

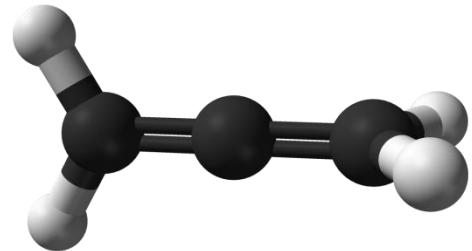
Allenes: Preparation and Recent Application

Nik Savage
Dong Group Literature Talk
September 19, 2012

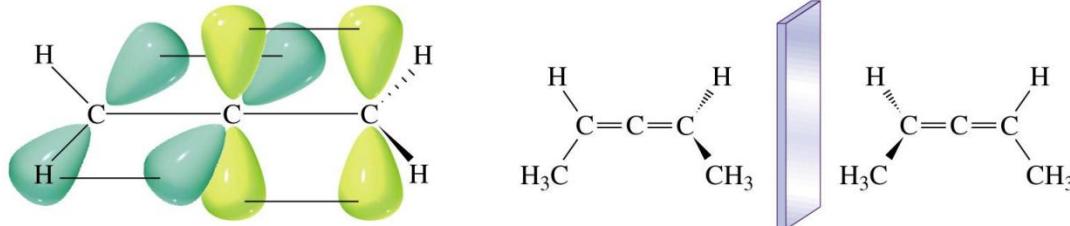
Outline

- Introduction
 - Overview
- Methods of Preparation
 - Elimination, Isomerization, Substitution
 - Homologation and Rearrangements
 - Transition Metal
- Application in Catalytic and Asymmetric Syntheses
 - Selected Examples Toward Natural Product Synthesis

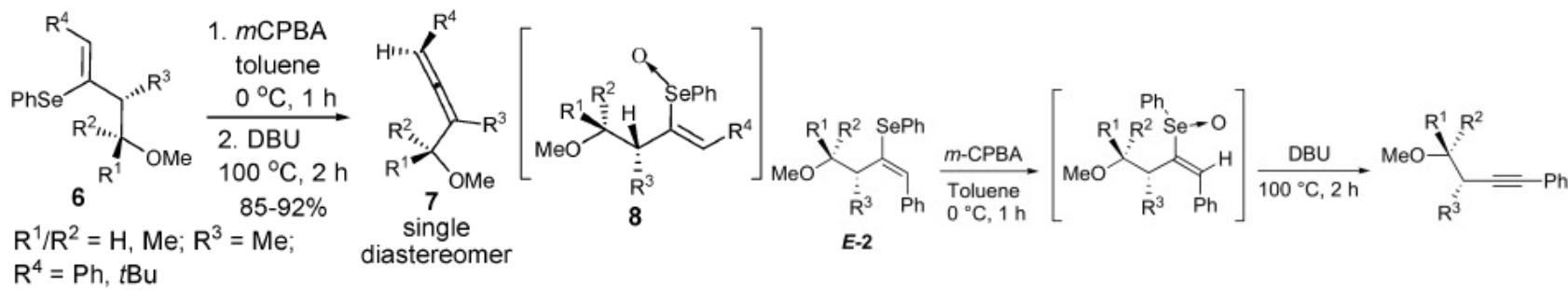
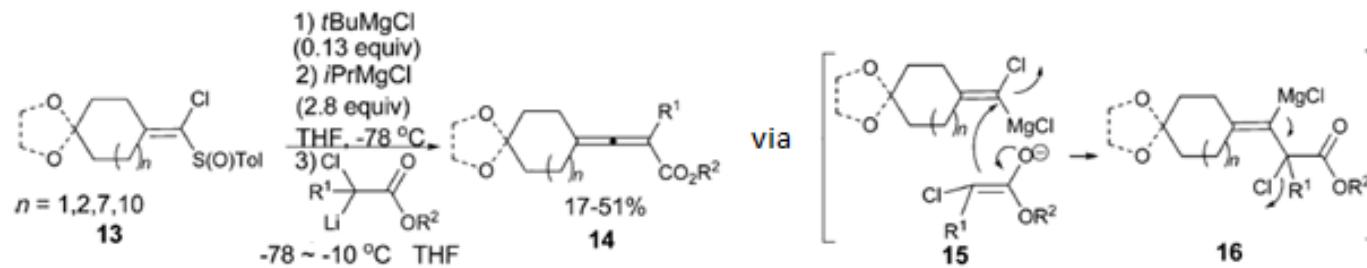
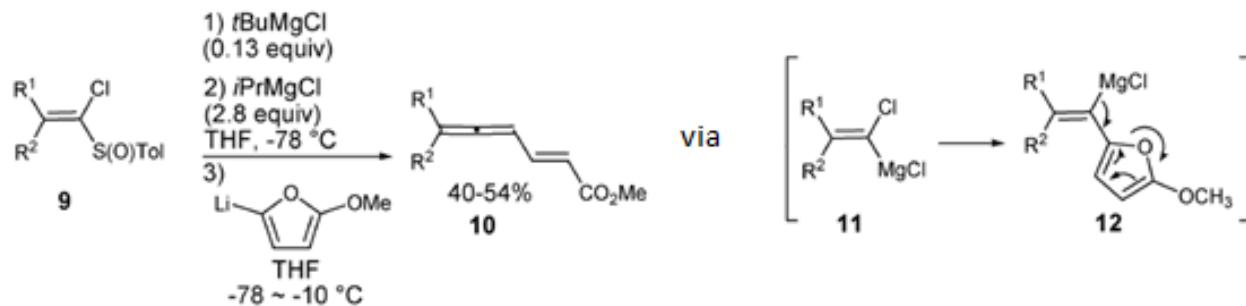
What are Allenes?



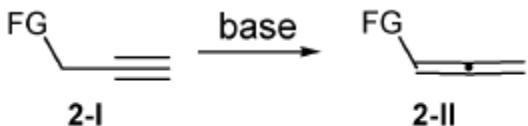
- Simplest class of cumulenes
- Unsaturation over three atoms allows unique flexibility for synthesis
- Higher reactivity than alkenes and alkynes
- Intrinsic axial chirality (elongated tetrahedral)



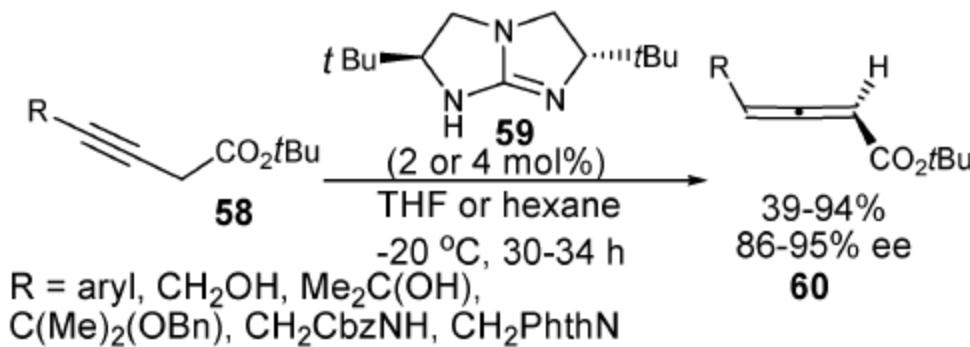
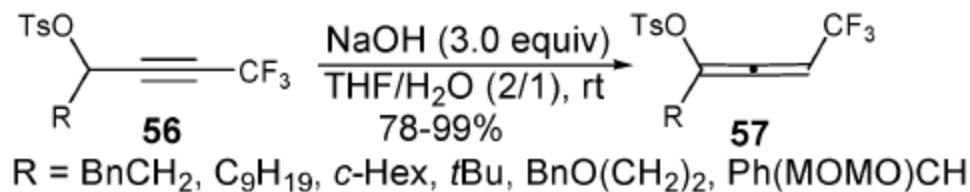
Preparation from Alkenes



