

Through-Space Conjugated Polymers and Conjugated Main-Chain Metallopolymers

Wednesday seminar

Gang Li

12/2/2015

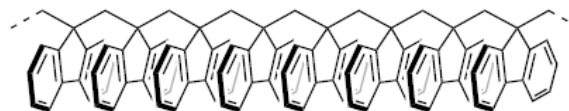


Conjugated Polymer Synthesis. Edited by Yoshiki Chujo
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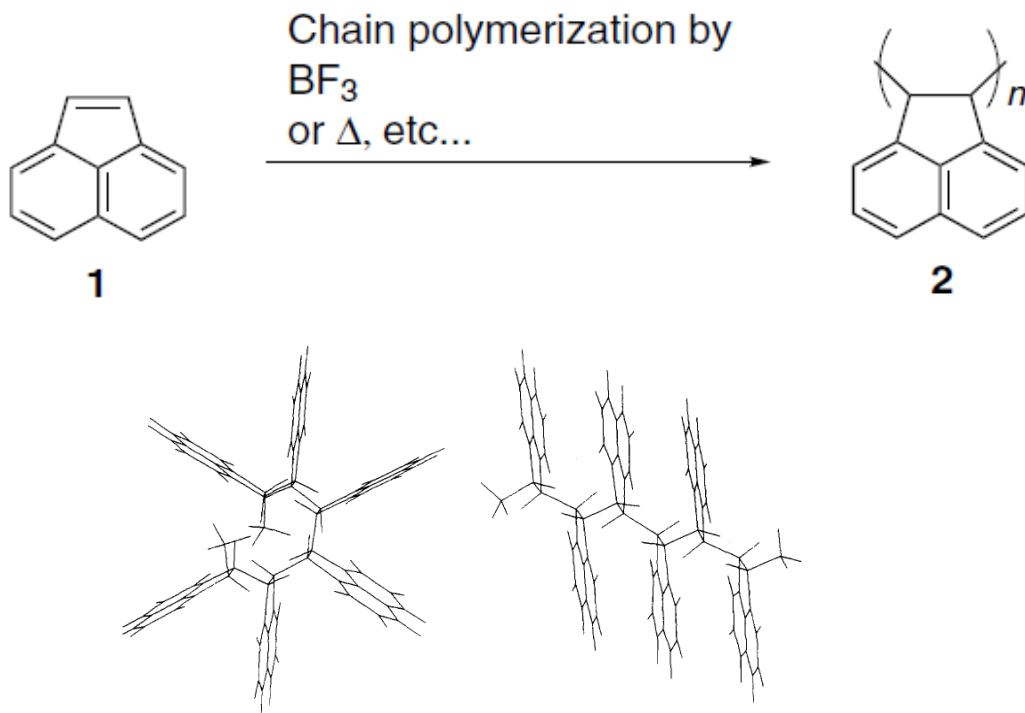
Through-Space Conjugated Polymers

- Yasuhiro Morisaki and Yoshiki Chujo

- Through-space conjugated polymers are π -stacked polymers in which π -electron systems such as aromatic rings or π -electron systems are layered in the single polymer chain.
- It is worth noting that layered π -electron systems are commonly found in nature. Such as layered aromatic rings in DNA .



Synthesis of Polyacenaphthylene

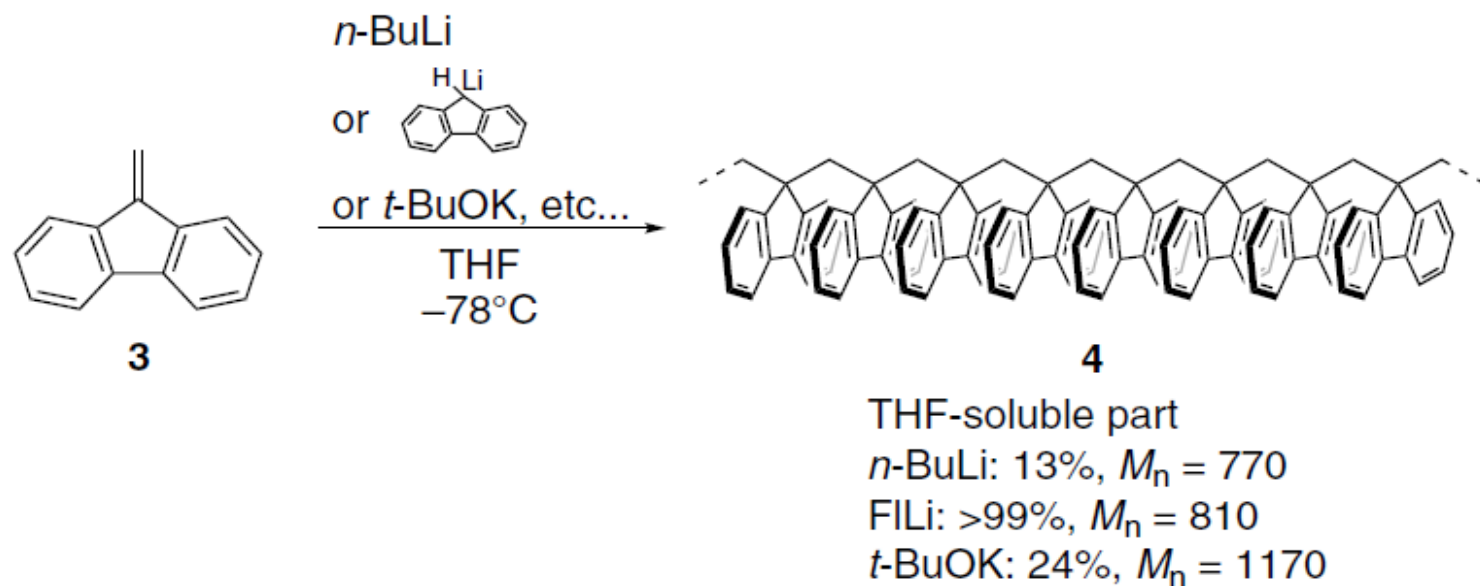


•The polyacenaphthylene in diluted solution showed photoluminescence from monomer units and intramolecular excimers, depending on the stereoregularity

Wang, Y.-C. Morawetz, H. *Makromol. Chem. Suppl.*, **1975**, 1, 283–295

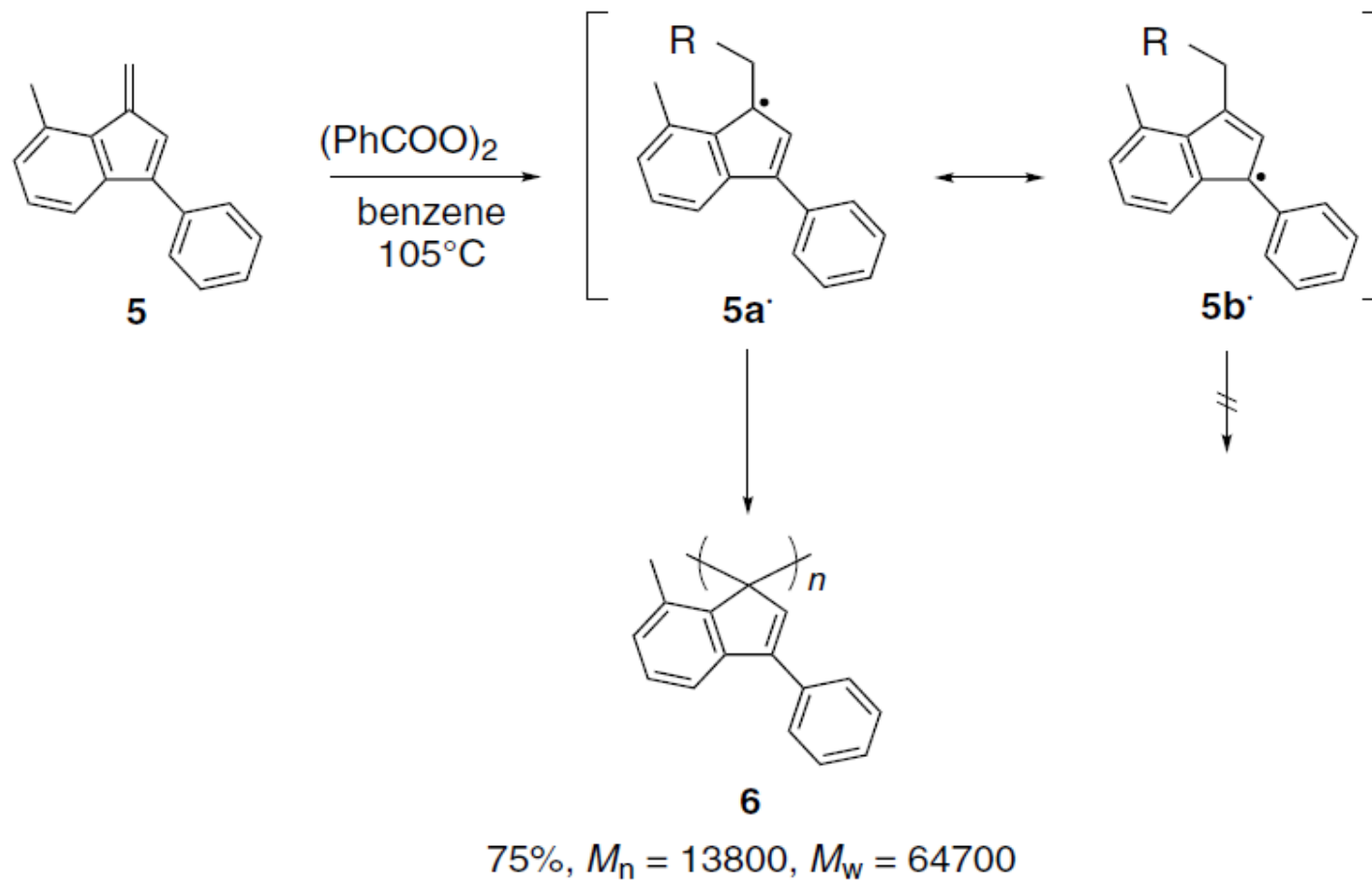
Mendicuti, F., Kulkarni, R., Patel, B., Mattice, W.L. *Macromolecules*, **1990**, 23, 2560–2566

Through-Space Conjugated Polymers with the Layered π -Electron Systems in the Side Chain

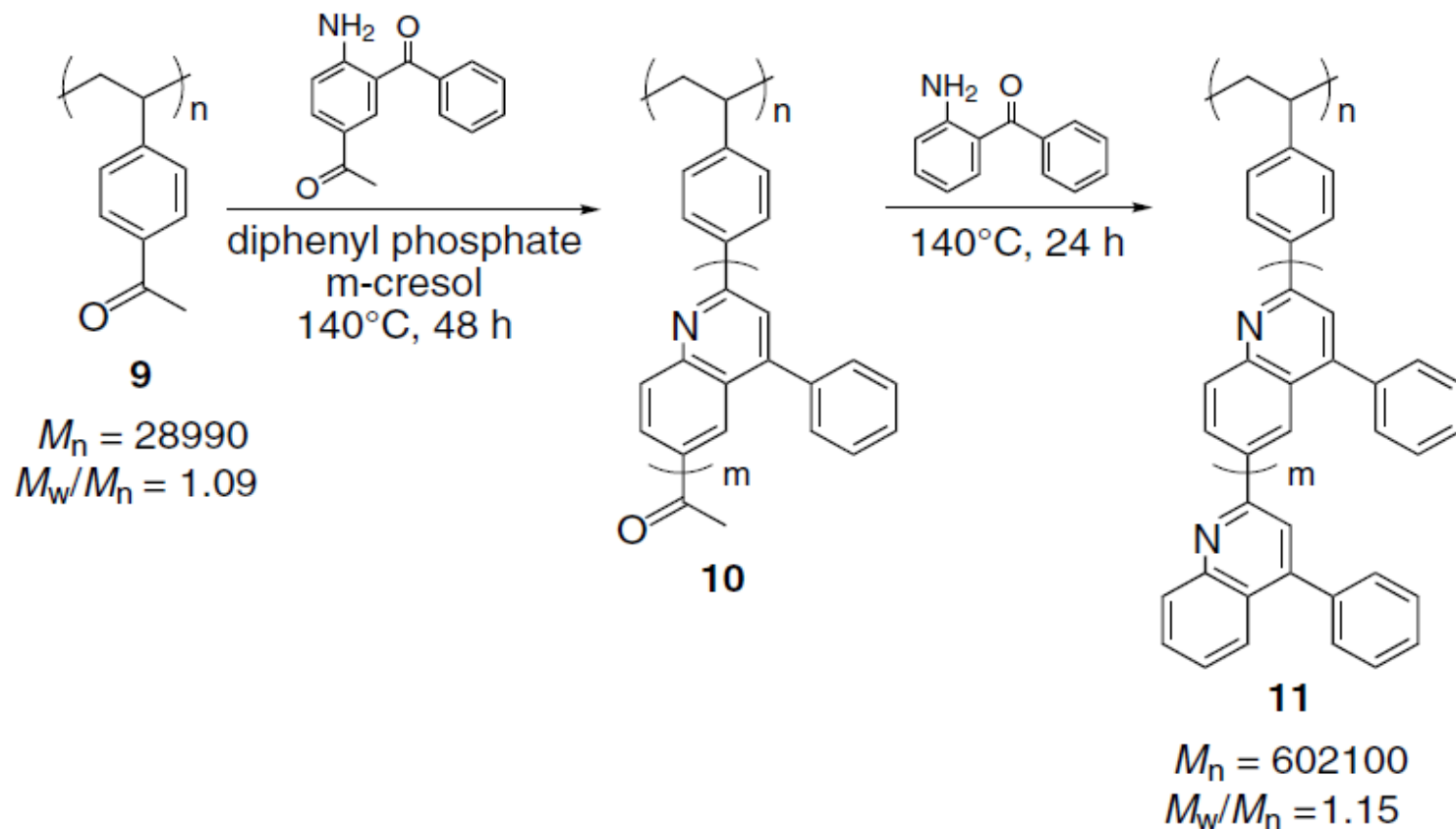


Nakano, T., Takewaki, K., Yade, T., and Okamoto, Y. *J. Am. Chem. Soc.*, **2001**, *123*, 9182–9183; Nakano, T. and Yade, T. *J. Am. Chem. Soc.*, **2001**, *125*, 15474–15484;

