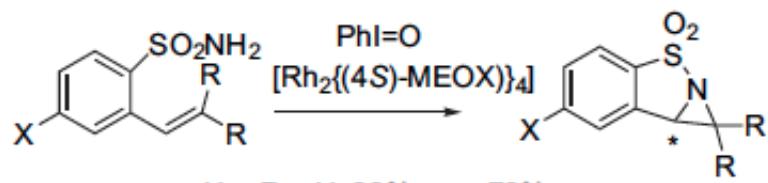
H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

# *Rh-catalyzed aziridinations*

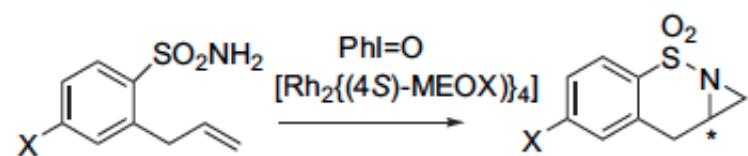


Fruit, C.; Robert-Peillard, F.; Bernardinelli, G.; Muller, P.; Dodd, R. H.; Dauban, P.  
*Tetrahedron: Asymmetry* **2005**, *16*, 3484–3487.  
 Yamawaki, M.; Tanaka, M.; Abe, T.; Anada, M.; Hashimoto, S. *Heterocycles* **2007**, *72*, 709–721  
 H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

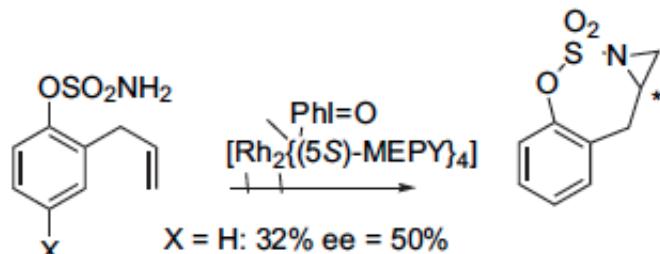
# *Rh-catalyzed intramolecular aziridinations*



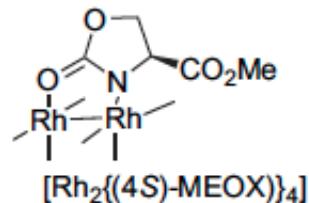
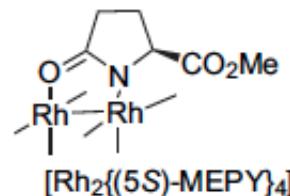
$X = R = H$ : 90% ee = 76%  
 $X = \text{Me}, R = H$ : 71% ee = 74%  
 $X = \text{Br}, R = H$ : 74% ee = 75%  
 $X = H, R = \text{Me}$ : 81% ee = 64%  
 $X = \text{Cl}, R$ : 95% ee = 63%



$X = \text{Me}$ : 73% ee = 55%  
 $X = \text{Cl}$ : 68% ee = 67%  
 $X = H$ : 71% ee = 57%



$X = H$ : 32% ee = 50%  
 $X = \text{H}$ : 10% ee = 52%  
 $X = \text{F}$ : 63% ee = 17%  
 $X = \text{OMe}$ : 18% ee = 39%

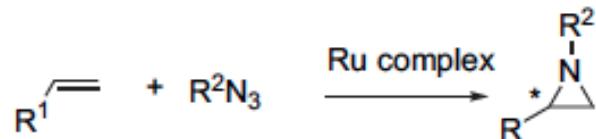


Liang, J.-L.; Yuan, S.-X.; Hong Chan, P.W.; Che, C.-M. *Tetrahedron Lett.* **2003**, *44*, 5917–5920.

Hayes, C. J.; Beavis, P. W.; Humphries, L. A. *Chem. Commun.* **2006**, 4501–4502.