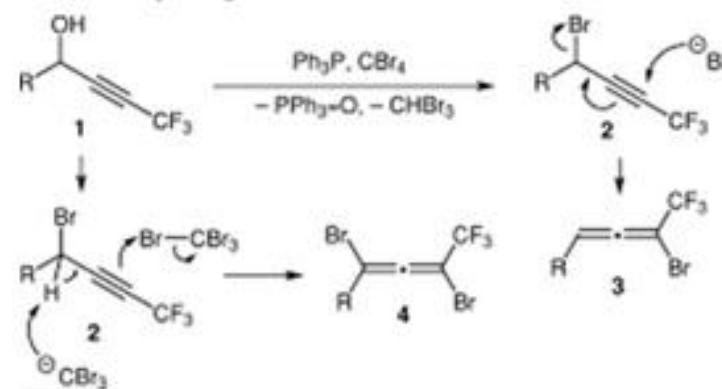
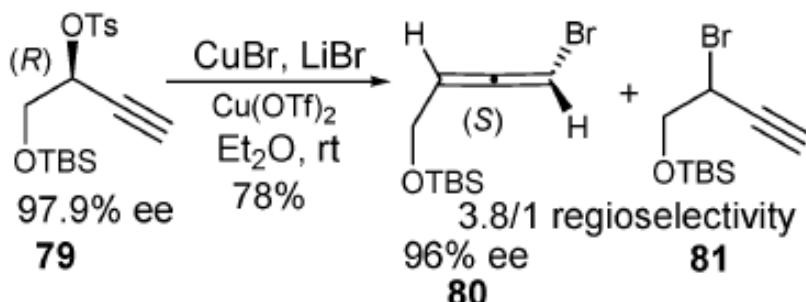
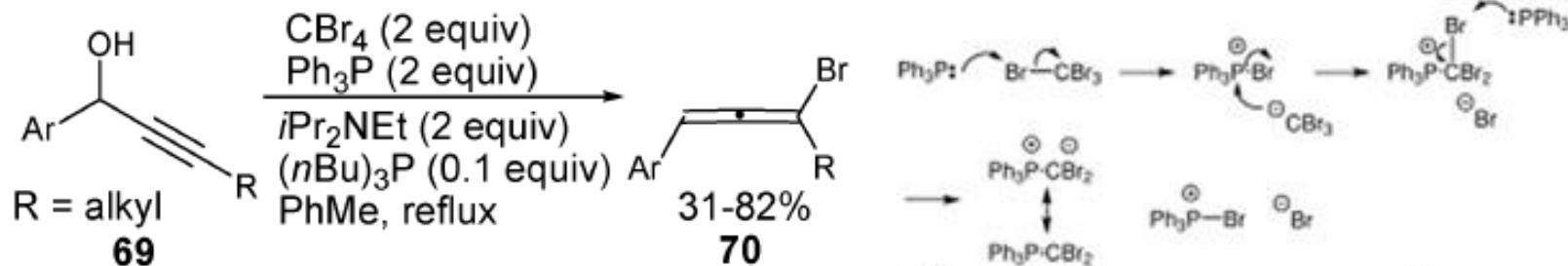
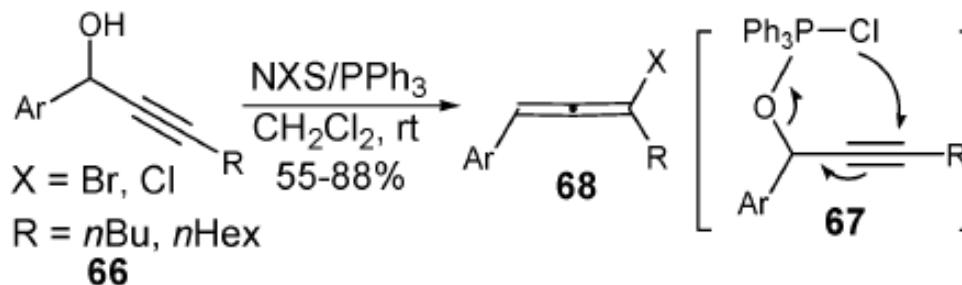
Preparation from Alkynes



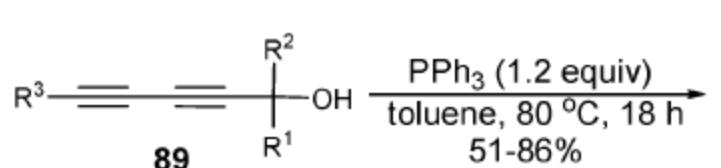
FG: alkenyl, alkynyl, aryl, carbonyl,
 RO^- , R_2N^- , RS^- , $\text{PO}(\text{OEt})_2^-$



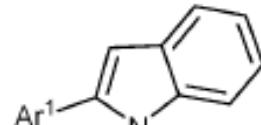
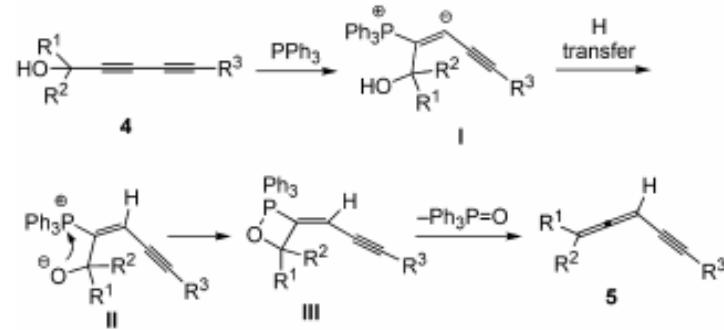
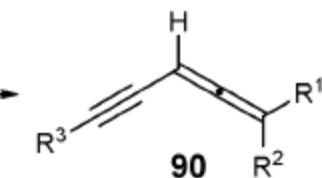
Preparation from Alkynes



Alkynes - Substitution

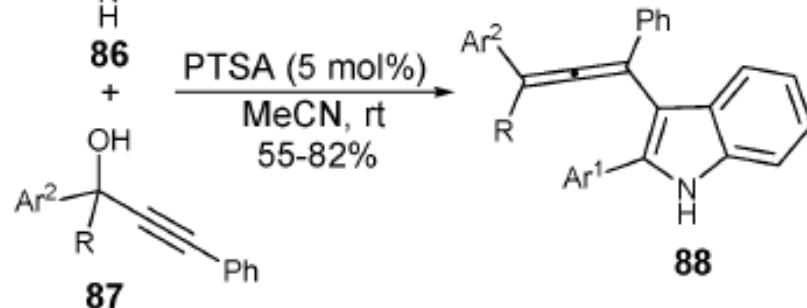


$\text{R}^1, \text{R}^2 = \text{Me, Et, -(CH}_2)_5\text{-, -(CH}_2)_6\text{-, Ph, }n\text{C}_6\text{H}_{13}$

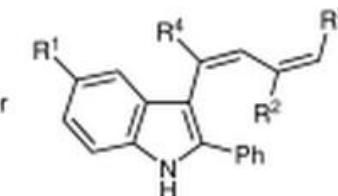
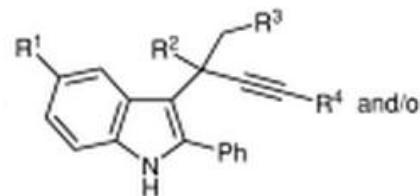


$\xrightarrow[\text{MeCN, rt}]{\text{PTSA (5 mol\%)}}$

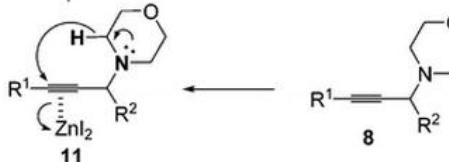
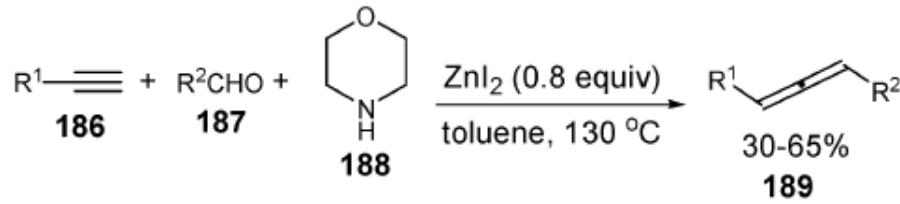
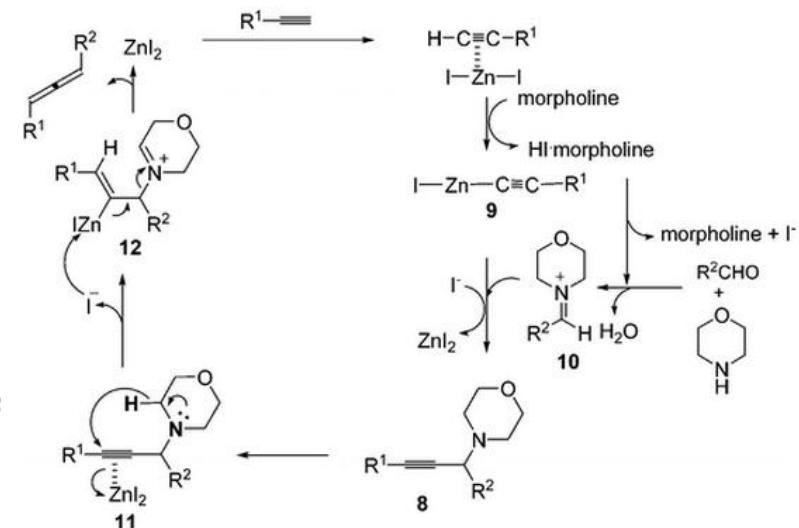
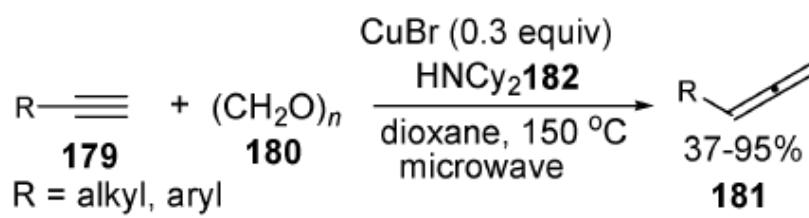
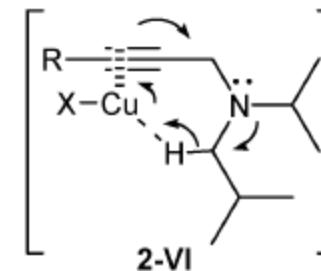
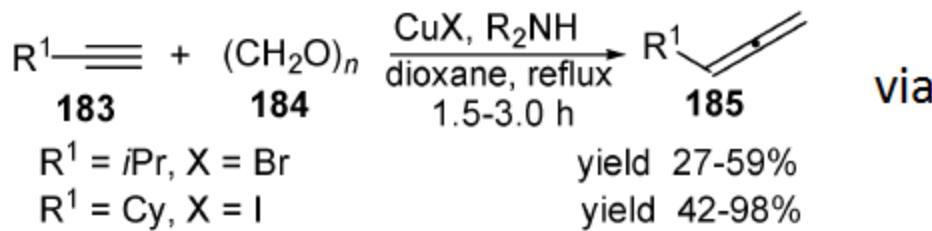
55-82%