Synthesis of Polybenzofulvene



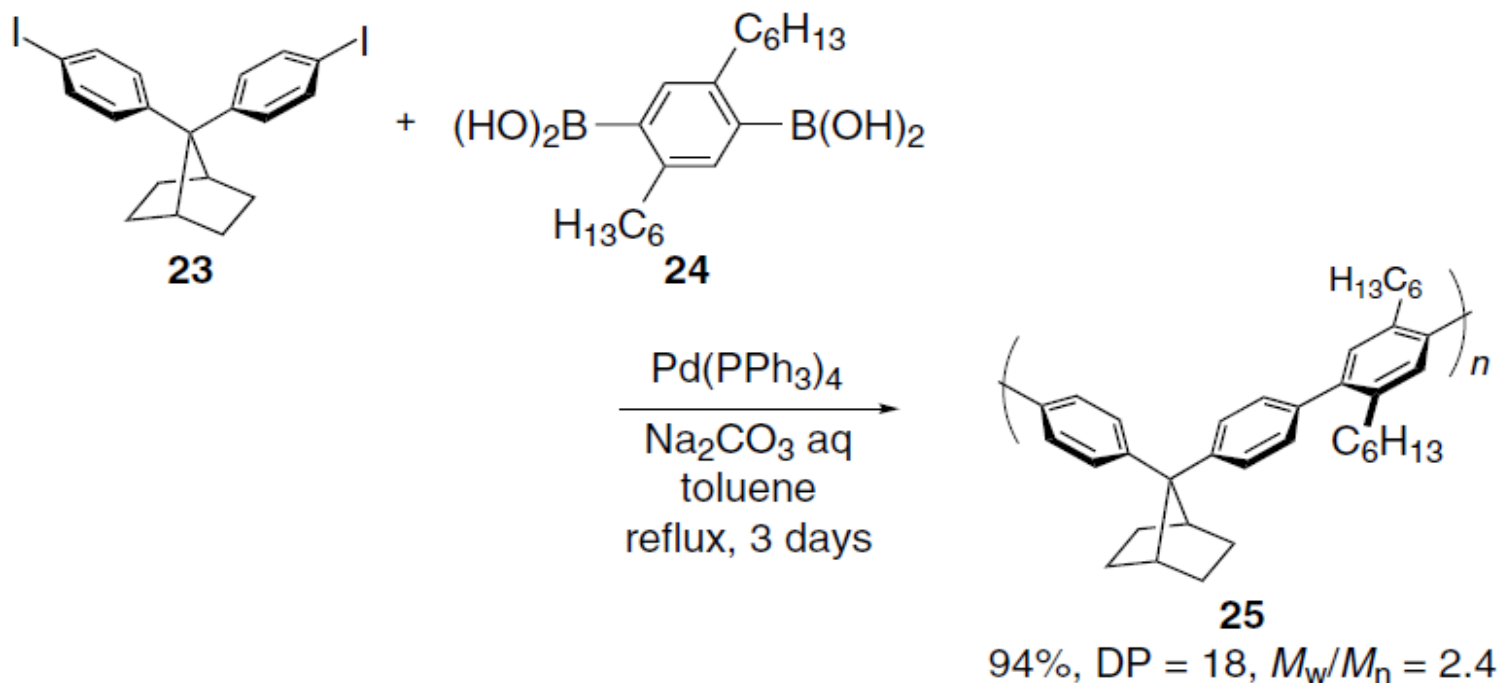
Polystyrene-graft-Poly(4-Phenylquinoline)



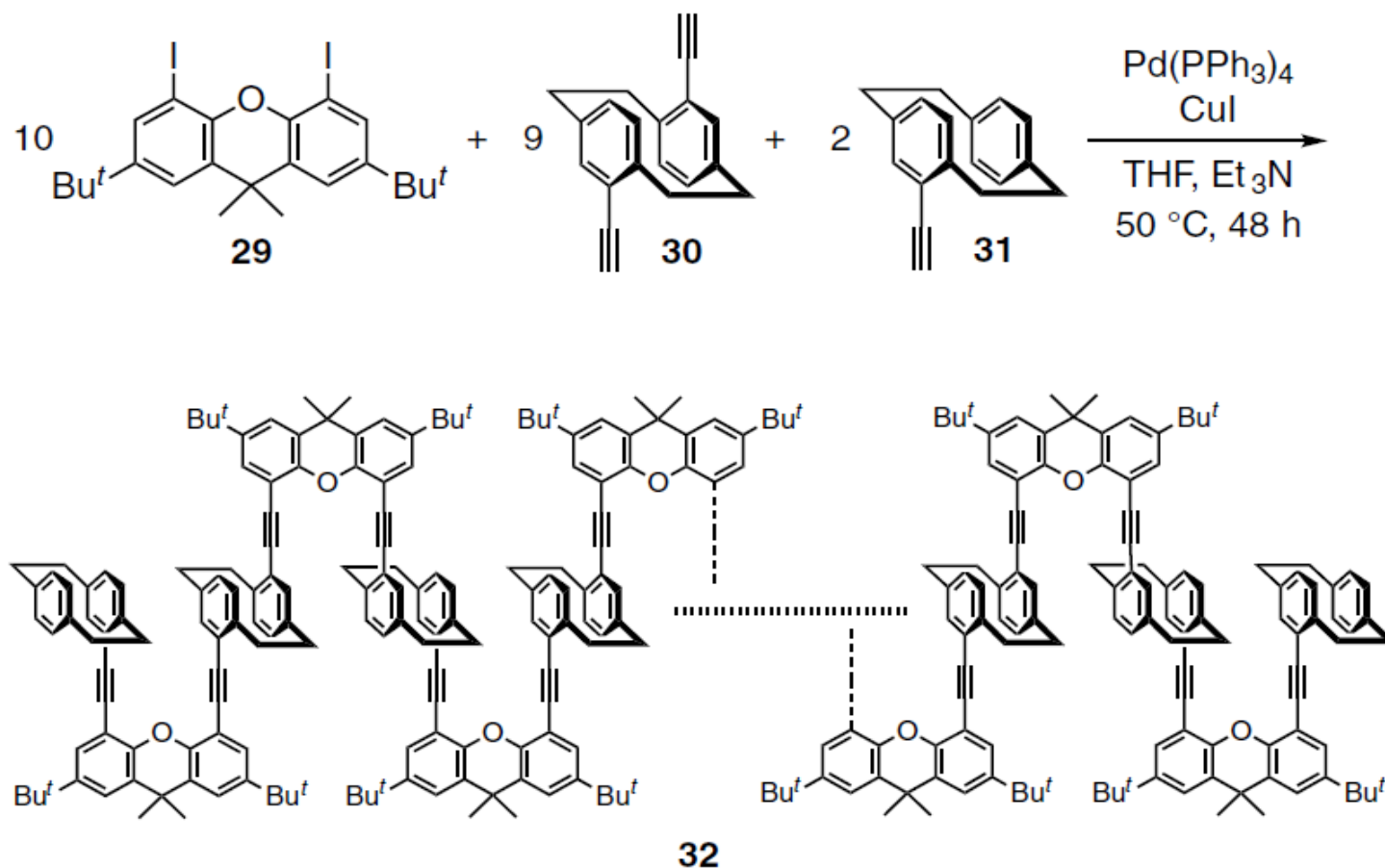
•An electroluminescence (EL) device using polymer 11 as an emitting layer showed increasing efficiency and brightness

Through-Space Conjugated Polymers with the Layered π -Electron Systems in the main Chain

•Phenylene-Layered Polymer Based on a Norbornane Scaffold



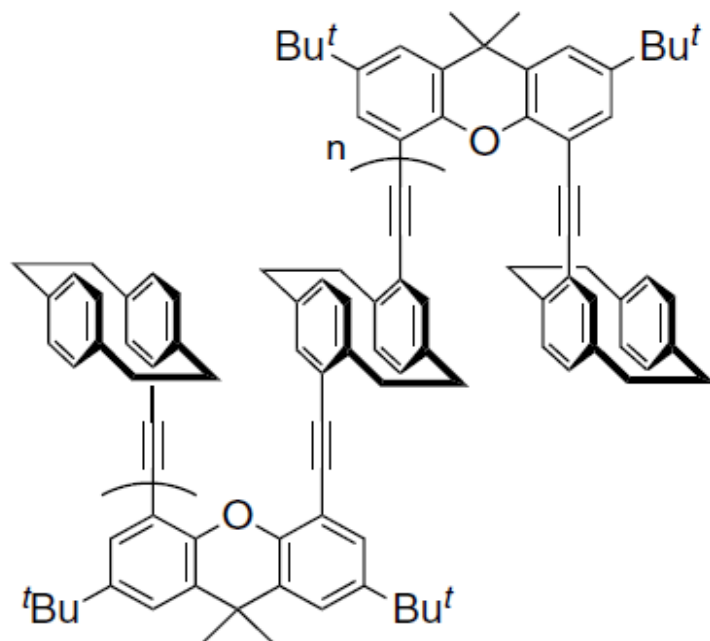
Aromatic Ring-Layered Polymers Based on a Xanthene Scaffold



Mn: 4100. Sixteen benzenes were layered in the polymer main chain.

Morisaki, Y., Murakami, T., Chujo, Y. *Macromolecules*, **2008**, *41*, 5960–5963

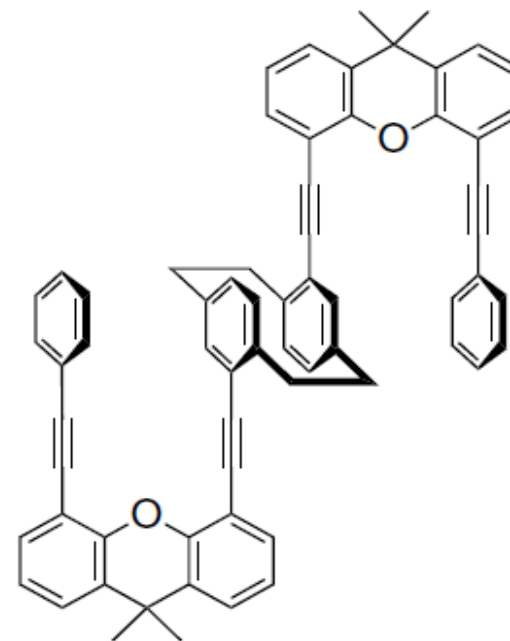
Aromatic Ring-Layered Polymers Based on a Xanthene Scaffold



32a ($M_n = 4100$): $\lambda_{\text{abs,max}} = 330 \text{ nm}$

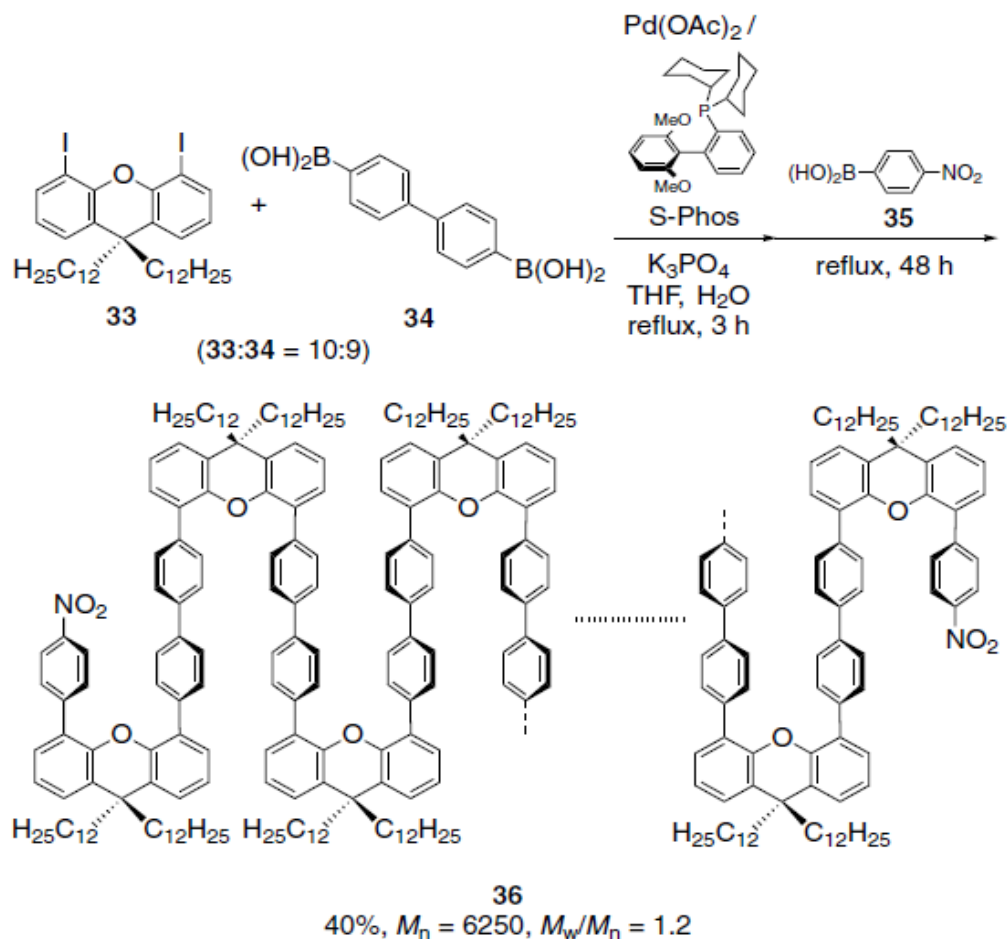
32b ($M_n = 3100$): $\lambda_{\text{abs,max}} = 330 \text{ nm}$

32c ($M_n = 2100$): $\lambda_{\text{abs,max}} = 330 \text{ nm}$



33 $\lambda_{\text{abs,max}} = 320 \text{ nm}$

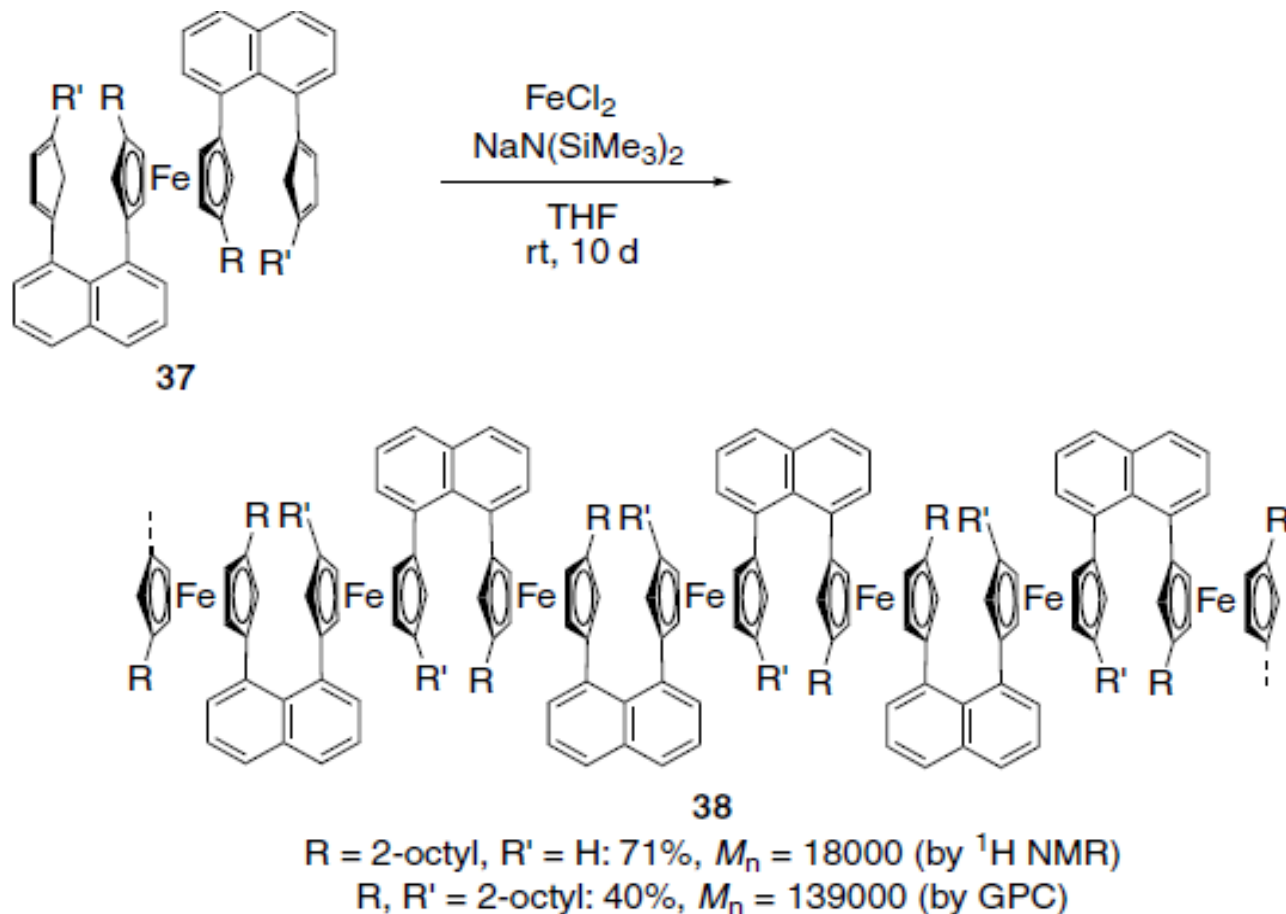
Oligophenylene-Layered Polymers



- The two end capping nitrobenzene units in polymer 36 effectively quenched the photoluminescence of the layered biphenylenes