H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

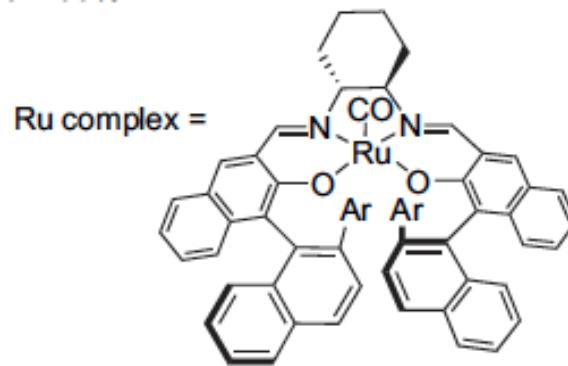
# *Ru-catalyzed aziridinations*



90% ee = 87%

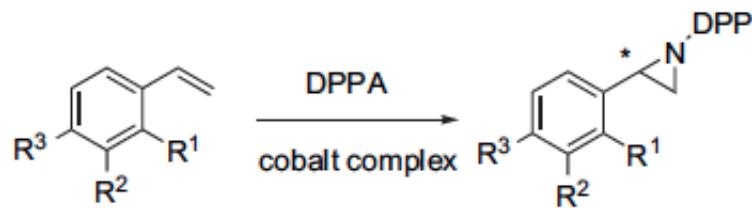
$\text{R}^1=p\text{-BrC}_6\text{H}_4, \text{R}^2=p\text{-Ns}, \text{Ar}=3,5\text{-Cl}_2\text{-4-(Me)}_2\text{SiC}_6\text{H}_2:$

93% ee = 83%



- Kawabata, H.; Omura, K.; Katsuki, T. *Tetrahedron Lett.* **2006**, 47, 1571–1574  
Kawabata, H.; Omura, K.; Uchida, T.; Katsuki, T. *Chem. Asian. J.* **2007**, 2, 248–256  
H. Pellissier, *Tetrahedron*, **2010**, 66, 1509-1555

# *Co-catalyzed aziridinations*

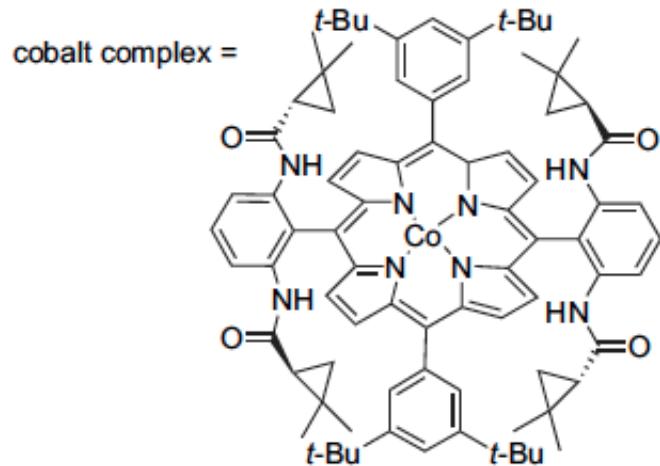


$R^1 = R^2 = R^3 = H$ : 88% ee = 37%

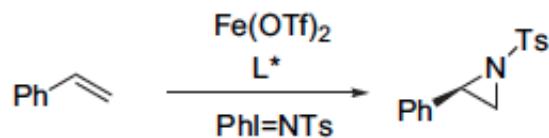
$R^1 = \text{Me}$ ,  $R^2 = R^3 = H$ : 35% ee = 46%

$R^1 = R^3 = H$ ,  $R^2 = \text{Me}$ : 52% ee = 44%

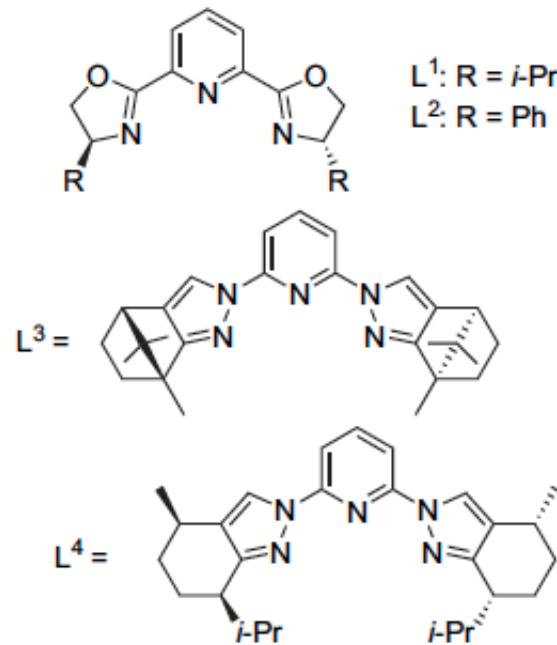
$R^1 = R^2 = H$ ,  $R^3 = \text{Me}$ : 58% ee = 37%