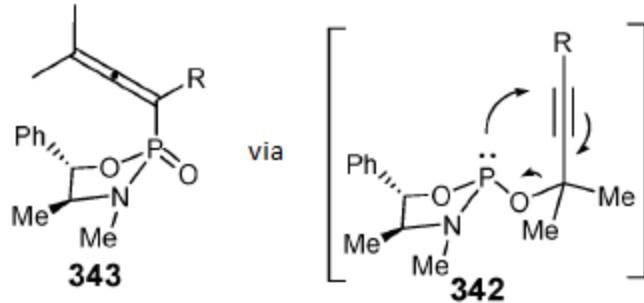
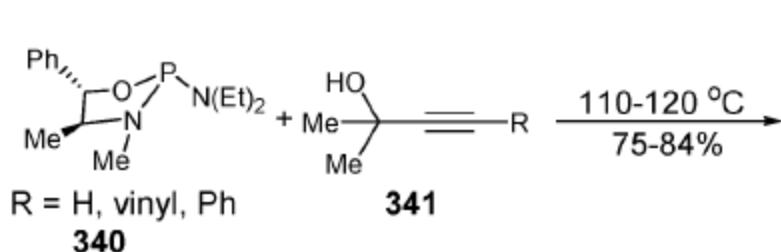
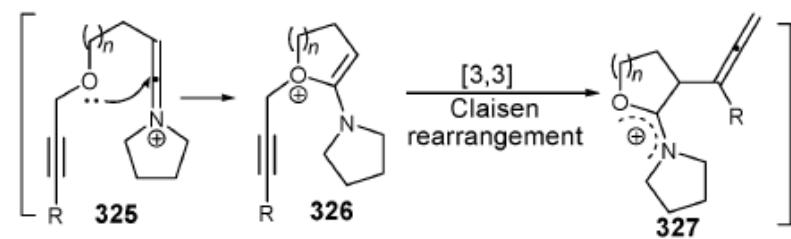
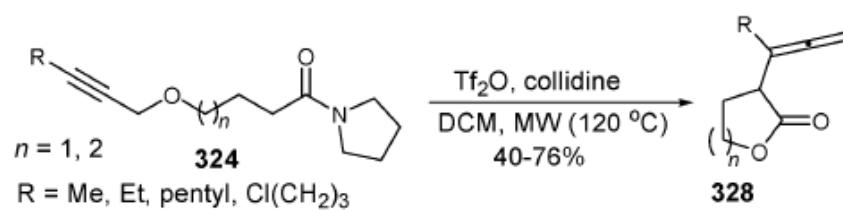
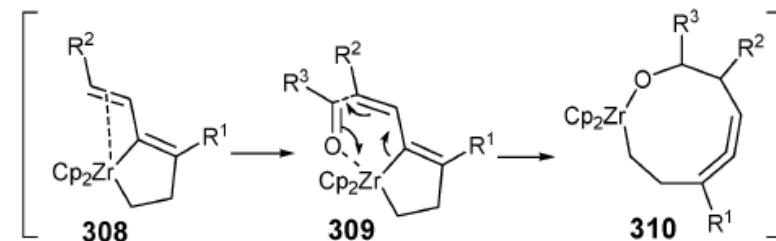
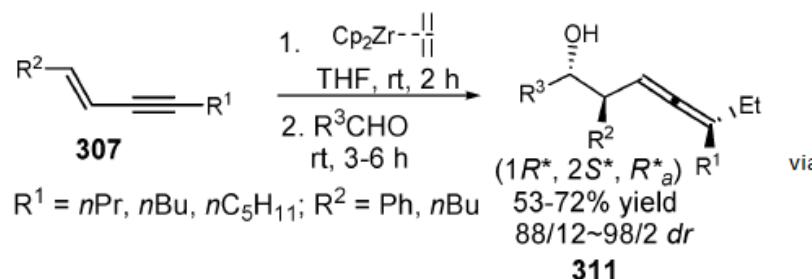
$\text{R} = i\text{Pr, }c\text{-C}_3\text{H}_5, c\text{-C}_4\text{H}_7, c\text{-C}_5\text{H}_9, c\text{-C}_6\text{H}_{11}$