Ferrocene-Layered Polymers Based on a Naphthalene Scaffold

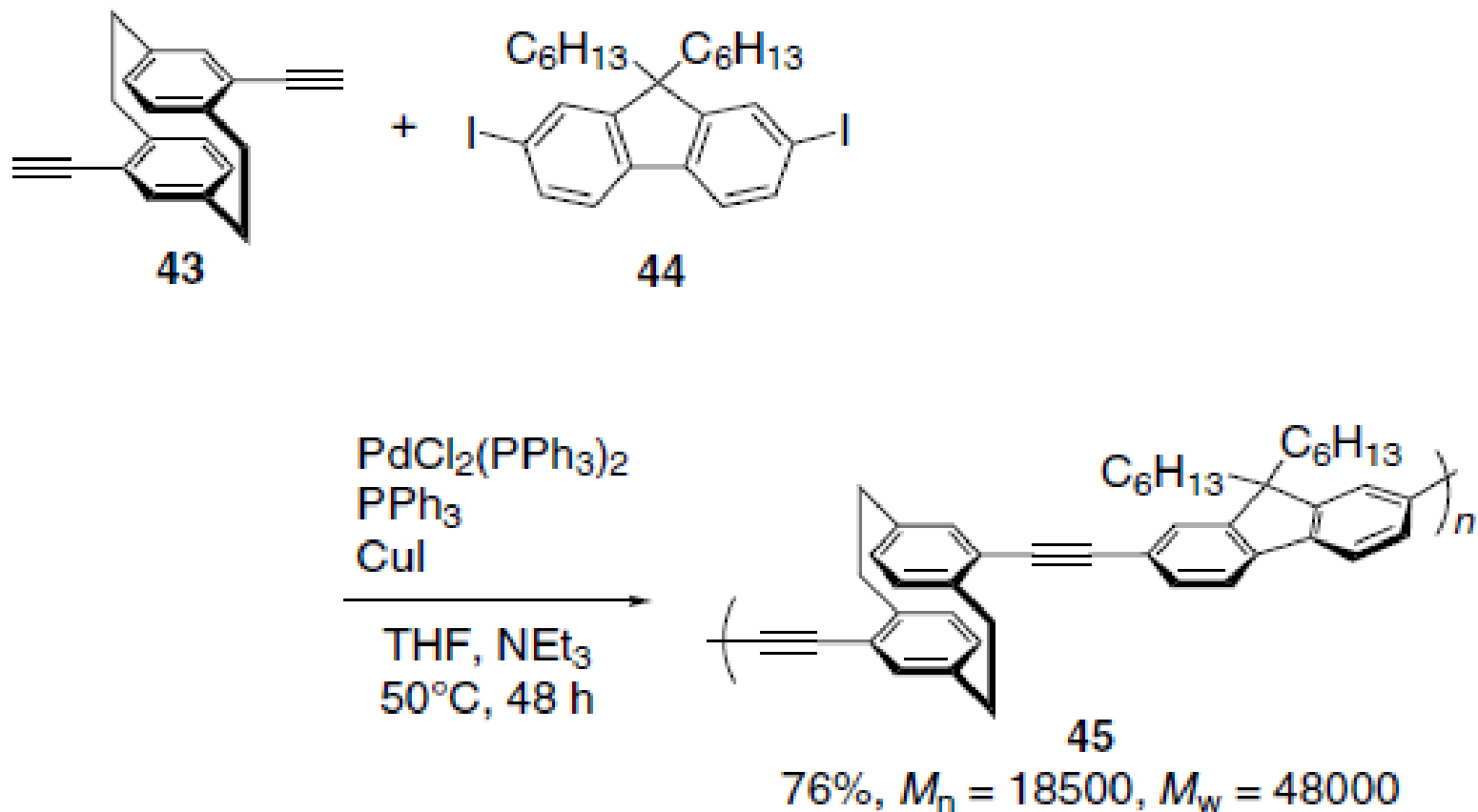
Synthesis of Face- to-Face Metallocene Polymers



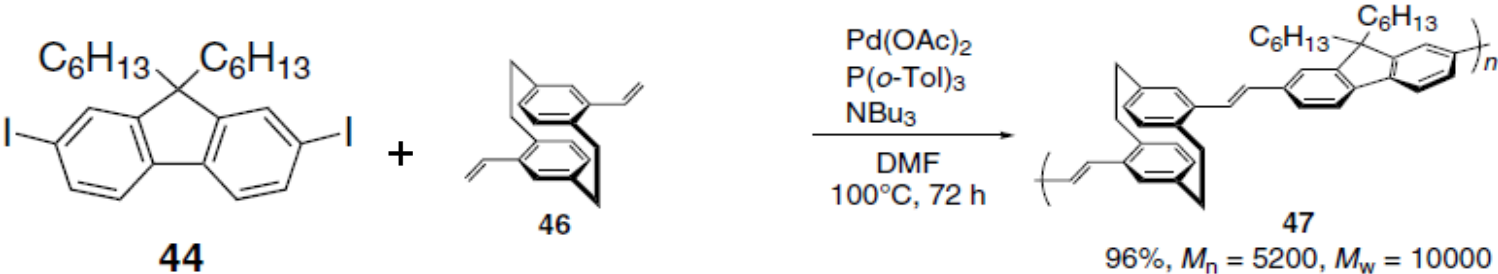
Nugent, H.M., Rosenblum, M., and Klemarczky, P. *J. Am. Chem. Soc.*, **1993**, *115*, 3848–3849;
Rosenblum, M., et al. *Macromolecules*, **1995**, *28*, 6330–6342

[2.2]Paracyclophane-Containing Through-Space Conjugated Polymers

•Synthesis of [2.2]paracyclophane-containing poly(p-arylene-ethynylene) **45** by the Sonogashira–Hagihara reaction.

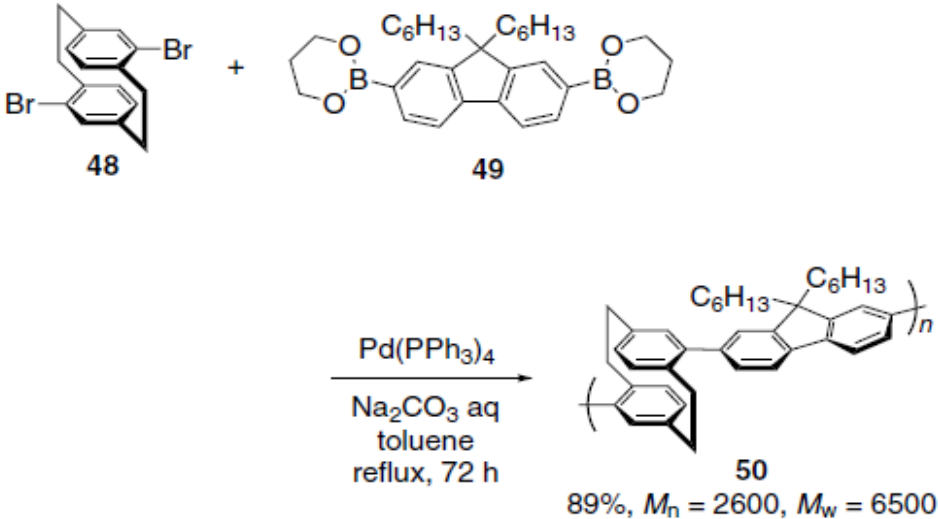


•by the Heck reaction.



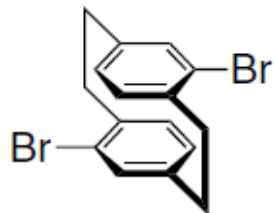
Mitchell, R.H. *J. Am. Chem. Soc.*, **2002**, 124, 2352–2357.

•by the Suzuki-Miyaura reaction

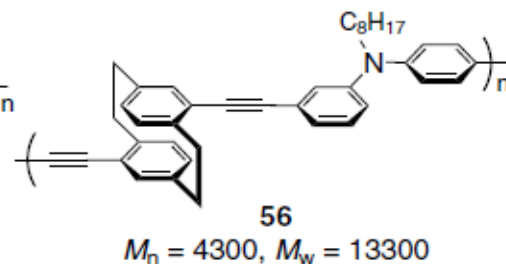
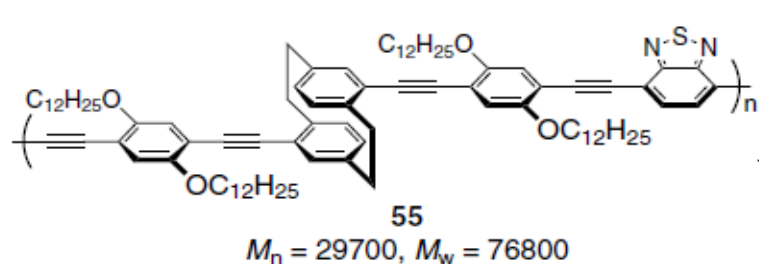
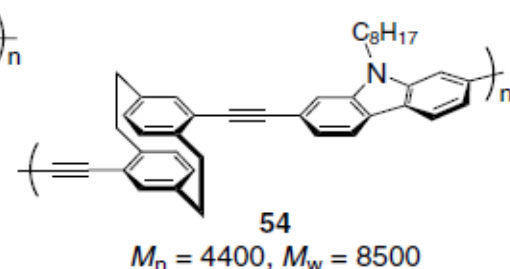
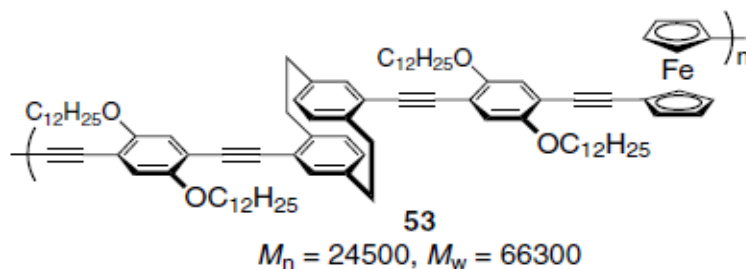
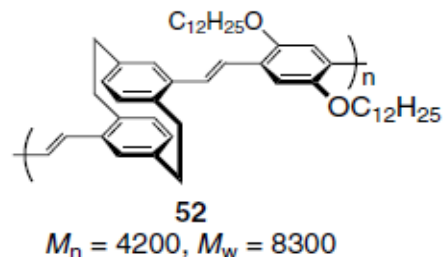
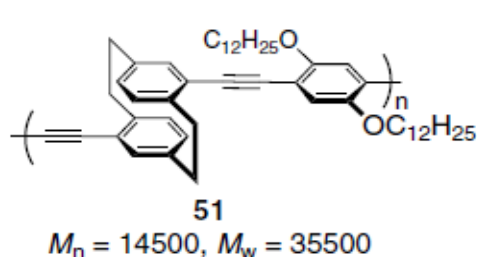


Morisaki, Y. and Chujo, Y. *Bull. Chem. Soc. Jpn.*, **2005**, 78, 288–293

Several examples

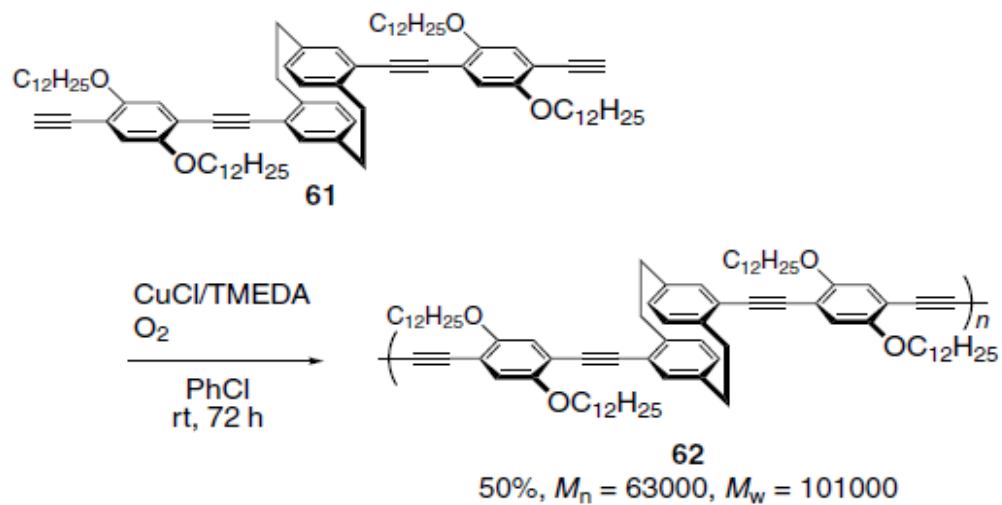


The key monomer

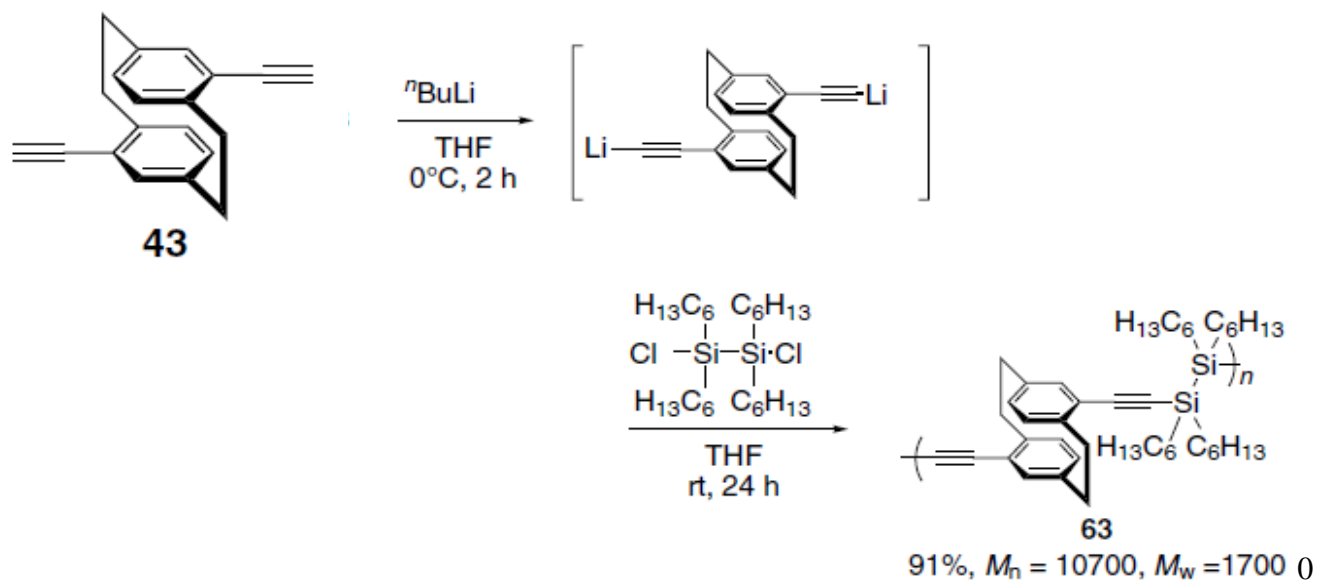


a) Morisaki, Y. and Chujo, Y. *Macromolecules*, 2002 35, 587–589; (b) Morisaki, Y., Ishida, T., and Chujo, Y. *Macromolecules*, **2002**, 35, 7872–7877; (c) Morisaki, Y. and Chujo, Y. *Macromolecules*, **2003**, 36, 9319–9324