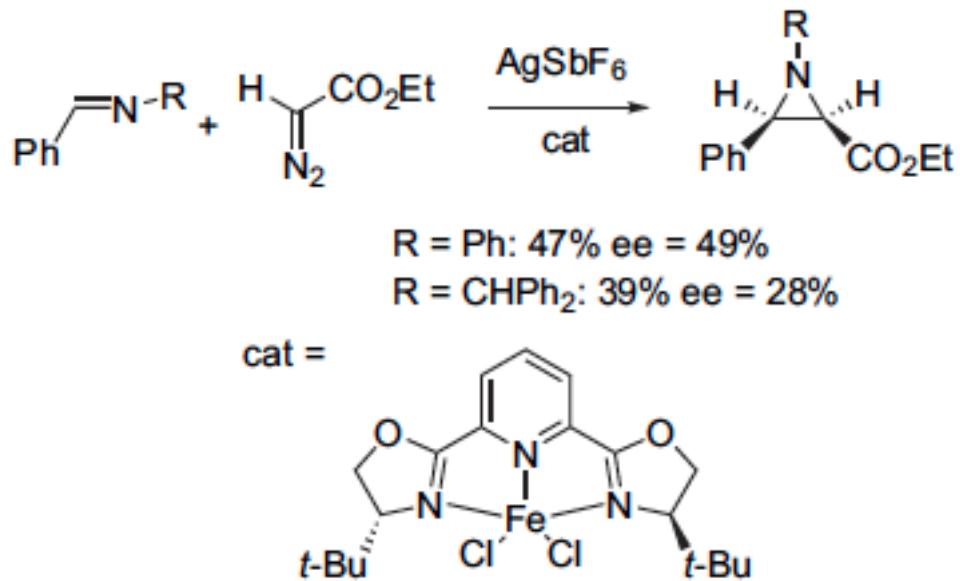
# *Fe-catalyzed aziridinations*



L $^*$  = L $^1$ : 72% ee = 40%  
L $^*$  = L $^2$ : 51% ee = 25%  
L $^*$  = L $^3$ : 60% ee = 20%  
L $^*$  = L $^4$ : 40% ee = 6%

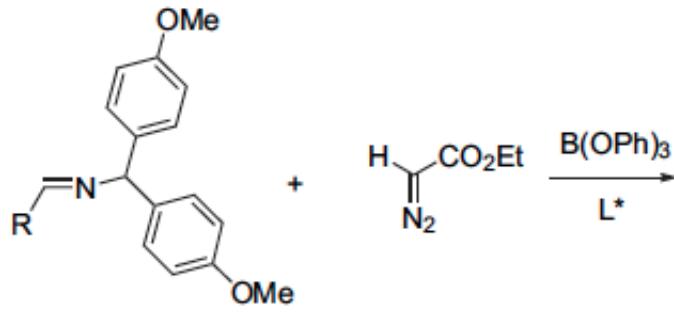


## *Carbene methods*



Redlich, M.; Hossain, M. M. *Tetrahedron Lett.* **2004**, *45*, 8987–8990.  
H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509-1555

# *Carbene methods*



with L\* = VAPOL:

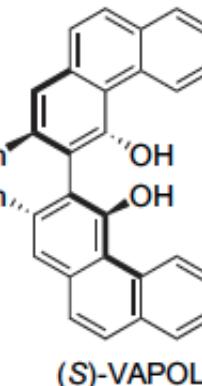
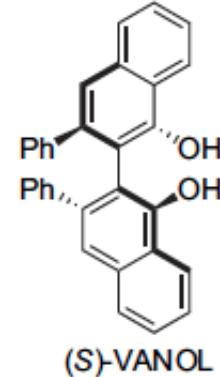
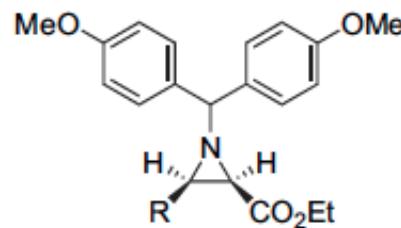
R = Cy: 86% *cis:trans* = 97:3 ee = 84%

R = *t*-Bu: 70% *cis:trans* = 98:2 ee = 75%

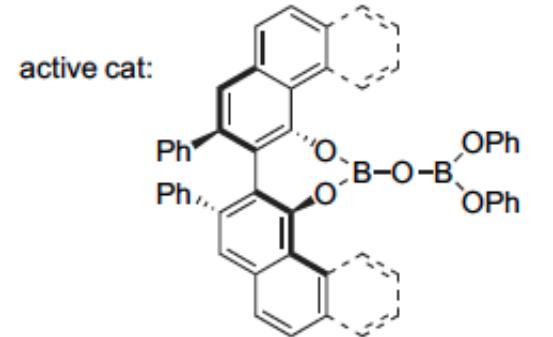
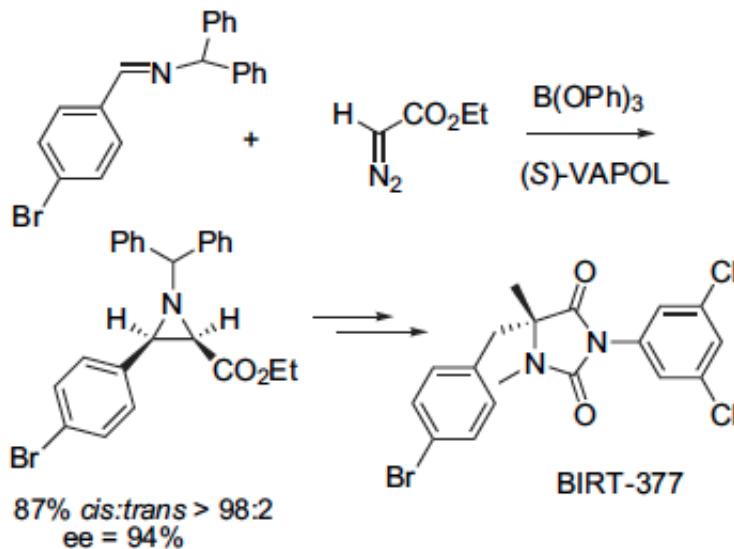
with L\* = VANOL:

R = Cy: 69% *cis:trans* = 98:2 ee = 77%

R = *t*-Bu: 77% *cis:trans* = 98:2 ee = 87%

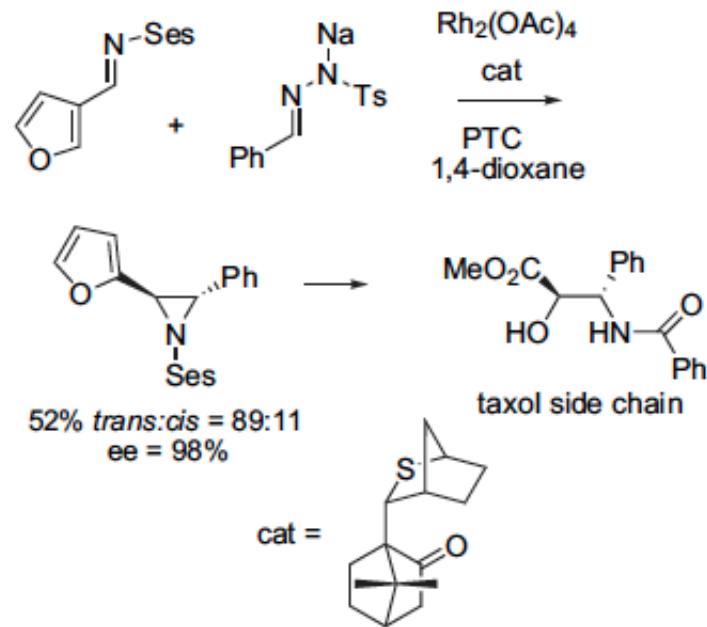


# *Carbene methods*



Lu, Z.; Zhang, Y.; Wulff, W. D. *J. Am. Chem. Soc.* **2007**, *129*, 7185–7194  
Zhang, Y.; Desai, A.; Lu, Z.; Hu, G.; Ding, Z.; Wulff, W. D. *Chem.-Eur. J.* **2008**, *14*, 3785–3803.  
H. Pellissier, *Tetrahedron*, **2010**, *66*, 1509–1555

# *Sulfur ylide-mediated methods*

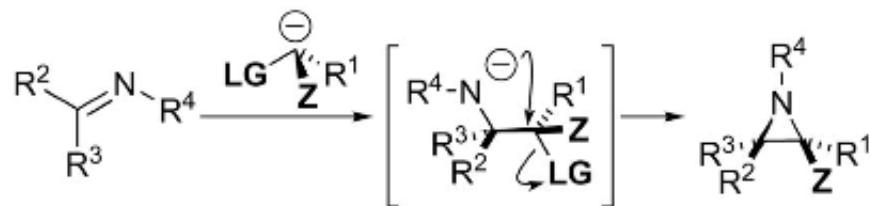


Aggarwal, V. K.; Winn, C. L. *Acc. Chem. Res.* **2004**, 37, 611–620.

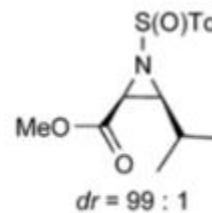
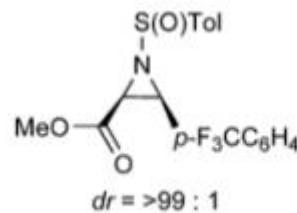
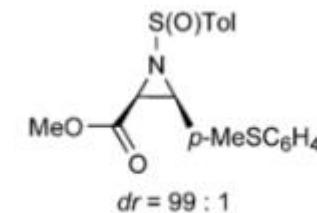
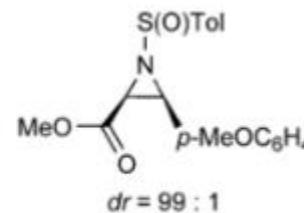
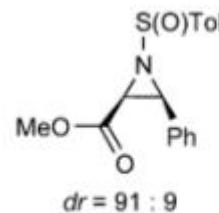
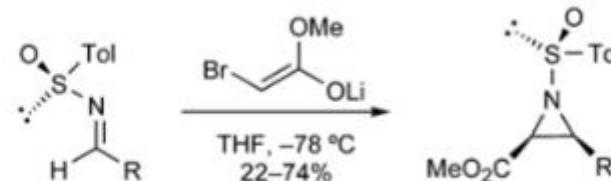
Aggarwal, V. K.; Vasse, J.-L. *Org. Lett.* **2003**, 5, 3987–3990.

H. Pellissier, *Tetrahedron*, **2010**, 66, 1509–1555

# Aza-Darzens Reaction



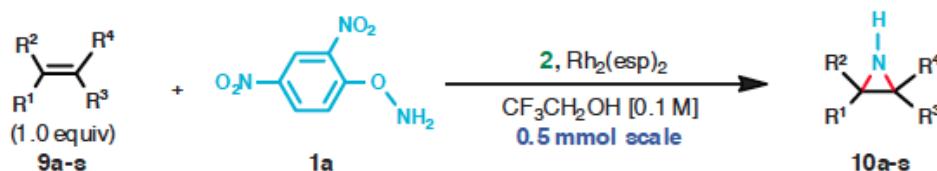
$\text{LG}^{\oplus}$  = Cl, Br, I, SR<sub>2</sub>, S(O)R<sub>2</sub>, N<sub>2</sub>  
 $\text{Z}^{\oplus}$  = C(O)<sub>n</sub>R, P(O)<sub>n</sub>R<sub>2</sub>, S(O)<sub>n</sub>R, CN