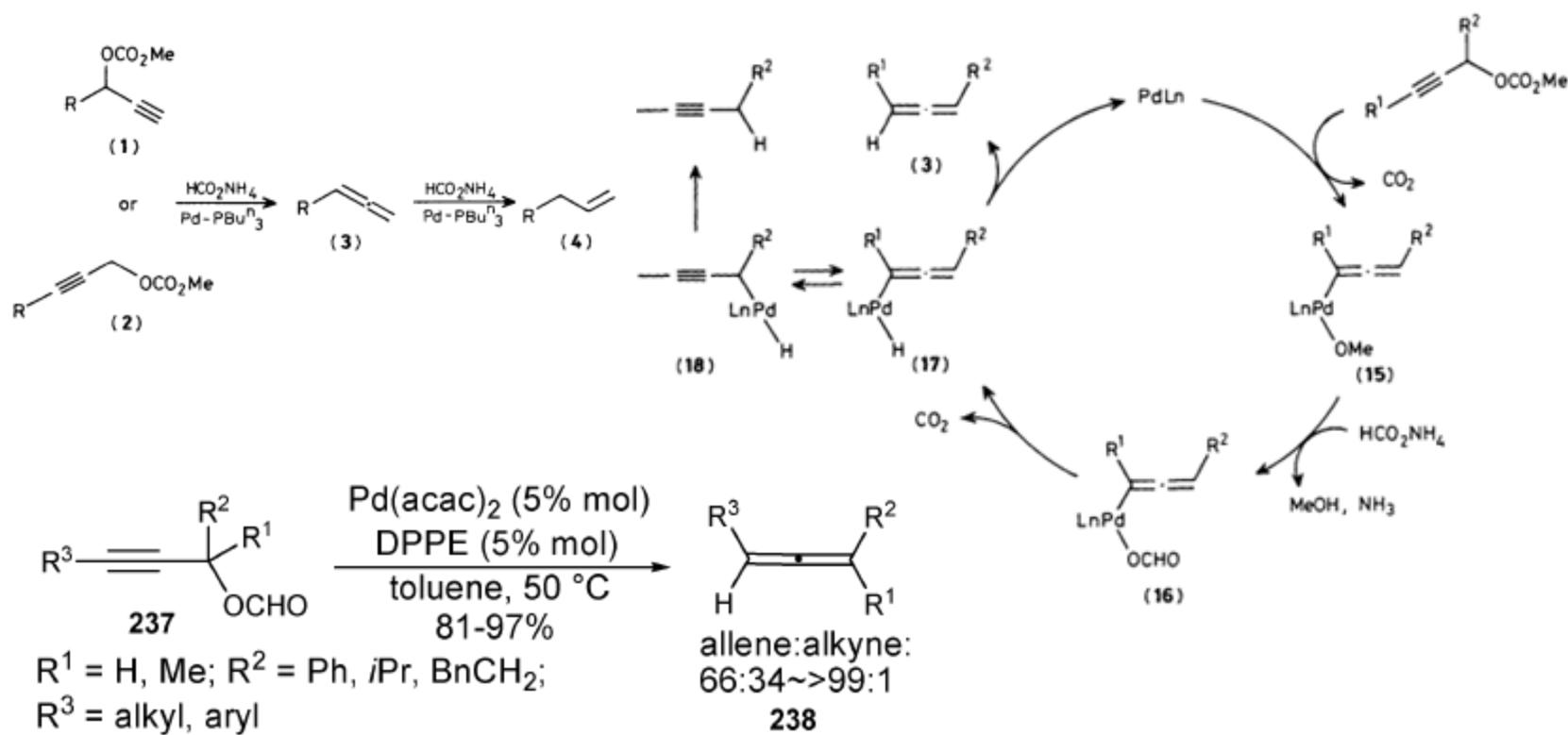
Preparation from Alkynes



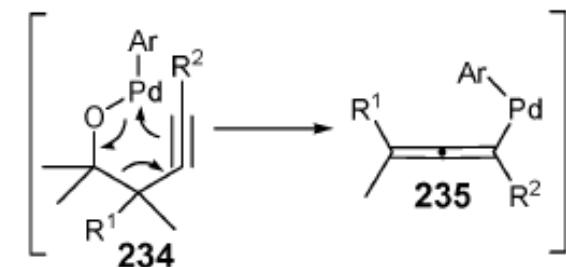
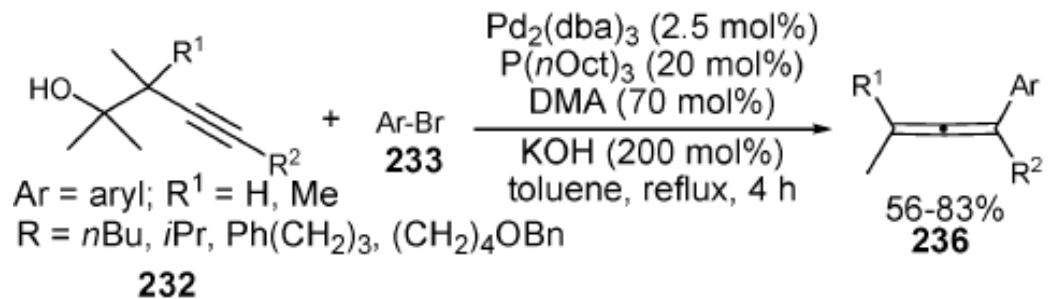
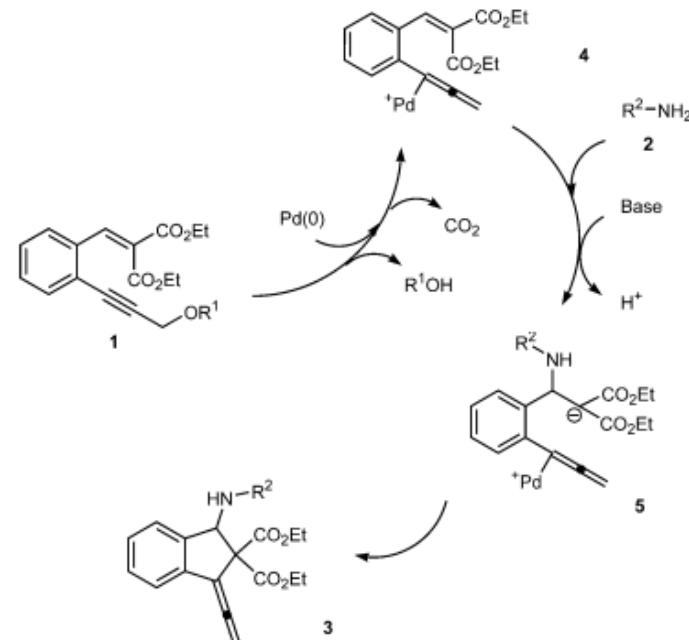
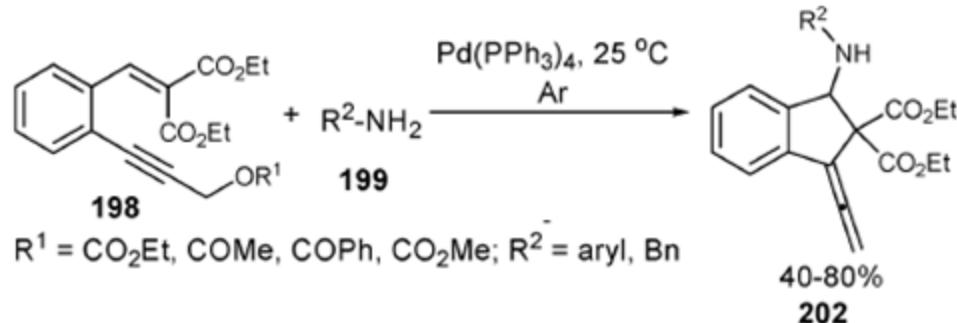
Preparation from Rearrangements



Pd-Catalyzed Allene Formation



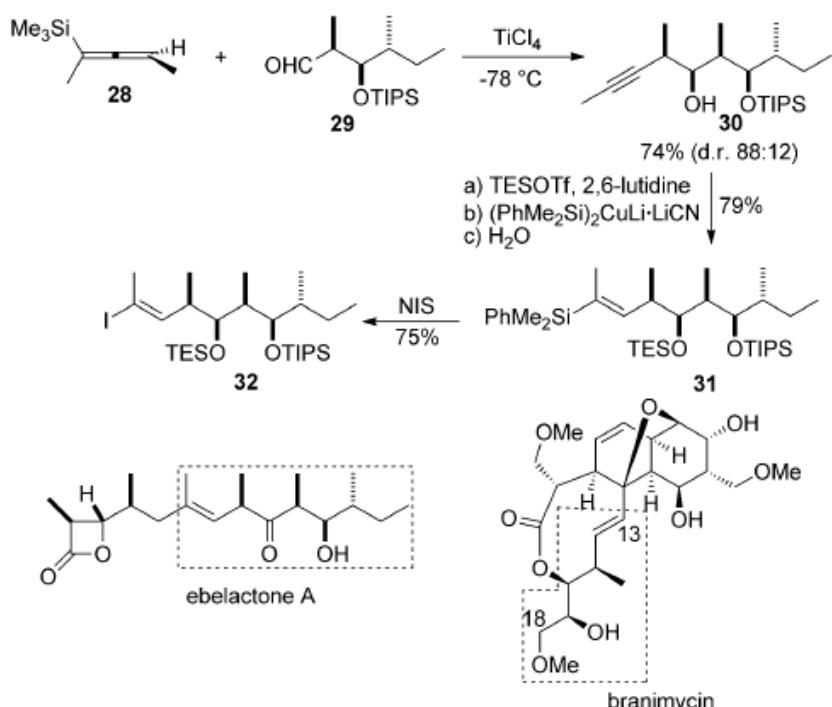
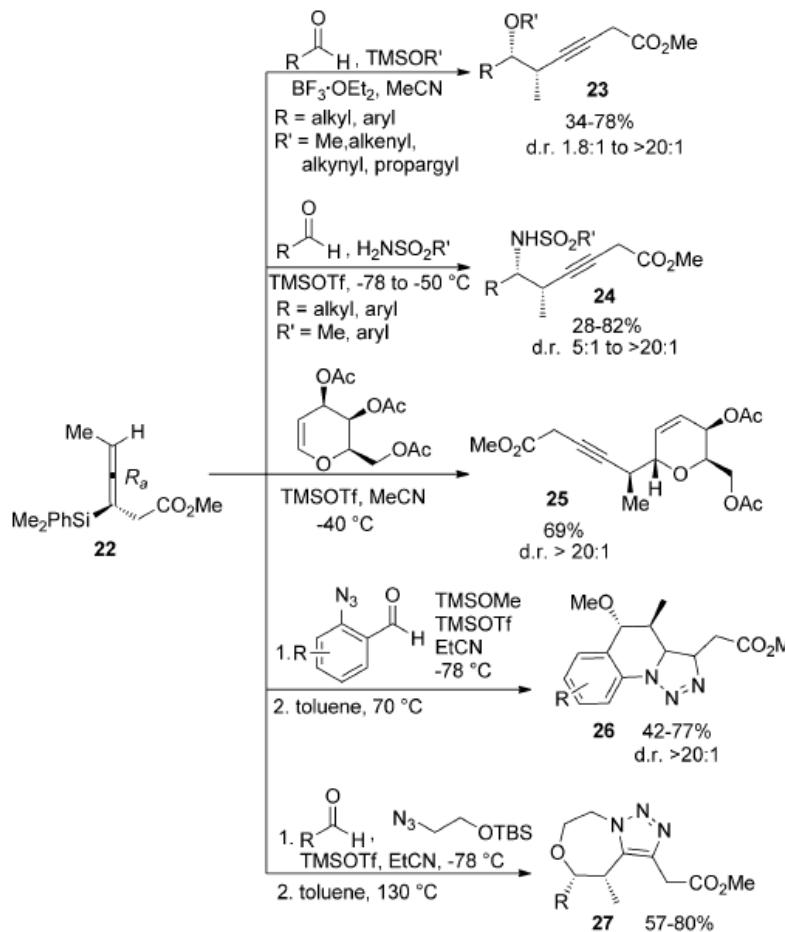
Pd Examples