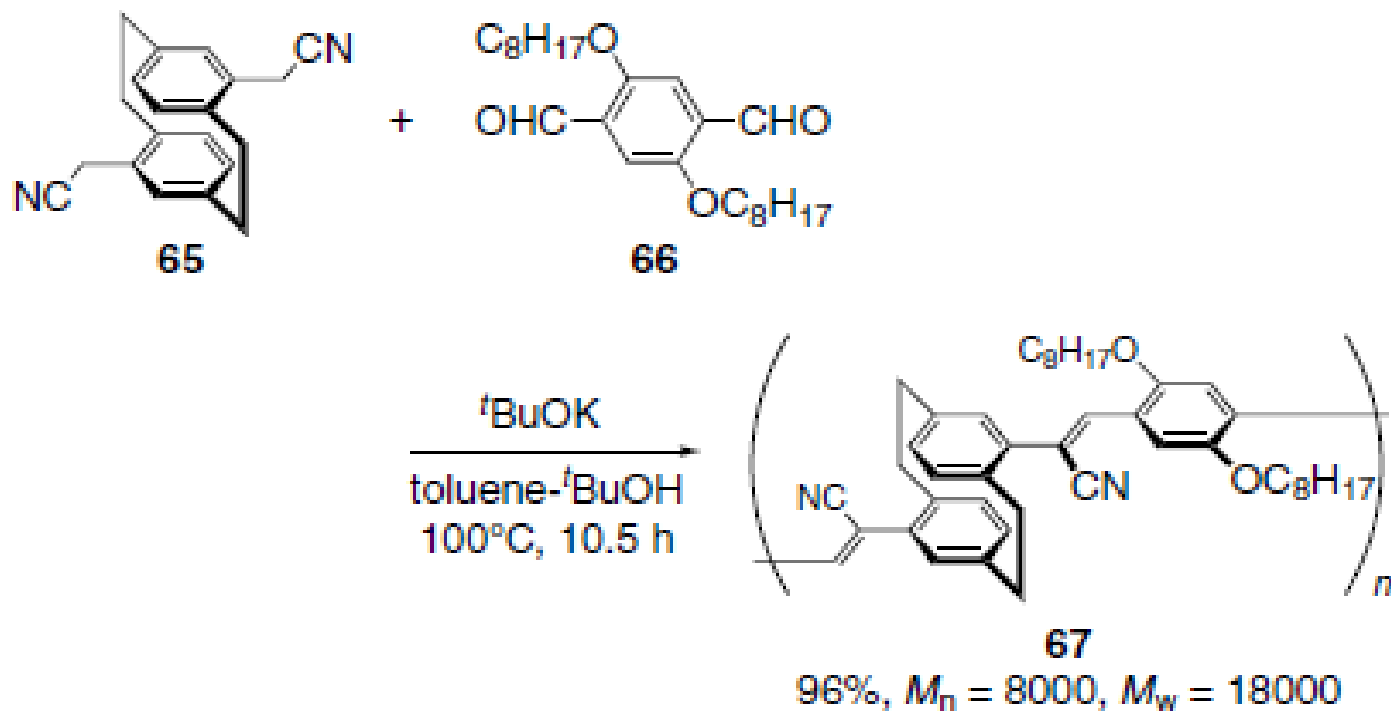
The oxidative coupling reaction



Conjugated polymer comprising disilylene and [2.2]paracyclophane units

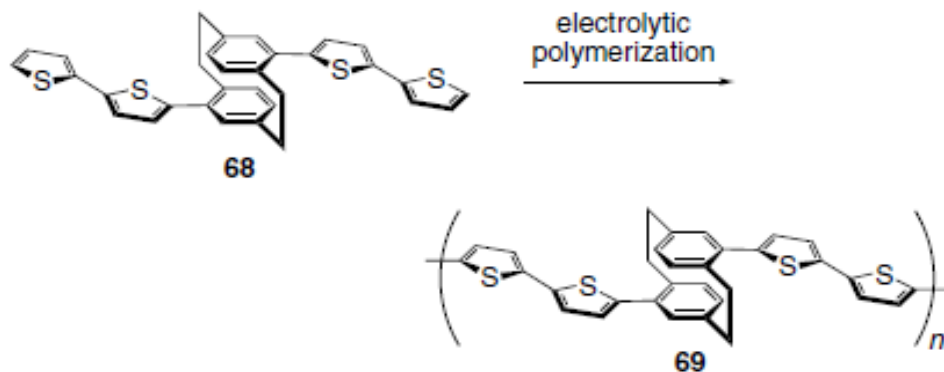


Knoevenagel reaction

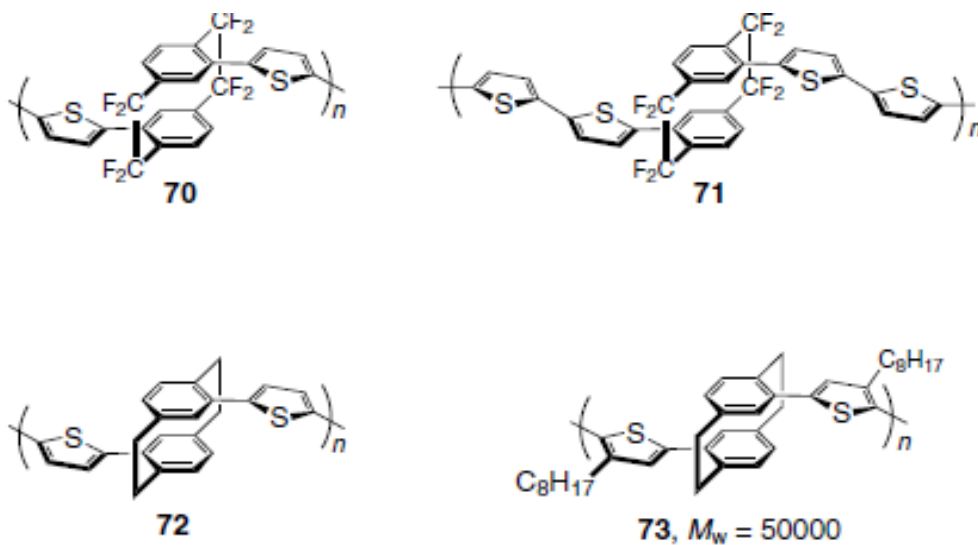


(a) Morisaki, Y., Lin, L., and Chujo, Y. *Chem. Lett.*, (2009) 38, 734–735; (b) Morisaki, Y., Lin, L., and Chujo, Y. *J. Polym. Sci. Part A: Polym. Chem.*, (2009) 47, 5979–5988.

•Electrochemical polymerization



Guyard, L. and Audebert, P. *Electrochem. Commun.*, **2001**, 3, 164–167



Salhi, F. and Collard, D.M. *Adv. Mater.*, **2003**, 15, 81–85.

Applications:

- Through-space conjugated polymers can potentially be used as optoelectronic devices such as photovoltaic devices, and electroluminescence devices, as well as in a single molecule device . . .

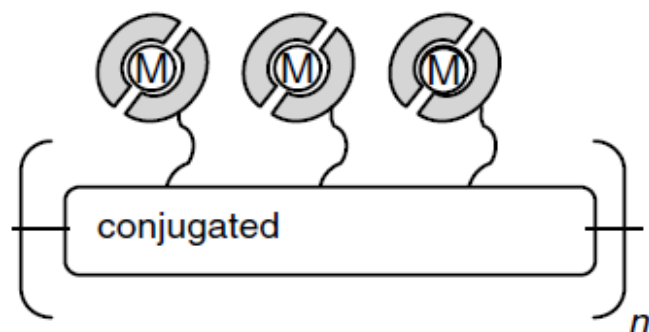
- **Synthetic Strategies to Conjugated Main-Chain Metallopolymers**

Andreas Wild, Andreas Winter, Martin D. Hager, and Ulrich S. Schubert

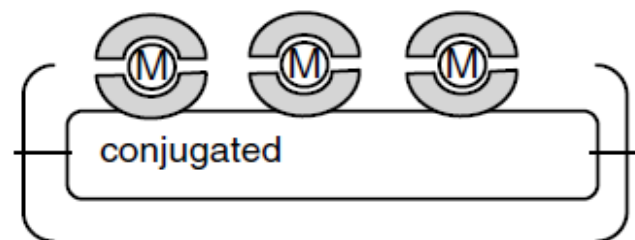
- **The great interest in these materials is mainly attributed to their special properties, which represent a combination of the physical, electronic and optical properties of the organic polymer as well as the physical, electronic and optical properties of the incorporated metal complexes**
- **Having many advantages compared to inorganic semiconductors, such as low cost, and the suitability for solution processing and large area application.**

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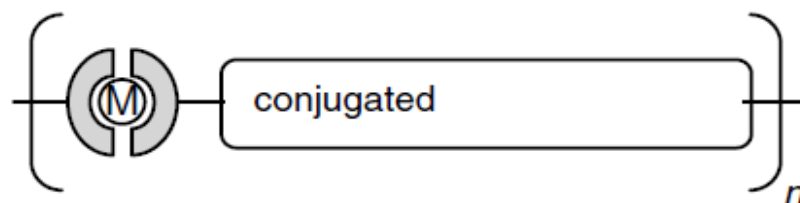
Three Different Types of Conjugated Metallopolymers



Type I

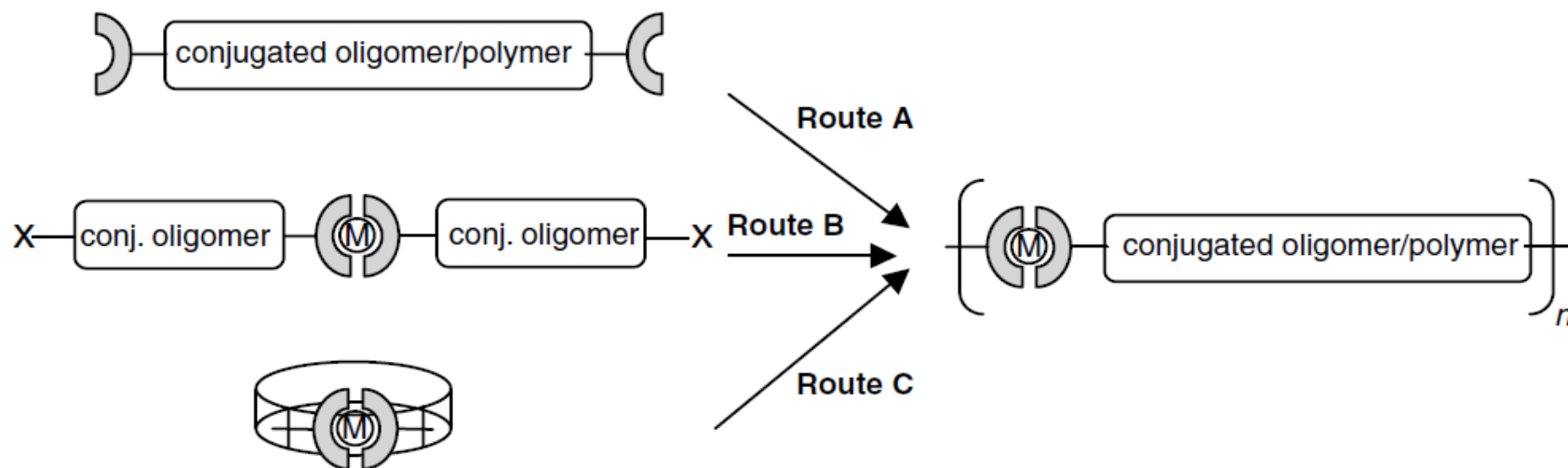


Type II



Type III

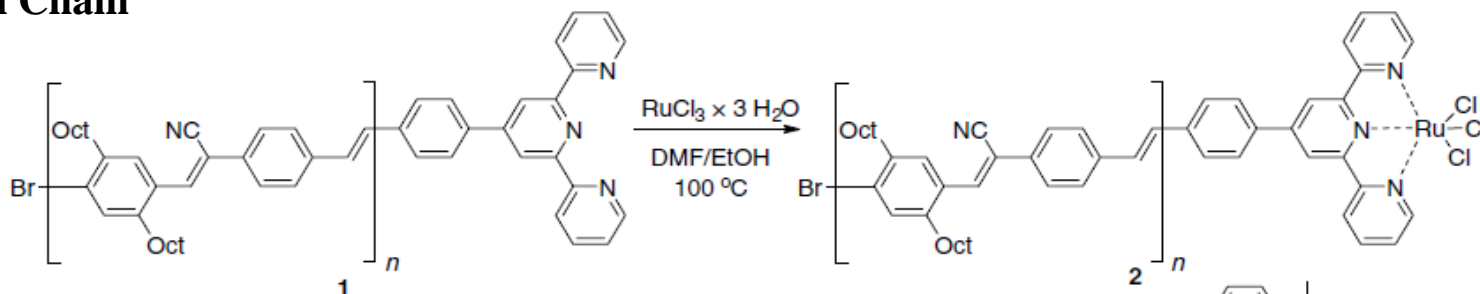
Synthetic Routes



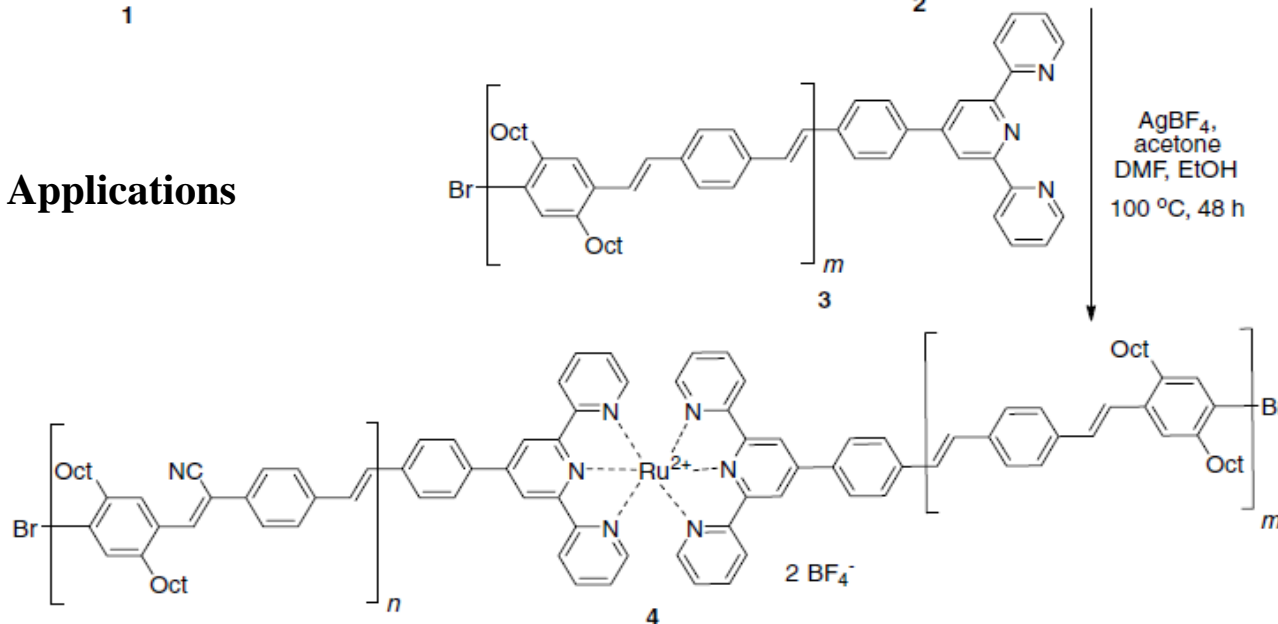
•Due to the broad field of different metallopolymer, my topic will mainly focus on conjugated metallopolymer with terpyridines, porphyrins and metallaynes