# *Conclusion*

- Aziridines leads to building blocks for heterocycle chemistry and natural products.
- The direct aziridination of olefins is more difficult than the corresponding epoxidation.
- Most of these methods rely either on the transfer of substituted nitrenes to the C=C bond of olefins or the transfer of substituted carbenes to the C=N bond of imines.
- Normally, the result is an aziridine bearing a strongly electron-withdrawing N-protecting group (e.g., Ts:para-toluenesulfonyl; Ns: para-nitrophenylsulfonyl).
- In addition, the high reactivity of N-protected nitrenes might give rise to nonproductive allylic C-H amination products, as well as the loss of stereospecificity.

# Why



## Structure of N-H Aziridines

(Entry): Compound #; T (°C), t (h), Isolated Yield (%), Regio - and diastereoselectivity [ratios]

### N-H Aziridination of aliphatic mono-, di-, and tri-substituted olefins:

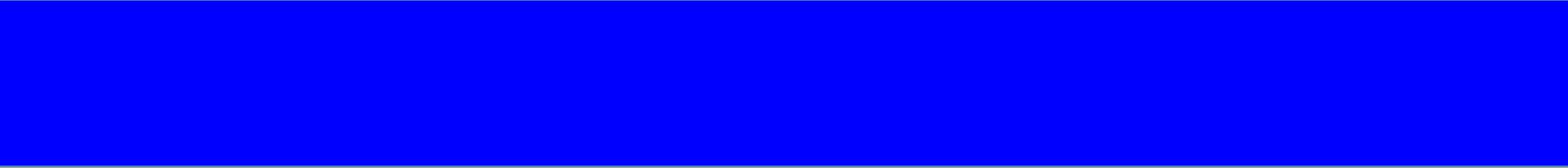
 (1): <b>10a</b>	 (2): <b>10b</b>	 (3): <b>10c</b>	 (4): <b>10d</b>	 (5): <b>10e</b>
<b>1a</b> (1.2 equiv), <b>2</b> (5 mol%) 25 °C, 2 h; 59%	<b>1a</b> (1.2 equiv), <b>2</b> (5 mol%) 25 °C, 2 h; 72%	<b>1a</b> (2 equiv), <b>2</b> (5 mol%) 25 °C, 5 h; 77%	<b>1a</b> (1.2 equiv), <b>2</b> (2x1 mol%) 25 °C, 3 h; 64%	<b>1a</b> (1.2 equiv), <b>2</b> (3x1 mol%) 25 °C, 36 h; 78%
 (6): <b>10f</b>	 (7): <b>10g</b>	 (8): <b>10h</b>	 (9): <b>10i</b>	 (10): <b>10j</b>
<b>1a</b> (1.2 equiv), <b>2</b> (1 mol%) 25 °C, 4 h; 72%	<b>1a</b> (1.2 equiv), <b>2</b> (2 x 1 mol%) 25 °C, 29 h; 55%	<b>1a</b> (1.2 equiv), <b>2</b> (1 mol%) 25 °C, 2 h; 91%	<b>1a</b> (1.2 equiv), <b>2</b> (3x1 mol%) 25 °C, 6 h; 82% [1:1 dr]	<b>1a</b> (1.2 equiv), <b>2</b> (1 mol%) 25 °C, 1 h; 86% + 4% of <b>10jj</b>