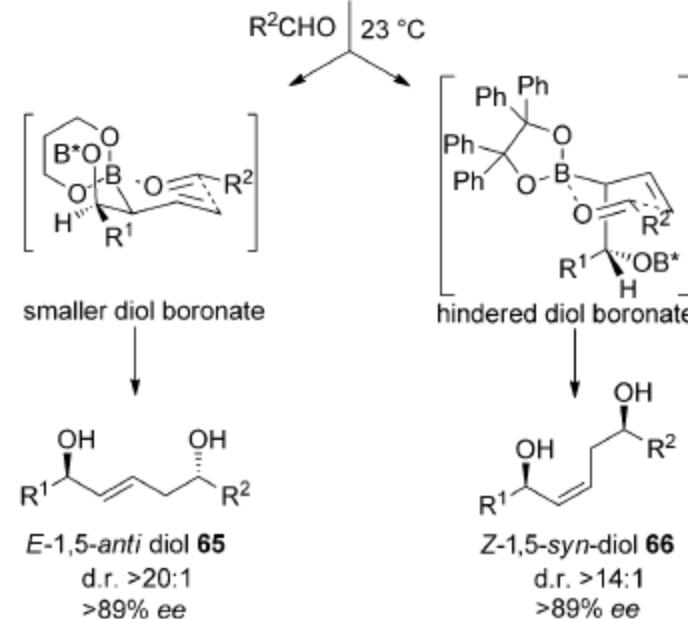
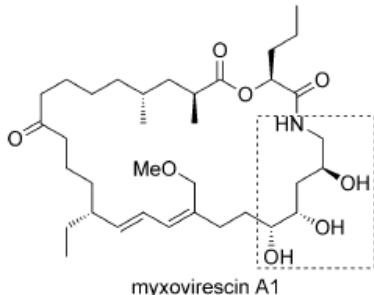
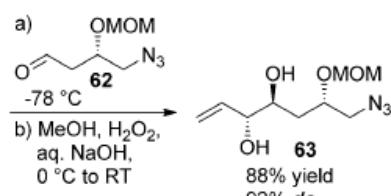
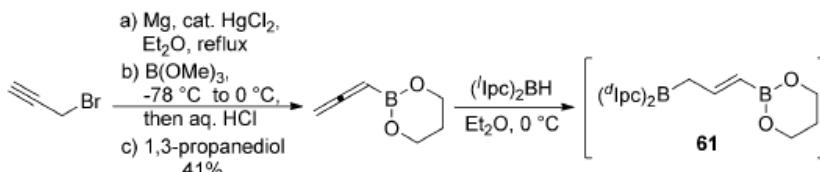
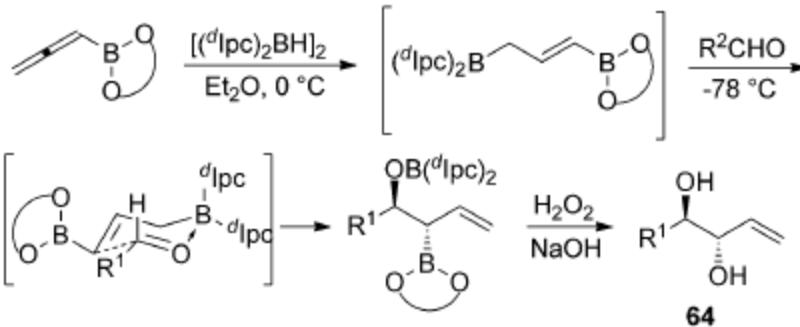
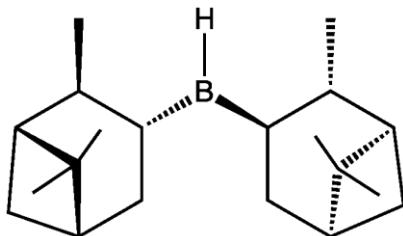
Why Bother Making Them?

- Recent explosion (> 2003) of catalytic and asymmetric reactions involving allenes (oftentimes applied toward making a natural product)
- Starting allenes may/may not have been discussed previously

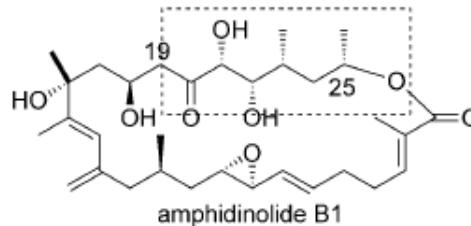
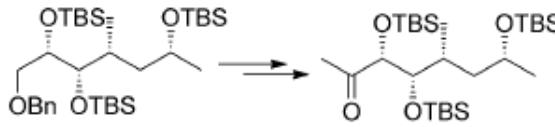
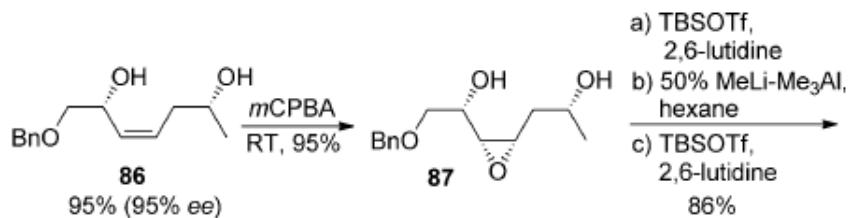
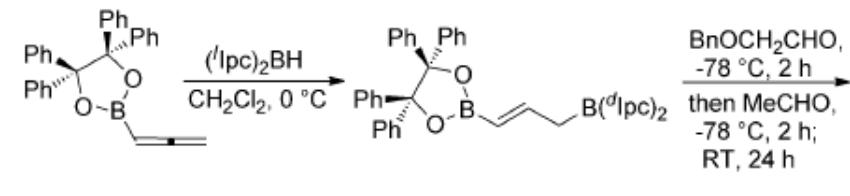
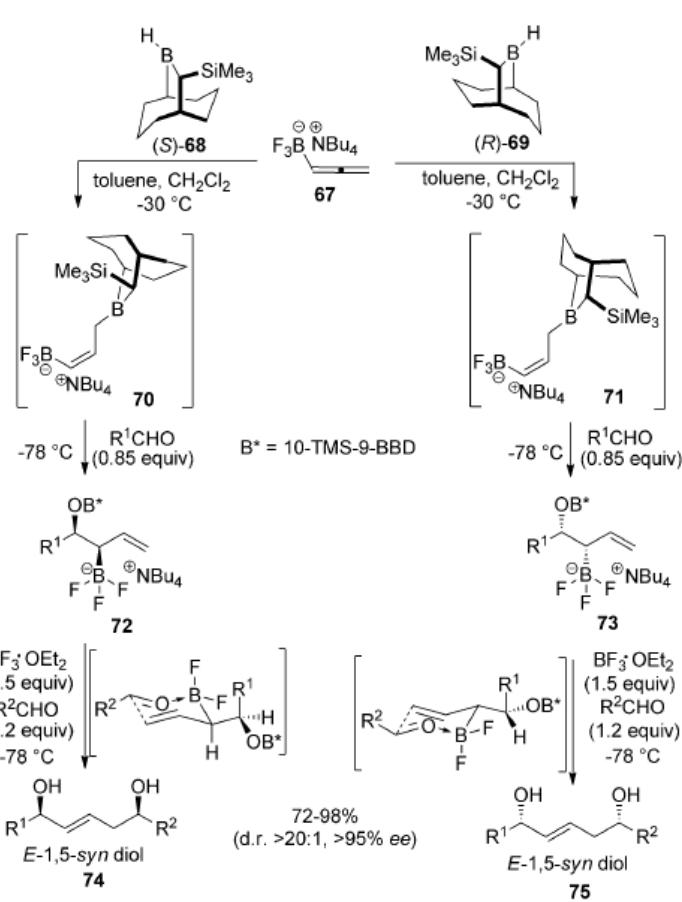
Reactions With Allenylsilane Reagents