Conjugated Metallopolymers with Terpyridines

• π -Conjugated Polymers with Terpyridine Units and Other Tridentate Ligands as Part of the Main Chain

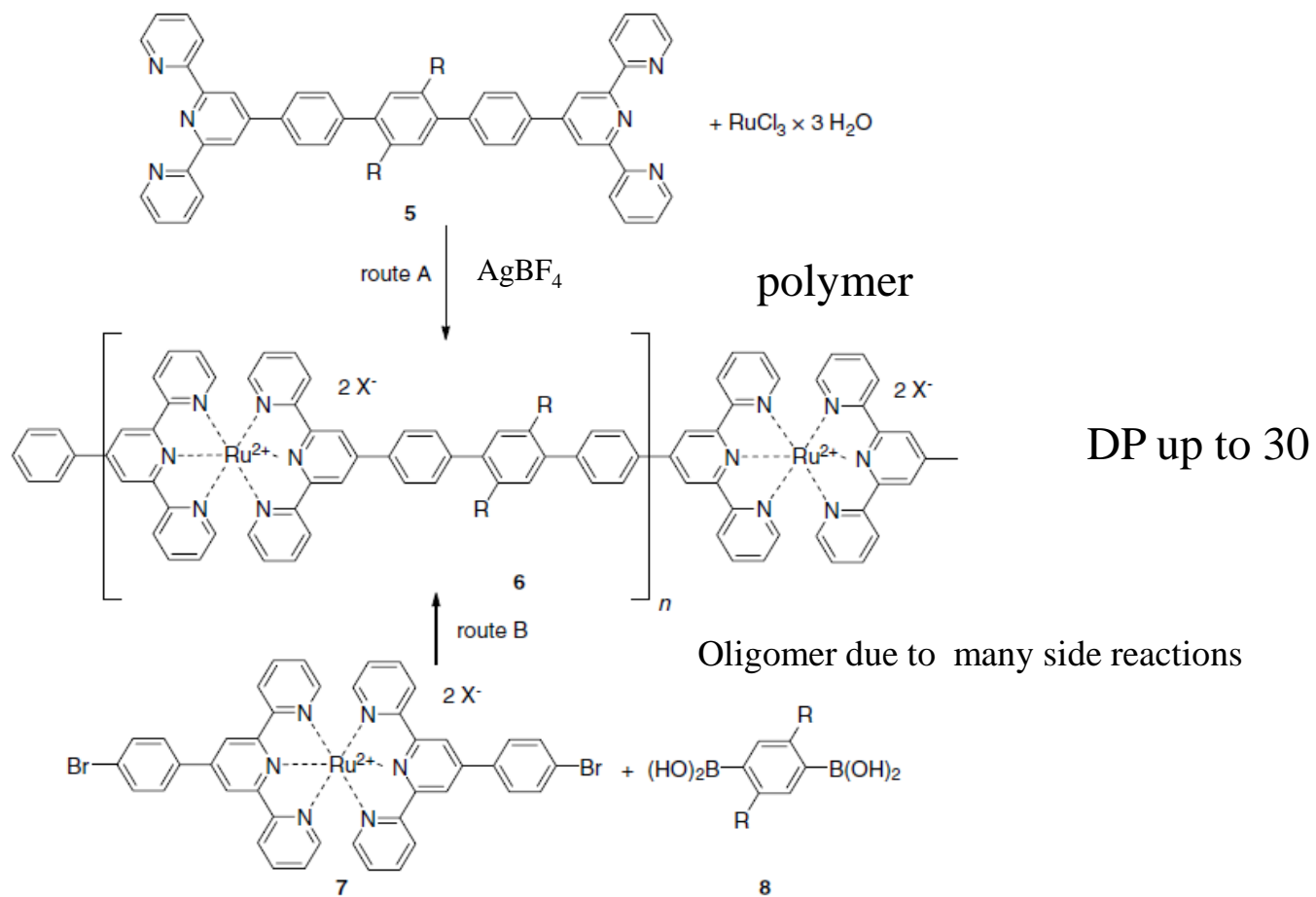


For Photovoltaic Applications



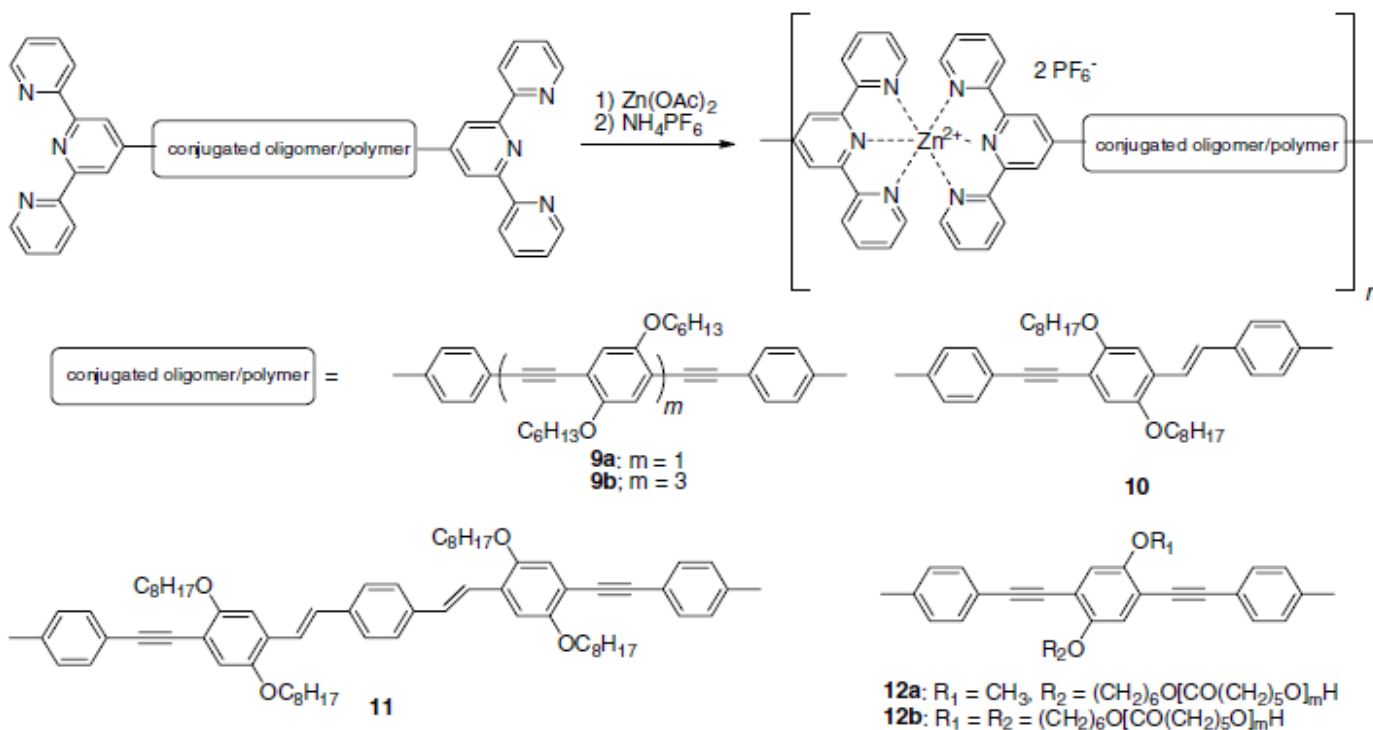
In this case, terpyridine plays only structural role.

Conjugated Metallopolymers with Terpyridines



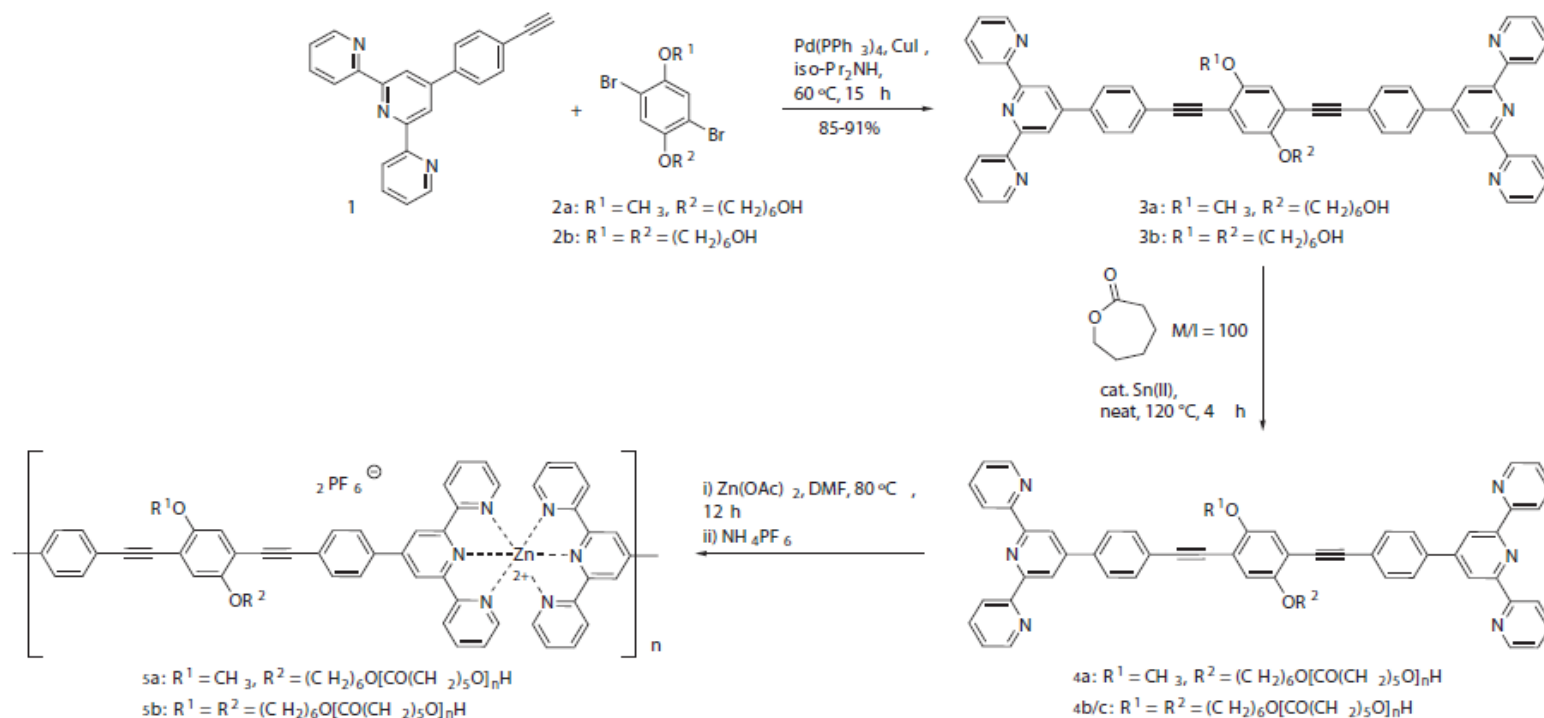
Conjugated Metallopolymers with Terpyridines

Introducing solubilizing groups increase solubility of metallopolymer-----



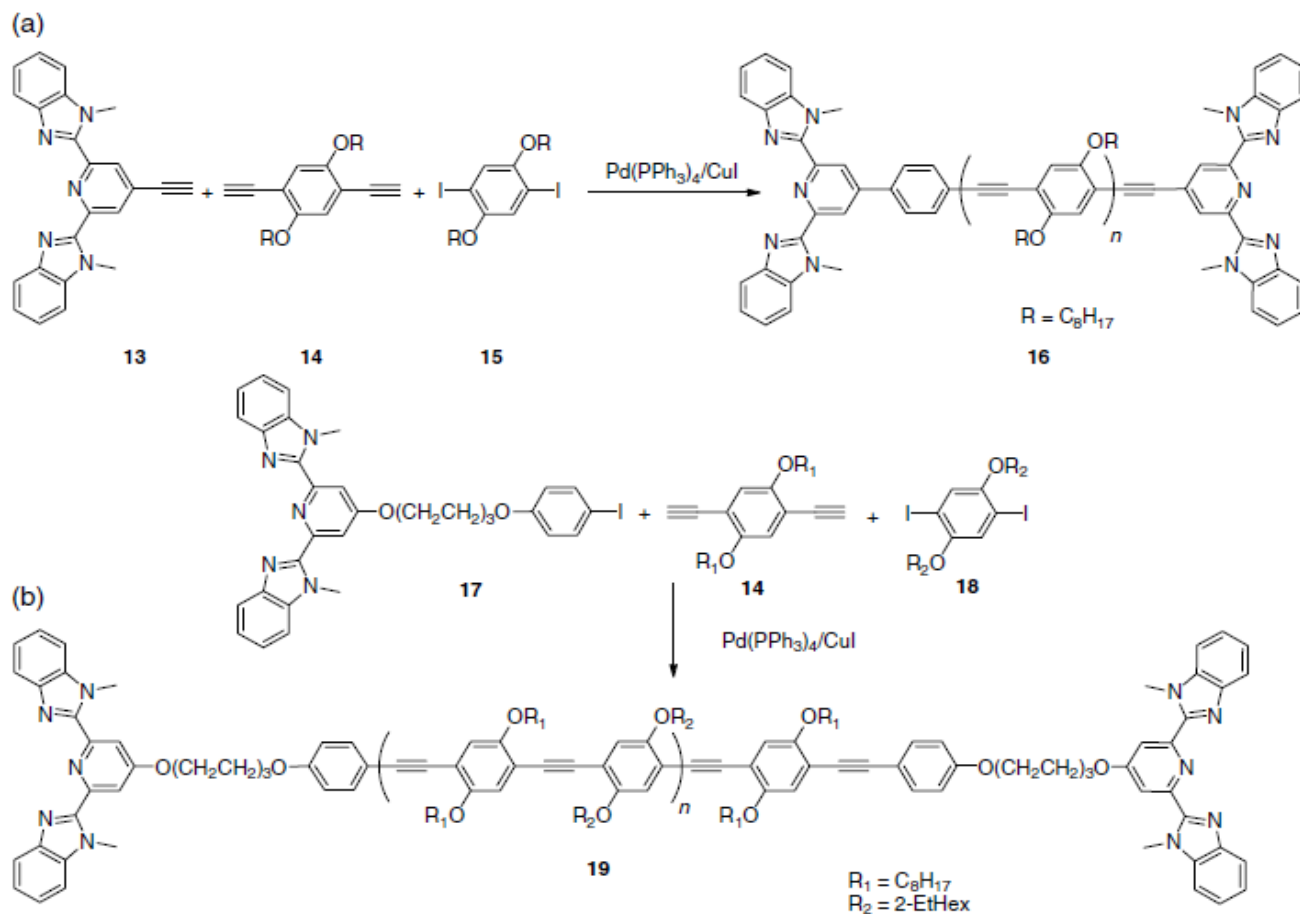
Conjugated Metallopolymers with Terpyridines