### B N-Me Aziridination of di- and tri-substituted aliphatic olefins and styrenes:

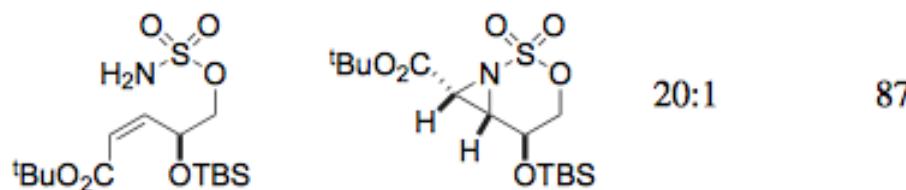
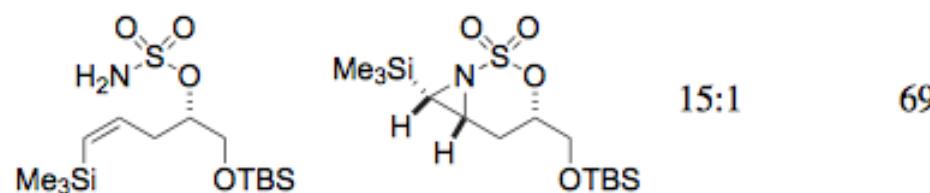
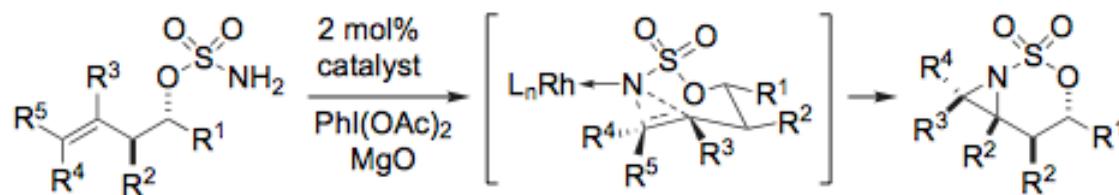
 (29): <b>13a</b>	 (30): <b>13b</b>	 (31): <b>13c</b>	 (32): <b>13d</b>
<b>1b</b> (1.4 equiv) <b>2</b> (1.5 mol%) 25 °C, 4 h; 78%	<b>1b</b> (1.6 equiv) <b>2</b> (2 mol%) 25 °C, 6 h; 80%	<b>1b</b> (1.4 equiv), <b>2</b> (1.25 mol%) 25 °C, 4 h; 83% [>1:30 regioisomers favoring 6,7]	<b>1b</b> (1.4 equiv) <b>2</b> (1.25 mol%) 25 °C, 4 h; 81%



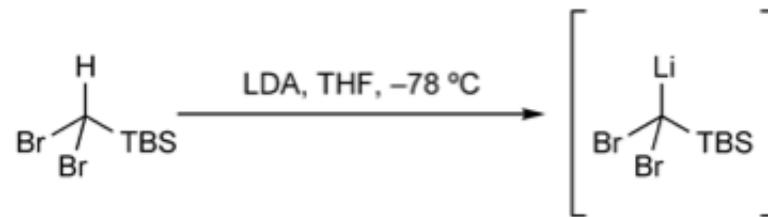
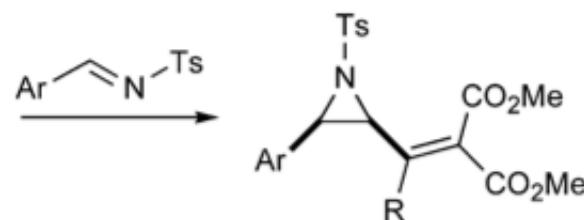
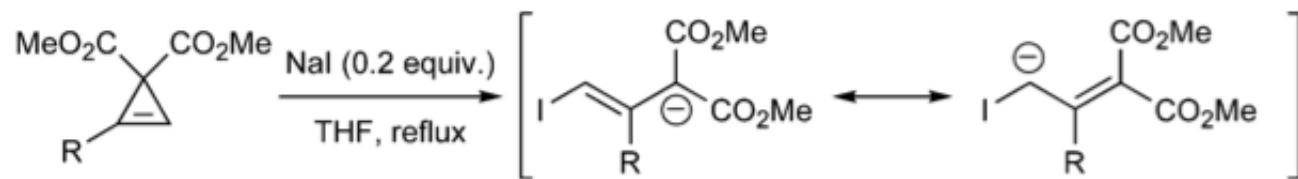
*Thank you*



# Quiz !



# Quiz !



# Quiz !