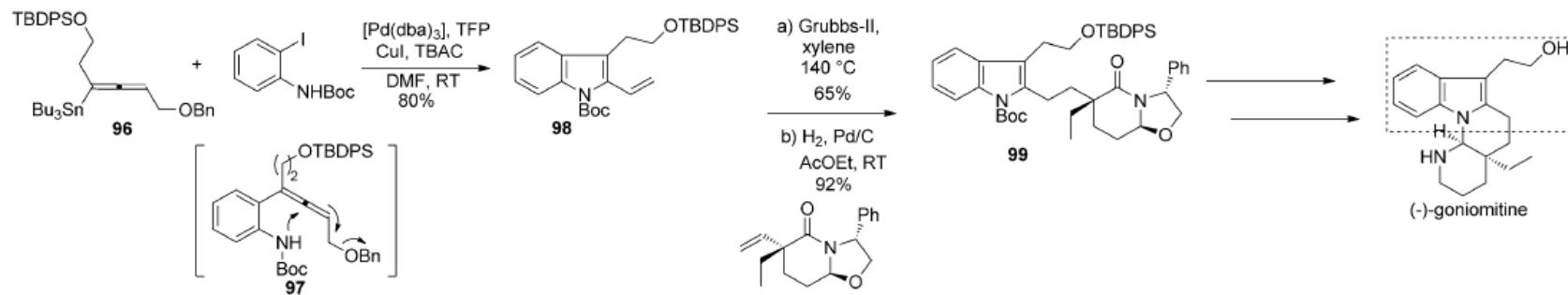
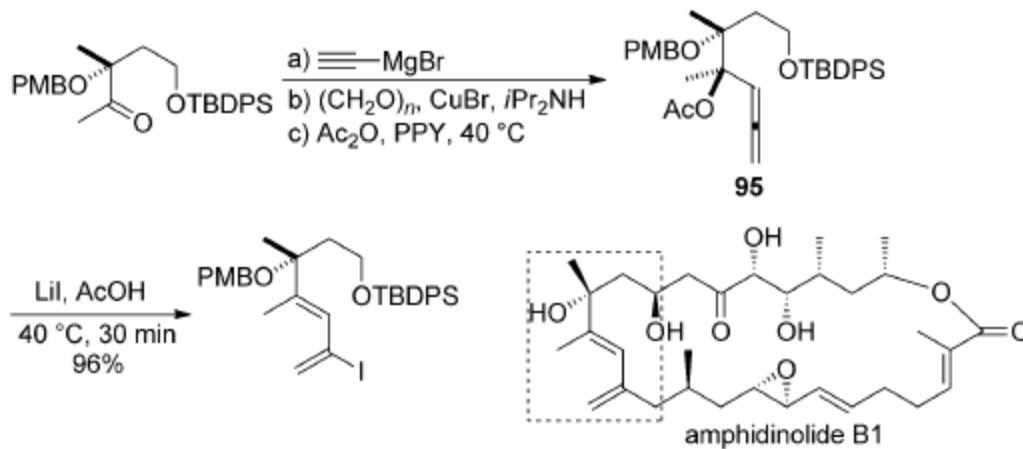
Allenylboranes



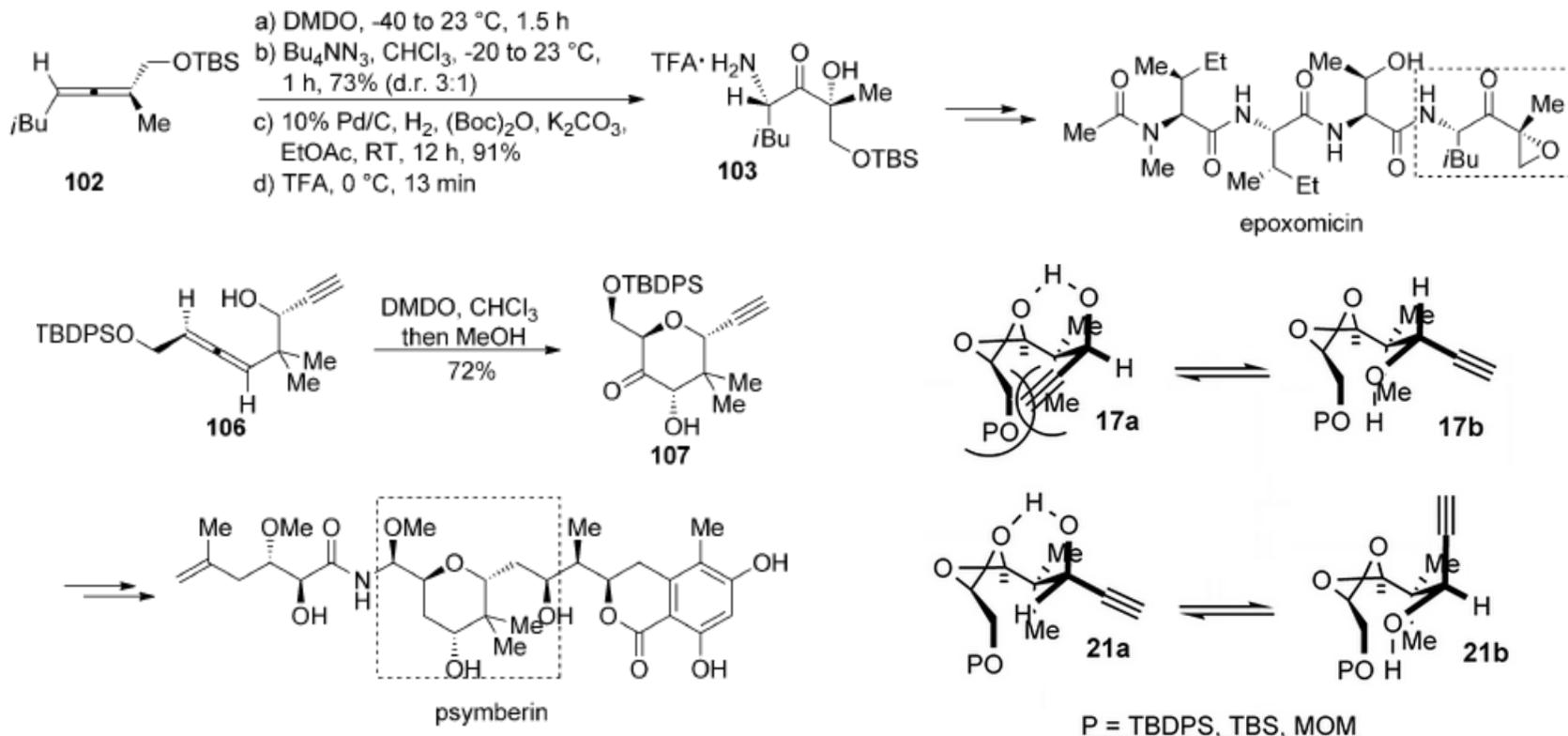
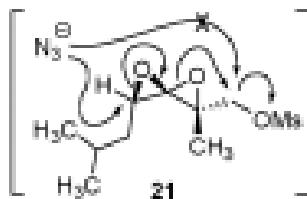
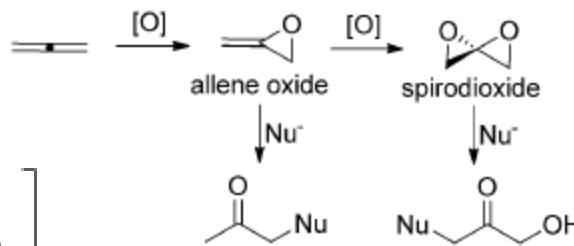
Improvements to 1,5 Diols

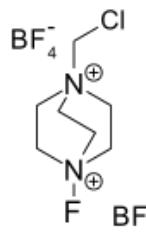


Allenes with an α -Functional Group

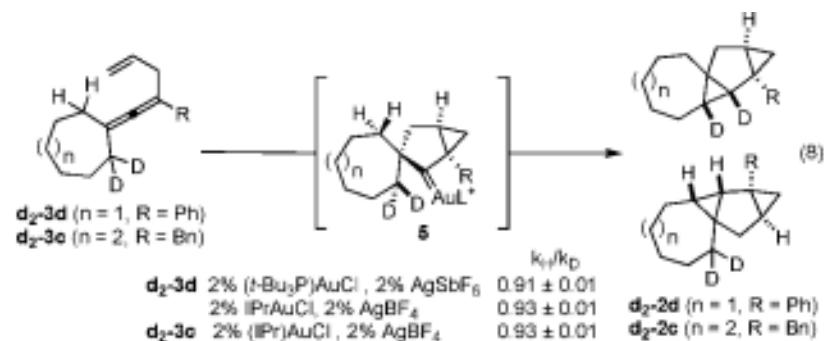
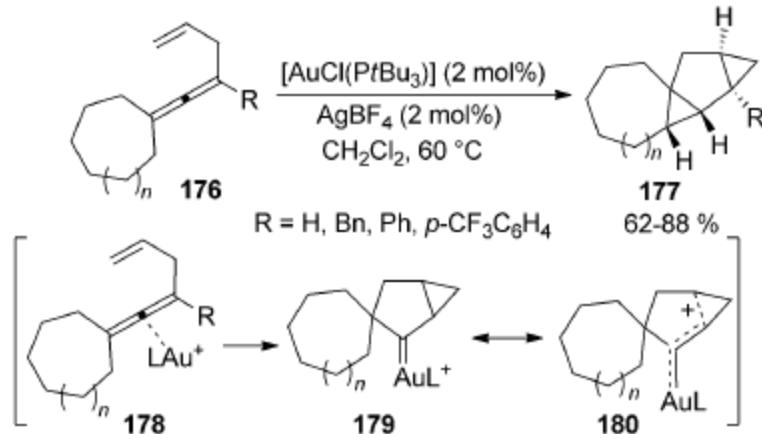
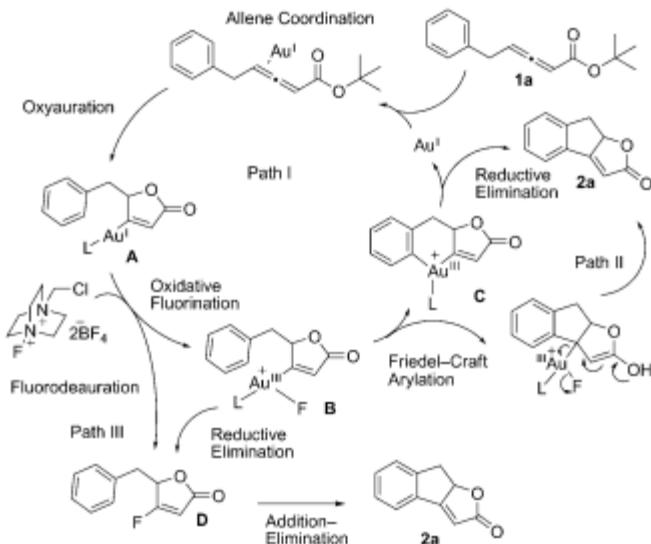
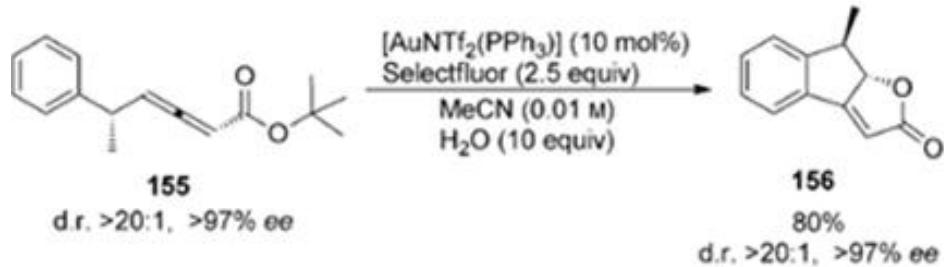