The Sonogashira reaction



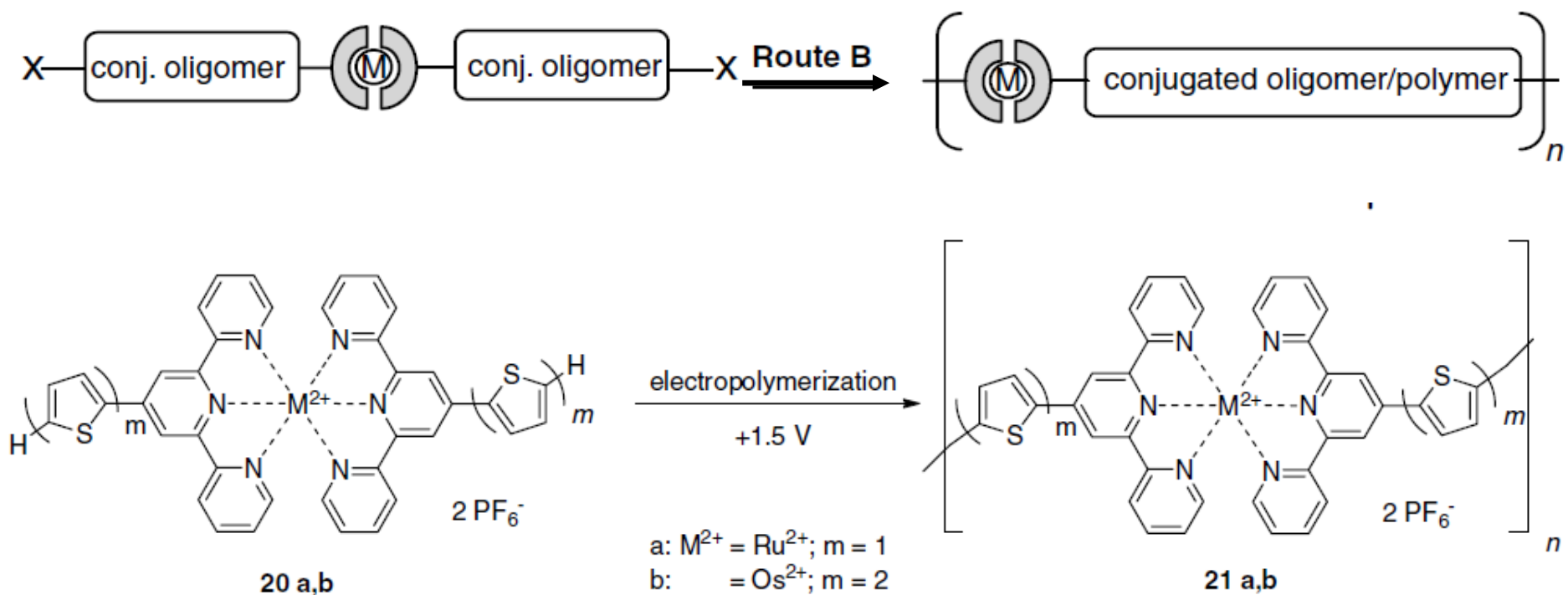
Terpyridines-like Conjugated Metallopolymers

- The synthesis of the conjugated macroligand and the decoupled macroligand



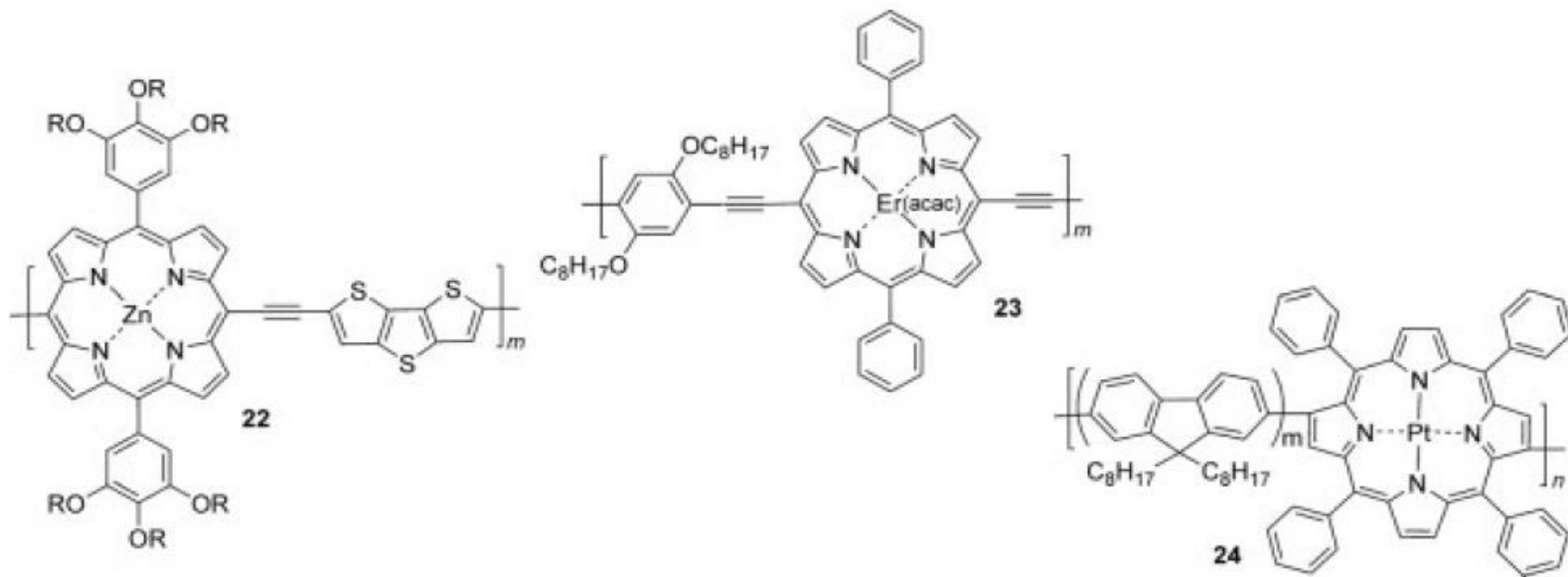
Weder et al. *Macromolecules* **2006**, 39, 651-657
Weder et al. *Macromolecules* **2008**, 41, 2157-2163

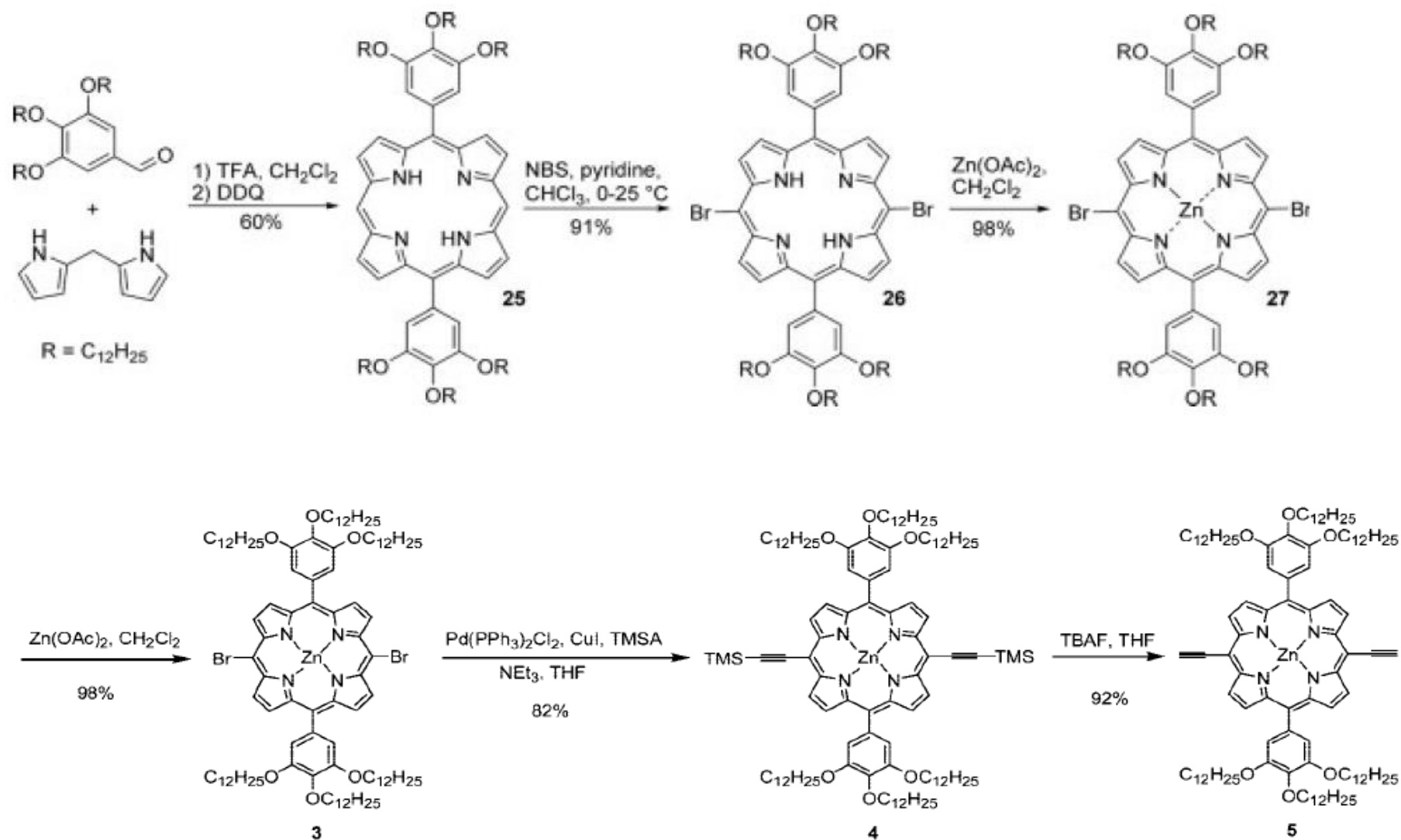
Conjugated Metallopolymers with Terpyridines



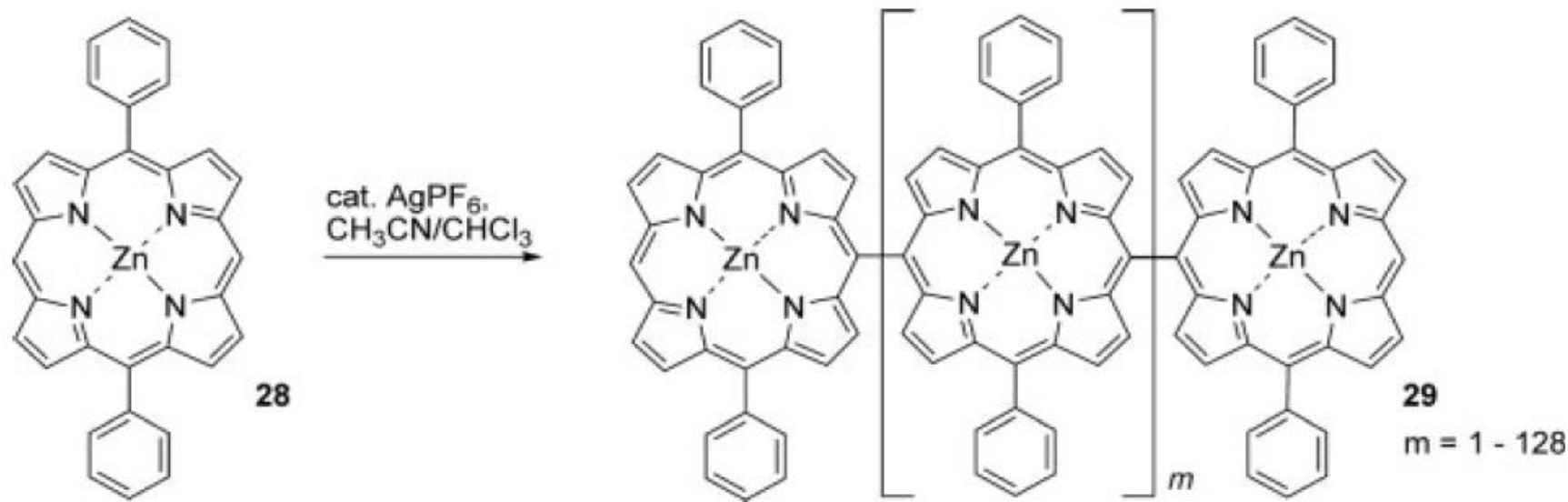
π -Conjugated Polymers with Porphyrin Units as Part of the Main Chain

- Several examples of conjugated polymers with porphyrin:



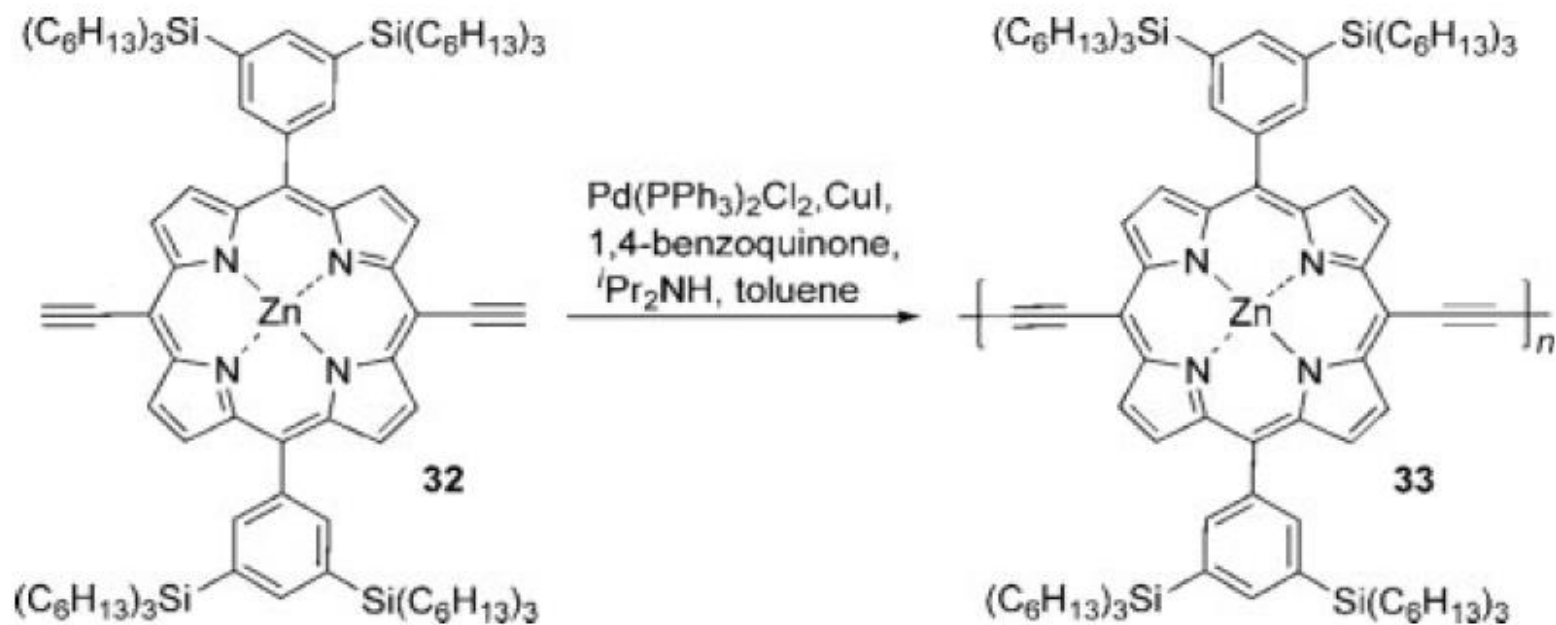


•Homopolymerization of porphyrin derivatives



DP up to 128, Mw up to 133 kg/mol by MALDI TOF MS

•Straightforward (homo)polymerization a bisethynyl-functionalized porphyrin

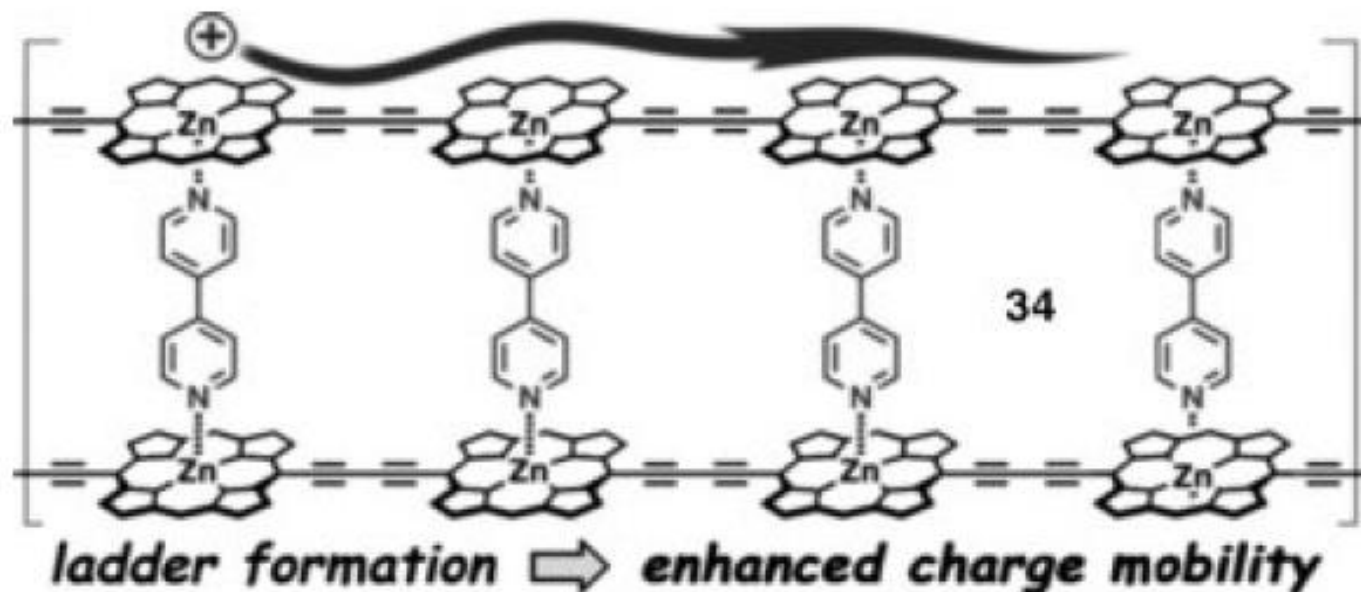


Mn: 83 700 da

DP: 50

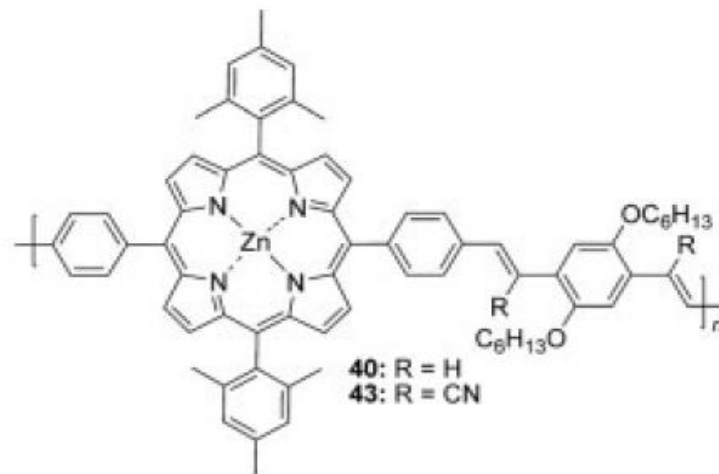
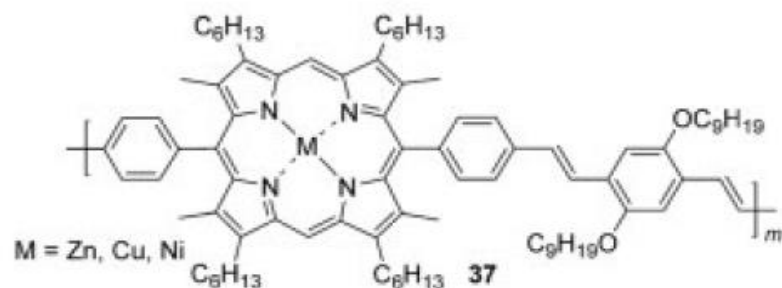
PDI: 1.88

Double-strand ladder-type polyporphyrin

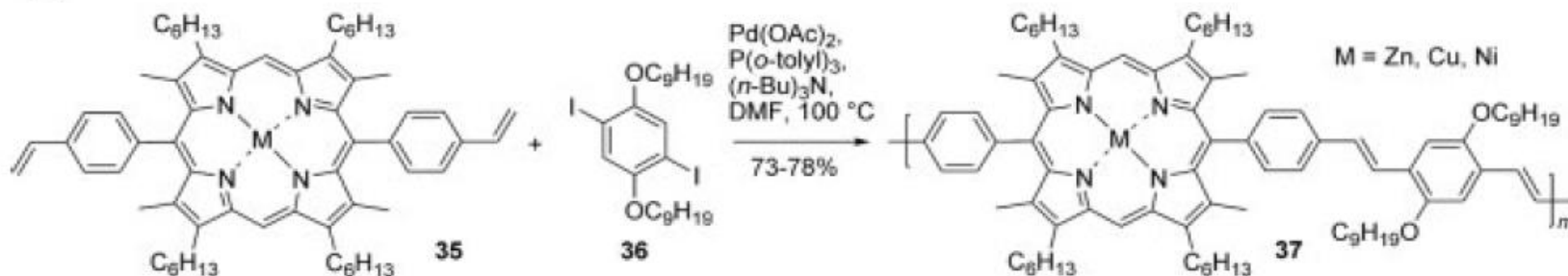


- In these ladder-type structures the effective conjugation length could be increased remarkably, enhancing charge mobility.

(a)



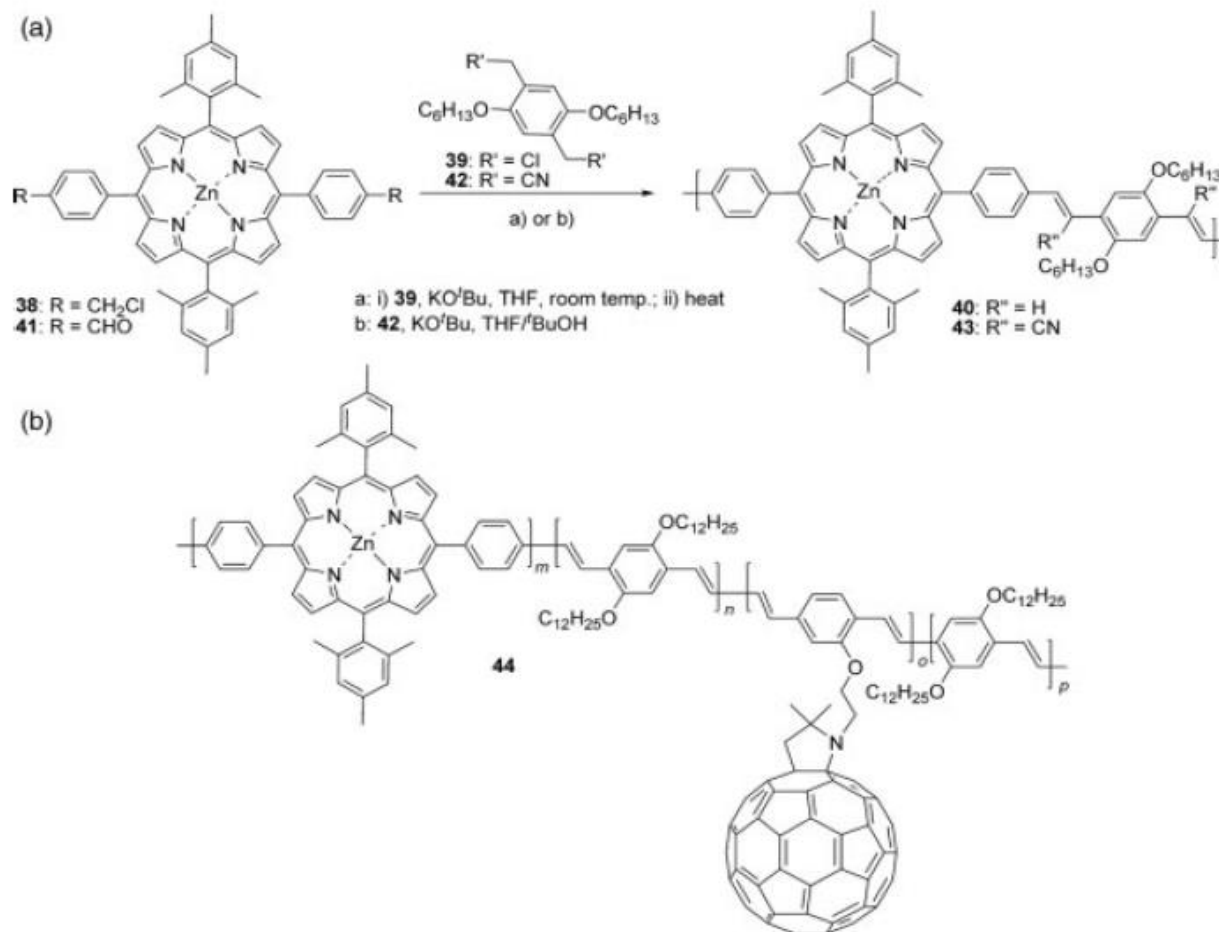
(b)



• Different types of conjugated porphyrin-pphenylenevinylene copolymers (a) and synthesis of the metalloporphyrin-containing PPV by Heck reaction (b)

• Under the reaction conditions utilized (i.e., DMF, 100 C) no exchange of the coordinated metal ions by the Pd(II) species could be observed

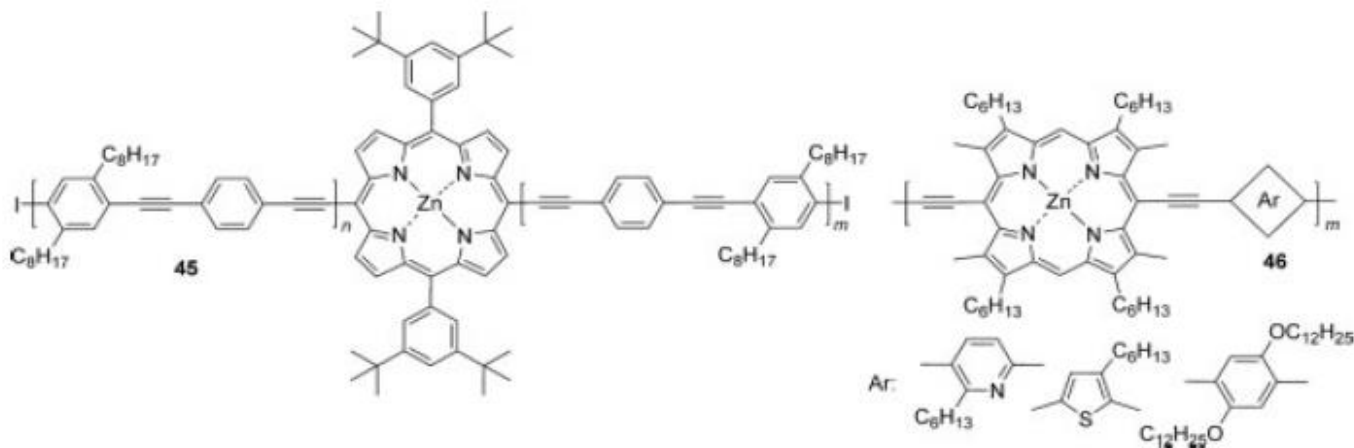
Two examples



- Heck reaction appears to be somehow superior to the conventional organic condensation reactions
- A promising candidate for photo induced electron-transfer systems

Conjugated Metallopolymers with Porphyrin

- Two different types of metalloporphyrin-containing poly(p-henyleneethyne)s (PPEs) by Sonogashira cross-coupling reactions

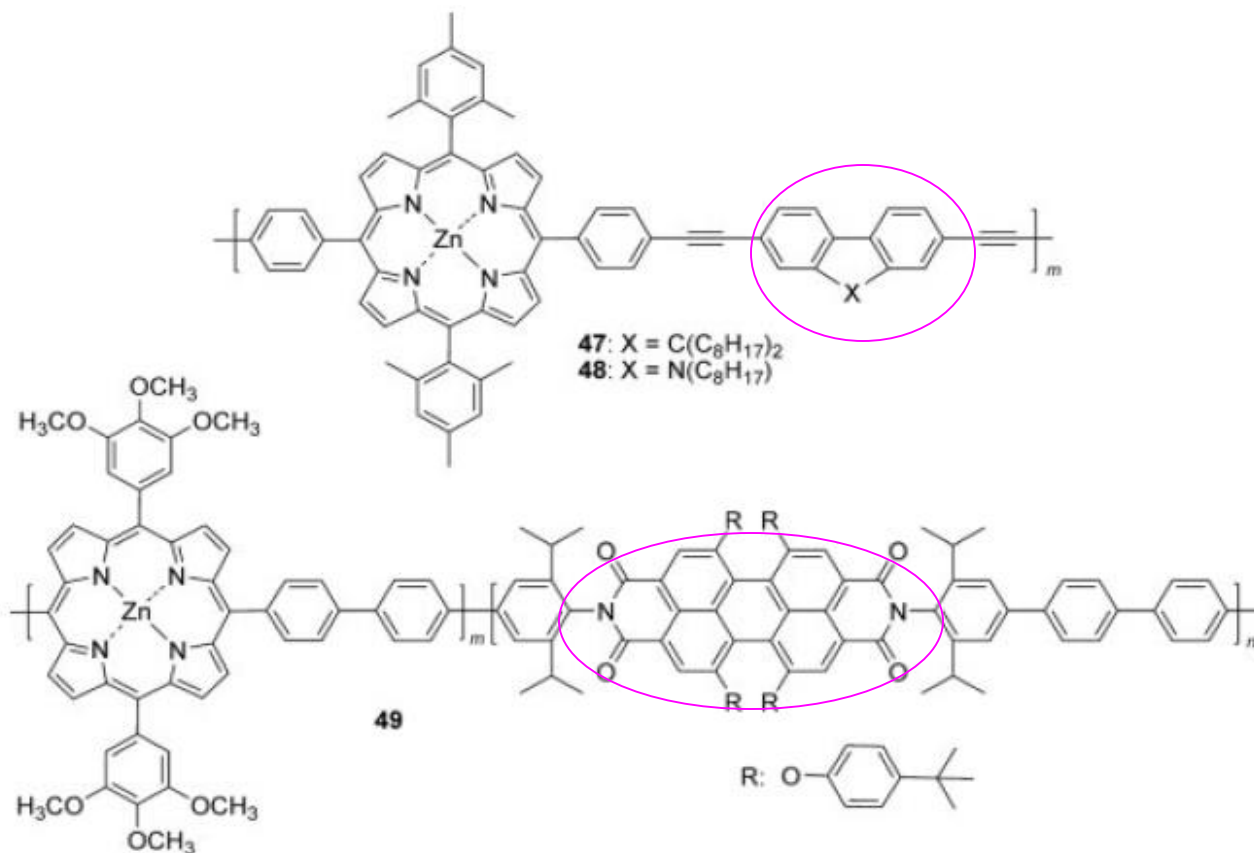


Nielsen, K.T., Spanggaard, H., and Krebs, F.C. *Displays*, **2004**, 25 (5), 231–235.