Epoxidations

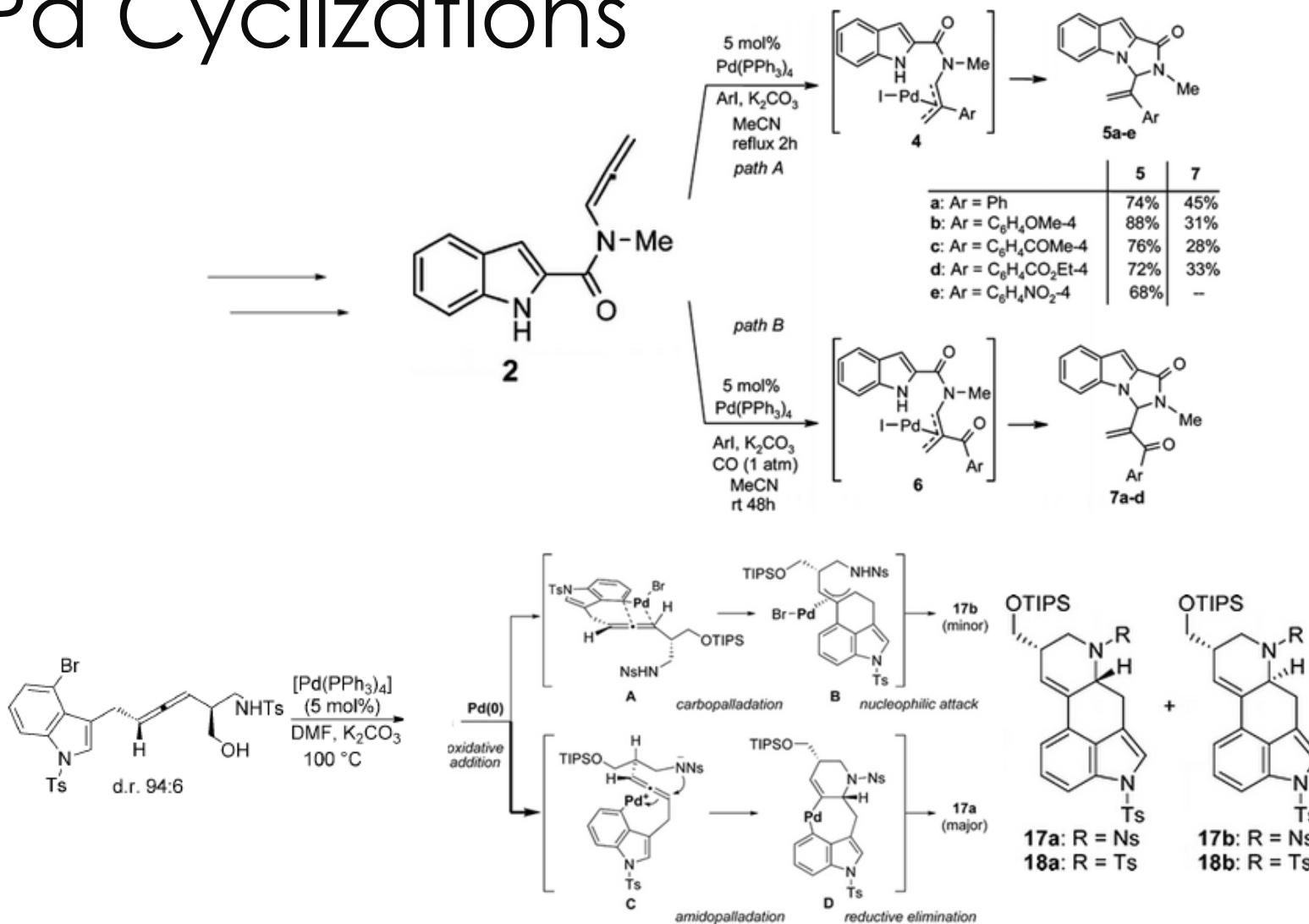




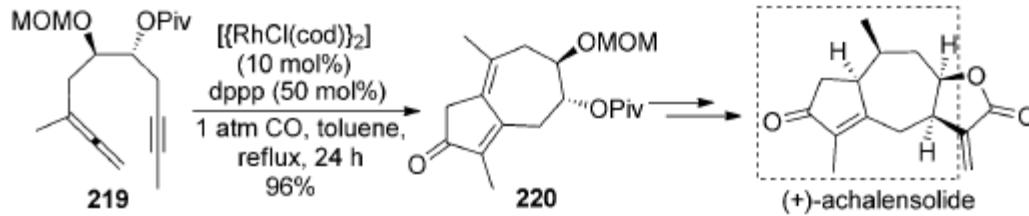
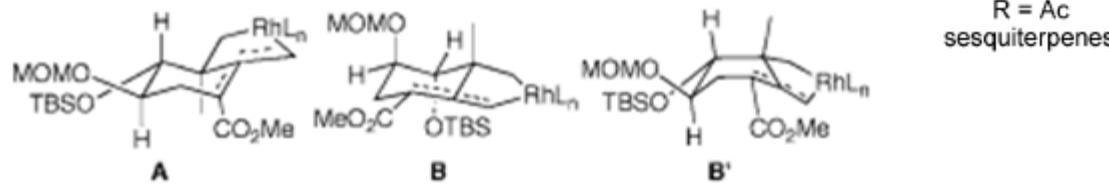
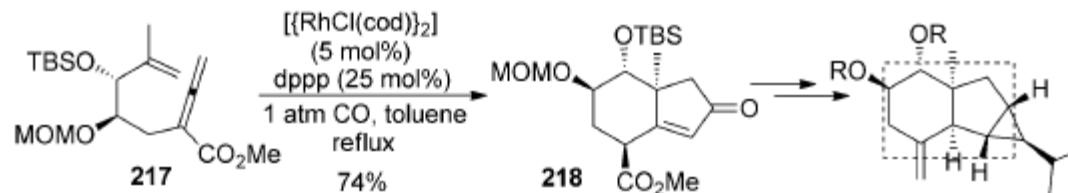
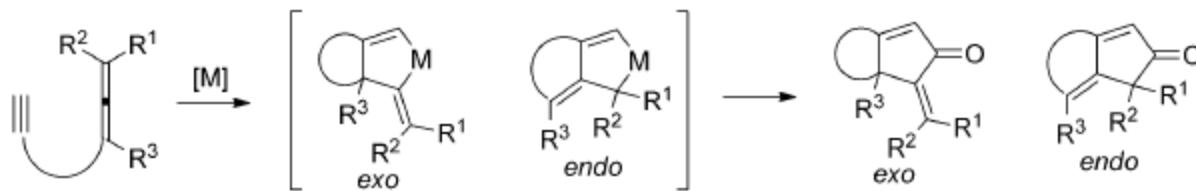
Au Cyclizations



Pd Cyclizations

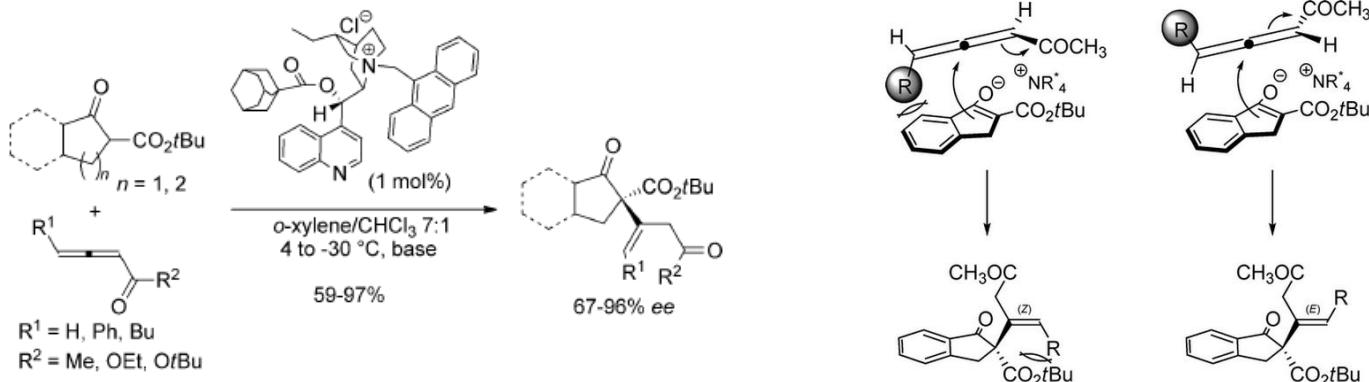


Pauson Khand Cyclizations

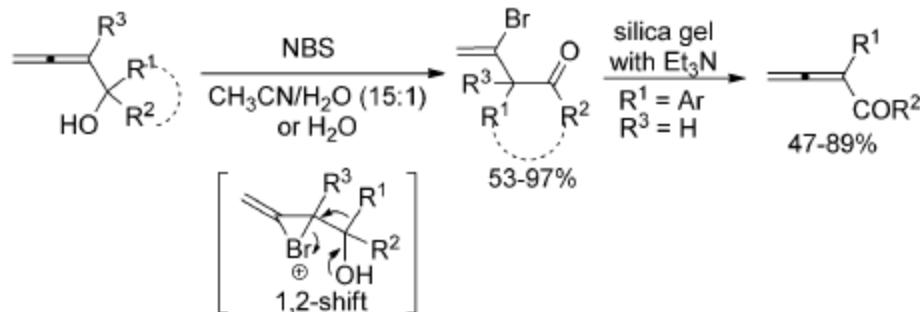


Future Work

- Nucleophilic Addition

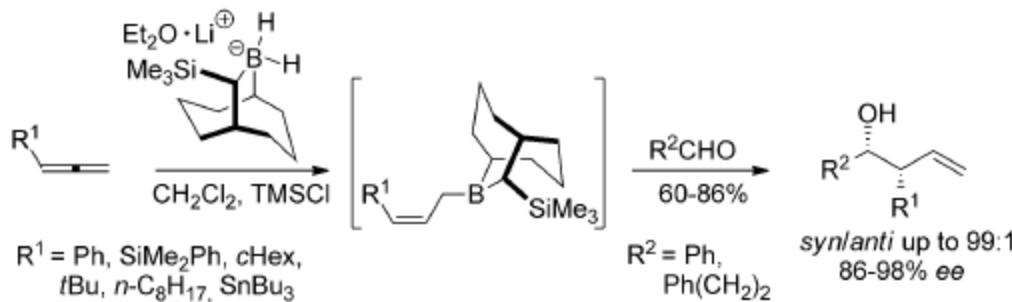


- Electrophilic Addition

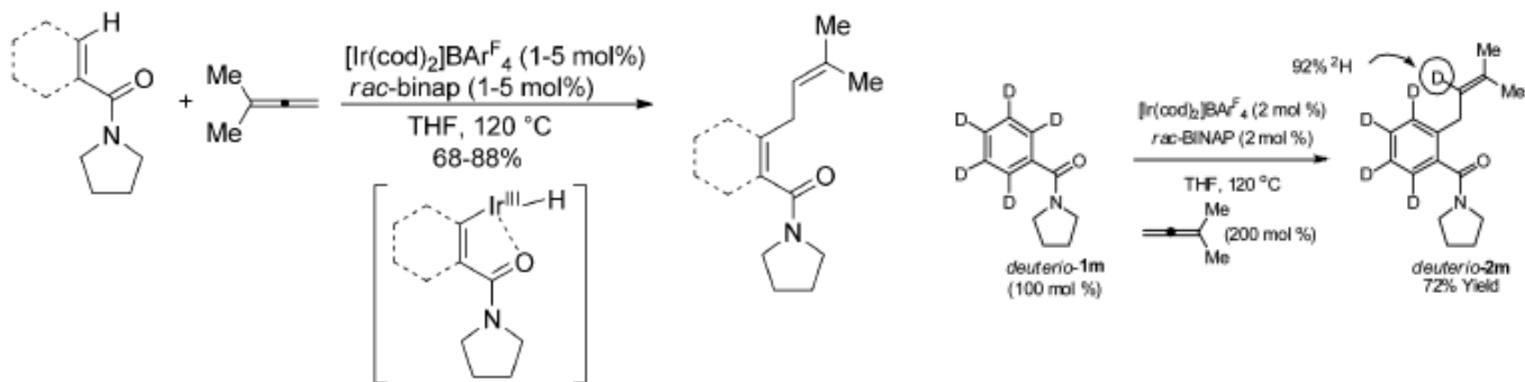


Future Work

○ Hydroborations



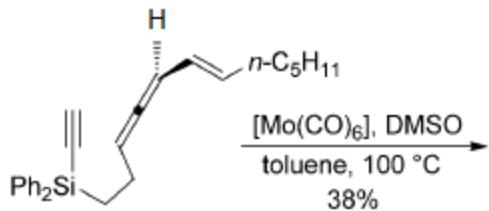
○ C-H Bond Activation



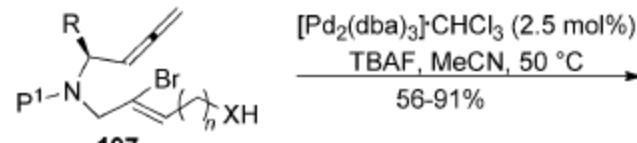
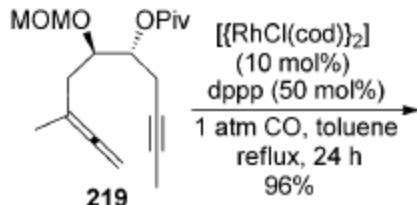
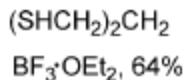
Conclusions

- Quickly becoming an irreplaceable class of compounds; no longer “chemist curiosities”
- Adaptation of “classical reactions” with wider substrate scope and milder reaction conditions.
- Remaining challenges:
 - Diverse reactivity still unpredictable (substituent effect)
 - Limited number of asymmetric reactions
 - Extensive structural and mechanistic studies still needed to elucidate reactivity and stereocontrol

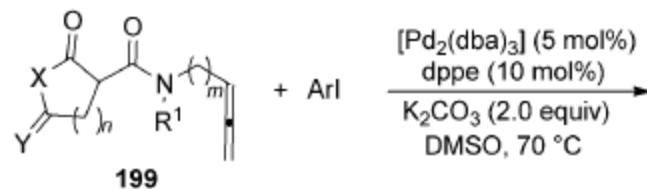
Thank You! Questions!



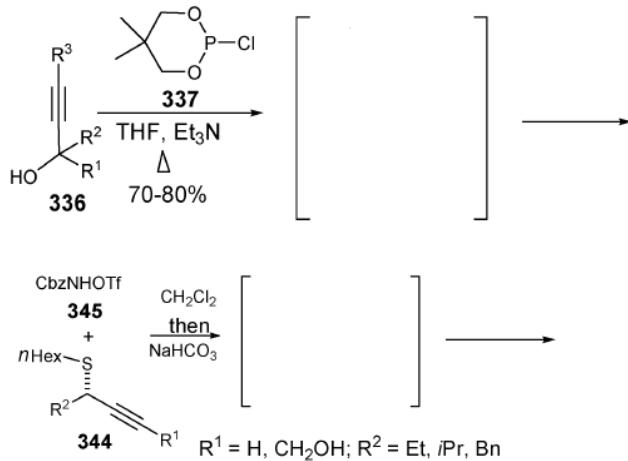
then



$\text{R} = \text{H, Me, } i\text{Pr}; \quad \text{P}^1 = \text{Mts, Ts; } n = 1, 2$
 $\text{X} = \text{NTs, NNs, O, C}(\text{CO}_2\text{Et})_2, o\text{-C}_6\text{H}_5\text{NHTs}$



$\text{R}^1 = \text{Me, Et, Bn, Pr, allyl}$
 $n = 1 \text{ to } 4, m = 1 \text{ to } 3$
 $\text{X} = \text{CH}_2, \text{NBn, NMe, NBoc; } Y = \text{O, H}$



Answers