Yamamoto, T., et al. *Macromolecules*, **2000**, 33 (16), 5988–5994

Conjugated Metallopolymers with Porphyrin

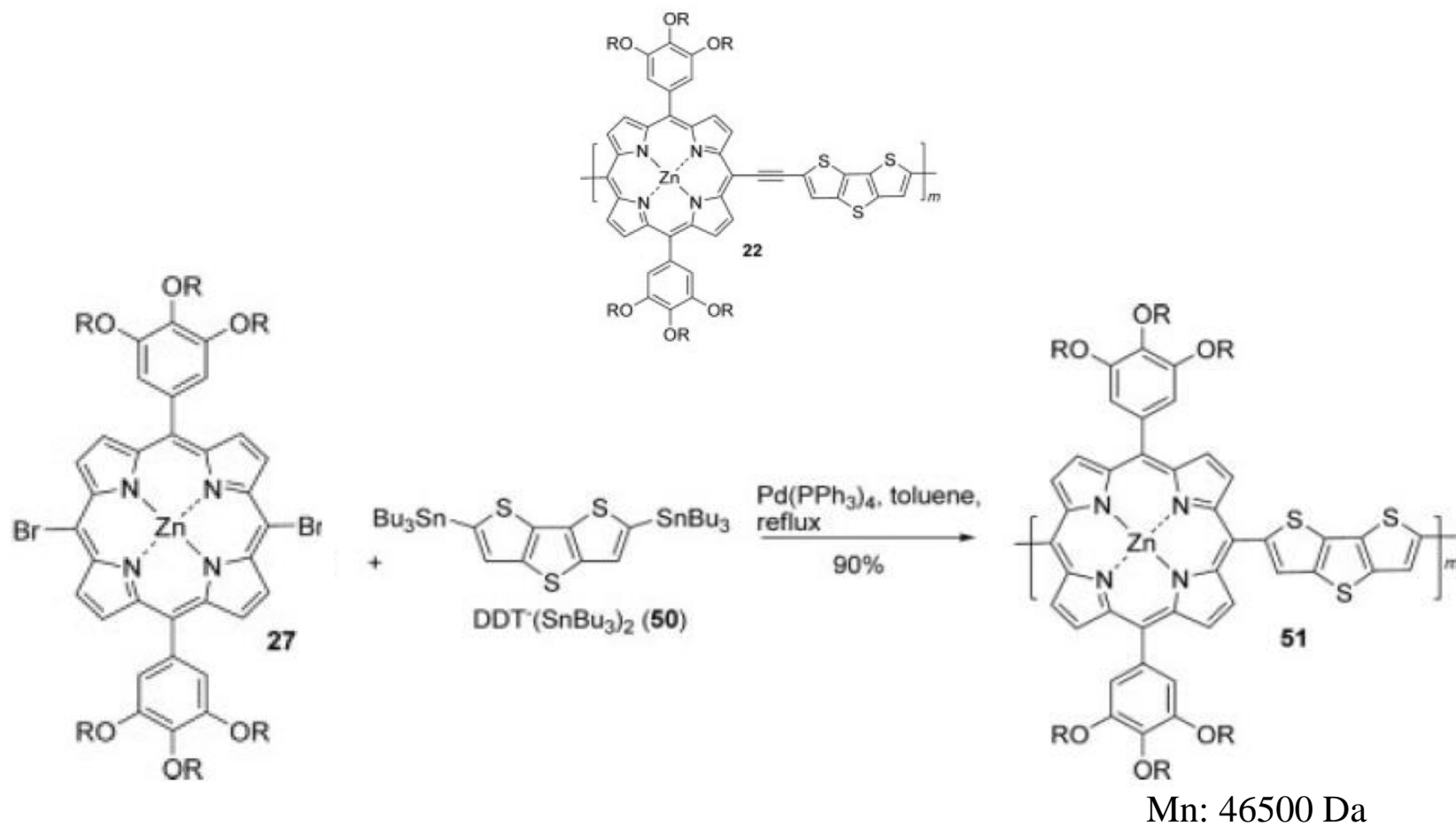
- Connecting with chromophores



Zhao, J.L., Li, B.S., and Bo, Z.S. *Chin. Sci. Bull.*, **2006**, 51, 1287–1295.

Zhu, D.B. et al. *J. Polym. Sci. Polym. Chem.*, **2006**, 44, 5863–5874.

Conjugated Metallopolymers with Porphyrin

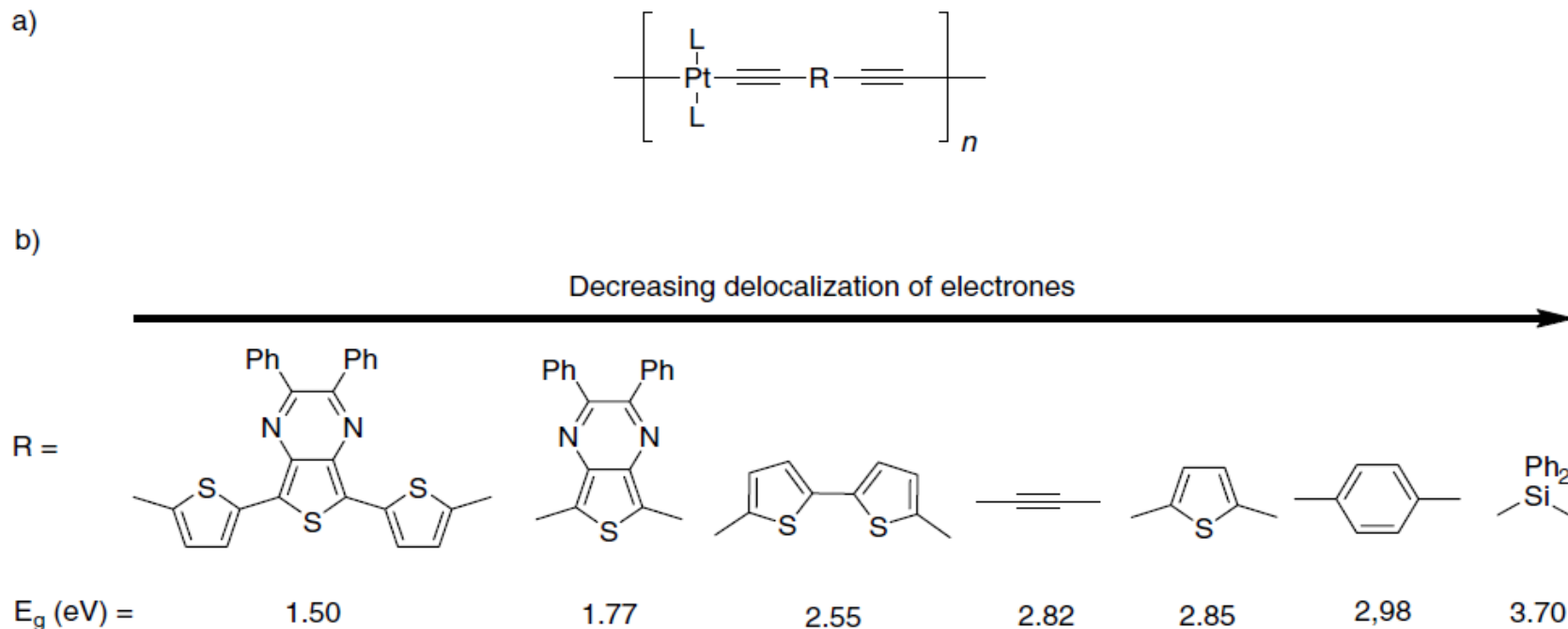


Huang, X.B., Zhu, C.L., Zhang, S.M., Li, W.W., Guo, Y.L., Zhan, X.W., Liu, Y.Q., and Bo, Z.Z. (). *Macromolecules*, **2008**, *41* (19), 6895–6902.

Rigid-Rod Polymetallaynes

- Rigid-rod transition metal acetylide polymers represent a class of metallopolymers due to their unique structures and optoelectronic properties

Rigid-Rod Polymetallaynes

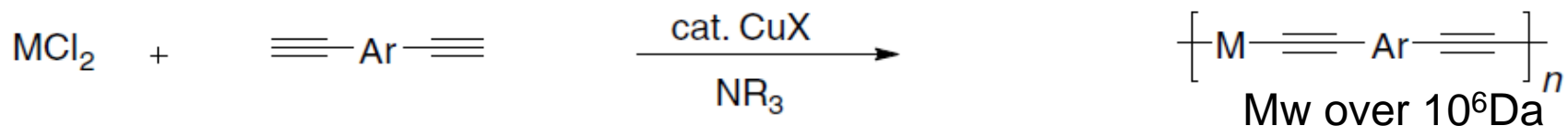


- (a) Schematic representation of the general structure of a metal acetylide polymer;
 (b) trend in optical band gaps for platinum(II) polyynes with different spacer groups [170, 180].

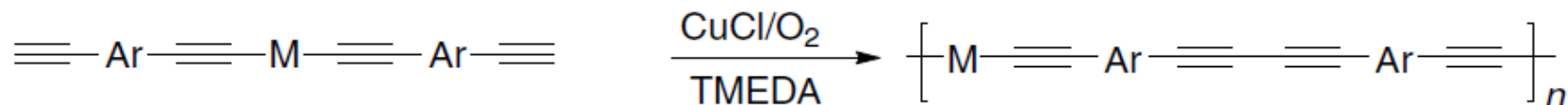
Rigid-Rod Polymetallaynes

- Three synthetic routes for the synthesis of d^{10} metal alkynyl polymers.

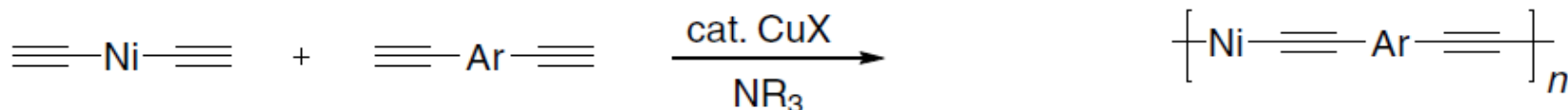
a) Dehydrohalogenation



b) Oxidative coupling



c) Alkynyl ligand exchange



M = *trans*-Pt(PR₃)₂, *trans*-Pd(PR₃)₂

Ni = *trans*-Ni(PR₃)₂

Ar = none or aromatic spacer

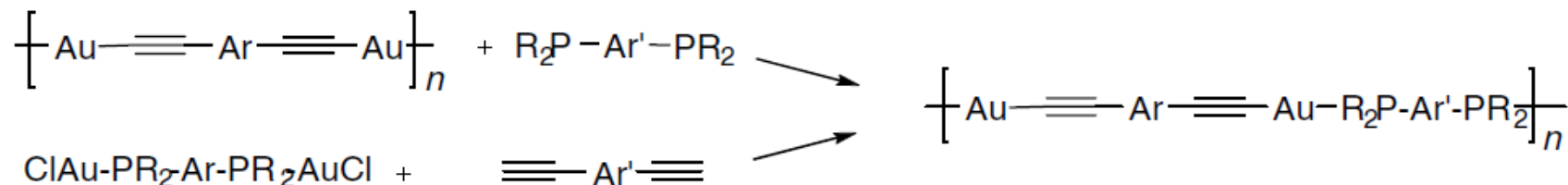
Rigid-Rod Polymetallaynes

•Gold(I)-acetylides

a)



b) Diphosphinoarene bridge

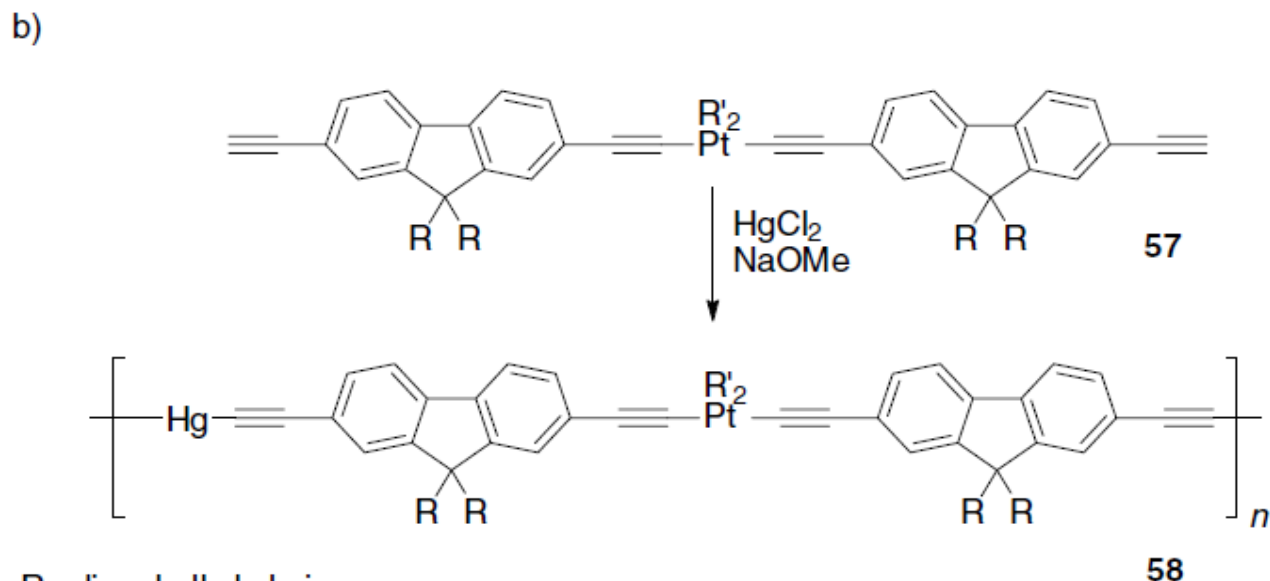
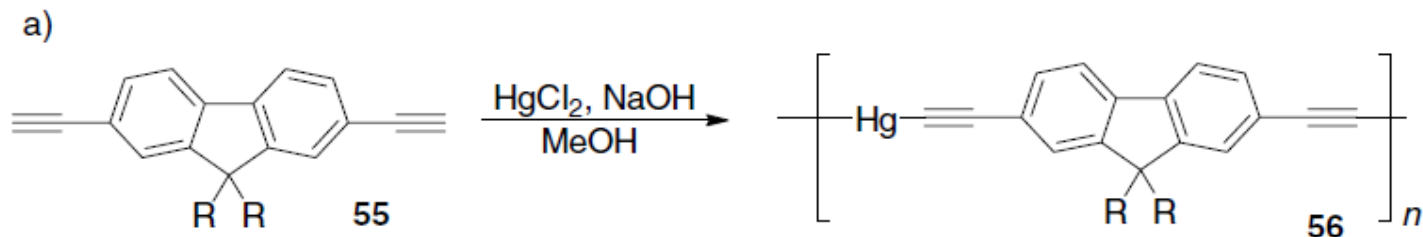


Ar, Ar' = aromatic spacer

R = phenyl or *i*-Pr

Rigid-Rod Polymetallaynes

•Mercury polyynes polymers



R = lineal alkyl chain
 R' = n -Bu

rigid-rod heteronuclear s-alkynyl polymer

Wong, W.Y., Liu, L., and Shi, J.X. *Angew. Chem. Int. Ed.*, **2003**, 42, 4064–4068.

Zhou, G.J., Wong, W.Y., Lin, Z.Y., and Ye, C. *Angew. Chem. Int. Ed.*, **2006**, 45, 6189–6193

Zhou, G.J., Wong, W.Y., Ye, C., and Lin, Z.Y. *Adv. Func. Mater.*, **2007**, 17, 963–975

Conclusion

•All synthetic routes for metallopolymers are uncontrolled polymerizations. The stoichiometry as well as the concentration is very important for the polymerization reaction. Therefore, the reproducibility of polymerizations is often a problem. However, the resulting polymers feature interesting properties can be used for emerging applications.

Thanks for your attention!

